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Detection of deep stratospheric intrusions by cosmogenic ³⁵S

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The extent to which stratospheric intrusions on synoptic scales influence the tropospheric ozone (O₃) levels remains poorly understood, because quantitative detection of stratospheric air has been challenging. Cosmogenic ³⁵S mainly produced in the stratosphere has the potential to identify stratospheric air masses at ground level, but this approach has not yet been unambiguously shown. Here, we report unusually high ³⁵S concentrations (7,390 atoms m^{-3} ; ~16 times greater than annual average) in fine sulfate aerosols (aerodynamic diameter less than 0.95 µm) collected at a coastal site in southern California on May 3, 2014, when groundlevel O3 mixing ratios at air quality monitoring stations across southern California (43 of 85) exceeded the recently revised US National Ambient Air Ouality Standard (daily maximum 8-h average: 70 parts per billion by volume). The stratospheric origin of the significantly enhanced ³⁵S level is supported by in situ measurements of air pollutants and meteorological variables, satellite observations, meteorological analysis, and box model calculations. The deep stratospheric intrusion event was driven by the coupling between midlatitude cyclones and Santa Ana winds, and it was responsible for the regional O₃ pollution episode. These results provide direct field-based evidence that ³⁵S is an additional sensitive and unambiguous tracer in detecting stratospheric air in the boundary layer and offer the potential for resolving the stratospheric influences on the tropospheric O₃ level.

stratosphere-troposphere exchange | surface ozone | National Ambient Air Quality Standard | radioactive isotope of sulfur | Santa Ana wind

igh ground-level ozone (O_3) mixing ratios exert adverse impacts on human health, vegetation, and materials (1, 2). In the free troposphere, O_3 is an important greenhouse gas contributing to global warming. It also controls the lifetime of other reactive greenhouse gases through oxidation processes (3), serves as the dominant precursor of the hydroxyl radical, and enhances the oxidizing capacity of the troposphere (4). Tropospheric O₃ formation involves a series of photochemical reactions related to anthropogenic emissions of O₃ precursors [e.g., nitrogen oxides (NO_x), carbon monoxide (CO), and volatile organic compounds (VOCs)], biomass burning, and lightning (5). In addition, elevated levels of tropospheric O_3 may be caused by the intrusion of O_3 -rich stratospheric air masses (4, 6–8). Detection of such stratospheric intrusion events by field-based measurements has been a major scientific concern since the 1970s (9). Concurrent measurement of ground-level O₃, CO, and humidity is the most common method (10, 11), but it is ambiguous and only useful in extreme events and background sites. Ozonesondes, lidar, and aircraft measurements provide high-resolution information on vertical O₃ distributions (12-14), but they are relatively expensive and not widely available. Therefore, it is crucial to find an additional and unambiguous stratospheric tracer at ground level to assist in such investigations.

 35 S (half-life = 87 d) is a cosmogenic isotope naturally produced by the interaction of high-energy cosmic rays with 40 Ar in the atmosphere. The flux of cosmic rays and the production rate of 35 S depend on both latitude and altitude, with higher values at the

polar region and in the stratosphere (and lower at the equatorial region and in the boundary layer) (15). Cosmogenic ³⁵S oxidizes to ³⁵SO₂ in ~1 s after production and is further oxidized to ³⁵SO₄²⁻ before wet and dry removal. Therefore, the variation of ³⁵SO₄²⁻ concentrations at ground level is controlled by the SO₂ oxidation and sulfate removal rates as well as air masses originating from the higher atmosphere. Because of the higher production rate of ³⁵S in the stratosphere (one to two orders of magnitude greater than in the troposphere) (15), significant enhancement of ${}^{35}\text{SO}_4{}^{2-}$ concentration at ground level may offer a new tool to quantify the impact of deep stratospheric intrusions on tropospheric O_3 . A unique advantage of ³⁵S is that it exists in both gas (SO₂) and particle (sulfate) phases and has an ideal half-life (87 d) for studying atmospheric processes on synoptic scales, providing additional information on potential impacts of stratospheric intrusions on gas to particle (SO₂ to SO₄²⁻) conversion rates. The development of the optimized low-level liquid scintillation counting technique (16) gave rise to a growing number of aerosol ³⁵SO₄²⁻ measurements in recent years. Based on simple and unconstrained 1D box model calculations, slightly elevated ${}^{35}\text{SO}_4{}^{2-}$ concentrations in early studies were linked to the polar vortex activity (17), Santa Ana winds, and shallow stratospheretroposphere exchange events in southern California (18). Recent studies applied mesoscale meteorology models to investigate the possible downward transport processes of aged stratospheric air (19, 20). However, the reliability of 35 S as a stratospheric tracer remains uncertain and debated, because the magnitudes of ³⁵SO₄²⁻ enhancements were relatively small and

