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### **Title**

Development of Transition Metal-Catalyzed Reactions for the Synthesis and Biological Evaluation of Enantioenriched Diarylalkanes and Homopropargylic Sultams

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Development of Transition Metal-Catalyzed Reactions for the Synthesis and Biological  
Evaluation of Enantioenriched Diarylalkanes and Homopropargylic Sultams

DISSERTATION

Submitted in partial satisfaction of the requirements  
for the degree of

DOCTOR OF PHILOSOPHY

in Chemistry

by

Charlotte A. Osborne

Dissertation Committee:  
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2015

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## **DEDICATION**

For my family

for their unwavering love and support

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“Aromatic Sulfide Compounds and Methods and Use Thereof.” Yonova, I. M.; **Osborne, C. A.**; Morrissette, N. S.; Jarvo, E. R. **2015**, U. S. Patent Appl. No. 14/666,088.

“Stereospecific Cross-Coupling Reactions of Aryl-Substituted Tetrahydrofurans, Tetrahydropyrans, and Lactones.” Tollefson, E. J.; Dawson, D. D.; **Osborne, C. A.**; Jarvo, E. R. *J. Am. Chem. Soc.* **2014**, *136*, 14951.

“Diaryl and Heteroaryl Sulfides: Synthesis via Sulfenyl Chlorides and Evaluation as Selective Anti-Breast-Cancer Agents.” Yonova, I. M.; **Osborne, C. A.**; Morrissette, N. S.; Jarvo, E. R. *J. Org. Chem.* **2014**, *79*, 1947.

“Stereospecific Nickel-Catalyzed Cross-Coupling Reactions of Alkyl Grignard Reagents and Identification of Selective Anti-Breast-Cancer Agents.” Yonova, I. M.; Johnson, A. G.; **Osborne, C. A.**; Moore, C. E.; Morrissette, N. S.; Jarvo, E. R. *Angew. Chem. Int. Ed.* **2014**, *53*, 2422.

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## **PRESENTATIONS**

**Osborne, C. A.**; Yonova, I. M.; Johnson, A. G.; Tollefson, E. J.; Morrissette, N. S.; Jarvo, E. R.

*Identification of Enantioenriched Diarylalkanes as Selective Anti-Cancer Agents*

Chao Family Comprehensive Cancer Center Scientific Retreat (Poster), September 2014

**Osborne, C. A.**; Tollefson, E. J.; Yonova, I. M.; Johnson, A. G.; Morrissette, N. S.; Jarvo, E. R.

*Identification of Selective Anticancer Agents Synthesized by Stereospecific Nickel-Catalyzed Cross-Coupling Reactions*

248<sup>th</sup> National Meeting of the American Chemical Society (Oral Presentation), August 2014

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247<sup>th</sup> National Meeting of the American Chemical Society (Poster Presentation), March 2014

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## **ABSTRACT OF DISSERTATION**

Development of Transition Metal-Catalyzed Reactions for the Synthesis and Biological  
Evaluation of Enantioenriched Diarylalkanes and Homopropargylic Sultams

by

Charlotte A. Osborne

Doctor of Philosophy in Chemistry

University of California, Irvine, 2015

Prof. Elizabeth R. Jarvo, Chair

Methodology for the synthesis of enantioenriched compounds is critical to the discovery of new pharmaceuticals. To access single enantiomers of biologically relevant molecules for evaluation, the Jarvo laboratory has developed both enantiospecific and enantioselective reactions. Herein, we report the development of transition metal-catalyzed methodology to generate diarylalkanes and homopropargylic sultams in high enantiomeric excess.

Diaryl sulfides and 1,1-diarylalkanes are known to exhibit a wide variety of therapeutic applications, including treatment of breast cancer. These compound classes were synthesized for biological evaluation against a range of cancer cell lines. Heteroaromatic diaryl thioethers were prepared by the reaction of sulfenyl chlorides with arylzinc bromides, while enantioenriched 1,1-diarylalkanes were synthesized through the nickel-catalyzed cross-coupling reactions of alkyl Grignard reagents. Several lead compounds were identified that selectively inhibited breast

cancer cell proliferation in the low micromolar range. In particular, a diarylalkane containing a thiophene moiety exhibited selective activity against a triple-negative breast cancer line.

Stereospecific nickel-catalyzed cross-coupling methodology has also been applied to the synthesis of tubulin-binding diarylethane derivatives. Upon evaluation for inhibition of tubulin polymerization, the (*S*)-enantiomer of a trimethoxyphenyl-containing diarylethane exhibited higher levels of tubulin disruption than the (*R*)-enantiomer. These results show proof of concept that enantioenriched diarylethanes exhibit marked differences in tubulin-binding activity.

We developed a different type of nickel-catalyzed reaction for the stereospecific ring-opening of *O*-heterocycles in order to provide acyclic alcohols with controlled formation of a new C–C bond. Aryl-substituted tetrahydropyrans underwent nickel-catalyzed Kumada coupling reactions with a range of Grignard reagents to furnish acyclic alcohols with high diastereoselectivity. A furan-substituted lactone underwent a Negishi cross-coupling reaction to provide the corresponding carboxylic acid in high enantiospecificity. Biological evaluation of the products identified several lead compounds with selective activity against a triple-negative breast cancer line.

In the final Chapter of this dissertation, we describe the synthesis of enantioenriched homopropargylic  $\alpha$ -chiral sulfonamides via the enantioselective silver-catalyzed propargylation of *N*-sulfonyl ketimines. This reaction proceeded in high yield and excellent enantiomeric excess, and was compatible with a wide variety of diaryl and alkyl ketimines. Synthetic transformations of homopropargylic products via enyne ring-closing metathesis and reduction reactions proceeded with high stereochemical fidelity. Both allenyl and propargyl borolane reagents could be used to obtain propargylic products, most consistent with a mechanism involving transmetallation of the silver catalyst with the borolane reagent.

## *Chapter One*

# **Biological Evaluation of Diaryl Sulfides and Enantioenriched Diarylalkanes as Anti-Breast-Cancer Agents**

### **1.1 Introduction**

Breast cancer is an extremely heterogeneous disease.<sup>1,2</sup> In order to provide varied treatment options for different breast cancer patients, numerous types of pharmaceuticals have been developed. One class of pharmaceuticals includes colchicine and combretastatin A-4 (Figure 1.1a), which act by inhibiting microtubule polymerization so cancer cells cannot divide and proliferate.<sup>3,4</sup> The Jarvo laboratory's goal was to synthesize combretastatin A-4 analogues containing the diaryl motif (**1.3**, Figure 1.1b) in order to investigate their anti-breast-cancer activity. Specifically, we wanted to target diaryl sulfides and diarylalkanes (Figure 1.1c), as these compound classes have been reported to exhibit anti-breast-cancer activity.<sup>5,6</sup> In this Chapter, we disclose the biological evaluation of diaryl sulfides and diarylalkanes synthesized using cross-coupling methods developed by the Jarvo laboratory.

---

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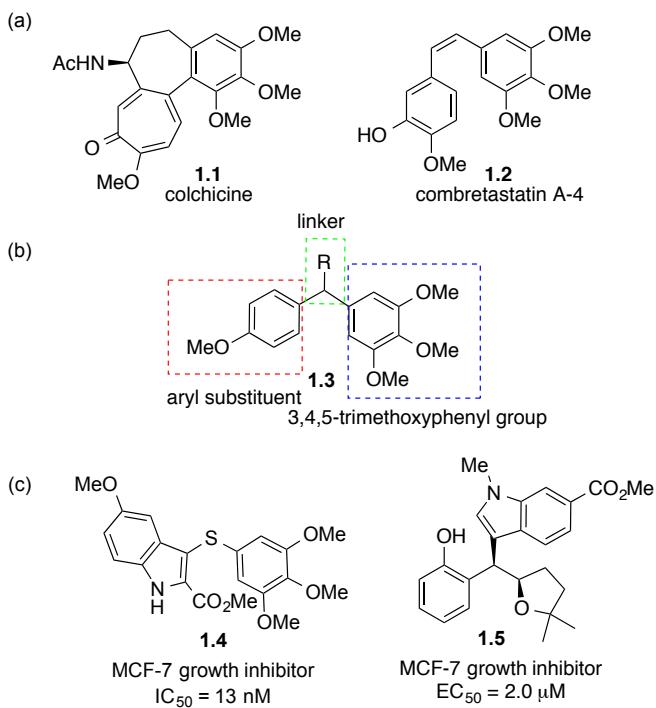
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<sup>4</sup> Tron, G. C.; Pirali, T.; Sorba, G.; Pagliai, F.; Busacca, S.; Genazzani, A. A. *J. Med. Chem.* **2006**, *49*, 3033.

<sup>5</sup> Representative diaryl sulfide possessing anti-breast-cancer activity: De Martino, G.; La Regina, G; Coluccia, A; Edler, M. C.; Barbera, M. C.; Brancale, A.; Wilcox, E.; Hamel, E.; Artico, M.; Silvestri, R. *J. Med. Chem.* **2004**, *47*, 6120.

<sup>6</sup> Representative diarylalkane possessing anti-breast-cancer activity: (a) Pathak, T. P.; Gligorich, K. M.; Welm, B. E.; Sigman, M. S. *J. Am. Chem. Soc.* **2010**, *132*, 7870; (b) Pathak, T. P.; Osiak, J. G.; Vaden, R. M.; Welm, B. E.; Sigman, M. S. *Tetrahedron* **2012**, *68*, 5203.



**Figure 1.1.** Compounds exhibiting anti-breast-cancer activity: (a) colchicine and combretastatin A-4, (b) representative diarylethane, and (c) representative diaryl sulfide and diarylalkane.

Diaryl sulfides abound as potential therapeutic treatments for a variety of diseases, including breast cancer,<sup>5</sup> inflammatory diseases,<sup>7,8</sup> diabetes,<sup>9</sup> HIV,<sup>10</sup> and Alzheimer's disease<sup>11</sup> (Figure 1.2). Thioethers possessing N-heterocyclic functionalities are especially prevalent; however, these sensitive moieties were poorly tolerated in previous syntheses of diaryl sulfides that relied on harsh methodology. We postulated that developing milder reaction conditions

<sup>7</sup> Liu, G.; Link, J. T.; Pei, Z.; Reilly, E. B.; Leitza, S.; Nguyen, B.; Marsh, K. C.; Okasinski, G. F.; von Geldern, T. W.; Ormes, M.; Fowler, K.; Gallatin, M. *J. Med. Chem.* **2000**, *43*, 4025.

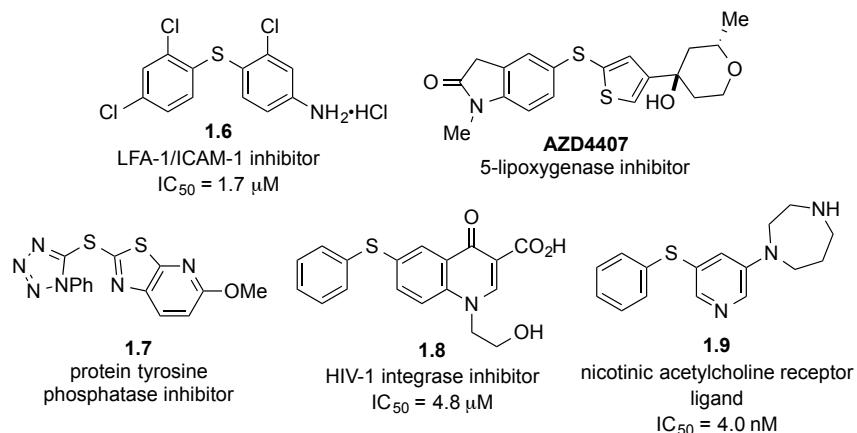
<sup>8</sup> Alcaraz, M. L.; Atkinson, S.; Cornwall, P.; Foster, A. C.; Gill, D. M.; Humphries, L. A.; Keegan, P. S.; Kemp, R.; Merifield, E.; Nixon, R. A.; Noble, A. J.; O'Beirne, D.; Patel, Z. M.; Perkins, J.; Rowan, P.; Sadler, P.; Singleton, J. T.; Tornos, J.; Watts, A. J.; Woodland, I. A. *Org. Process Res. Dev.* **2005**, *9*, 555.

<sup>9</sup> Tang, P. C.; Ramphal, J. Y.; Harris, G. D. Jr.; Nematalla, A. S. Patent WO/1998/27092 A1, June 25, 1998.

<sup>10</sup> Pasquini, S.; Mugnaini, C.; Tintori, C.; Botta, M.; Trejos, A.; Arvela, R. K.; Larhed, M.; Witvrouw, M.; Michiels, M.; Christ, F.; Debyser, Z.; Corelli, F. *J. Med. Chem.* **2008**, *51*, 5125.

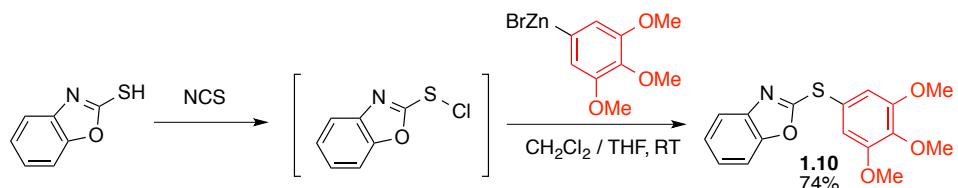
<sup>11</sup> Nielsen, S. F.; Nielsen, E. O.; Olsen, G. M.; Liljefors, T.; Peters, D. *J. Med. Chem.* **2000**, *43*, 2217.

would allow for the synthesis and biological evaluation of functionalized diaryl thioethers structurally analogous to combretastatin A-4. To this end, we recently reported the coupling of in situ-generated sulfenyl chlorides with arylzinc reagents to provide diaryl thioethers in high yield (Scheme 1.1).



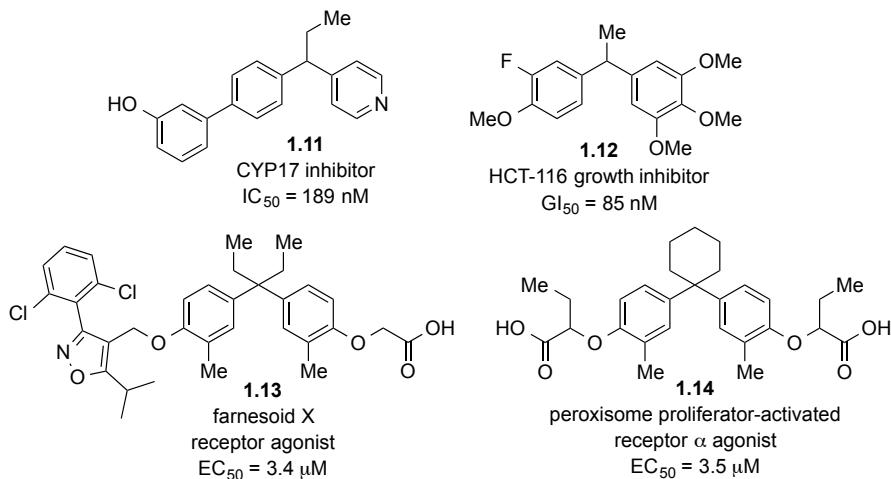
**Figure 1.2.** Representative bioactive diaryl sulfides.

**Scheme 1.1.** Coupling reaction developed by the Jarvo laboratory to synthesize diaryl sulfides for biological evaluation.



In contrast to diaryl sulfides, diarylalkanes are chiral and can be synthesized and evaluated as single enantiomers. The Jarvo laboratory has previously developed nickel-catalyzed stereospecific cross-coupling methodology for the construction of enantioenriched

diaryleethanes.<sup>12</sup> We hypothesized that developing a method to synthesize enantioenriched 1,1-diarylalkanes, pharmacophores found in a range of bioactive molecules (Figure 1.3),<sup>6,13</sup> would provide improved lead compounds for the identification of anti-cancer drugs. Synthesis of both enantiomers of diarylalkanes would also allow us to determine whether one enantiomer exhibits greater activity than the other. Recently, the Jarvo laboratory developed the stereospecific nickel-catalyzed Kumada cross-coupling reaction of benzylic ethers to generate diarylalkanes in high enantiomeric excess (ee) (Scheme 1.2).

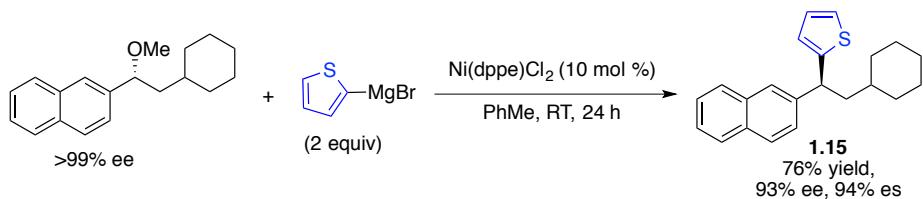


**Figure 1.3.** Representative bioactive diarylalkanes.

<sup>12</sup> (a) Taylor, B. L. H.; Swift, E. C.; Waetzig, J. D.; Jarvo, E. R. *J. Am. Chem. Soc.* **2011**, *133*, 389; (b) Greene, M. A.; Yonova, I. M.; Williams, F. J.; Jarvo, E. R. *Org. Lett.* **2012**, *14*, 4293; (c) Wisniewska, H. M.; Swift, E. C.; Jarvo, E. R. *J. Am. Chem. Soc.* **2013**, *135*, 9083.

<sup>13</sup> Representative examples: (a) as ligands for nuclear receptors, see: Kainuma, M; Kasuga, J.-i.; Hosoda, S; Wakabayashi, K.-i.; Tanatani, A; Nagasawa, K.; Miyachi, H.; Makishima, M; Hashimoto, Y. *Bioorg. Med. Chem. Lett.* **2006**, *16*, 3213; (b) as combretastatin analogues for colon cancer, see: Messaoudi, S.; Hamze, A.; Provost, O.; Tréguier, B.; Rodrigo De Losada, J.; Bignon, J.; Liu, J.-M.; Wdzieczak-Bakala, J.; Thoret, S.; Dubois, J.; Brion, J.-D.; Alami, M. *ChemMedChem.* **2011**, *6*, 488; (c) prostate cancer: Hu, Q. Z.; Yin, L. N.; Jagusch, C.; Hille, U. E.; Hartmann, R. W. *J. Med. Chem.* **2010**, *53*, 5049; (d) diabetes: Kim, R. M.; Parmee, E. R.; Tan, Q.; Yang, C.; Lins, A. R. U. S. Patent 12/227,030, May 11, 2007.

**Scheme 1.2.** Nickel-catalyzed cross-coupling reaction developed by the Jarvo laboratory to synthesize enantioenriched diarylalkanes for biological evaluation.



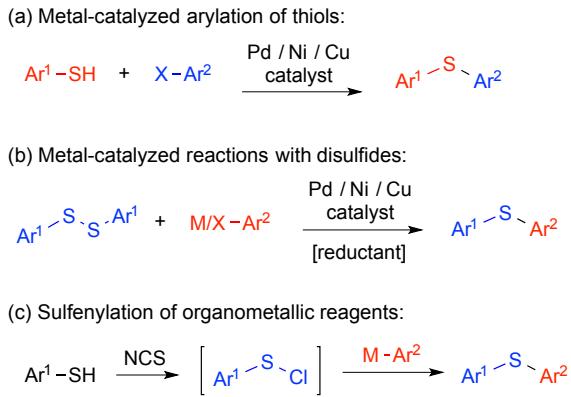
In this Chapter, we report the anti-breast-cancer activity of diaryl sulfides and diarylalkanes synthesized using the methodology in Schemes 1.1 and 1.2. Functionalized diaryl thioethers were formed from in situ-generated sulfenyl chlorides and mild organozinc reagents. A series of heterocyclic diaryl thioethers were designed and prepared as combretastatin A-4 analogues; two of these compounds demonstrated micromolar activity against the MCF-7 breast cancer cell line. We also report the discovery of several enantioenriched diarylalkanes as selective inhibitors of breast cancer cell proliferation.