Significance

The recently revised stricter US National Ambient Air Quality Standard for ground-level ozone (O₃) requires precise methods to screen "exceptional events," such as naturally occurring deep stratospheric intrusions. However, existing approaches used in detecting stratospheric intrusions and evaluating their contributions to ground-level O₃ enhancement are not satisfactory. Here, we introduce the use of cosmogenic ³⁵S to assist in such quantifications. We measured the highest ³⁵S concentration in natural sulfate aerosols ever reported in the literature during deep stratospheric intrusions. The downward transport of stratospheric O₃ is confirmed by air quality data and meteorological analysis, showing the high sensitivity of ³⁵S as a stratospheric tracer and its utility to understand atmospheric transport and chemistry processes.

The authors declare no conflict of interest.

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Table 1. ${}^{35}SO_4{}^{2-}$ concentrations in size-segregated aerosol samples collected at the University of California, San Diego in spring of 2014

Date	Total airflow (m³)	$^{35}SO_4^{2-}$ (atoms m ⁻³)		
		>7.2 µm	0.95–7.2 μm	<0.95 µm
March 27 to April 4	11,647	12 ± 7	33 ± 5	90 ± 4
April 6 and 7	1,543	n.d.	n.d.	170 ± 40
April 12–18	9,503	15 ± 8	77 ± 6	230 ± 10
May 3–7	7,176	4 ± 6	75 ± 8	7,390 ± 50
May 10–14	7,571	n.d.	50 ± 10	370 ± 20
May 19–23	7,727	3 ± 7	51 ± 7	130 ± 10

n.d., Not detectable.

other stratospheric signatures (e.g., high O_3 level and low humidity) were not observed in suspected aged stratospheric air masses (18, 19). Measurements of ${}^{35}SO_4{}^{2-}$ during deep stratospheric intrusions, which directly entrain fresh stratospheric air to the boundary layer, have never been made.

Climatological studies showed that the western United States is one of the global hotspots for deep stratospheric intrusions, which is likely because of the east Pacific storm track and the high-altitude orography (4, 21). Such deep stratospheric intrusion events compromise high-altitude regions of the western United States in attaining US National Ambient Air Quality Standard (NAAQS) for O₃ (6). Additional field-based observation studies indicated that this O3-rich stratospheric air can even be transported to lowaltitude regions, such as Los Angeles at the southern California coast (8, 22, 23). Consequently, this region is a natural laboratory for studying the potential of ³⁵S as a tracer for stratospheric intrusions. In October of 2015, the NAAQS for O3 was revised to 70 ppbv (parts per billion by volume) [daily maximum 8-h average (MDA8)] from the previous standard of 75 ppbv by the US Environmental Protection Agency (EPA). This new standard has been effective since December 28, 2015, and hence, identifying and excluding such naturally occurring "exceptional events" become increasingly important (24). The EPA recommends identifying exceptional events with supporting evidence, emphasizing the urgent need to find a new and sensitive stratospheric tracer, such as ³⁵S, to offer an unambiguous diagnostic for stratospheric intrusions. In this study, we measure ³⁵S concentrations during deep stratospheric intrusions to show the sensitivity of ³⁵S to stratospheric air in the boundary layer.

Field-Based ³⁵S Measurements

Table 1 summarizes ³⁵S concentrations in size-segregated sulfate aerosol $({}^{35}SO_4{}^{2-})$ collected on the rooftop of Pacific Hall on the campus of the University of California, San Diego [32.876° N, 117.242° W; 120 m above sea level (a.s.l.)] in spring of 2014. Most of ³⁵S concentrations agree well with our previous measurements at the Scripps Pier (10 m a.s.l. and 1.7 km from the sampling site in this study) (18), but an unusually high ${}^{35}SO_4{}^{2-}$ concentration of 7,390 atoms m^{-3} was found in the fine aerosol sample (with aerodynamic diameter less than 0.95 μ m) collected on May 3, 2014, ~16 times greater than the annual mean of 460 atoms m⁻ (18, 25) (Fig. 1). In fact, this value is the highest ${}^{35}SO_4{}^{2-}$ concentration ever reported for natural aerosol samples in the literature, which may be explained by stratospheric influence. It is noted that comparably high ³⁵S levels were measured in two rainwater samples collected in Korea in spring and winter (200 and 400 mBq L^{-1} , respectively), two seasons with frequent stratospheric intrusions in East Asia (26), which were significantly greater (more than a factor of ~6) than other rainwater samples (4–60 mBq L⁻¹) (27). Using the ${}^{35}SO_4{}^{2-}$ scavenging ratio obtained in Japan (28), the atmospheric ${}^{35}SO_4{}^{2-}$ concentrations in Korea were calculated to be 150-2,200 atoms m⁻³ in most cases and

7,500 and 15,000 atoms m⁻³ in two episodes. Although this estimation is subject to large uncertainties (~60% relative SD) (28), it matches the most recent ${}^{35}SO_4{}^{2-}$ measurement directly made on aerosol samples collected in East Asia (90–1,130 atoms m⁻³ in most samples and an increased ${}^{35}SO_4{}^{2-}$ concentration of 3,150 atoms m⁻³ in a sample affected by aged stratospheric air from the free troposphere) (20). The aforementioned high rainwater ${}^{35}S$ activities (27) were likely affected by stratospheric origin of our unusually high ${}^{35}SO_4{}^{2-}$ concentration directly measured on sulfate aerosol is examined.

Before discussing the impacts of stratospheric intrusions, other potential factors should be carefully evaluated. Apart from natural cosmogenic production, the ${}^{35}\text{Cl}[n,p]{}^{35}\text{S}$ reaction between neutrons escaping from Fukushima and ${}^{35}\text{Cl}$ in the coolant seawater is the only identified anthropogenic source of ${}^{35}\text{S}$ (25, 28). It is worthwhile to note that core elements reactors do not emit 35 S. It is high levels of ³⁵Cl in seawater that react with neutrons and allow ³⁵S productions. Given the highly specific reaction condition, there is no evidence showing that the observed ${}^{35}SO_4{}^{2-}$ spike in this study is caused by the ${}^{35}Cl[n_p]{}^{35}S$ reaction in the Fukushima or any nuclear plant. In addition, it was proposed that atmospheric ${}^{35}SO_4{}^{2-}$ removed by precipitation or dry deposition processes might reenter the boundary layer through the atmosphere and land surface interaction (biomass burning or wind-blown terrestrial dust) within ~ 66 d and elevate ${}^{35}\text{SO}_4{}^{2-}$ concentrations in the boundary layer (27). Although this hypothesis remains to be proven, we carefully screen out this potential scenario. Large wildfires (>300 acres; defined by the California Department of Forestry and Fire Protection) were not recorded during the sampling period (cdfdata.fire.ca.gov/ pub/cdf/images/incidentstatsevents 253.pdf). The absence of large wildfires is also supported by the satellite observations (Fig. S1). Therefore, any significant contribution of ³⁵S from biomass burning is implausible in this study period. The low ${}^{35}SO_4^{-2}$ concentration in coarse particle observed in the same set of aerosol samples (Table 1) also suggests that ³⁵SO₄²⁻ in resuspended terrestrial soil or dust cannot account for the notable enhancement of ${}^{35}\text{SO}_4{}^{2-}$ in this study. To date, there is no evidence or theory showing that other sources/processes can lead to significant variations or productions of ³⁵S. After considering all potential factors, the impact of air masses from the stratosphere, where the natural cosmogenic production rate of 35 S is approximately two orders of magnitude greater than the Earth's surface (15), is the most likely candidate to explain the elevated ³⁵S concentration.

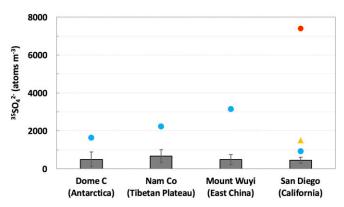


Fig. 1. The ${}^{35}SO_4{}^{2-}$ concentration in the fine aerosol sample collected on May 3, 2014 (red circle) and the comparison with annual means (gray bars; error bars stand for 1 SD) and the highest values (blue circles) measured at different sampling sites in previous studies (17–20). The orange triangle represents the sample affected by the trans-Pacific transport of ${}^{35}S$ produced from the ${}^{35}Cl[n,p]{}^{35}S$ reaction in Fukushima (25).