## 1.2 Synthesis of Diaryl Sulfides via Sulphenyl Chlorides

Synthetic methods for the formation of diaryl thioethers have received significant attention in recent years. Numerous metal-catalyzed reactions for the generation of carbon–sulfur bonds, including copper- and palladium-catalyzed transformations, have been disclosed (Figure 1.4a).<sup>14</sup> However, these reactions typically require elevated temperatures. A milder copper-catalyzed synthesis of diaryl sulfides that proceeds at 0 °C was published in 2013.<sup>15</sup>

<sup>14</sup> (a) For a review, see: Eichman, C. C.; Stambuli, J. P. *Molecules* **2011**, *16*, 590; (b) for a representative example of a copper-catalyzed reaction, see: Kwong, F. Y.; Buchwald, S. L. *Org. Lett.* **2002**, *4*, 3517; (c) for a representative example of a palladium-catalyzed reaction, see: Fernández-Rodríguez, M. A.; Hartwig, J. F. *J. Org. Chem.* **2009**, *74*, 1663.

<sup>15</sup> Uyeda, C.; Tan, Y.; Fu, G. C.; Peters, J. C. *J. Am. Chem. Soc.* **2013**, *135*, 9548.



**Figure 1.4.** Strategies for the synthesis of diaryl sulfides.

Electrophilic sulfur reagents have also been employed (Figure 1.4b and c). Reactions of disulfides with aryl iodides, boronic acids, or silanes proceed with stoichiometric reducing reagents to generate the desired diaryl sulfides (Figure 1.4b).<sup>16</sup> In addition, sulphenyl chlorides formed in situ from aryl thiols and *N*-chlorosuccinimide can be utilized as highly reactive starting materials. Schlosser and co-workers have reported reactions of sulphenyl chlorides with indoles.<sup>17</sup> Recently, a related transformation for the sulfenylation of Grignard reagents, employing in situ-generated sulphenyl chlorides, was reported by Lee and co-workers (Figure 1.4c).<sup>18</sup>

Based on our previous work with *N*-chloroamines,<sup>19</sup> our laboratory concurrently developed the sulfenylation of organozinc reagents as a functional-group tolerant<sup>20</sup> strategy for formation of diaryl sulfides. Our approach provides a synthesis for diaryl thioethers containing

<sup>16</sup> (a) For a review, see: Wladislaw, B.; Marzorati, L.; Di Vitta, C. *Org. Prep. Proc. Int.* **2007**, *39*, 447; (b) for representative examples with aryl iodides, see: Taniguchi, N.; Onami, T. *J. Org. Chem.* **2004**, *69*, 915; (c) for representative examples with aryl boronic acids, see: Taniguchi, N. *Synlett* **2006**, 1351; (d) for representative examples with aryl trimethoxysilanes, see: Luo, P.-S.; Yu, M.; Tang, R.-Y.; Zhong, P.; Li, J.-H. *Tetrahedron Lett.* **2009**, *50*, 1066.

<sup>17</sup> Schlosser, K. M.; Krasutski, A. P.; Hamilton, H. W.; Reed, J. E.; Sexton, K. *Org. Lett.* **2004**, *6*, 819.

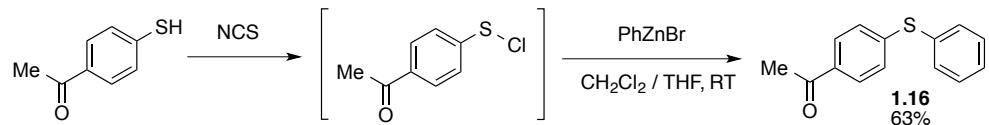
<sup>18</sup> While this work was in progress, Lee and co-workers disclosed a method for the synthesis of diaryl sulfides utilizing in situ-formed sulphenyl chlorides and organomagnesium reagents: Cheng, J.-H.; Ramesh, C.; Kao, H.-L.; Wang, Y.-J.; Chan, C.-C.; Lee, C.-F. *J. Org. Chem.* **2012**, *77*, 10369.

<sup>19</sup> (a) Barker, T. J.; Jarvo, E. R. *J. Am. Chem. Soc.* **2009**, *131*, 15598; (b) Barker, T. J.; Jarvo, E. R. *Angew. Chem., Int. Ed.* **2011**, *50*, 8325.

<sup>20</sup> For demonstration of functional group tolerance of organozinc reagents, see: Bernhardt, S.; Manolikakes, G.; Kunz, T.; Knochel, P. *Angew. Chem., Int. Ed.* **2011**, *50*, 9205.

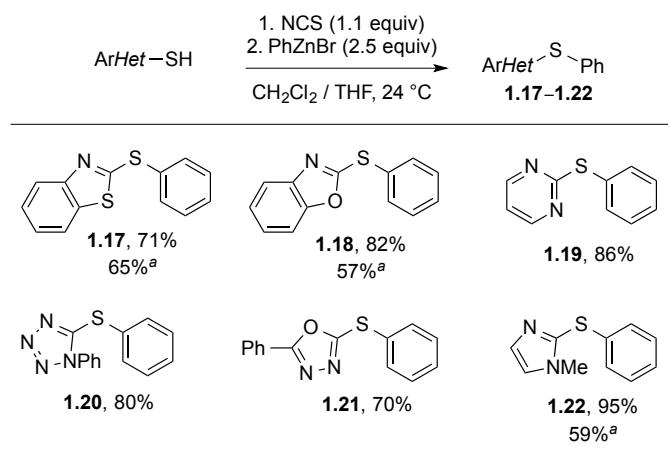
sensitive functional groups, including ketones and N-heterocycles. For example, 1-(4-mercaptophenyl)-ethanone reacted smoothly to provide desired diaryl sulfide **1.16** with no observed competitive addition to the ketone (Scheme 1.3).

**Scheme 1.3.** Formation of ketone-containing diaryl sulfide **1.16**.



A broad range of heterocycles react with phenylzinc bromide to provide good to excellent yields of the corresponding sulfides, including benzothiazole, benzoxazole, pyrimidine, tetrazole, oxadiazole, and imidazole functional groups (Scheme 1.4). Notably, when phenylmagnesium bromide was used instead of phenylzinc bromide, compounds **1.17**, **1.18**, and **1.22** were obtained in diminished yields (65%, 57%, and 59%, respectively), highlighting the improved functional group tolerance of organozinc reagents.

**Scheme 1.4.** Reactivity of heteroaryl thiols.<sup>21</sup>

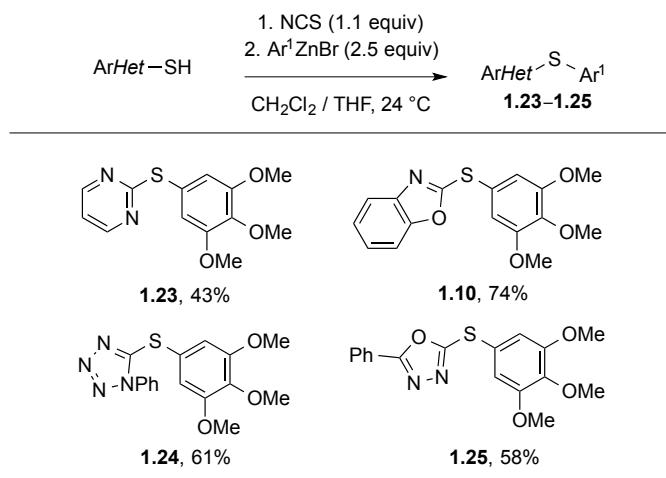


<sup>a</sup>Yield obtained using PhMgBr.

<sup>21</sup> Products **1.17–1.22** were synthesized using phenylzinc bromide by Ivelina Yonova; see: Yonova, I. M.; Osborne, C. A.; Morrisette, N. S.; Jarvo, E. R. *J. Org. Chem.* **2014**, *79*, 1947.

We sought to synthesize combretastatin A-4 analogues<sup>3,4</sup> using our method since it tolerates a diverse range of heterocycles and would further SAR studies of these compounds. Diaryl sulfide analogues of combretastatin containing N-heterocyclic moieties have been reported to be active against MCF-7 breast cancer cell lines (e.g., **1.4**).<sup>5,22–25</sup> We examined reactions of a variety of heteroaryl sulfides with 3,4,5-trimethoxyphenylzinc bromide, biasing our small library of analogues toward inclusion of the 3,4,5-trimethoxyphenyl scaffold, a privileged motif commonly found in anti-cancer compounds that target microtubules.<sup>13b, 26</sup> We were gratified to see that the corresponding arylzinc bromide reacts with a variety of heteroaryl sulfenyl chlorides to afford the respective trimethoxyphenyl-substituted thioethers in modest to good yields (Scheme 1.5).

**Scheme 1.5.** Combretastatin A-4 analogues.<sup>27</sup>



<sup>22</sup> Barbosa, E. G.; Bega, L. A. S.; Beatriz, A.; Sarkar, T.; Hamel, E.; do Amaral, M. S.; de Lima, D. P. *Eur. J. Med. Chem.* **2009**, *44*, 2685.

<sup>23</sup> La Regina, G.; Bai, R.; Rensen, W. M.; Di Cesare, E.; Coluccia, A.; Piscitelli, F.; Famiglini, V.; Reggio, A.; Nalli, M.; Pelliccia, S.; Da Pozzo, E.; Costa, B.; Granata, I.; Porta, A.; Maresca, B.; Soriani, A.; Iannitto, M. L.; Santoni, A.; Li, J.; Cona, M. M.; Chen, F.; Ni, Y.; Brancale, A.; Dondio, G.; Vultaggio, S.; Varasi, M.; Mercurio, C.; Martini, C.; Hamel, E.; Lavia, P.; Novellino, E.; Silvestri, R. *J. Med. Chem.* **2013**, *56*, 123.

<sup>24</sup> Lu, Y.; Li, C.-M.; Wang, Z.; Chen, J.; Mohler, M. L.; Li, W.; Dalton, J. T.; Miller, D. D. *J. Med. Chem.* **2011**, *54*, 4678.

<sup>25</sup> Lee, H.-Y.; Chang, J.-Y.; Nien, C.-Y.; Kuo, C.-C.; Shih, K.-H.; Wu, C.-H.; Chang, C.-Y.; Lai, W.-Y.; Liou, J.-P. *J. Med. Chem.* **2011**, *54*, 8517.

<sup>26</sup> Jordan, A.; Hadfield, J. A.; Lawrence, N. J.; McGown, A. T. *Med. Res. Rev.* **1998**, *18*, 259.

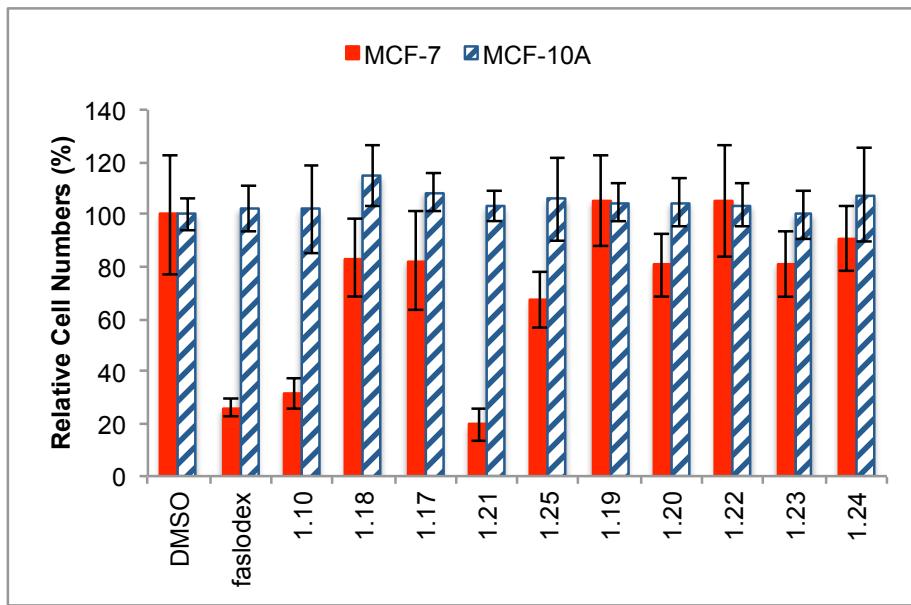
<sup>27</sup> Products **1.10** and **1.23–1.25** were synthesized by Ivelina Yonova; see reference 21.

### 1.3 Biological Evaluation of Diaryl Sulfides

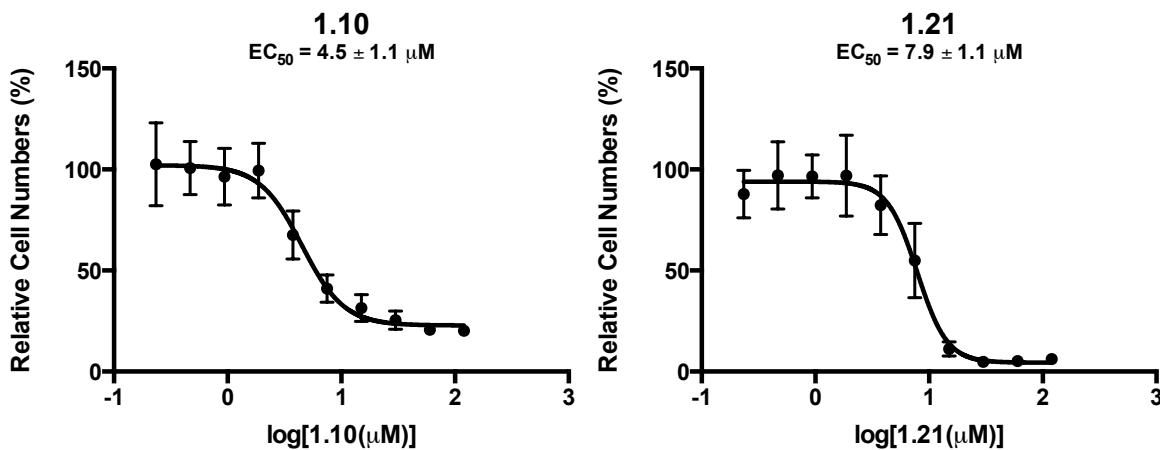
We set out to evaluate these combretastatin A-4 analogues for anti-breast-cancer activity. Select products from Schemes 1.4 and 1.5 were tested for anti-cancer activity against the MCF-7 breast cancer cell line relative to the normal MCF-10A stromal cell line using a proliferation-based procedure (Figure 1.5).<sup>6a</sup> Results are compared to activity of the estrogen receptor antagonist, faslodex (ICI 182,780).<sup>28</sup> Two compounds demonstrated selective inhibition of cancer cell proliferation. Diaryl sulfide **1.10**, containing benzoxazole and 3,4,5-trimethoxyphenyl moieties, was a potent inhibitor of MCF-7 cell proliferation ( $EC_{50} = 4.5 \mu\text{M}$ , Figure 1.6a). In comparison, the simple phenyl analogues **1.18** and **1.17** were inactive. In contrast, trimethoxyphenyl-containing thioether **1.25** performed poorly, while its phenyl analogue **1.21** was a more potent cell proliferation inhibitor ( $EC_{50} = 7.9 \mu\text{M}$ , Figure 1.6b). Further studies to evaluate the biological activity of **1.10** and **1.21** against a broad panel of cancer cell lines are ongoing.

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<sup>28</sup> For a discussion of faslodex and estrogen receptor antagonists, see: (a) Howell, A. *Endocr. Relat. Cancer* **2006**, *13*, 689; (b) Wakeling, A. E.; Dukes, M.; Bowler, J. *Cancer Res.* **1991**, *51*, 3867.



**Figure 1.5.** Evaluation of diaryl sulfides for anti-breast-cancer activity. Compounds were evaluated at 10 μM against breast cancer (MCF-7) and normal breast cell lines (MCF-10A). Cell proliferation is represented as relative cell numbers after treatment, where a low percentage indicates potent anti-proliferative activity for that compound. All data are normalized to the DMSO vehicle control.



**Figure 1.6.** Dose response curves to determine half-maximal effective concentration (EC<sub>50</sub>) values for (a) compound 1.10 and (b) compound 1.21.

## 1.4 Biological Evaluation of Enantioenriched Diarylalkanes

Another class of anti-cancer agents is diarylalkanes. Compounds containing this scaffold have demonstrated bioactivity against a wide range of indications, including breast cancer.<sup>6,13</sup> To access enantioenriched 1,1-diarylalkanes for biological evaluation, our laboratory developed the stereospecific cross-coupling reaction of benzylic ethers with arylmagnesium reagents (Scheme 1.2).<sup>29,30</sup> The Jarvo laboratory has previously developed nickel-catalyzed cross-coupling reactions of aryl Grignard reagents with benzhydryl alcohol derivatives to provide triarylmethanes;<sup>31</sup> however, this method failed to afford satisfactory yields with benzylic alcohol derivatives containing  $\beta$ -hydrogens. Changing catalysts from  $\text{Ni}(\text{cod})_2$  and dppo to  $\text{Ni}(\text{dppe})\text{Cl}_2$  allowed for incorporation of a variety of substituted aryl Grignard reagents to generate the compounds shown in Scheme 1.6.

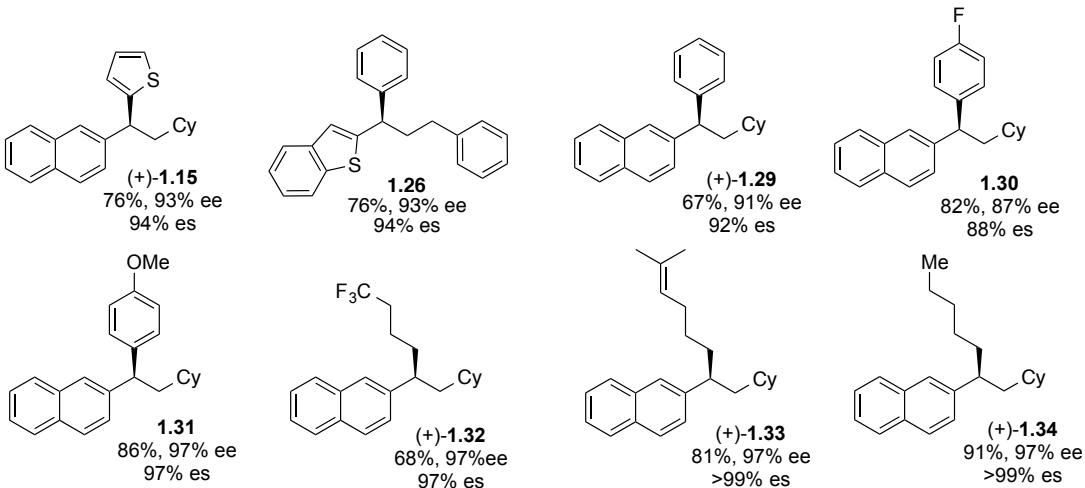
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<sup>29</sup> For a similar disconnection, see: (a) Lopez-Perez, A.; Adrio, J.; Carretero, J. C. *Org. Lett.* **2009**, *11*, 5514; (b) Imao, D.; Glasspoole, B. W.; Laberge, V. S.; Crudden, C. M. *J. Am. Chem. Soc.* **2009**, *131*, 5024; (c) Li, J.; Burke, M. D. *J. Am. Chem. Soc.* **2011**, *133*, 13774; (d) Maity, P.; Shacklady-McAtee, D. M.; Yap, G. P. A.; Sirianni, E. R.; Watson, M. P. *J. Am. Chem. Soc.* **2013**, *135*, 280.