Exceptional Event of O₃ Enrichment

On May 3, 2014, when the ³⁵S-rich aerosol sample was collected, a regional O₃ pollution event was observed over southern California (Fig. 2). Two stations in Los Angeles were in the category of "unhealthy" [Air Quality Index (AQI): 151-200 or MDA8: 86-105 ppby], and 80% of stations in southern California (68 of 85) were in the categories of "unhealthy for sensitive groups" (AQI: 101-150 or MDA8: 71-85 ppbv) or "moderate" (AQI: 51-100 or MDA8: 55-70 ppby). Although the other 15 stations were in the category of "good" (AQI: 0-50 or MDA8: 0-54 ppbv) on this day, the O_3 mixing ratios in 14 stations were still higher than annual means, and five of them were significantly greater (>80th percentile) than normal days. The relatively low O₃ mixing ratios (compared with other stations) are because of substantial NO emissions from vehicles in surrounding areas that lower ambient O₃ mixing ratios via the "titration effect" (NO + $O_3 \rightarrow NO_2 + O_2$) (29). For example, the MDA8 of 54 ppbv in the Otay Mesa station (located at the United States-Mexico border and affected by the busy crossing of heavy-duty trucks) on May 3, 2014 was the annual highest value.

Fig. 3 A and B shows the time series of relative humidity (RH), temperature, and wind speed recorded in San Diego from April 27 to May 8, 2014. RH dramatically dropped down from 67% at 2100 hours Pacific Standard Time (PST) on April 28 to 7% at 1500 hours PST on April 29 accompanied with enhanced temperature (~30 °C) and wind speed (>15 m/s). The wind speed reached a maximum of 28 m s⁻¹ on April 30, with wind direction shift from variable to northwesterly (Fig. 3B). These abnormal meteorological conditions are typical signatures of Santa Ana winds, which are highly dry, hot, and strong winds that descend from inland desert regions to the Pacific coastal region in southern California (18, 22, 30-32). These foehn-like katabatic winds result from a strong pressure gradient between a high pressure over the Great Basin and an offshore low pressure. The high pressure can compress sinked air, force the air temperature to rise, and reduce its RH. Although the wind speed returned back to normal on May 2, 2014, low RH and high temperature persisted until May 4, suggesting that the Santa Ana wind event lasted for 5 d (April 29 to May 3, 2014).

Santa Ana wind events are often behind a cold front associated with an upper-level trough (31), which cannot only exacerbate the katabatic winds but also, can lead to the formation of deep tropopause folds and stratospheric intrusions (22). A recent study suggested that the coupling between Santa Ana winds and stratospheric intrusions might pose serious O₃ pollution threats across the coast of southern California (22). In this study, groundlevel O3 mixing ratios during the Santa Ana event increased significantly (Fig. 3A). The annual highest MDA8 in the Alpine station in San Diego (81 ppbv) was recorded on May 3, with a maximum 1-h O₃ mixing ratio of 88 ppbv at 1200 hours PST. Solar radiation and temperature were stronger during the Santa Ana episodes, but the ratio of NO₂ photolysis to NO + O_3 reaction rates $(j/k = [NO][O_3]/[NO_2])$, with the photostationary state assumption) (33) showed only slight enhancement from April 29 to May 1, suggesting that photochemical production of O₃ in the O₃-NO-NO₂ cycle was not a major factor leading to the elevated O₃ levels during the entire Santa Ana period. Although wildfires occur commonly during Santa Ana wind events, which can significantly increase ground-level O₃ mixing ratios (22, 30), no significant wildfire occurred in our study period (Fig. S1), suggesting that emissions from wildfires were also not a major contributor. A closer look into the O₃ diurnal variations reveals significant enhancements of nighttime O₃ mixing ratios during the Santa Ana period, which were 20-29 ppbv greater than on normal days (Fig. S2). Because photochemical O_3 production ceases at nighttime, this result suggests a larger contribution of long-range transports (including stratospheric intrusions) to the enhanced O₃ mixing ratios in the Santa Ana period. A negative correlation between O3 and CO (a tracer of anthropogenic emission) may suggest O₃

originating from the stratosphere, where CO is depleted and O₃ is rich (34, 35). To rule out the potential impact of nighttime titration effect, which can also lead to negative O₃–CO correlation (36, 37), only daytime data (0700–1800 hours) were considered in this study. A significant negative correlation between O₃ and CO in the Santa Ana event (r = -0.60, confidence level > 99.9%) compared with normal days (r = -0.04, confidence level = 40.9%) implies that the elevated O₃ levels during the Santa Ana event were likely related to the vertical transport of O₃ from high altitudes.

In summary, the concurrently enhanced ground-level ³⁵SO₄²⁻ and O₃ concentrations and negative O₃-CO correlation indicate that the O₃ episode on May 3, 2014 was likely affected by a deep stratospheric intrusion event, a naturally occurring exceptional event. Because most Santa Ana winds only entrain air masses from the free troposphere to the boundary layer and would not lead to a significant enhancement of ground-level O₃ mixing ratios (18, 30), a stratospheric intrusion event that transports O_3 rich stratospheric air to the free troposphere before or during the Santa Ana event is required to result in the observed groundlevel ${}^{35}SO_4{}^{2-}$ and O_3 concentrations. The vertical ozone profile retrieved from Global Ozone Monitoring Experiment-2 satellite observation on April 30, 2014 revealed significant enhancements of the tropopause O₃ levels and total O₃ columns at the upstream region of southern California during the Santa Ana event, indicating stratospheric air masses mixing into the troposphere (Fig. S3). Under the influences of Santa Ana winds, such O₃-rich stratospheric air may be transported downward to the boundary layer and westerly to coastal southern California.

Meteorological Model Analysis

To quantitatively estimate the probability that air masses sampled in this Santa Ana wind event partly originate from the stratosphere, an inert stratospheric tracer was simulated by a mesoscale meteorology model [Weather Research and Forecasting Model (WRF)] (20). The time series of the WRF-simulated stratospheric tracer at the boundary layer show a small peak (0.6%) at 1200 hours PST on April 30, 2014, 1 d after the onset of the Santa Ana wind event, and the highest peak (1.5%) at 0500 hours PST on May 1, 2014 (Fig. 3*C*). The WRF-simulated stratospheric tracer gradually decreased from the highest peak to baseline (\sim 0%) from May 1 to 5, 2014 (Fig. 3*C*). Stratospheric O₃ mixing ratios at the boundary layer simulated by an independent Lagrangian particle dispersion model [FLEXible PARTicle dispersion model (FLEXPART);

Ozone AQI Values by site on 05/03/2014



Fig. 2. Distribution of ozone AQI and levels of health concern in California recorded by the US EPA on May 3, 2014 (www3.epa.gov/airdata).

EARTH, ATMOSPHERIC, ND PLANETARY SCIENCE

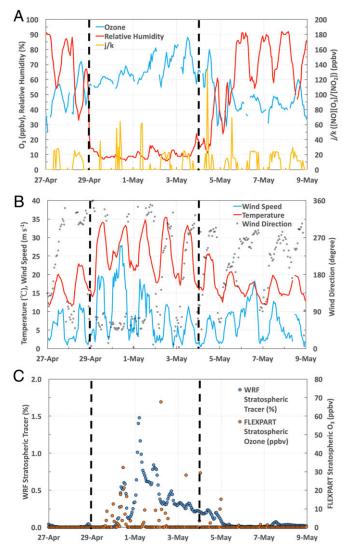


Fig. 3. Time series of hourly (A) O_3 , RH, and *j/k* measured in San Diego (the Alpine monitoring station); (B) temperature, wind speed, and direction measured in San Diego (the Kearny Mesa station); and (C) the simulated WRF stratospheric tracer and FLEXPART stratospheric O_3 at the boundary layer in San Diego. The vertical black dashed lines define the period of the Santa Ana event (April 29 to May 3, 2014) based on abnormal RH and temperature.

driven by the WRF output] (19) show a consistent trend (Fig. 3C), further supporting the stratospheric origins of ^{35}S and O_3 .