<sup>30</sup> For representative alternative strategies for enantioselective synthesis of 1,1-diarylalkanes, see: (a) reference 6a; (b) Saini, V.; Liao, L.; Wang, Q.; Jana, R.; Sigman, M. S. *Org. Lett.* **2013**, *15*, 5008; (c) Wang, X.; Guram, A.; Caille, S.; Hu, J.; Preston, J. P.; Ronk, M.; Walker, S. *Org. Lett.* **2011**, *13*, 1881; (d) Fessard, T. C.; Andrews, S. P.; Motoyoshi, H.; Carreira, E. M. *Angew. Chem., Int. Ed.* **2007**, *46*, 9331.

<sup>31</sup> (a) Taylor, B. L. H.; Harris, M. R.; Jarvo, E. R. *Angew. Chem., Int. Ed.* **2012**, *51*, 7790; stereospecific Suzuki couplings of arylboronic esters: (b) Harris, M. R.; Hanna, L. E.; Greene, M. A.; Moore, C. E.; Jarvo, E. R. *J. Am. Chem. Soc.* **2013**, *135*, 3303; (c) Zhou, Q.; Srinivas, H. D.; Dasgupta, S.; Watson, M. P. *J. Am. Chem. Soc.* **2013**, *135*, 3307.

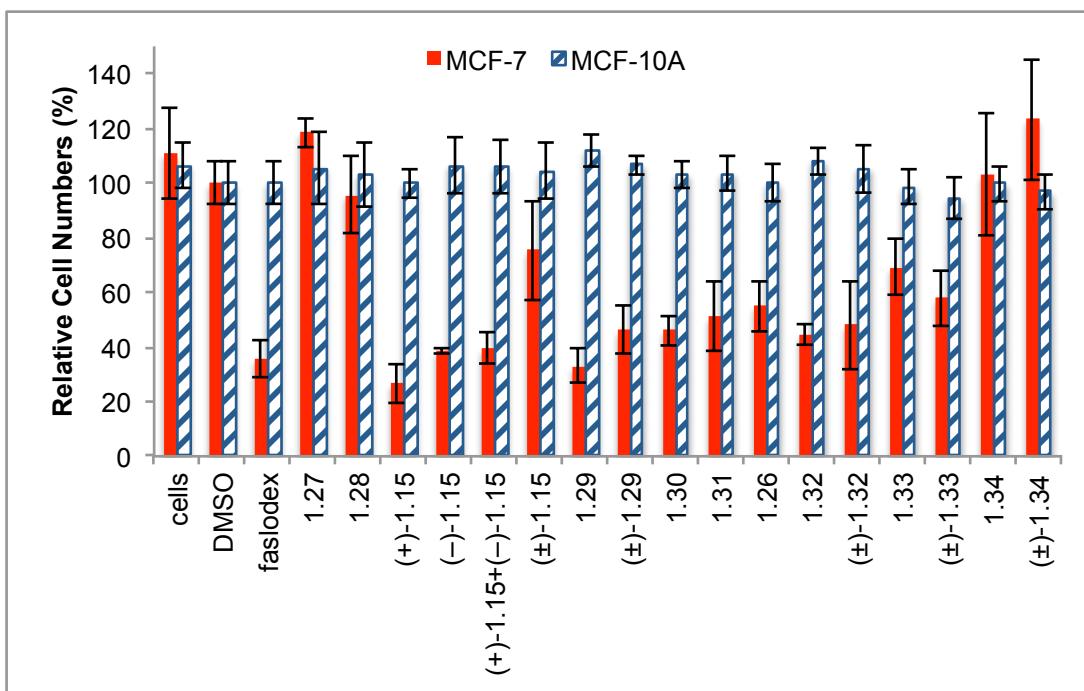
**Scheme 1.6.** 1,1-Diarylalkanes synthesized by stereospecific nickel-catalyzed cross-coupling reactions.<sup>32</sup>



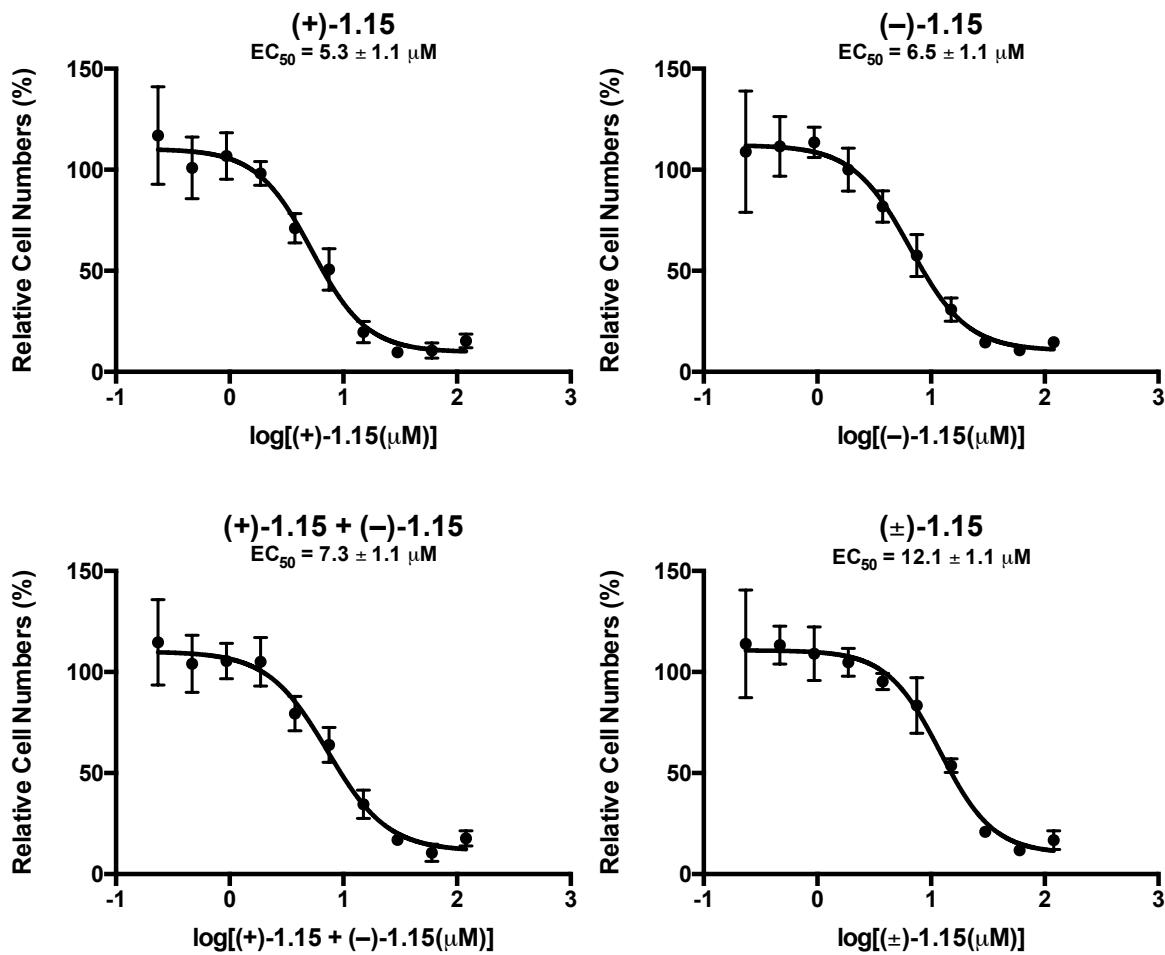
Having obtained a variety of enantioenriched alkanes and diarylalkanes, we set out to evaluate these compounds for biological activity. The cross-coupling products in Scheme 1.6 were tested for selective anti-breast-cancer activity against the MCF-7 breast cancer cell line relative to the normal MCF-10A stromal cell line using a proliferation-based procedure.<sup>6a</sup> Selected results of the broad compound screen are shown in Figure 1.7. Several compounds demonstrated selectivity for the inhibition of breast cancer cell proliferation; results were compared to those obtained with estrogen receptor antagonist faslodex (ICI 182,780).<sup>28</sup> Thiophene-containing diarylalkane (+)-1.15 inhibited MCF-7 cell proliferation with an EC<sub>50</sub> of 5.3 μM (Figure 1.8a). We observed that (−)-1.15 (EC<sub>50</sub> = 6.5 μM, Figure 1.8b) and the racemic mixture (EC<sub>50</sub> = 7.3 μM, Figure 1.8c) were both nearly as efficacious as the (+)-enantiomer. Interestingly, the isomeric diarylalkane 1.26 exhibited a similar level of inhibition. Control experiments confirmed that thiophene (1.27) and benzothiophene (1.28) did not inhibit cell

<sup>32</sup> Compounds in Scheme 1.6 were synthesized by Ivelina Yonova and A. George Johnson; see: Yonova, I. M.; Johnson, A. G.; Osborne, C. A.; Moore, C. E.; Morrissette, N. M.; Jarvo, E. R. *Angew. Chem., Int. Ed.* **2014**, *53*, 2422.

growth. Replacing the thiophene moiety with different aryl groups, such as phenyl (**1.29**), *para*-methoxyphenyl (**1.30**), or *para*-fluorophenyl (**1.31**) resulted in similar selective inhibition of cancer cell proliferation. Furthermore, compounds containing hydrocarbon chains (**1.33** and **1.34**) were much less potent. These results provide new lead compounds with selective inhibition of breast cancer cell growth.



**Figure 1.7.** Evaluation of diarylalkanes for anti-breast-cancer activity. Compounds were evaluated at 10  $\mu\text{M}$  against breast cancer (MCF-7) and normal breast cell lines (MCF-10A). Cell proliferation is represented as relative cell numbers after treatment, where a low percentage indicates potent anti-proliferative activity for that compound. All data are normalized to the DMSO vehicle control.

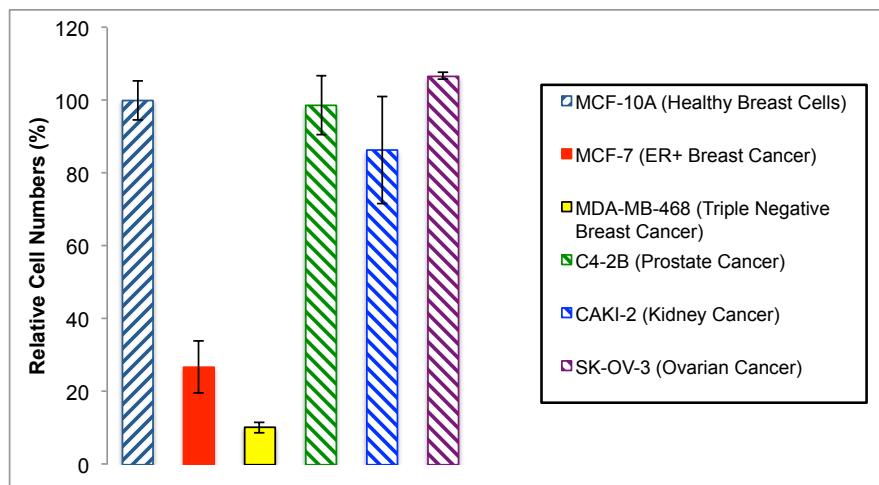


**Figure 1.8.** Dose response curves to determine half-maximal effective concentration (EC<sub>50</sub>) values for (a) enantiomer (+)-1.15, (b) enantiomer (-)-1.15, and (c) the racemic mixture (+)-1.15 + (-)-1.15.

Given the potency of (+)-1.15, we sought to pursue further biological testing of this lead compound. (+)-1.15 was evaluated against a panel of cancer cell lines including prostate, kidney, ovarian, and triple-negative breast cancer cells using an MTT assay.<sup>33</sup> We found (+)-1.15 to be highly selective for anti-breast-cancer activity: it was inactive against prostate, kidney, or ovarian

<sup>33</sup> MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) assay: Alley, M. C.; Scudiero, D. A.; Monks, A.; Hursey, M. L.; Czerwinski, M. J.; Fine, D. L.; Abbott, B. J.; Mayo, J. G.; Shoemaker, R. H.; Boyd, M. R. *Cancer Res.* **1988**, *48*, 589.

cancer cell lines. In contrast, thiophene (+)-**1.15** strongly inhibited proliferation of MDA-MB-468 triple-negative breast cancer cells (Figure 1.9). This activity is significant because triple-negative breast cancer is a particularly aggressive and difficult cancer to treat: it does not express genes for the estrogen receptor (ER), progesterone receptor (PR), or human epidermal growth factor receptor (Her2).<sup>34</sup>



**Figure 1.9.** Evaluation of (+)-**1.15** for anti-cancer activity at 10  $\mu\text{M}$ . Cell proliferation is represented as relative cell numbers after treatment, where a low percentage indicates potent anti-proliferative activity. All data are normalized to the DMSO vehicle control.

We hypothesize that the mechanism of action for our lead compound is independent of estrogen receptor alpha (ER $\alpha$ ) since (+)-**1.15** inhibits both ER+ (MCF-7) and ER- (MDA-MB-468) breast cancer cell lines. However, while the triple-negative MDA-MB-468 line does not express ER $\alpha$ , it still expresses the other protein isoform estrogen receptor beta (ER $\beta$ ).<sup>35</sup> Our working hypothesis is that (+)-**1.15** targets ER $\beta$ , the protein isoform expressed by both breast

<sup>34</sup> (a) Irvin, W. J., Jr.; Carey, L. A. *Eur. J. Cancer* **2008**, *44*, 2799; (b) Carey, L. A. *The Oncologist* **2011**, *16*, 71.

<sup>35</sup> Skliris, G. P.; Leygue, E.; Watson, P. H.; Murphy, L. C. *J. Steroid Biochem. Mol. Biol.* **2008**, *109*, 1.

cancer cell lines we tested. Ongoing studies will refine our understanding of the mechanism by which thiophene (+)-**1.15** demonstrates selectivity for breast cancer.

## 1.5 Biological Evaluation of Triarylmethane Analogues

Having established a robust method for the biological evaluation of diaryl sulfides and diarylalkanes, we sought to evaluate triarylmethane analogues synthesized by the Jarvo laboratory. Triarylmethanes have been established as potent anti-cancer agents.<sup>36</sup> For example, triarylmethanes are analogues of tamoxifen, an estrogen receptor antagonist that has been used to treat breast cancer for more than 30 years.<sup>37</sup> Despite showing promise as anti-breast-cancer agents, access to single enantiomers of diarylmethanes has been limited.

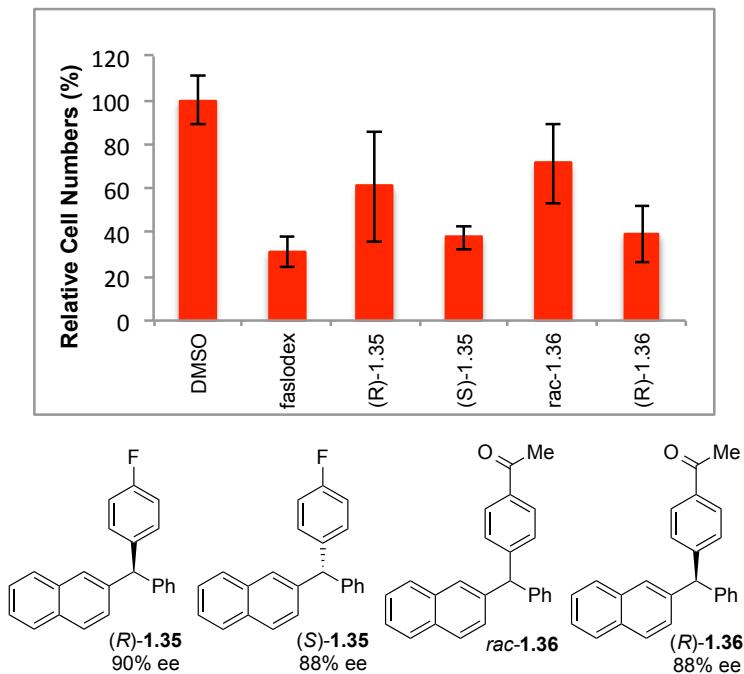
The Jarvo laboratory has developed stereospecific nickel-catalyzed Kumada and Suzuki cross-coupling methodology for the construction of enantioenriched triarylmethanes. The Kumada cross-coupling reaction of benzylic ethers with arylmagnesium reagents generated triarylmethanes in high yield and high es.<sup>31a</sup> The Suzuki cross-coupling reaction of benzylic alcohol derivatives with arylboronic esters allowed for greater functional group tolerance. Notably, this reaction afforded either enantiomer of product from a single chiral intermediate, by judicious choice of ligand.<sup>31b</sup>

We evaluated both enantiomers of triarylmethane analogues for anti-breast-cancer activity using the proliferation-based protocol.<sup>6a</sup> MCF-7 cell growth with 10 µM of compound was compared to activity of the estrogen receptor antagonist faslodex (ICI 182,780).<sup>28</sup> Two compounds exhibited different levels of breast cancer cell inhibition based on which enantiomer was evaluated. (*S*)-**1.35** proved more potent than (*R*)-**1.35**, while (*R*)-**1.36** was more efficacious

<sup>36</sup> Shagupta; Srivastava, A. K.; Sharma, R.; Mishra, R.; Balapure, A. K.; Murthy, P. S. R.; Panda, G. *Bioorg. Med. Chem.* **2006**, *14*, 1497.

<sup>37</sup> Jordan, V. C.; Collins, M. M.; Rowsby, L.; Prestwich, G. J. *Endocrinol.* **1977**, *75*, 305.

than *rac*-**1.36** (Figure 1.10). These preliminary data provide incentive to further develop methodology for the synthesis of enantioenriched triarylmethanes. Future work will determine EC<sub>50</sub> values for these triarylmethanes, as well as confirm their selectivity for the MCF-7 breast cancer line compared to the MCF-10A normal breast line.



**Figure 1.10.** Evaluation of enantioenriched triarylmethanes for anti-breast-cancer activity.<sup>38</sup> Compounds were evaluated at 10 μM against the MCF-7 breast cancer cell line. Cell proliferation is represented as relative cell numbers after treatment, where a low percentage indicates potent anti-proliferative activity for that compound. All data are normalized to the DMSO vehicle control.

<sup>38</sup> Compounds **1.35** and **1.36** were synthesized by Michael Harris; see references 31a and 31b.

## **1.6 Conclusions**

We have developed a mild and efficient protocol for the synthesis of diaryl thioethers. This method tolerates a wide array of heterocyclic moieties and is amenable to the construction of highly functionalized diaryl and diheteroaryl sulfides. Biological studies of select compounds have identified two promising inhibitors of MCF-7 breast cancer cell proliferation. Future efforts will focus on using this methodology to create a larger library of functionalized heterocyclic sulfides and investigating their biological activity against a broad range of cancer cell lines.

The Jarvo laboratory has developed a stereospecific nickel-catalyzed Kumada cross-coupling reaction for the synthesis of enantioenriched 1,1-diarylalkanes. Biological testing of compounds synthesized using this methodology identified several promising leads that exhibit selective inhibition of breast cancer cell proliferation in the low micromolar range.