The process of how this plume intruded to the troposphere and reached coastal southern California is investigated by the horizontal distributions of the WRF-simulated stratospheric tracers and weather systems at various altitudes. As clearly shown in Fig. 4A, the stratospheric intrusion episode leading to the elevation of the WRF-simulated stratospheric tracer was directly triggered by a cutoff low-pressure system, which is accompanied by strong convective motions and tropopause folding and a typical synoptic condition resulting in active stratosphere to troposphere exchange over western United States (10, 22). The development of this synoptic situation is shown in Fig. S4. At the beginning, a strong midlevel short-wave trough developed into a closed low-pressure area as the system occluded over the central high plains since April 27, 2014. An associated mid- to low-level cyclone then reached its peak intensity while a trailing cold front move eastward across eastern Kansas, eastern Oklahoma, and northern Texas on April 29, 2014. Meanwhile, the strong northeasterly flow on the southwest flank of the cyclone swept through southwestern United States and caused a significant late season Santa Ana event. Later on, the low-pressure system started to abate, stretched southward from Canada to New Mexico and from northern California to the Gulf of Mexico on April 30, and eventually, dissipated after May 2, 2014.

Exchange processes between the free troposphere and the boundary layer are vital to bring ³⁵S- and O₃-rich stratospheric air to the sampling site. Fig. 4B shows that the impacts of stratospheric air on the boundary layer were mostly confined in the regions affected by the Santa Ana wind. Stratospheric air masses in the free troposphere started to penetrate into the boundary layer in Utah at 1000 hours PST on April 30, 2014 behind the trough line and were subsequently transported to southern California via the northeast Santa Ana wind on May 1 (Fig. S5). Zonal cross-sections of potential vorticity (PV) and the WRFsimulated stratospheric tracer clearly show the distinctive tropopause folding associated with the cutoff low between ~105° W and ~115° W and a tongue of stratospheric air sloping downward at ~115° W (Fig. 5), highlighting the pathway of the stratospheric air into the boundary layer. The MDA8 O₃ mixing ratios recorded in Las Vegas (35.786° N, 115.357° W; 924 m a.s.l.), the upstream region of southern California, were 59, 69, and 56 ppbv on April 30, May 1, and May 2, 2014, respectively (airquality.clarkcountyny. gov/cgi-bin/aqi map.pl). The increases of MDA8 on May 1 in part support our model results. Because the stratospheric air masses were continuously entrained into the free troposphere by the occluded low-pressure system, these southwestward-transported air masses might slope downward to the boundary layer in southern California as well (SI Text and Fig. S4).

³⁵S Box Model Calculation

Priyadarshi et al. (18) used a 1D four-box model, which was developed to calculate the ${}^{35}SO_4{}^{2-}$ concentration in fine aerosol collected at the Scripps Pier, to quantify air mixing during the Santa Ana wind events and shallow stratosphere-troposphere exchange events. The model parameters (Table S1) and uncertainties were thoroughly described in refs. 18 and 25. Specifically, it was suggested that, in a shallow stratosphere-troposphere exchange event, $\sim 7\%$ of the total air masses in the free troposphere were originated in the low stratosphere per day, whereas during the Santa Ana wind event, ~41% of air mass sampled in the marine boundary layer recently originated from the free troposphere per day (18). Here, we use the same box model and combine the mixing effects of stratospheric intrusions and Santa Ana winds (i.e., $7\% \times 41\% = -3\%$ of total air sampled in the boundary layer originated from the stratosphere in 1 d) to simulate the coupling between Santa Ana winds and stratospheric intrusions in this study. The averaged ${}^{35}SO_4{}^{2-}$ concentration during May 3–7, 2014 calculated by the model was 7,100 atoms m^{-3} , reasonably agreeing with our measurement (7,390 atoms m^{-3}). The model predicts that the averaged ${}^{35}SO_2$ and ${}^{35}SO_4^{2-}$ concentrations during the episode period (April 29 to May 3) were 2,100 and 11,000 atoms m^{-3} , respectively, which however, cannot be verified in this stage, because samples were not collected from April 29 to May 2, 2014 because of operational issues. To date, the highest ³⁵SO₂ concentration (1,800 atoms m⁻³) was measured at New Haven by Tanaka and Turekian (38) in April of 1992. The highest atmospheric ${}^{35}SO_4{}^{2-}$ concentration was estimated to be 15,000 atoms m⁻³ from rainwater samples as discussed previously (27). These field-based measurements suggest that our modelestimated ³⁵S concentrations during deep stratospheric intrusions are plausible. If only the mixing of stratospheric intrusion (or the Santa Ana wind) was considered, the estimated averaged ${}^{35}SO_4{}^2$ -3 concentration during May 3-7, 2014 was 3,400 (or 742) atoms m⁻ significantly lower than our field-based data and previous calculations, implying that the coupling between stratospheric intrusions and Santa Ana winds is crucial to lead to the observed high 35 S and probably, O₃ levels.

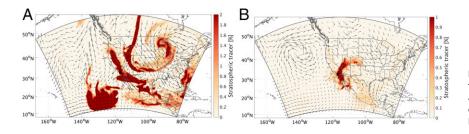


Fig. 4. Spatial distribution of the WRF stratospheric tracer at (*A*) 500 hPa at 0000 hours PST on April 30 and (*B*) 390 m above ground level at 0500 hours PST on May 1. The black star indicates the location of San Diego.

Previous studies suspected that the mixing between O₃-rich stratospheric air masses and polluted plumes might accelerate the oxidation of SO_2 (20). In this study, significant enhancements of j/k are observed during the post-Santa Ana period (May 4 and 5) (Fig. 3A), indicating stronger photochemical production of O_3 . This phenomenon may be because the downward-transported stratospheric O₃ actively participated in the O₃-NO_x-VOCs chemistry (39, 40) when the O₃-rich air masses were mixed with polluted low-altitude air masses (41). The decreased O_3 mixing ratio (Fig. 3A) suggested that the photochemically produced O_3 might rapidly participate in other reactions (e.g., the formation of secondary aerosols, including heterogeneous productions of sulfate) as a sink. If the oxidation lifetime of SO_2 in the boundary layer during the post-Santa Ana period (May 4 and 5) is reduced from 4 to 0.5 d, an averaged ${}^{35}SO_4{}^{2-}$ concentration during May 3– 7 of 7,400 atoms m⁻³ is obtained, perfectly matching the obser-vational data (7,390 atoms m⁻³). Although this hypothesis has yet to be tested by high-temporal resolution ${}^{35}SO_2/{}^{35}SO_4^{2-}$ measurement and proper chemistry modeling, the suspected enhanced aerosol formation rate is partly supported by higher PM_{2.5} (particulate matter with an aerodynamic diameter less than $2.5 \,\mu m$) concentrations (mean $\pm \sigma$) during the post-Santa Ana period $(10.8 \pm 3.3 \ \mu g \ m^{-3})$ than the Santa Ana period $(6.3 \pm 2.4 \ \mu g \ m^{-3})$; 2014 annual mean: $8.1 \pm 3.6 \ \mu g \ m^{-3}$). This potential influence is particularly important in the regions heavily impacted by SO₂ emissions and stratospheric intrusions, such as East Asia (3, 20, 41, 42).

Conclusions and Implications

In summary, our result is encouraging, because it shows the high sensitivity of 35 S to stratospheric intrusions and reveals the crucial role of coupling between Santa Ana wind and stratospheric intrusions in bringing fresh stratospheric air to the southern California coast. The absolute amount of radiation (or activity) in the 35 S-rich sample is small (0.68 mBq m⁻³) and not a concern for human health, but our highly sensitive measurement technique renders 35 S a sensitive tracer of stratospheric intrusions, an important process in nature for which there are gaps in understanding.

There is an urgent need to identify and screen the exceptional events for ground-level O3 caused by stratospheric intrusions. Our study reveals that field-based measurement of cosmogenic ³⁵S at ground level can serve as an additional valuable diagnostic for the occurrence of deep stratospheric intrusions. This method has three advantages. (i) The optimized aerosol sample handling procedures and low-level liquid scintillation spectroscopy method enable measuring low ³⁵S activities (0.2 disintegration per minute) in a simple, economical, effective, and highly sensitive way (16). (ii) The half-life of ${}^{35}S$ (87 d) is ideal for studying atmospheric processes on synoptic timescales, and it also permits a relatively long storage time of aerosol samples, which are routinely collected by the EPA, until the sampling period is suspected. (iii) Radiogenic ^{35}S has the potential to provide additional information on the impacts of stratospheric intrusions on gas to particle conversion rates and thereafter, possible particulate matter pollution events.