## **1.7 Experimental Details**

### ***General Procedures***

All reactions were carried out under an atmosphere of N<sub>2</sub> using glassware that was either oven- or flame-dried prior to use. Dichloromethane (CH<sub>2</sub>Cl<sub>2</sub>) and tetrahydrofuran (THF) were degassed with argon and then passed through two 4 x 36 inch columns of anhydrous neutral A-2 alumina (8 x 14 mesh; LaRoche Chemicals; activated under a flow of argon at 350 °C for 12 h) to remove H<sub>2</sub>O. <sup>1</sup>H NMR spectra were recorded on Bruker CRYO-500 (500 MHz <sup>1</sup>H, 125.7 MHz <sup>13</sup>C) or DRX-400 (400 MHz <sup>1</sup>H, 100 MHz <sup>13</sup>C) spectrometers. Proton chemical shifts are reported in ppm ( $\delta$ ) relative to internal tetramethylsilane (TMS,  $\delta$  0.00). Data are reported as follows: chemical shift (multiplicity [singlet (s), broad singlet (br s), doublet (d), doublet of doublets (dd), triplet (t), doublet of triplets (dt), quartet (q), multiplet (m), apparent singlet (ap s), and apparent

doublet (ap d)], coupling constants [Hz], integration. Carbon chemical shifts are reported in ppm ( $\delta$ ) relative to TMS with the respective solvent resonance as the internal standard ( $\text{CDCl}_3$ ,  $\delta$  77.16 ppm). Unless otherwise indicated, NMR data were collected at 25 °C. Infrared spectra were obtained on a Mattson Instruments *Galaxy 5000* spectrometer (thin film) and are reported in terms of frequency of absorption ( $\text{cm}^{-1}$ ). Melting points (m.p.) were obtained using a Mel-Temp melting point apparatus and are uncorrected. High resolution mass spectrometry was performed by the University of California, Irvine Mass Spectrometry Center. Analytical thin-layer chromatography (TLC) was performed using Silica Gel 60 F<sub>254</sub> pre-coated plates (0.25 mm thickness). Visualization was accomplished by irradiation with a UV lamp and/or staining with KMnO<sub>4</sub> solution. Flash chromatography was performed using Silica Gel 60 Å (170-400 mesh) from Fisher Scientific.

Phenylmagnesium bromide<sup>39</sup> and phenylzinc bromide<sup>40</sup> were prepared according to reported procedures. Molarities of organomagnesium and organozinc reagents were determined by titration.<sup>41</sup> N-Chlorosuccinimide (NCS) was recrystallized from benzene and stored in an amber vial for up to two weeks.

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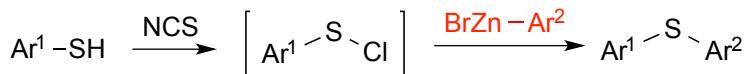
<sup>39</sup> Bollmann, A.; Blann, K.; Dixon, J. T.; Hess, F. M.; Killian, E.; Maumela, H.; McGuinness, D. S.; Morgan, D. H.; Neveling, A.; Otto, S.; Overett, M.; Slawin, A. M. Z.; Wasserscheid, P.; Kuhlmann, S. *J. Am. Chem. Soc.* **2004**, *126*, 14712.

<sup>40</sup> Berman, A. M.; Johnson, J. S. *Synlett* **2005**, 1799.

<sup>41</sup> Krasovskiy, A.; Knochel, P. *Synthesis* **2006**, 890.

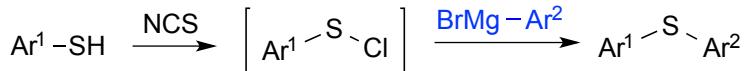
## ***General Procedures for Sulfenylation Reactions***

### ***General Procedure A for Sulfenylation of Arylzinc Reagents***



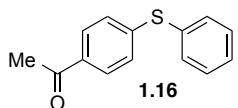
To a solution of NCS (0.073 g, 0.55 mmol) in DCM (1.0 mL) was added thiol (0.50 mmol) and the solution was stirred for 30 min in the absence of direct light. The solution was taken up using a Teflon needle and added drop-wise to a solution of arylzinc reagent in THF (1.25 mmol). Upon completion, as judged by TLC, the reaction mixture was quenched with MeOH, concentrated in vacuo, and the residue was adsorbed onto 3 mL of silica gel and purified by flash column chromatography.

### ***General Procedure B for Sulfenylation of Arylmagnesium Reagents***

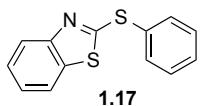


To a solution of NCS (0.073 g, 0.55 mmol) in DCM (1.0 mL) was added thiol (0.50 mmol) and the solution was stirred for 30 min in the absence of direct light. The solution was taken up using a Teflon needle and added drop-wise to a solution of arylmagnesium reagent in THF (1.25 mmol). Upon completion, as judged by TLC, the reaction mixture was quenched with MeOH, concentrated in vacuo, and the residue was adsorbed onto 3 mL of silica gel and purified by flash column chromatography.

### Characterization Data for Products



**4-Phenylsulfanylacetophenone (1.16)** was prepared according to general procedure A from 1-(4-sulfanylphenyl)ethan-1-one (60  $\mu$ L, 0.5 mmol), NCS (0.073 g, 0.55 mmol), and PhZnBr (1.3 mmol, 2.6 mL). Purification by flash column chromatography (5% EtOAc in hexanes) afforded the title compound as a pale yellow solid (0.072 g, 63%). Spectral data were consistent with reported values.<sup>42</sup> **TLC**  $R_f$  = 0.3 (5% EtOAc in hexanes);  **$^1H$  NMR** (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.81 (d,  $J$  = 8.5 Hz, 2H), 7.50–7.47 (m, 2H), 7.40–7.38 (m, 3H), 7.20 (d,  $J$  = 8.5 Hz, 2H), 2.54 (s, 3H);  **$^{13}C$  NMR** (125 MHz, CDCl<sub>3</sub>)  $\delta$  197.3, 145.0, 134.5, 134.0, 132.1, 129.8, 129.0, 128.9, 127.5, 26.5; **IR** (neat) 2922, 1677, 1589, 690 cm<sup>-1</sup>.

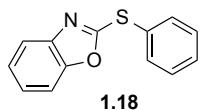


**2-Phenylthio-benzothiazole (1.17)** was prepared according to general procedure A from 2-mercaptopbenzothiazole (0.084 g, 0.50 mmol), NCS (0.073 g, 0.55 mmol), and PhZnBr (1.3 mmol, 1.7 mL). Purification by flash column chromatography (15% EtOAc in hexanes) afforded the title compound as a colorless oil (0.087 g, 71%). Compound **1.17** was also prepared from PhMgBr according to general procedure B to afford 65% yield (determined by  $^1H$  NMR in comparison to the internal standard phenyltrimethylsilane). Spectral data were consistent with reported values.<sup>43</sup> **TLC**  $R_f$  = 0.5 (30% EtOAc in hexanes);  **$^1H$  NMR** (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.87 (d,  $J$  = 8.4 Hz, 1H), 7.72 (m, 2H), 6.63 (d,  $J$  = 8.0 Hz, 1H), 7.52–7.43 (m, 3H), 7.38 (m, 1H),

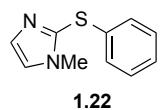
<sup>42</sup> Park, N.; Park, K.; Jang, M.; Lee, S. *J. Org. Chem.* **2011**, *76*, 4371.

<sup>43</sup> Zhou, A.-X.; Liu, X.-Y.; Yang, K.; Zhao, S.-C.; Liang, Y.-M. *Org. Biomol. Chem.* **2011**, *9*, 5456.

7.25 (m, 1H); **<sup>13</sup>C NMR** (125 MHz, CDCl<sub>3</sub>) δ 169.8, 154.0, 135.6, 135.4, 130.6, 130.01, 129.98, 126.2, 124.4, 122.0, 120.9.



**2-Phenylthio-benzoxazole (1.18)** was prepared according to general procedure A from 2-mercaptopbenzoxazole (0.076 g, 0.50 mmol), NCS (0.073 g, 0.55 mmol), and PhZnBr (1.3 mmol, 1.7 mL). Purification by flash column chromatography (15% EtOAc in hexanes) afforded the title compound as a colorless oil (0.093 g, 82%). Compound **1.18** was also prepared from PhMgBr according to general procedure B to afford 57% yield (determined by <sup>1</sup>H NMR in comparison to the internal standard phenyltrimethylsilane). Spectral data were consistent with reported values.<sup>43</sup> **TLC** R<sub>f</sub> = 0.5 (30% EtOAc in hexanes); **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>) δ 7.70 (m, 2H), 7.59 (m, 1H), 7.47–7.42 (m, 3H), 7.39 (m, 1H), 7.27–7.21 (m, 2H); **<sup>13</sup>C NMR** (100 MHz, CDCl<sub>3</sub>) δ 163.4, 152.0, 142.1, 134.5, 130.0, 129.8, 127.3, 124.5, 124.4, 119.2, 110.2.



**1-Methyl-2-(phenylthio)-1H-imidazole (1.22)** was prepared according to general procedure A from 2-mercaptop-1-methylimidazole (0.057 g, 0.50 mmol), NCS (0.073 g, 0.55 mmol), and PhZnBr (1.3 mmol, 1.7 mL). Purification by flash column chromatography (10% EtOAc in hexanes) afforded the title compound as a colorless oil (0.081 g, 95%). Compound **1.22** was also prepared from PhMgBr according to general procedure B to afford 59% yield (determined by <sup>1</sup>H NMR in comparison to the internal standard phenyltrimethylsilane). Spectral data were

consistent with reported values.<sup>43</sup> **TLC**  $R_f$  = 0.2 (30% EtOAc in hexanes); **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.25 (m, 2H), 7.18–7.13 (m, 4H), 7.06 (d, *J* = 1.0 Hz, 1H), 3.62 (s, 3H); **<sup>13</sup>C NMR** (125 MHz, CDCl<sub>3</sub>)  $\delta$  138.1, 135.0, 130.2, 129.3, 128.0, 126.6, 123.9, 33.9.

### ***General Procedures for Biological Experiments with Fluorescence Assay***

Fluorescence assay experiments were performed according to a modified procedure by Sigman and co-workers.<sup>6a</sup>

### ***Materials***

The following reagents were obtained from commercial sources as indicated: Dulbecco's Modified Eagle's Medium (DMEM)/high glucose containing 4.5 g/L glucose and 4.0 mM L-glutamine (HyClone); fetal bovine serum (FBS), heat-inactivated (Omega Scientific); L-glutamine, 200 mM (Gibco); penicillin/streptomycin solution 50X (Mediatech); DMEM/Ham's Nutrient Mixture F12 containing 2.5 mM L-glutamine, 3151 mg/L dextrose, and 55 mg/L sodium pyruvate (Sigma-Aldrich); horse serum (Sigma-Aldrich); 50  $\mu$ M hydrocortisone solution (Sigma-Aldrich); human insulin solution (Sigma-Aldrich); cholera toxin (Sigma-Aldrich); human Epidermal Growth Factor (EGF), recombinant (Sigma-Aldrich); 0.25% Trypsin-EDTA (Gibco); nuclease-free sterile water (Fisher Scientific); molecular biology grade DMSO (Sigma-Aldrich); ICI 182,780 (faslodex) (Tocris Bioscience).

### ***Cell Lines and Culture Conditions***

MCF-7 cells were maintained in DMEM/high glucose supplemented with 10% FBS, L-glutamine, and penicillin/streptomycin. Experiments with MCF-7 cells were performed in

DMEM/high glucose supplemented with 2% FBS, L-glutamine, and penicillin/streptomycin. MCF-10A cells were maintained in standard medium according to a modified recipe by Brugge and co-workers:<sup>44</sup> DMEM/F12 supplemented with 5% horse serum, 10 µg/mL human insulin, 0.5 µg/mL hydrocortisone, 10 ng/mL EGF, 100 ng/mL cholera toxin, and penicillin/streptomycin. Experiments with MCF-10A cells were performed in the same medium.

#### *Evaluation of Compounds Against MCF-7 Cells*

MCF-7 cells were centrifuged in 1X PBS for 20 min, then the pellet was resuspended in DMEM supplemented with 10% FBS and filtered through a 40 µm nylon cell strainer (Fisher Scientific) to prevent clumping. The cells were seeded at 1,500 cells per well in 96-well flat bottom plates suitable for fluorimetry, using 175 µL per well DMEM supplemented with 10% FBS, and grown for 24 h in 5% CO<sub>2</sub> at 37 °C. The compounds (including the faslodex positive control) were dissolved in molecular biology grade DMSO to achieve a 3.5 mM stock solution, then sterile filtered through a 0.45 µm PVDF syringe filter unit (Fisher Scientific). The 3.5 mM stock solutions were subsequently diluted to a final concentration of 10 µM in DMEM supplemented with 2% FBS. Additionally, the corresponding DMSO vehicle control was diluted using the same medium.

After 24 h growth, the cells were treated by replacing the normal media with fresh media containing the individual compounds or vehicle control (day 0). The outer rows of wells were not used to eliminate the possibility of effects due to evaporation of media. The cells were incubated with compound for 48 h then treated again by aspirating the media and adding fresh media containing the compounds and controls (day 2). This procedure was repeated after an additional

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<sup>44</sup> Debnath, J.; Muthuswamy, S. K.; Brugge, J. S. *Methods* **2003**, 30, 256.

48 h (day 4). After incubating a final 24 h, the 96-well plates were rinsed with 1X PBS, blotted dry, and then frozen at –78 °C overnight (day 5). On day 6, cell proliferation was measured using the fluorescence-based CyQUANT Cell Proliferation Assay Kit (Invitrogen).

Fluorimetry analysis was performed according to a modified procedure by McGowan and co-workers.<sup>45</sup> Cells were stained with 200 µL/well of 1X CyQUANT GR dye in cell lysis buffer for 10 min in the dark at room temperature and quantified by fluorimetry at 535 nm with 485 nm excitation. The fluorescence values were normalized to the DMSO vehicle control. The normalized values were plotted as an average ± standard deviation of 6 wells per compound.

#### *Evaluation of Compounds Against MCF-10A Cells*

MCF-10A cells were centrifuged in 1X PBS for 20 min, then the pellet was resuspended in DMEM/F12 and filtered through a 40 µm nylon cell strainer (Fisher Scientific) to prevent clumping. The cells were seeded at 9,000 cells per well in 96-well flat bottom plates suitable for fluorimetry, using 175 µL per well DMEM/F12, and grown for 24 h in 5% CO<sub>2</sub> at 37 °C. The 3.5 mM stock solutions of compound in DMSO were subsequently diluted to a final concentration of 10 µM in DMEM/F12. Additionally, the corresponding DMSO vehicle control was diluted using the same medium.

Addition of compounds was performed as specified above for days 0 through 6. Fluorimetry analysis was performed as specified above for MCF-7 cells, with the exception of staining MCF-10A cells with 200 µL/well of 5X CyQUANT GR dye in cell lysis buffer for 10 min in the dark at room temperature before quantification by fluorimetry. The fluorescence values were

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<sup>45</sup> McGowan, E. M.; Alling, N.; Jackson, E. A.; Yagoub, D.; Haass, N. K.; Allen, J. D.; Martinello-Wilks, R. *PLoS ONE* **2011**, *6*, e20623.

normalized to the DMSO vehicle control. The normalized values were plotted as an average ± standard deviation of 6 wells per compound.

#### *Dose Response of Compounds 1.10 and 1.21*

MCF-7 cells were centrifuged in 1X PBS for 20 min, then the pellet was resuspended in DMEM supplemented with 10% FBS and filtered through a 40 µm nylon cell strainer (Fisher Scientific) to prevent clumping. The cells were seeded at 1,500 cells per well in 96-well flat bottom plates suitable for fluorimetry, using 175 µL per well DMEM supplemented with 10% FBS, and grown for 24 h in 5% CO<sub>2</sub> at 37 °C. The compounds **1.10** and **1.21** were dissolved in molecular biology grade DMSO to achieve a 42 mM stock, then sterile filtered through a 0.45 µm PVDF syringe filter unit (Fisher Scientific). The 42 mM stock solutions in DMSO were subsequently diluted to 120 µM in DMEM supplemented with 2% FBS, and then serially diluted to achieve 10 different concentrations. Additionally, the corresponding DMSO vehicle controls for each concentration were serially diluted using the same medium.

Addition of compounds was performed as specified above for days 0 through 6. Fluorimetry analysis was performed as specified above for the evaluation of compounds against MCF-7 cells. The fluorescence values were normalized to the DMSO vehicle controls corresponding to each concentration. The normalized values were plotted as an average ± standard deviation of 4 wells per concentration and these data were analyzed using the dose response nonlinear regression fitting function (log[inhibitor] vs. response with variable slope (four parameters)) with GraphPad Prism 6.

*Dose Response of Compounds (+)-**1.15**, (-)-**1.15**, (+)-**1.15** + (-)-**1.15**, and (±)-**1.15***

MCF-7 cells were centrifuged in 1X PBS for 20 min, then the pellet was resuspended in DMEM supplemented with 10% FBS and filtered through a 40 µm nylon cell strainer (Fisher Scientific) to prevent clumping. The cells were seeded at 1,500 cells per well in 96-well flat bottom plates suitable for fluorimetry, using 175 µL per well DMEM supplemented with 10% FBS, and incubated with 5% CO<sub>2</sub> at 37 °C for 24 h. The compounds (+)-**1.15**, (-)-**1.15**, (+)-**1.15** + (-)-**1.15**, and (±)-**1.15**<sup>46</sup> were dissolved in molecular biology grade DMSO to achieve a 42 mM stock, then sterile filtered through a 0.45 µm PVDF syringe filter unit (Fisher Scientific). The 42 mM stock solutions in DMSO were subsequently diluted to 120 µM in DMEM supplemented with 2% FBS, and then serially diluted to achieve 10 different concentrations. Additionally, the corresponding DMSO vehicle controls for each concentration were serially diluted using the same medium.

Addition of compounds was performed as specified above for days 0 through 6. Fluorimetry analysis was performed as specified above for the evaluation of compounds against MCF-7 cells. The fluorescence values were normalized to the DMSO vehicle controls corresponding to each concentration. The normalized values were plotted as an average ± standard deviation of 4 wells per concentration and these data were analyzed using the dose response nonlinear regression fitting function (log[inhibitor] vs. response with variable slope (four parameters)) with GraphPad Prism 6.

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<sup>46</sup> (+)-**1.15** and (-)-**1.15** are samples of each single enantiomer with 93% ee and 92% ee, respectively; (±)-**1.15** is a sample of the racemic standard; (+)-**1.15** + (-)-**1.15** is a sample containing an equimolar ratio of the two enantiomers, combined in DMSO prior to biological testing.

### ***General Procedures for Biological Experiments with MTT Assay***

MTT assay experiments were performed according to a modified procedure by Alley and co-workers.<sup>33</sup>

### ***Materials***

The following reagents were obtained from commercial sources as indicated: RPMI 1640 medium (HyClone); McCoy's 5A medium (HyClone); fetal bovine serum (FBS), heat-inactivated (Omega Scientific); L-glutamine, 200 mM (Gibco); penicillin/streptomycin solution 50X (Mediatech); amphotericin B (HyClone); DMEM/Ham's Nutrient Mixture F12 containing 2.5 mM L-glutamine, 3151 mg/L dextrose, and 55 mg/L sodium pyruvate (Sigma-Aldrich); horse serum (Sigma-Aldrich); 50 µM hydrocortisone solution (Sigma-Aldrich); human insulin solution (Sigma-Aldrich); cholera toxin (Sigma-Aldrich); human Epidermal Growth Factor (EGF), recombinant (Sigma-Aldrich); 0.25% Trypsin-EDTA (Gibco); nuclease-free sterile water (Fisher Scientific); molecular biology grade DMSO (Sigma-Aldrich); MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) (Sigma-Aldrich).

### ***Cell Lines and Culture Conditions***

**MCF-10A** cells were maintained in standard medium according to a modified recipe by Brugge and co-workers.<sup>44</sup> DMEM/F12 supplemented with 5% horse serum, 10 µg/mL human insulin, 0.5 µg/mL hydrocortisone, 10 ng/mL EGF, 100 ng/mL cholera toxin, and penicillin/streptomycin. Experiments with MCF-10A cells were performed in the same medium.

**MCF-7** cells were maintained in RPMI 1640 supplemented with 10% FBS, L-glutamine, and penicillin/streptomycin. Experiments with MCF-7 cells were performed in the same medium.

**MDA-MB-468** cells were maintained in RPMI 1640 supplemented with 10% FBS, L-glutamine, penicillin/streptomycin, and amphotericin B. Experiments with MDA-MB-468 cells were performed in the same medium.

**C4-2B** cells were maintained in RPMI 1640 supplemented with 10% FBS, L-glutamine, and penicillin/streptomycin. Experiments with C4-2B cells were performed in the same medium.

**CAKI-2** cells were maintained in McCoy's 5A supplemented with 10% FBS, L-glutamine, and penicillin/streptomycin. Experiments with CAKI-2 cells were performed in the same medium.

**SK-OV-3** cells were maintained in RPMI 1640 supplemented with 10% FBS, L-glutamine, and penicillin/streptomycin. Experiments with SK-OV-3 cells were performed in the same medium.

### ***Evaluation of Compounds Against Cell Lines***

#### ***Preparation of Cell Lines***

**MCF-10A** cells were centrifuged in DMEM/F12 at 1.0 rcf for 8 min, then the pellet was resuspended in DMEM/F12. The cells were seeded at 20,000 cells per well in 24-well flat bottom plates, using 500 µL per well DMEM/F12, and grown for 24 h in 5% CO<sub>2</sub> at 37 °C.

**MCF-7** cells were centrifuged in RPMI at 1.0 rcf for 8 min, then the pellet was resuspended in RPMI. The cells were seeded at 20,000 cells per well in 24-well flat bottom plates, using 500 µL per well RPMI, and grown for 24 h in 5% CO<sub>2</sub> at 37 °C.

**MDA-MB-468** cells were centrifuged in RPMI with amphotericin B at 1.0 rcf for 8 min, then the pellet was resuspended in RPMI with amphotericin B. The cells were seeded at 20,000 cells per well in 24-well flat bottom plates, using 500 µL per well RPMI with amphotericin B, and grown for 24 h in 5% CO<sub>2</sub> at 37 °C.

**C4-2B** cells were centrifuged in RPMI at 1.0 rcf for 8 min, then the pellet was resuspended in RPMI. The cells were seeded at 20,000 cells per well in 24-well flat bottom plates, using 500 µL per well RPMI, and grown for 24 h in 5% CO<sub>2</sub> at 37 °C.

**CAKI-2** cells were centrifuged in McCoy's 5A at 1.0 rcf for 8 min, then the pellet was resuspended in McCoy's 5A. The cells were seeded at 12,500 cells per well in 24-well flat bottom plates, using 500 µL per well McCoy's 5A, and grown for 24 h in 5% CO<sub>2</sub> at 37 °C.

**SK-OV-3** cells were centrifuged in RPMI at 1.0 rcf for 8 min, then the pellet was resuspended in RPMI. The cells were seeded at 20,000 cells per well in 24-well flat bottom plates, using 500 µL per well RPMI, and grown for 24 h in 5% CO<sub>2</sub> at 37 °C.

#### *Preparation of Compounds*

The compounds were dissolved in molecular biology grade DMSO to achieve a 100 mM stock solution, and then sterile filtered through a 0.45 µm PVDF syringe filter unit (Fisher Scientific). The 100 mM stock solutions were subsequently diluted to 20 mM stock solutions, then diluted to a final concentration of 10 µM in the corresponding medium for each cell line. Additionally, the corresponding DMSO vehicle control was diluted using the same medium.

After 24 h growth, the medium was carefully aspirated from each of the wells containing cells. The cells were treated by replacing the normal medium with 600 µL fresh medium per well containing the individual compounds or vehicle control (day 0). The plates were gently agitated, then the cells were incubated with compound at 37 °C with 5% CO<sub>2</sub> for 72 h. On day 3, cell proliferation was measured using the MTT absorbance assay.

#### *MTT Absorbance Assay*

Quantitative analysis was performed according to a modified procedure by Alley and co-workers.<sup>33</sup> After 72 h incubation with compound, 200 µL of MTT dye solution (3 mg MTT/ 1 mL PBS) was carefully added to each well containing cells. The plates were incubated at 37 °C for 1 h 15 min, after which the MTT dye was aspirated from each well. 300 µL of MTT dissolve solution (4% 1M HCl in IPA) was added to each well and the plates were agitated 15 min. From each well, 200 µL of cell lysate solution was carefully transferred to a new well in a clear-bottom 96-well plate suitable for UV-Vis spectroscopy. Cells were quantified by absorbance at 570 nm. The absorbance values were normalized to the DMSO vehicle control. The normalized values were plotted as an average ± standard deviation of 3 wells per compound.

## *Chapter Two*

### **Synthesis and Biological Evaluation of Enantioenriched Diarylethanes as Tubulin Polymerization Disruptors**

#### **2.1 Introduction**

Microtubules are essential for the cell cycle: they facilitate mitosis, transport, and maintenance of cell structures.<sup>1</sup> Monomers of the protein tubulin assemble together to form microtubule polymers, which are dynamic, growing or shrinking as required to perform tasks around the cell.<sup>2</sup> Due to their significance in cell division, microtubules are an important focus for anti-cancer therapeutics. Drugs targeting microtubules prevent cancer cells from proliferating and because cancer cells divide more rapidly than healthy cells, they are more impacted by anti-mitotic therapeutics.

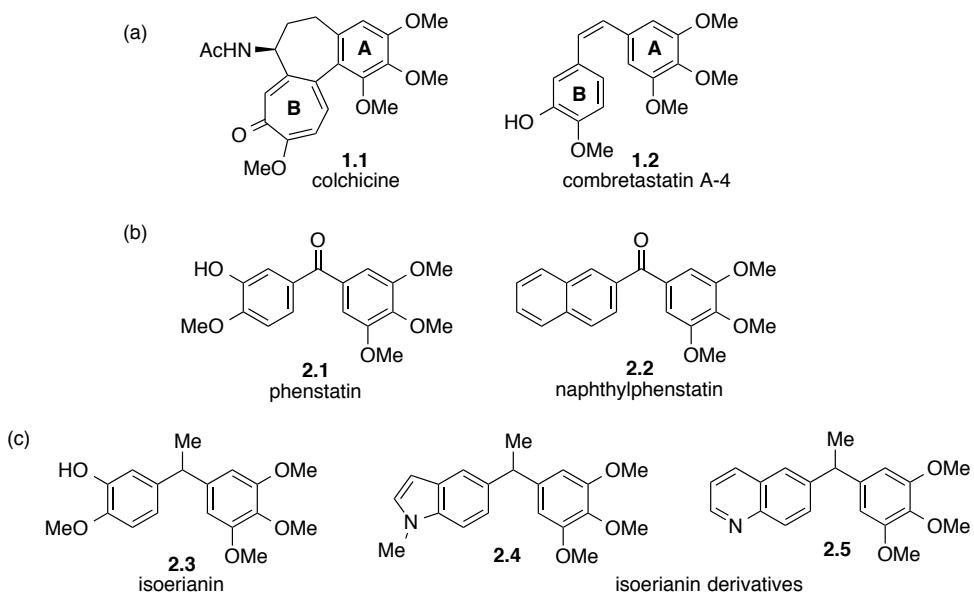
There are three main classes of anti-mitotic drugs: the vinca alkaloids, the taxanes, and compounds related to colchicine. These mitotic inhibitors limit tumor proliferation by preventing cancer cell division, but each drug category targets microtubules in a different manner. The vinca alkaloids, for example, destroy the mitotic spindles to prevent cell division from taking place. The taxane family includes natural products such as paclitaxel (Taxol), which act by stabilizing microtubules so mitosis cannot occur. Compounds that interact with the colchicine binding site of tubulin prevent assembly of tubulin subunits into functional microtubules, and thus also inhibit cell division. In this Chapter we will focus on synthesis of compounds in the third class of anti-mitotic drugs, colchicine analogues.

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<sup>1</sup> Jordan, A.; Hadfield, J. A.; Lawrence, N. J.; McGown, A. T. *Med. Res. Rev.* **1998**, *18*, 259.

<sup>2</sup> Jordan, M. A.; Wilson, L. *Nat. Rev. Cancer* **2004**, *4*, 253.

Multiple tubulin-binding agents that target the colchicine binding site have been identified. The ancient anti-gout remedy colchicine (**1.1**) was one of the first molecules discovered to bind to tubulin. Like a number of related diaryl natural products, including combretastatin A-4 (**1.2**, Figure 2.1a), it inhibits tubulin polymerization by interacting with  $\beta$ -tubulin at a site now known as the colchicine binding pocket.<sup>3</sup> 3,4,5-Trimethoxyphenyl ring A is necessary to retain activity as a tubulin disruptor, but the identity of ring B is more modular; for example, phenstatin and naphthylphenstatin (Figure 2.1b) are well tolerated in the colchicine binding site.<sup>4</sup> Diaryl analogues containing ethylene<sup>5</sup> or ethane<sup>6</sup> linkers are also active tubulin disruptors. In particular, racemic isoerianin derivatives (Figure 2.1c) are potent inhibitors of cancer cell division.<sup>6</sup>



**Figure 2.1.** Tubulin-binding agents that interact with the colchicine binding site.

<sup>3</sup> Álvarez, R.; Álvarez, C.; Mollinedo, F.; Sierra, B. G.; Medarde, M.; Peláez, R. *Bioorg. Med. Chem.* **2009**, *17*, 6422.

<sup>4</sup> Álvarez, C.; Álvarez, R.; Corchete, P.; Pérez-Melero, C.; Peláez, R.; Medarde, M. *Bioorg. Med. Chem.* **2008**, *16*, 8999.

<sup>5</sup> Hamze, A.; Giraud, A.; Messaoudi, S.; Provot, O.; Peyrat, J.-F.; Bignon, J.; Liu, J.-M.; Wdzieczak-Bakala, J.; Thoret, S.; Dubois, J.; Brion, J.-D.; Alami, M. *Chem. Med. Chem.* **2009**, *4*, 1912.

<sup>6</sup> Messaoudi, S.; Hamze, A.; Provot, O.; Tréguier, B.; Rodrigo De Losada, J.; Bignon, J.; Liu, J.-M.; Wdzieczak-Bakala, J.; Thoret, S.; Dubois, J.; Brion, J.-D.; Alami, M. *Chem. Med. Chem.* **2011**, *6*, 488.

While the 1,1-diarylethane motif has been established as a potent tubulin polymerization inhibitor, this class of compound has yet to be synthesized and examined as single enantiomers. It is well known that the difference between enantiomers of pharmaceuticals can dramatically affect their binding ability and efficacy as drugs.<sup>7</sup> Methodology developed in the Jarvo laboratory would allow efficient access to both enantiomers of bioactive diarylethanes. We hypothesized that use of enantioenriched diarylethanes would further refine three-dimensional structure-activity relationships and provide improved lead compounds for the identification of anti-cancer drugs.

The Jarvo laboratory has previously demonstrated the stereospecific synthesis of several 1,1-diarylethanes via the nickel-catalyzed Kumada cross-coupling reactions of diaryl alcohol derivatives. The cross-coupling reaction of enantioenriched diaryl ether (*S*)-**2.6** with methylmagnesium iodide provided (*R*)-**2.7**, a known tubulin-binding compound,<sup>4</sup> in high enantiomeric excess (ee) and enantiospecificity (es) (Scheme 2.1a).<sup>8,9</sup> Less reactive substrates containing methoxyphenyl substituents, such as (*S*)-**2.8**, underwent smooth cross-coupling reactions when a methoxyethyl ether leaving group was used in the transformation (Scheme 2.1b).<sup>10</sup> We postulated that we could apply this cross-coupling methodology to prepare a series of trimethoxyphenyl-substituted diarylethane analogues for tubulin-binding studies.

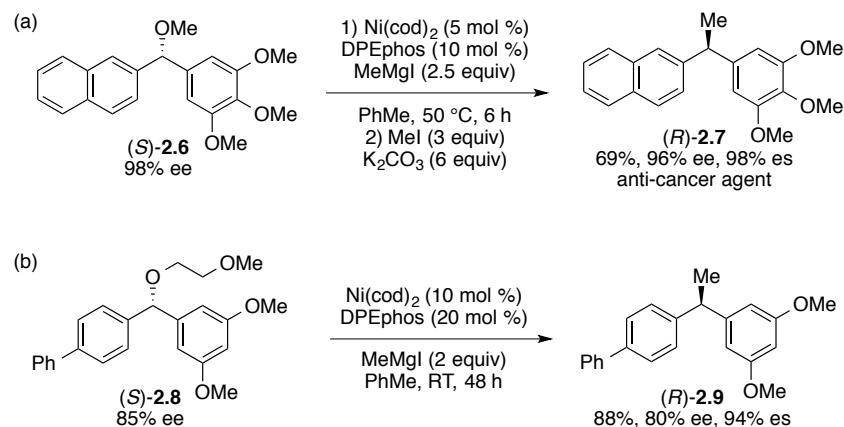
<sup>7</sup> Ariëns, E. J. *Eur. J. Clin. Pharmacol.* **1984**, *26*, 663.

<sup>8</sup> Taylor, B. L. H.; Swift, E. C.; Waetzig, J. D.; Jarvo, E. R. *J. Am. Chem. Soc.* **2011**, *133*, 389.

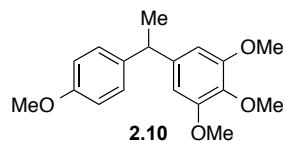
<sup>9</sup> es = ee<sub>product</sub>/ee<sub>starting material</sub>; see: Denmark, S. E.; Vogler, T. *Chem.-Eur. J.* **2009**, *15*, 11737.

<sup>10</sup> Greene, M. A.; Yonova, I. M.; Williams, F. J.; Jarvo, E. R. *Org. Lett.* **2012**, *14*, 4293.

**Scheme 2.1.** Enantiospecific nickel-catalyzed cross-coupling methods for the synthesis of methoxyphenyl-substituted 1,1-diarylethanes.



In this Chapter, we report the synthesis of both enantiomers of naphthyl-substituted diarylethane **2.7** using nickel-catalyzed Kumada cross-coupling methods. Subsequent biological evaluation for tubulin polymerization disruption was performed to determine relative tubulin-binding affinities. We hypothesized that one enantiomer would exhibit greater potency as a tubulin polymerization disruptor. In addition, we report the synthesis and biological evaluation of diarylethane **2.10**. We chose to target this more challenging diarylethane in order to expand our cross-coupling methodology, and because this compound is an anti-cancer agent: **2.10** has demonstrated cytotoxic activity toward colon carcinoma cells<sup>6</sup> and has been identified as a smallpox anti-viral agent.<sup>11</sup> Finally, we report the first synthesis of indole-substituted diarylethane **2.4** achieved using a cross-coupling reaction.



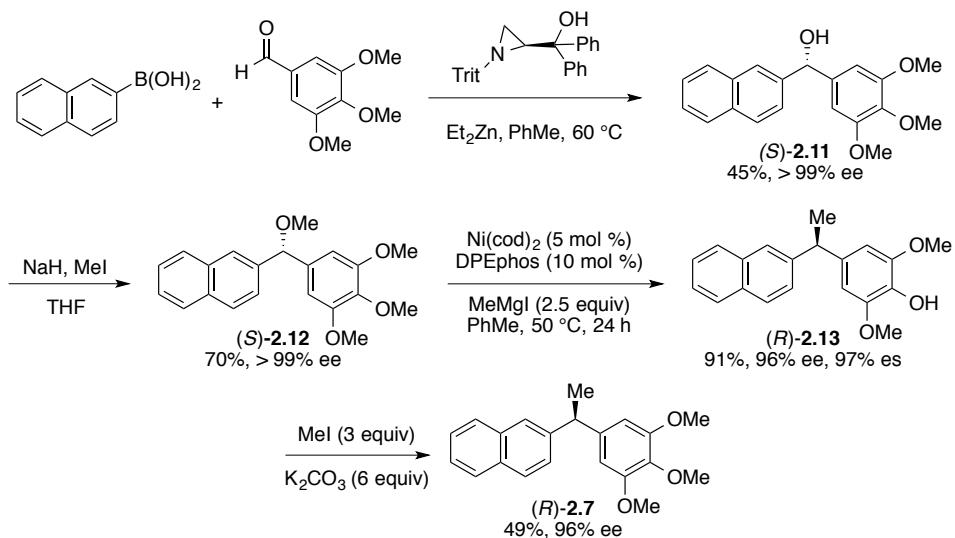
**Figure 2.2.** Structure of diarylethane **2.10**.

<sup>11</sup> Cheltsov, A. V.; Aoyagi, M.; Aleshin, A.; Yu, E. C.-W.; Gilliland, T.; Zhai, D.; Bobkov, A. A.; Reed, J. C.; Liddington, R. C.; Abagyan, R. *J. Med. Chem.* **2010**, *53*, 3899.

## 2.2 Synthesis of Enantioenriched Naphthyl-Substituted Diarylethanes

The synthesis of naphthyl-substituted diarylethane (*R*)-**2.7** was undertaken according to the procedure shown in Scheme 2.2.<sup>8</sup> The stereogenic center was installed via an asymmetric arylation reaction of 3,4,5-trimethoxybenzaldehyde with 2-naphthylboronic acid using a chiral aziridine catalyst.<sup>12</sup> Protection of diaryl alcohol (*S*)-**2.11** with iodomethane proceeded in 70% yield with retention of stereochemistry to generate diaryl methyl ether (*S*)-**2.12** for the Kumada cross-coupling reaction.

**Scheme 2.2.** Synthesis of diarylethane (*R*)-**2.7**.



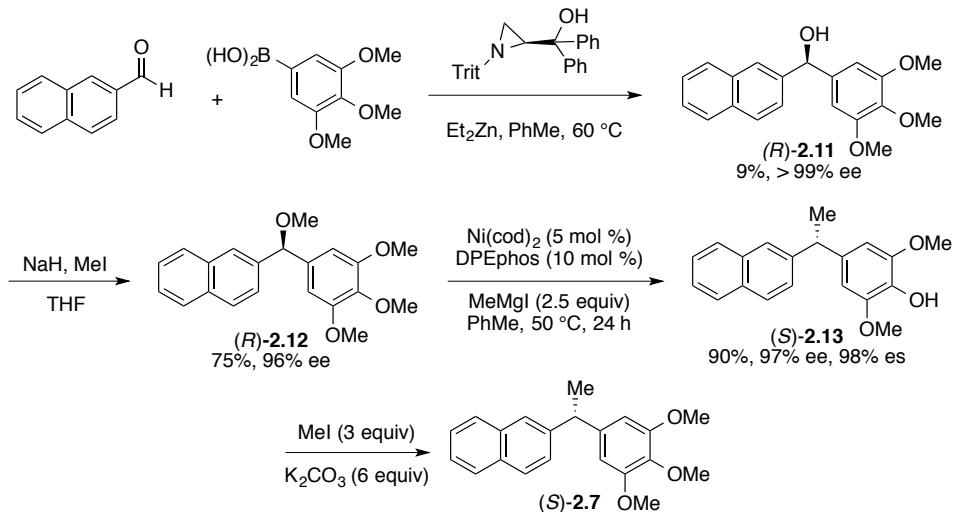
Upon subjecting (*S*)-**2.12** to the cross-coupling conditions, the reaction formed demethylated compound (*R*)-**2.13** in 91% yield and 96% ee; this phenol could be remethylated by treatment with iodomethane in the presence of potassium carbonate to furnish the intended diarylethane (*R*)-**2.7** (Scheme 2.2). It was proposed that during the course of the cross-coupling reaction, methoxy substituents *ortho* to one another chelate excess magnesium ions, activating the *para*-methoxyphenyl group for cleavage by methylmagnesium iodide.<sup>8</sup> While previous work in the Jarvo laboratory showed only partial demethylation after 6 hours, we observed quantitative

<sup>12</sup> Braga, A. L.; Paixão, M. W.; Westermann, B.; Schneider P. H.; Wessjohann, L. A. *J. Org. Chem.* **2008**, *73*, 2879.

demethylation after longer reaction times, an observation consistent with the coordination of excess magnesium ions in solution.