Although this box model shows the ability to reproduce the observed ${}^{35}SO_4{}^{2-}$ concentration, we should mention that the box model result still possesses uncertainties, because most

parameters in the model are not constrained by field-based measurements (18). The low temporal resolution of ³⁵S measurements in this study limits the use of field-based ³⁵S measurement in evaluating the model result and improving the model. In the future, a more strategic and comprehensive study can be designed to fully resolve the impacts of deep stratospheric intrusions on the tropospheric sulfur cycle and ground-level O₃ concentrations. The Realtime Air Quality Modeling System (RAQMS) model has been widely used in predicting and analyzing the stratospheric intrusion events (22, 23, 43). Although the RAQMS model underestimated the ground-level O₃ concentration in this episode (May 3, 2014), it showed capability to forecast the east Pacific storm track and stratospheric O₃ intrusions in the higher atmosphere (ragms-ops.ssec.wisc.edu/previous products/). The forecast result of the RAQMS can be used to design intensive aerosol and SO_2 sampling for ³⁵S measurements ~2 d before the occurrences of stratospheric intrusion events. With high temporal resolution (0.5-1 d), the evolution of ³⁵S during deep stratospheric intrusions can be resolved.

Aircraft field missions showed that concentrations of cosmogenic beryllium isotopes (⁷Be and ¹⁰Be; half-life = 53 d and 1.38 My, respectively) in the lower stratosphere can be ~40–110 times greater than in the boundary layer (44, 45). Similar measurements for ³⁵S are crucial to constrain box model results. More efforts on modeling works (e.g., updating ³⁵S production rate and incorporating ³⁵S into a 3D chemistry transport model with O₃ and sulfur chemistry) can advance quantifying the impacts of stratospheric intrusions on ground-level ³⁵S and O₃ at high temporal and spatial resolutions. The extent to which stratospheric intrusions may affect the gas to particle conversion rate can also be quantified by coupled measurements of ⁷Be and ³⁵S and proper modeling (38).

Climatological studies revealed that the western United States and the Himalayas are two global hotspots for deep stratospheric intrusions (4). In particular, a global chemistry–climate model showed strong contributions of stratospheric intrusions to MDA8 ground-level O_3 in Nevada (6, 7, 10). Our previous measurements showed high ³⁵S concentrations in the San

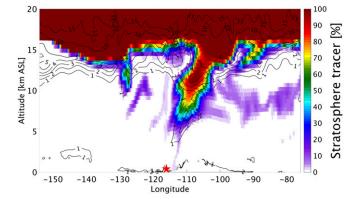


Fig. 5. Zonal cross-section of the WRF stratospheric tracer with PV (unit: PV unit) contours superimposed at 0000 hours PST on May 1, 2014. The red star indicates the location of San Diego.

Fernando Valley in California (a sampling site close to Nevada) (16) and at Mount Everest in the Himalayas (19), supporting the model results. These measurements imply that the spatial distribution of 35 S may provide invaluable information on regional variabilities of stratospheric intrusion strength and frequency to constrain model results. The high sensitivity of 35 S also allows for precise quantification of the contribution of aged stratospheric air to the background troposphere, which is crucial in understanding the O₃ budget (3, 7), the carbon cycle (46, 47), and the variations of other cosmogenic radionuclides, such as 10 Be, a primary proxy archive of past changes in solar activity, cosmic rays, and geomagnetic field intensity (45).

Materials and Methods

Size-segregated aerosol samples were collected using a high volume air sampler (HVP-4300AFC; Hi-Q) operated at a flow rate of ~1.13 $m^3 min^{-1}$. Soluble sulfate extracted from glass-fiber filter papers was subject to ³⁵S analysis

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using an ultralow-level liquid scintillation counting spectrometer (Wallac 1220 Quantulus) technique (16). Data on air pollutants (O_3 , NO_2 , NO, CO, and $PM_{2,5}$) and meteorological variables (temperature, RH, solar radiation, wind speed, and direction) were provided by the Air Pollution Control District County of San Diego (www.sdapcd.org). Field-based measurements were supported by a mesoscale meteorology model (the WRF), which permitted investigations of stratospheric intrusion processes (20). Detailed sampling, chemical processing, quality assurance, and control procedures as well as modeling approach can be found in *SI Text* and Figs. S6 and S7.

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