The synthetic sequence to obtain enantiomer (*S*)-**2.7** has never been reported, and was performed according to Scheme 2.3. By switching the identity of the arylboronic acid and aldehyde, the same enantiomer of chiral aziridine catalyst can be used to generate the other enantiomer of **2.11**.<sup>12</sup> The low yield of (*R*)-**2.11** after asymmetric arylation can be accounted for by the poor solubility of 3,4,5-trimethoxyphenylboronic acid in toluene; additional recrystallizations were required to obtain material of suitable ee. Methylation to generate benzylic ether (*R*)-**2.12** followed by the nickel-catalyzed cross-coupling reaction formed phenol (*S*)-**2.13** in 80% yield and 97% ee. As with (*R*)-**2.13**, subsequent remethylation using iodomethane resulted in the desired diarylethane (*S*)-**2.7**.<sup>13</sup>

**Scheme 2.3.** Synthesis of diarylethane (*S*)-**2.7**.



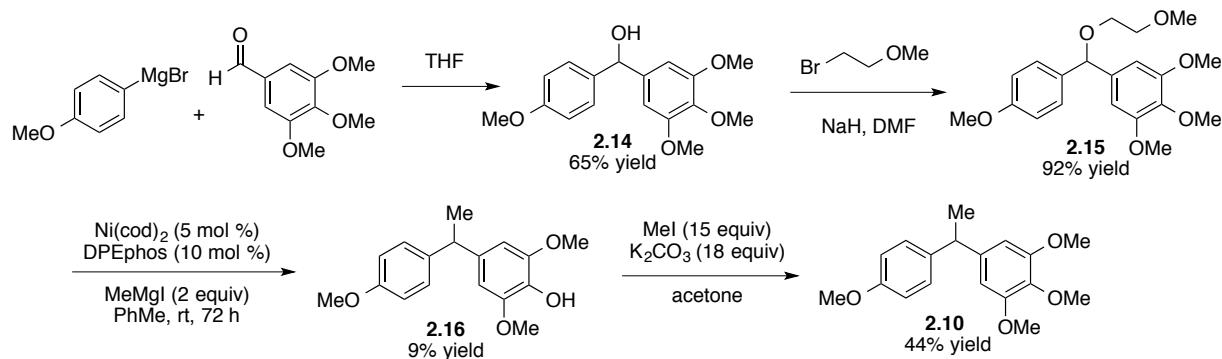
<sup>13</sup> Remethylation of (*S*)-**2.13** to obtain (*S*)-**2.7** was performed by LuisRuben Martinez.

### 2.3 Synthesis of *para*-Methoxyphenyl-Substituted Diarylethane **2.10**

We also prepared diarylethane **2.10**, an analogue of diarylethane **2.7**, for tubulin-binding studies. Guided by the Jarvo laboratory's previous reaction design for substrates lacking an extended aromatic system (Scheme 2.1b), we postulated that the cross-coupling reaction to obtain **2.10** would require a more activated leaving group.<sup>10</sup> As with (*S*)-**2.8**, we designed substrate **2.15** to contain a methoxyethyl ether leaving group instead of a simple methyl ether. We hypothesized that the methoxyethyl ether group would chelate excess magnesium ions to provide increased activation for oxidative addition.

The racemic synthesis of **2.10** involved addition of *para*-methoxyphenylmagnesium bromide to 3,4,5-trimethoxybenzaldehyde in order to generate alcohol **2.14** (Scheme 2.4). Alkylation using 2-(bromoethyl)methyl ether formed **2.15** in 92% yield.<sup>10</sup> Upon subjecting **2.15** to the cross-coupling conditions, the reaction yielded demethylated compound **2.16**, which could be remethylated to provide **2.10** (vide supra).

**Scheme 2.4.** Racemic synthesis of diarylethane **2.10**.



Initial attempts to develop the cross-coupling reaction resulted in low yield and significant background reaction, which lead us to conclude that the reaction of electron-rich substrates would be challenging (Scheme 2.4 and Table 2.1, entries 1 and 2). Electron-donating *para*-methoxyphenyl groups can facilitate carbocation formation, which leads to racemization of

starting material. We reasoned that investigating other catalyst systems with increased nickel loading would enable the cross-coupling reaction to outcompete the background reaction. Nickel catalysts ligated by *rac*-BINAP and CyDPEphos produced the highest yields at room temperature (entries 5 and 6). In order to improve reactivity, the cross-coupling reaction was performed at elevated temperatures. Heating the reaction to 40 °C afforded higher yields while keeping background reactivity low (entry 7); however, at 60 °C we observed background reactivity comparable to that of the nickel-catalyzed reaction (entry 11). The optimized reaction conditions, 24 hours at 40 °C with DPEphos, produced phenol **2.16** in 49% yield (entry 8).

**Table 2.1.** Ligand screen and effect of increased temperature and reaction time on the cross-coupling reaction of **2.15**.

entry	ligand	ligand loading (%)	time (h)	temp (°C)	SM yield (%) <sup>a</sup>	yield <b>2.16</b> (%) <sup>a</sup>
1 <sup>b</sup>	none, no Ni	--	72	rt	31	8
2 <sup>b</sup>	DPEphos	10	72	rt	37	9
3	DPEphos	10	48	rt	65	16
4	Xantphos	10	48	rt	61	22
5	<i>rac</i> -BINAP	10	48	rt	62	24
6	CyDPEphos	10	48	rt	56	25
7	<b>none, no Ni</b>	--	6	40	71	5
8	<b>DPEphos</b>	<b>20</b>	<b>24</b>	<b>40</b>	<b>28</b>	<b>49</b>
9	<i>rac</i> -BINAP	20	6	40	15	31
10	CyDPEphos	20	6	40	27	33
11	none, no Ni	--	6	60	50	54
12	<i>rac</i> -BINAP	20	6	60	37	nd
13	CyDPEphos	20	6	60	29	51

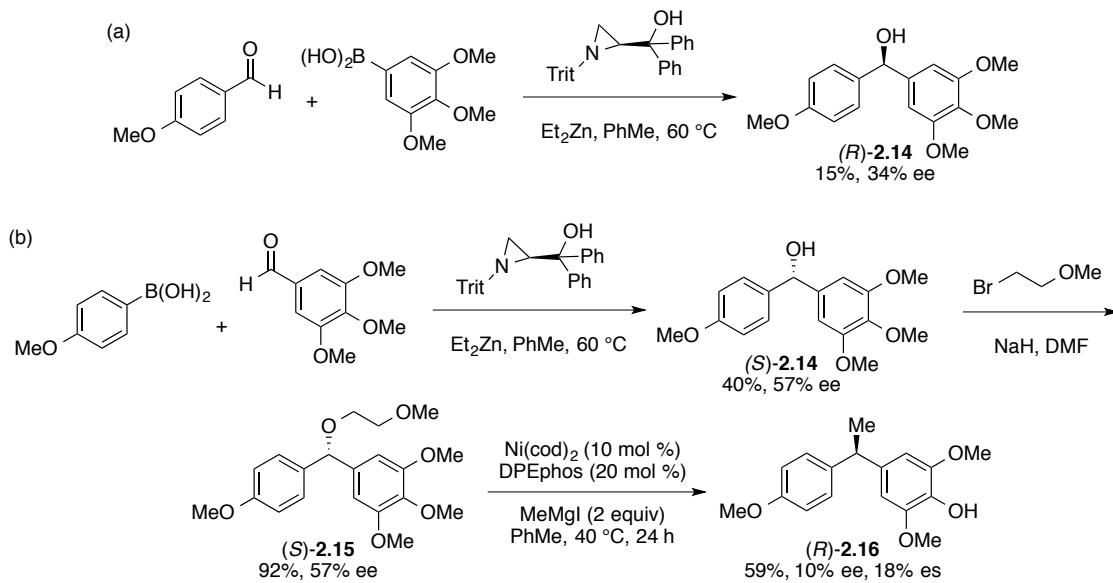
<sup>a</sup>Yield determined by <sup>1</sup>H NMR spectroscopy by comparison to an internal standard (PhTMS).

<sup>b</sup>Employed 5 mol % Ni(cod)<sub>2</sub>.

With optimized reaction conditions in hand, our next goal was to determine the enantiospecificity of the reaction. Synthesis of enantioenriched substrate **2.15** was accomplished

using the asymmetric aziridine-catalyzed arylation reaction (Scheme 2.5).<sup>12</sup> The reaction to generate (*R*)-**2.14** alcohol gave only 15% yield and 34% ee after recrystallization (Scheme 2.5a); the low yield can be accounted for by the low solubility of 3,4,5-trimethoxyphenylboronic acid in toluene, *vide supra*). In contrast, the reaction to make (*S*)-**2.14** alcohol provided 40% yield and 57% ee (Scheme 2.5b). We chose to move forward with the latter alcohol and synthesized the ether (*S*)-**2.15** in 92% yield. Subjecting (*S*)-**2.15** to the cross-coupling reaction conditions generated **2.16** in 59% yield; however, the resulting diarylethane was formed in only 10% ee and 18% es. We attribute the loss of stereochemistry to a background racemization pathway caused by the electron-rich *para*-methoxyphenyl group.

**Scheme 2.5.** Attempted enantioenriched synthesis of diarylethane **2.10**.



To circumvent having the electron-donating *para*-methoxyphenyl substituent in the ether substrate, we devised a synthetic plan for the cross-coupling reaction to incorporate the *para*-methoxyphenyl moiety as the transmetallating partner instead. The proposed transformation would involve the Kumada cross-coupling reaction of ether **2.17** with *para*-methoxyphenylmagnesium bromide to generate diarylethane **2.10** (Table 2.2). Our reaction

design was inspired by previous work in the Jarvo laboratory using the Ni(dppe)Cl<sub>2</sub> catalyst to minimize  $\beta$ -hydride elimination.<sup>14</sup> We attempted the Kumada cross-coupling reaction of substrate **2.17** with *para*-methoxyphenylmagnesium bromide using Ni(dppe)Cl<sub>2</sub>, but the reaction yielded < 2% desired product and 7%  $\beta$ -hydride elimination product (entry 1). Incorporating one equivalent of magnesium iodide increased the yield of desired product to 18%, but generated additional  $\beta$ -hydride elimination product (entry 2).<sup>15</sup>

**Table 2.2.** Attempted Kumada cross-coupling reaction using Ni(dppe)Cl<sub>2</sub>.

entry	SM yield (%) <sup>a</sup>	yield <b>2.10</b> (%) <sup>a</sup>	elimination yield (%) <sup>a</sup>	additive
1	68	< 2	7	--
2	49	18	16	MgI <sub>2</sub>

<sup>a</sup>Yield determined by <sup>1</sup>H NMR spectroscopy by comparison to an internal standard (PhTMS).

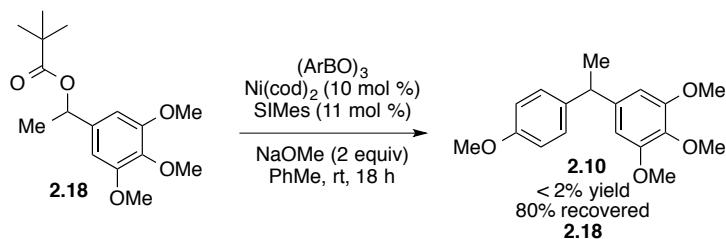
An alternative strategy to access diarylethane **2.10** involved the Suzuki cross-coupling reaction of ester **2.18** with *para*-methoxyphenylboronic ester (Scheme 2.6). Based on previous methods developed in the Jarvo laboratory, we postulated that incorporation of a pivovyl leaving group in the substrate would allow for use of an arylboroxine as the transmetallating agent.<sup>16</sup> When subjected to the reaction conditions, substrate **2.18** provided only recovered starting material.

<sup>14</sup> Yonova, I. M.; Johnson, A. G.; Osborne, C. A.; Moore, C. E.; Morrissette, N. S.; Jarvo, E. R. *Angew. Chem., Int. Ed.* **2014**, *53*, 2422.

<sup>15</sup> Greene, M. A. Diastereoselective Synthesis of Seven Membered Ring *trans*-Alkenes and Development of Stereospecific Nickel-Catalyzed Cross-Coupling Reactions. Ph.D. Thesis, The University of California, Irvine, May 2013.

<sup>16</sup> Harris, M. R.; Hanna, L. E.; Greene, M. A.; Moore, C. E.; Jarvo, E. R. *J. Am. Chem. Soc.* **2013**, *135*, 3303.

**Scheme 2.6.** Attempted Suzuki cross-coupling reaction to synthesize **2.10**.



#### 2.4 Synthesis of Indole-Substituted Diarylethane **2.4**

We were interested in pursuing the synthesis of indole-substituted diarylethane **2.4** because it potently inhibits tubulin polymerization and demonstrates bioactivity against colon, lung, breast, and triple-negative breast cancer cell lines.<sup>6,17</sup> Typically, synthesis of diarylethane **2.4** is achieved by hydrogenation of the diarylethylene precursor. Our proposed synthetic route to prepare **2.4** would incorporate a stereospecific nickel-catalyzed Negishi cross-coupling reaction to generate single enantiomers of **2.4**. Herein, we report the first synthesis of diarylethane **2.4** achieved using a cross-coupling reaction; these results have set the stage for the enantiospecific synthesis and subsequent biological evaluation of (*R*)- and (*S*)-**2.4**.

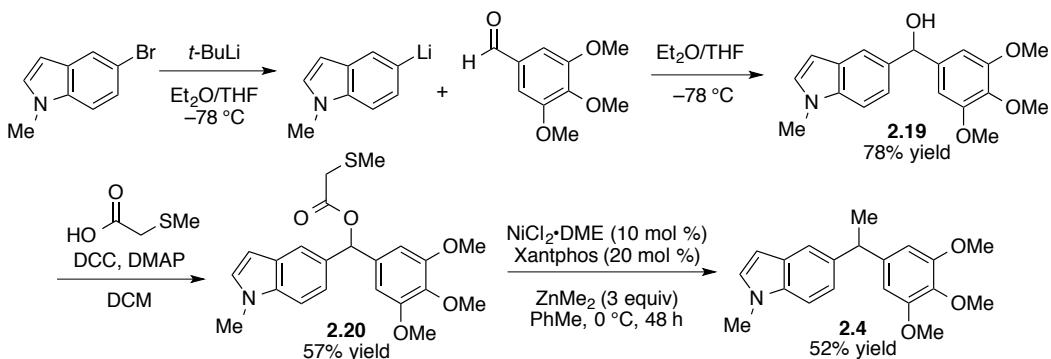
The Jarvo laboratory has previously developed the nickel-catalyzed Negishi cross-coupling reactions of diaryl 2-(methylthio)esters with dimethylzinc to generate diarylethanes.<sup>18</sup> To access diarylethane **2.4**, the requisite 2-(methylthio)acetate **2.20** was obtained by a DCC coupling reaction of alcohol **2.19** with 2-(methylthio)acetic acid (Scheme 2.7).<sup>19</sup> After evaluating several ligands for the transformation of **2.20** to **2.4**, we found that a combination of  $\text{NiCl}_2 \cdot \text{DME}$  and Xantphos provided the highest yields. Performing the cross-coupling reaction at 0 °C for 48 h furnished diarylethane **2.4** in 52% yield.

<sup>17</sup> Álvarez, R.; Puebla, R.; Fernando Díaz, J.; Bento, A. C. García-Navas, R.; de la Iglesia-Vicente, J.; Mollinedo, F.; Andreu, J. M.; Medarde, M.; Peláez, R. *J. Med. Chem.* **2013**, *56*, 2813.

<sup>18</sup> Wisniewska, H. M.; Swift, E. C.; Jarvo, E. R. *J. Am. Chem. Soc.* **2013**, *135*, 9083.

<sup>19</sup> Stayshich, R. M.; Meyer, T. Y. *J. Polym. Sci., Part A: Polym. Chem.* **2008**, *46*, 4704.

**Scheme 2.7.** Synthesis of diarylethane **2.4**.



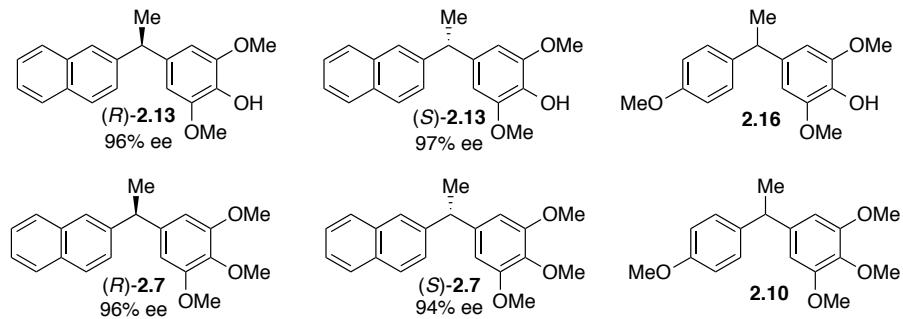
Initial attempts to generate either (*R*)- or (*S*)-**2.20** for the stereospecific cross-coupling reaction were unsuccessful: the asymmetric arylation reaction did not provide the desired enantioenriched diaryl alcohol **2.19**. Efforts to obtain (*R*)- or (*S*)-**2.19** via other synthetic pathways are in progress.

## 2.5 Biological Evaluation as Tubulin Polymerization Disruptors

Having utilized our cross-coupling methodology to synthesize the six diarylethanes shown in Figure 2.3, we performed a qualitative tubulin-binding assay.<sup>20</sup> Polymerized microtubules are visualized *in vivo* using fluorescence microscopy with a cell line containing green fluorescent protein (GFP)-tagged tubulin.<sup>21</sup> When the tubulin-binding agent is introduced, microtubule fibers are no longer visible by fluorescence microscopy because they have been depolymerized into tubulin monomers.

<sup>20</sup> Lyons-Abbott, S.; Sackett, D. L.; Wloga, D.; Gaertig, J.; Morgan, R. E.; Werbovetz, K. A.; Morissette, N. S. *Eukaryot. Cell* **2010**, *9*, 1825.

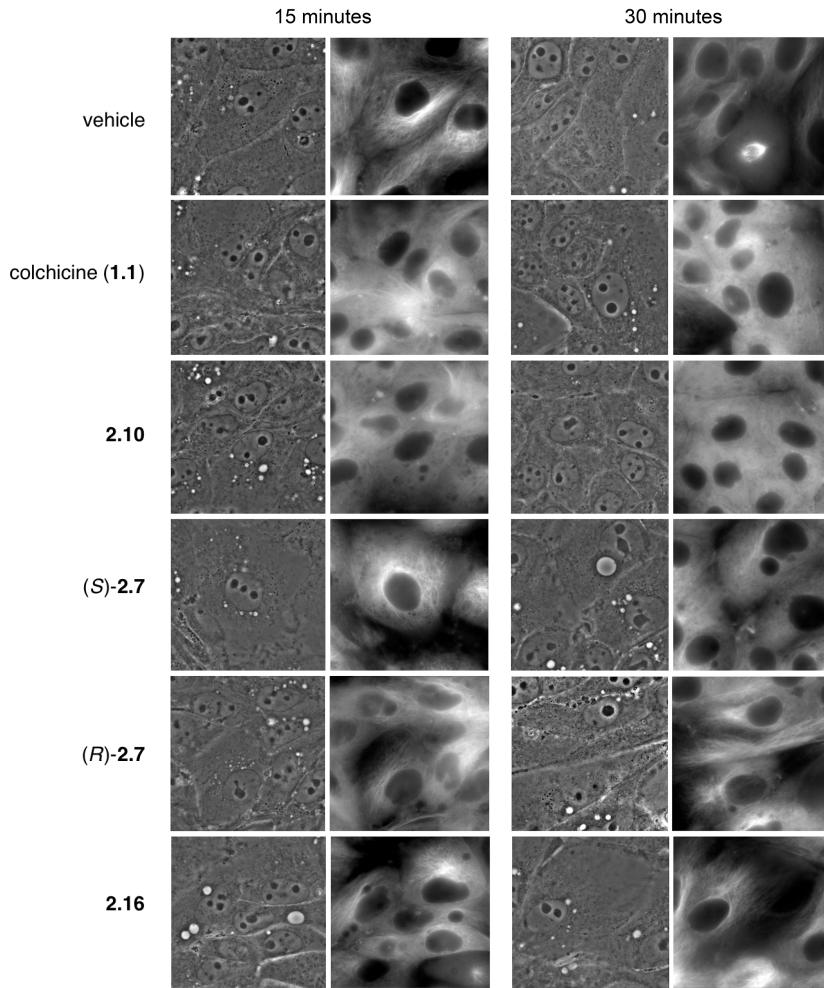
<sup>21</sup> Freitag, M.; Hickey, P. C.; Raju, N. B.; Selker, E. U.; Read, N. D. *Fungal Genet. Biol.* **2004**, *41*, 897.



**Figure 2.3.** Diarylethanes for biological testing.

In collaboration with the Morrisette laboratory at UC Irvine, our six diarylethanes were evaluated against an LLCPK cell line containing GFP-tagged tubulin. Our preliminary goal was to classify the substrates as active, inactive, or somewhat active tubulin disruptors. Samples were dissolved in DMSO and diluted to 10  $\mu$ M concentration in media before incubating with cells (0.1% DMSO). Fluorescence microscopy images were collected of the cells before compound exposure, after incubation for 15 minutes, then after incubation for 30 minutes.

The positive control, 10  $\mu$ M colchicine (**1.1**), showed significant tubulin disruption after 15 minutes, while the negative vehicle control of DMSO showed no microtubule disruption (Figure 2.4). Racemic compound **2.10** was the only substrate besides colchicine to demonstrate activity as a potent tubulin disruptor; its phenol counterpart **2.16** showed no tubulin disruption. Compound *(S)*-**2.7** demonstrated fair activity as a tubulin disruptor, while *(R)*-**2.7** showed no tubulin depolymerization. The remaining two phenol compounds, *(S)*-**2.13** and *(R)*-**2.13**, did not show activity as tubulin-binding agents. These results reflect the necessity of the trimethoxyphenyl motif for tubulin binding. The increased potency of compound *(S)*-**2.7** demonstrates the significance of stereochemical differentiation in the colchicine binding site.



**Figure 2.4.** Fluorescence microscopy images of LLCPK cells, containing GFP-tagged tubulin, upon incubation with diarylethane compounds.

## 2.6 Conclusions

The tubulin disruption studies confirm that **2.10** is a potent inhibitor of tubulin polymerization in the LLCPK cell line. Preliminary results show a marked difference in tubulin binding between *(S)*-**2.7** and *(R)*-**2.7**, and support our assertion that the colchicine binding site recognizes absolute stereochemistry. We have established that compounds containing a 2,6-dimethoxy-phenol motif (**2.16**, *(S)*-**2.13**, and *(R)*-**2.13**) do not exhibit activity as tubulin polymerization inhibitors, confirming the 3,4,5-trimethoxyphenyl ring is important for binding in

the colchicine site. Finally, we have completed the racemic synthesis of **2.4** utilizing a Negishi cross-coupling reaction.

Synthesis of enantioenriched (*R*)-**2.10**, (*S*)-**2.10**, (*R*)-**2.4**, and (*S*)-**2.4** is under way. These compounds will be evaluated for activity as tubulin polymerization disruptors to determine whether a single enantiomer demonstrates greater affinity for tubulin binding, and to assign IC<sub>50</sub> values to these substrates.

## 2.7 Experimental Details

### *General Procedures*

All reactions were carried out under an atmosphere of N<sub>2</sub>. All glassware was oven- or flame-dried prior to use. Tetrahydrofuran (THF), diethyl ether (Et<sub>2</sub>O), dichloromethane (CH<sub>2</sub>Cl<sub>2</sub>), and toluene (PhMe) were degassed with Ar and then passed through two 4 x 36 inch columns of anhydrous neutral A-2 alumina (8 x 14 mesh; LaRoche Chemicals; activated under a flow of argon at 350 °C for 12 h) to remove H<sub>2</sub>O. All other solvents used were purchased “anhydrous” commercially, or purified as described (vide infra). <sup>1</sup>H NMR spectra were recorded on Bruker DRX-400 (400 MHz <sup>1</sup>H, 100 MHz <sup>13</sup>C), GN-500 (500 MHz <sup>1</sup>H, 125.7 MHz <sup>13</sup>C), or CRYO-500 (500 MHz <sup>1</sup>H, 125.7 MHz <sup>13</sup>C) spectrometers. Proton chemical shifts are reported in ppm (δ) relative to internal tetramethylsilane (TMS, δ 0.00). Data are reported as follows: chemical shift [multiplicity [singlet (s), broad singlet (br s), doublet (d), doublet of doublets (dd), triplet (t), doublet of triplets (dt), doublet of doublet of triplets (ddt), triplet of triplets (tt), quartet (q), multiplet (m)], coupling constants [Hz], integration]. Carbon chemical shifts are reported in ppm (δ) relative to TMS with the respective solvent resonance as the internal standard (CDCl<sub>3</sub>, δ 77.16 ppm). Unless otherwise indicated, NMR data were collected at 25 °C. Infrared spectra

were obtained on a Thermo Scientific Nicolet iS5 FT-IR Spectrometer. Analytical thin-layer chromatography (TLC) was performed using Silica Gel 60 F<sub>254</sub> precoated plates (0.25 mm thickness). Visualization was accomplished by irradiation with a UV lamp and/or staining with KMnO<sub>4</sub>, ceric ammonium molybdate (CAM), or *p*-anisaldehyde (PAA) solutions. Flash chromatography was performed using Silica Gel 60Å (170-400 mesh) from Fisher Scientific. Melting points (m.p.) were obtained using a Mel-Temp melting point apparatus and are uncorrected. Optical rotations were measured on a Rudolph Research Analytical Autopol III Automatic Polarimeter. SFC determinations of enantiopurity were performed on a Berger Analytical instrument using a Daicel™ Chiralpak® column (OD-H, OJ-H, AD-H, or AS-H; 100 bar, 50 °C). High resolution mass spectrometry was performed by the University of California, Irvine Mass Spectrometry Center.

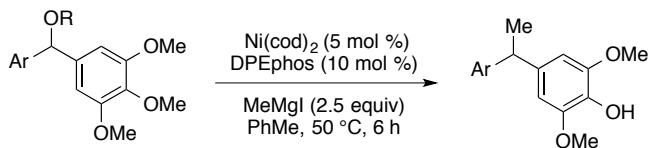
Ni(cod)<sub>2</sub> was purchased from Strem, stored in a glovebox freezer (-20 °C) under an atmosphere of N<sub>2</sub>, and used as received. All ligands were purchased from Strem, Aldrich, or Solvias and stored in a glovebox under an atmosphere of N<sub>2</sub>. Dimethyl zinc (ZnMe<sub>2</sub>) was purchased from Aldrich and stored under N<sub>2</sub> at 4 °C. Grignard reagents were freshly prepared from the halide precursor. All Grignard reagents and ZnMe<sub>2</sub> were titrated with iodine prior to use.<sup>22</sup> All other chemicals were purchased commercially and used as received.

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<sup>22</sup> Krasovskiy, A.; Knochel, P. *Synthesis* **2006**, 5, 890.

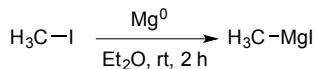
## **General Cross-Coupling Procedures**

### *Method A: Cross-Coupling with Methyl Grignard*



Performed according to a procedure reported by Jarvo and co-workers.<sup>8</sup> In a glovebox, a flame-dried bomb flask equipped with a stir bar was charged with Ni(cod)<sub>2</sub> (0.05 equiv) and DPEphos (0.10 equiv). A solution of substrate (1.0 equiv) in PhMe was added and the reaction mixture was stirred for 5 min. MeMgI (2.5 equiv) was added dropwise and the vial was sealed, removed from the glovebox, and stirred for 24 h at 40 °C or 50 °C. The reaction was quenched with saturated aqueous ammonium chloride, extracted with EtOAc (x 3), washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated in vacuo. Phenyltrimethylsilane (PhTMS) was added as internal standard and a <sup>1</sup>H NMR yield was obtained before purification by flash column chromatography.

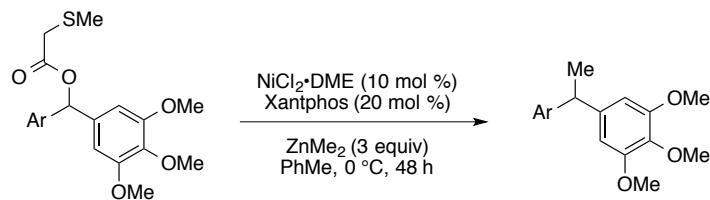
### *Preparation of Methyl Grignard Reagent*



Under a N<sub>2</sub> atmosphere, a 3-necked flask equipped with a stir bar, reflux condenser, and Schlenk filtration apparatus was charged with magnesium turnings (1.1 g, 45 mmol, 1.5 equiv). The flask and magnesium turnings were then flame-dried under vacuum and the flask was back-filled with N<sub>2</sub>. Anhydrous Et<sub>2</sub>O (7 mL) and a crystal of iodine (ca. 2 mg) were added to the flask. Freshly distilled iodomethane (1.9 mL, 31 mmol, 1.0 equiv) was slowly added over 30 min to maintain a gentle reflux. The mixture was stirred for 2 h at room temperature then filtered through the fritted Schlenk filter into the Schlenk bomb under N<sub>2</sub> atmosphere. The magnesium turnings were

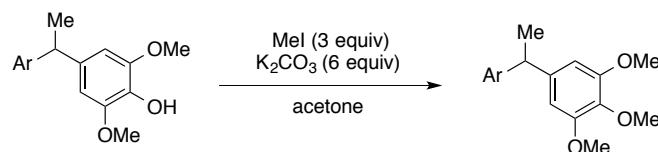
washed with Et<sub>2</sub>O (2 × 1.0 mL) then the Schlenk bomb was sealed, removed, and placed under an argon atmosphere. The resulting methyl Grignard reagent was typically between 2.4 and 3.0 M as titrated by Knochel's method<sup>22</sup> and could be stored (sealed under argon atmosphere or in a glovebox) for up to 4 weeks.

*Method B: Cross-Coupling with Dimethylzinc*



Performed according to a procedure reported by Jarvo and co-workers.<sup>18</sup> In a glovebox, a flame-dried 7 mL vial equipped with a stir bar was charged with NiCl<sub>2</sub>·DME (0.10 equiv) and Xantphos (0.20 equiv). A solution of substrate (1.0 equiv) in PhMe was added and the reaction mixture was stirred for 5 min before being sealed and removed from the glovebox. The vial was equipped with a N<sub>2</sub> line, cooled to 0 °C, and ZnMe<sub>2</sub> (3.0 equiv) was added, which resulted in an immediate color change from green to orange. The reaction was allowed to stir at 0 °C for 48 h, then it was quenched with isopropyl alcohol, filtered through a plug of silica gel (neat Et<sub>2</sub>O), and concentrated in vacuo. Phenyltrimethylsilane (PhTMS) was added as internal standard and a <sup>1</sup>H NMR yield was obtained before purification by flash column chromatography.

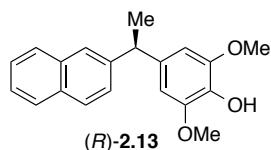
*Method C: Remethylation with Iodomethane*



Performed according to a procedure reported by Jarvo and co-workers.<sup>8</sup> Substrate (1.0 equiv)

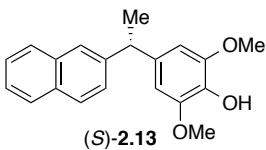
was dissolved in acetone and  $K_2CO_3$  (6.0 equiv) was added, followed by iodomethane (3.0 equiv). The reaction mixture was heated to reflux and stirred overnight, after which the reaction was cooled to room temperature, filtered, washed with acetone, and concentrated in vacuo.  $H_2O$  was added and the mixture was extracted with EtOAc ( $\times 3$ ). The combined organics were washed with brine, dried over  $Na_2SO_4$ , and concentrated in vacuo.

### **Characterization Data for Products**

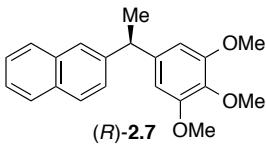


**2,6-Dimethoxy-4-((R)-1-naphthalen-3-yl)ethylphenol ((R)-2.13).** Prepared according to Method A, using the following amounts of reagents:  $Ni(cod)_2$  (6.5 mg, 0.024 mmol, 0.050 equiv), DPEphos (25 mg, 0.047 mmol, 0.10 equiv), ether (*S*-**2.12** (0.16 g, 0.47 mmol, 1.0 equiv), PhMe (5.0 mL), and MeMgI (0.59 mL, 1.2 mmol, 2.0 M in  $Et_2O$ , 2.5 equiv). The product was purified by flash column chromatography using 25% EtOAc/hexanes to afford the title compound as a colorless oil (0.13 g, 0.42 mmol, 91%). Analytical data are consistent with literature values.<sup>8</sup> **TLC**  $R_f$  = 0.3 (20% EtOAc/hexanes); **m.p.** 88–90 °C; **<sup>1</sup>H NMR** ( $CDCl_3$ , 500 MHz)  $\delta$  7.80–7.78 (m, 2H), 7.73 (d,  $J$  = 8.6 Hz, 1H), 7.66 (s, 1H), 7.45–7.41 (m, 2H), 7.30 (dd,  $J$  = 8.5, 1.8 Hz, 1H), 6.47 (s, 2H), 5.40 (s, 1H), 4.23 (q,  $J$  = 7.2 Hz, 1H), 3.81 (s, 6H), 1.70 (d,  $J$  = 7.2 Hz, 3H); **<sup>13</sup>C NMR** ( $CDCl_3$ , 125 MHz)  $\delta$  147.0, 144.0, 137.4, 133.6, 133.1, 132.2, 128.0, 127.8, 127.7, 126.8, 126.0, 125.5, 125.2, 104.6, 56.4, 44.9, 22.1; **IR** (neat) 3420, 2962, 1612, 1243, 1226, 1111  $cm^{-1}$ ; **HRMS** (TOF MS ES+)  $m/z$  calcd for  $C_{20}H_{20}O_3$  ( $M + Na$ )<sup>+</sup> 331.1310, found 331.1317; **[ $\alpha$ ]<sup>24</sup>D** +22.2 ( $c$  1.45,  $CHCl_3$ ), literature **[ $\alpha$ ]<sup>23</sup>D** +23.6 ( $c$  0.42,  $CHCl_3$ ); **SFC**

analysis (AD-H, 15% MeOH, 2.5 mL/min, 215 nm) indicated 96% ee:  $t_R$  (minor) = 12.11 min,  $t_R$  (major) = 13.25 min.

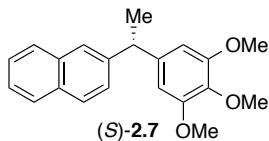


**2,6-Dimethoxy-4-((S)-1-naphthalen-3-yl)ethylphenol ((S)-2.13).** Prepared according to Method A, using the following amounts of reagents: Ni(cod)<sub>2</sub> (1.6 mg, 6.0  $\mu$ mol, 0.050 equiv), DPEphos (6.2 mg, 0.012 mmol, 0.10 equiv), ether (*R*-**2.12** (39 mg, 0.12 mmol, 1.0 equiv), PhMe (1.0 mL), and MeMgI (0.13 mL, 0.29 mmol, 2.3 M in Et<sub>2</sub>O, 2.5 equiv). The product was purified by flash column chromatography using 10–25% EtOAc/hexanes to afford the title compound as a colorless oil (32 mg, 0.11 mmol, 90%). Analytical data are consistent with the values listed for (*R*-**2.13** (vide supra).  $[\alpha]^{24}_D -26.6$  (*c* 1.40, CHCl<sub>3</sub>); SFC analysis (AD-H, 15% MeOH, 2.5 mL/min, 215 nm) indicated 97% ee:  $t_R$  (major) = 12.10 min,  $t_R$  (minor) = 13.33 min.

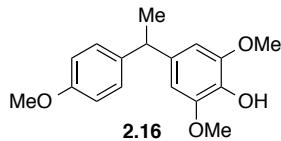


**2-((R)-1-(3,4,5-Trimethoxyphenyl)ethyl)naphthalene ((R)-2.7).** Prepared according to Method C, using the following amounts of reagents: phenol (*R*-**2.13** (55 mg, 0.18 mmol, 1.0 equiv), K<sub>2</sub>CO<sub>3</sub> (0.447 g, 3.24 mmol, 18.0 equiv), iodomethane (102  $\mu$ L, 1.62 mmol, 9.00 equiv), and acetone (3.0 mL). The product was purified by flash column chromatography using 15% Et<sub>2</sub>O/pentanes to afford the title compound as a colorless oil (28 mg, 0.088 mmol, 49%). Analytical data are consistent with literature values.<sup>8</sup> TLC R<sub>f</sub> = 0.4 (40% Et<sub>2</sub>O/hexanes); <sup>1</sup>H

**NMR** ( $\text{CDCl}_3$ , 400 MHz)  $\delta$  7.81–7.79 (m, 2H), 7.75 (d,  $J$  = 8.5 Hz, 1H), 7.68 (s, 1H), 7.46–7.42 (m, 2H), 7.32 (dd,  $J$  = 8.5, 1.8 Hz, 1H), 6.47 (s, 2H), 4.24 (q,  $J$  = 7.2 Hz, 1H), 3.83 (s, 3H), 3.80 (s, 6H), 1.71 (d,  $J$  = 7.2 Hz, 3H);  **$^{13}\text{C}$  NMR** ( $\text{CDCl}_3$ , 125 MHz)  $\delta$  153.2, 143.7, 142.1, 136.4, 133.6, 132.2, 128.1, 127.9, 127.7, 126.8, 126.1, 125.5, 125.3, 105.0, 60.9, 56.2, 45.2, 22.0; **IR** (neat) 2932, 1588, 1506, 1417, 1232, 1124, 1008  $\text{cm}^{-1}$ ; **HRMS** (TOF MS ES+)  $m/z$  calcd for  $\text{C}_{21}\text{H}_{22}\text{O}_3$  ( $\text{M} + \text{Na}$ ) $^+$  345.1467, found 345.1454;  $[\alpha]^{24}_D$  +26.9 ( $c$  1.29,  $\text{CHCl}_3$ ), literature  $[\alpha]^{23}_D$  +24.9 ( $c$  0.78,  $\text{CHCl}_3$ ); **SFC** analysis (AD-H, 15% MeOH, 2.5 mL/min, 215 nm) indicated 96% ee:  $t_R$  (minor) = 3.86 min,  $t_R$  (major) = 4.20 min.

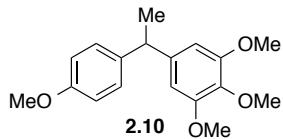


**2-((S)-1-(3,4,5-Trimethoxyphenyl)ethyl)naphthalene ((S)-2.7).** Prepared according to Method C by LuisRuben Martinez. The product was further purified by flash column chromatography using 40%  $\text{Et}_2\text{O}$ /hexanes to afford the title compound as a colorless oil. Analytical data are consistent with the values listed for (*R*)-2.7 (vide supra).  $[\alpha]^{24}_D$  -26.5 ( $c$  0.70,  $\text{CHCl}_3$ ); **SFC** analysis (AD-H, 15% MeOH, 2.5 mL/min, 215 nm) indicated 94% ee:  $t_R$  (major) = 3.83 min,  $t_R$  (minor) = 4.25 min.

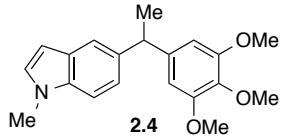


**2,6-Dimethoxy-4-(1-(4-methoxyphenyl)ethyl)phenol (2.16).** Prepared according to Method A, using the following amounts of reagents:  $\text{Ni}(\text{cod})_2$  (16 mg, 0.058 mmol, 0.10 equiv), DPEphos (63 mg, 0.12 mmol, 0.20 equiv), ether **2.15** (0.21 g, 0.58 mmol, 1.0 equiv),  $\text{PhMe}$  (6.0 mL), and

MeMgI (0.58 mL, 1.2 mmol, 2.0 M in Et<sub>2</sub>O, 2.0 equiv). The product was purified by flash column chromatography using 40% Et<sub>2</sub>O/hexanes to afford the title compound as a yellow oil (98 mg, 0.34 mmol, 59%). **TLC**  $R_f$  = 0.2 (40% Et<sub>2</sub>O/hexanes); **<sup>1</sup>H NMR** (CDCl<sub>3</sub>, 400 MHz)  $\delta$  7.12 (d,  $J$  = 8.6 Hz, 2H), 6.83 (d,  $J$  = 8.7 Hz, 2H), 6.42 (s, 2H), 5.37 (s, 1H), 4.02 (q,  $J$  = 7.2 Hz, 1H), 3.83 (s, 6H), 3.78 (s, 3H), 1.58 (d,  $J$  = 7.2 Hz, 3H); **<sup>13</sup>C NMR** (CDCl<sub>3</sub>, 125 MHz)  $\delta$  157.9, 147.0, 138.7, 138.0, 133.0, 128.4, 113.8, 104.4, 56.3, 55.3, 44.0, 22.4; **IR** (neat) 3430, 2962, 2836, 1509, 1240, 1212, 1111 cm<sup>-1</sup>; **HRMS** (TOF MS ES+)  $m/z$  calcd for C<sub>17</sub>H<sub>20</sub>O<sub>4</sub> (M + Na)<sup>+</sup> 311.1259, found 311.1265.



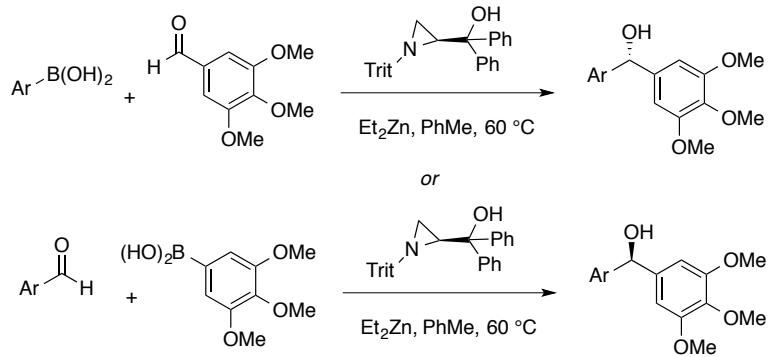
**1,2,3-Trimethoxy-5-(1-(4-methoxyphenyl)ethyl)benzene (2.10).** Prepared according to Method C, using the following amounts of reagents: phenol **2.16** (36 mg, 0.13 mmol, 1.0 equiv), K<sub>2</sub>CO<sub>3</sub> (0.310 g, 2.25 mmol, 18.0 equiv), iodomethane (94.0  $\mu$ L, 1.50 mmol, 12.0 equiv), and acetone (3.0 mL). The product was purified by flash chromatography (10–40% Et<sub>2</sub>O/pentanes) to afford the title compound as a colorless oil (17 mg, 0.055 mmol, 44%). Analytical data are consistent with literature values.<sup>6</sup> **TLC**  $R_f$  = 0.2 (10% Et<sub>2</sub>O/hexanes); **<sup>1</sup>H NMR** (CDCl<sub>3</sub>, 400 MHz)  $\delta$  7.14 (d,  $J$  = 8.6 Hz, 2H), 6.84 (d,  $J$  = 8.7 Hz, 2H), 6.42 (s, 2H), 4.04 (q,  $J$  = 7.2 Hz, 1H), 3.82 (s, 9H), 3.79 (s, 3H), 1.60 (d,  $J$  = 7.2 Hz, 3H); **<sup>13</sup>C NMR** (CDCl<sub>3</sub>, 125 MHz)  $\delta$  157.9, 153.1, 142.5, 138.4, 136.2, 128.4, 113.8, 104.6, 60.9, 56.1, 55.3, 44.3, 22.3; **IR** (neat) 2933, 2835, 1508, 1235, 1123, 1008 cm<sup>-1</sup>; **HRMS** (TOF MS ES+)  $m/z$  calcd for C<sub>18</sub>H<sub>22</sub>O<sub>4</sub> (M + Na)<sup>+</sup> 325.1416, found 325.1427.



**1-Methyl-5-(1-(3,4,5-trimethoxyphenyl)ethyl)-1*H*-indole (2.4).** Prepared according to Method B, using the following amounts of reagents:  $\text{NiCl}_2 \cdot \text{DME}$  (3.0 mg, 0.014 mmol, 0.10 equiv), Xantphos (16 mg, 0.027 mmol, 0.20 equiv), ester **2.20** (50 mg, 0.12 mmol, 1.0 equiv), PhMe (1.7 mL), and  $\text{ZnMe}_2$  (0.34 mL, 0.41 mmol, 1.2 M in pentanes, 3.0 equiv). The product was purified by flash column chromatography using 20–40% EtOAc/hexanes to afford the title compound as a colorless oil (20 mg, 0.062 mmol, 52%). Analytical data are consistent with literature values.<sup>6</sup> **TLC**  $\mathbf{R}_f = 0.5$  (20% EtOAc/hexanes); **<sup>1</sup>H NMR** ( $\text{CDCl}_3$ , 400 MHz)  $\delta$  7.48 (s, 1H), 7.24 (d,  $J = 8.5$  Hz, 1H), 7.08 (dd,  $J = 8.5, 1.7$  Hz, 1H), 7.02 (d,  $J = 3.0$  Hz, 1H), 6.49 (s, 2H), 6.43 (dd,  $J = 3.0, 0.9$  Hz, 1H), 4.20 (q,  $J = 7.1$  Hz, 1H), 3.81–3.80 (m, 9H), 3.76 (s, 3H), 1.68 (d,  $J = 7.1$  Hz, 3H); **<sup>13</sup>C NMR** ( $\text{CDCl}_3$ , 125 MHz)  $\delta$  153.1, 143.3, 137.4, 136.1, 135.5, 129.1, 128.6, 122.0, 119.1, 109.2, 104.8, 100.9, 61.0, 56.2, 45.2, 33.0, 22.7; **IR** (neat) 2932, 1417, 1233, 1123, 721  $\text{cm}^{-1}$ ; **HRMS** (TOF MS ES+)  $m/z$  calcd for  $\text{C}_{20}\text{H}_{23}\text{NO}_3$  ( $\text{M} + \text{Na}$ )<sup>+</sup> 348.1576, found 348.1567.

### General Procedures for Starting Materials Synthesis

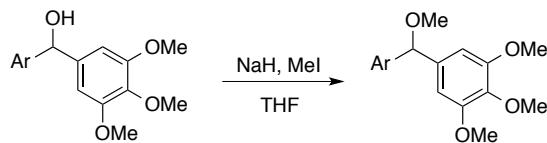
#### Method D: Asymmetric Aziridine-Catalyzed Arylation



Modified from a procedure reported by Braga and co-workers.<sup>12</sup> To a solution of arylboronic

acid (2.4 equiv) in anhydrous PhMe was added diethylzinc (1.0 M in PhMe, 7.2 equiv), and the solution was allowed to stir at 60 °C for 18 h. Upon cooling to room temperature, (*S*)- $\alpha,\alpha$ -diphenyl-1-(triphenylmethyl)-2-aziridinemethanol (0.10 equiv) was added as a solution in PhMe and the reaction mixture was allowed to stir for 10 minutes before the addition of a solution of arylaldehyde (1.0 equiv) in PhMe. After stirring 18 h at room temperature, 1M HCl was added and the product was extracted with EtOAc (x 3). The combined organics were washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated in vacuo.

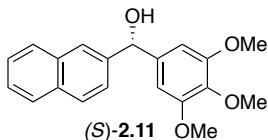
*Method E: Alkylation with Iodomethane*



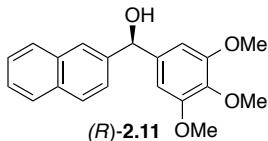
Modified from a procedure reported by Jarvo and co-workers.<sup>8</sup> To a suspension of NaH (1.5 equiv) in anhydrous THF was added a solution of substrate (1.0 equiv) in THF. The mixture was stirred for 1 h before addition of iodomethane (1.02 equiv) and then the reaction was stirred overnight. Excess NaH was quenched with saturated aqueous ammonium chloride, and the product was extracted with EtOAc (x 3). The combined organics were washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated in vacuo.

## Synthesis & Characterization Data for Starting Materials

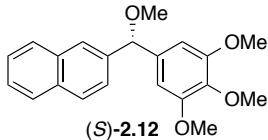
### Synthesis of Starting Materials for 2.7



**(S)-(3,4,5-Trimethoxyphenyl)(naphthalen-3-yl)methanol ((S)-2.11).** Prepared according to Method D, using the following amounts of reagents: 3,4,5-trimethoxybenzaldehyde (0.49 g, 2.5 mmol, 1.0 equiv), 2-naphthylboronic acid (1.0 g, 6.0 mmol, 2.4 equiv), diethylzinc (18.0 mL, 18.0 mmol, 1.00 M in toluene, 7.20 equiv), (*S*)- $\alpha,\alpha$ -diphenyl-1-(triphenylmethyl)-2-aziridinemethanol (0.12 g, 0.25 mmol, 0.10 equiv), and PhMe (43 mL). The product was purified by flash column chromatography using 25% EtOAc/hexanes, then recrystallized from 50% CH<sub>2</sub>Cl<sub>2</sub>/hexanes to afford the title compound as a white solid (0.36 g, 1.1 mmol, 45%). Analytical data are consistent with literature values.<sup>8</sup> **TLC** R<sub>f</sub> = 0.1 (40% Et<sub>2</sub>O/hexanes); **m.p.** 135–136 °C; **<sup>1</sup>H NMR** (CDCl<sub>3</sub>, 500 MHz) δ 7.85–7.77 (m, 4H), 7.49–7.41 (m, 3H), 6.62 (s, 2H), 5.88 (s, 1H), 3.81 (s, 3H), 3.79 (s, 6H), 2.63 (br s, 1H); **<sup>13</sup>C NMR** (CDCl<sub>3</sub>, 125 MHz) δ 153.3, 141.0, 139.4, 137.3, 133.3, 133.0, 128.4, 128.2, 127.8, 126.3, 126.1, 125.1, 124.8, 103.7, 76.4, 60.9, 56.1; **IR** (neat) 3357, 2938, 2838, 1590, 1233, 1128, 997 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>20</sub>H<sub>20</sub>O<sub>4</sub> (M + Na)<sup>+</sup> 347.1259, found 347.1261; **[α]<sup>26</sup>D** -1.5 (*c* 1.17, CHCl<sub>3</sub>), literature [α]<sup>23</sup>D -1.07 (*c* 0.95, CHCl<sub>3</sub>); **SFC** analysis (AD-H, 30% MeOH, 2.5 mL/min, 215 nm) indicated >99% ee: t<sub>R</sub> (minor) = 3.25 min, t<sub>R</sub> (major) = 3.55 min.

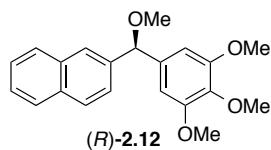


**(*R*)-(3,4,5-Trimethoxyphenyl)(naphthalen-3-yl)methanol ((*R*)-**2.11**).** Prepared according to Method D, using the following amounts of reagents: 2-naphthaldehyde (0.39 g, 2.5 mmol, 1.0 equiv), 3,4,5-trimethoxyphenylboronic acid (1.3 g, 6.0 mmol, 2.4 equiv), diethylzinc (18.0 mL, 18.0 mmol, 1.00 M in toluene, 7.20 equiv), (*S*)- $\alpha,\alpha$ -diphenyl-1-(triphenylmethyl)-2-aziridinemethanol (0.12 g, 0.25 mmol, 0.10 equiv), and PhMe (43 mL). The product was purified by flash column chromatography using 50% EtOAc/hexanes, then recrystallized twice from 50% CH<sub>2</sub>Cl<sub>2</sub>/hexanes to afford the title compound as a white solid (70 mg, 0.22 mmol, 9%). Analytical data are consistent with the values listed for (*S*)-**2.11** (vide supra).  $[\alpha]^{26}_D -0.7$  (*c* 0.20, CHCl<sub>3</sub>); SFC analysis (AD-H, 30% MeOH, 2.5 mL/min, 215 nm) indicated >99% ee: t<sub>R</sub> (major) = 3.34 min, t<sub>R</sub> (minor) = 3.67 min.



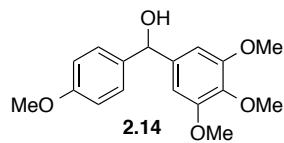
**2-((*S*)-Methoxy(3,4,5-trimethoxyphenyl)methyl)naphthalene ((*S*)-**2.12**).** Prepared according to Method E, using the following amounts of reagents: alcohol (*S*)-**2.11** (0.31 g, 0.96 mmol, 1.0 equiv), NaH (35 mg, 1.4 mmol, 1.5 equiv), iodomethane (60  $\mu$ L, 0.98 mmol, 1.02 equiv), and THF (5.5 mL). The product was purified by flash column chromatography using 40% Et<sub>2</sub>O/pentanes to afford the title compound as a white solid (0.24 g, 0.71 mmol, 70%). Analytical data are consistent with literature values.<sup>8</sup> TLC R<sub>f</sub> = 0.3 (40% Et<sub>2</sub>O/hexanes); m.p. 88–89 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  7.85–7.79 (m, 4H), 7.47–7.43 (m, 3H), 6.63 (s, 2H), 5.32 (s, 1H), 3.82 (s, 3H), 3.81 (s, 6H), 3.43 (s, 3H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz)  $\delta$  153.4, 139.3, 137.7,

137.3, 133.3, 133.0, 128.4, 128.1, 127.8, 126.3, 126.1, 125.8, 125.0, 104.0, 85.6, 60.9, 57.2, 56.2; **IR** (neat) 2937, 1587, 1230, 1126, 1088  $\text{cm}^{-1}$ ; **HRMS** (TOF MS ES+)  $m/z$  calcd for  $\text{C}_{21}\text{H}_{22}\text{O}_4$  ( $\text{M} + \text{Na}$ )<sup>+</sup> 361.1416, found 361.1419;  $[\alpha]^{26}_{\text{D}} +25.2$  ( $c$  1.35,  $\text{CHCl}_3$ ), literature  $[\alpha]^{23}_{\text{D}} +28.0$  ( $c$  0.57,  $\text{CHCl}_3$ ); **SFC** analysis (AD-H, 20% MeOH, 2.5 mL/min, 215 nm) indicated >99% ee:  $t_{\text{R}}$  (major) = 2.97 min,  $t_{\text{R}}$  (minor) = 3.23 min.



**2-((R)-Methoxy(3,4,5-trimethoxyphenyl)methyl)naphthalene ((R)-2.12).** Prepared according to Method E, using the following amounts of reagents: alcohol (*R*-**2.11** (67 mg, 0.2 mmol, 1.0 equiv), NaH (7.2 mg, 0.3 mmol, 1.5 equiv), iodomethane (13  $\mu\text{L}$ , 0.21 mmol, 1.02 equiv), and THF (2.0 mL). The product was purified by flash column chromatography using 40%  $\text{Et}_2\text{O}$ /pentanes to afford the title compound as a white solid (51 mg, 0.15 mmol, 75%). Analytical data are consistent with the values listed for (*S*)-**2.12** (vide supra).  $[\alpha]^{26}_{\text{D}} -25.3$  ( $c$  0.57,  $\text{CHCl}_3$ ); **SFC** analysis (AD-H, 20% MeOH, 2.5 mL/min, 215 nm) indicated 99% ee:  $t_{\text{R}}$  (minor) = 2.94 min,  $t_{\text{R}}$  (major) = 3.21 min.

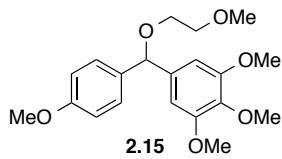
#### *Synthesis of Starting Materials for 2.10*



**(3,4,5-Trimethoxyphenyl)(4-methoxyphenyl)methanol (2.14).** THF (27 mL) was added to magnesium turnings (1.46 g, 60.0 mmol, 1.50 equiv) and catalytic  $\text{I}_2$ . 4-Bromoanisole (5.00 mL, 40.0 mmol, 1.00 equiv) was added slowly over 30 min at 0 °C, so as to maintain a gentle reflux.

The mixture was stirred for 20 min at 0 °C, then 40 min at room temperature. The resulting dark black Grignard reagent was titrated<sup>22</sup> and used immediately.

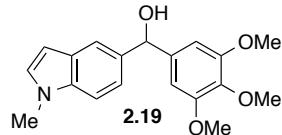
To a solution of 3,4,5-trimethoxybenzaldehyde (1.8 g, 9.3 mmol, 1.0 equiv) in THF (8.0 mL) at 0 °C was added *p*-methoxyphenylmagnesium bromide (16.0 mL, 13.9 mmol, 0.87 M in THF, 1.50 equiv). After stirring for 10 min at 0 °C, then 1.5 h at room temperature, the reaction was quenched with saturated aqueous ammonium chloride (10 mL). H<sub>2</sub>O (10 mL) was added and the mixture was extracted with EtOAc (3 × 10 mL). The combined organics were washed with brine (10 mL), dried over MgSO<sub>4</sub>, and concentrated in vacuo. The product was purified by flash column chromatography using 50% CH<sub>2</sub>Cl<sub>2</sub>/EtOAc to afford the title compound as a white solid (1.9 g, 6.1 mmol, 65%). **TLC R<sub>f</sub>** = 0.3 (50% CH<sub>2</sub>Cl<sub>2</sub>/EtOAc); **m.p.** 94–95 °C; **<sup>1</sup>H NMR** (CDCl<sub>3</sub>, 500 MHz) δ 7.24 (d, *J* = 8.8, 2H), 6.83 (d, *J* = 8.8, 2H), 6.56 (s, 2H), 5.64 (s, 1H), 3.79 (s, 3H), 3.77 (s, 6H), 3.75 (s, 3H), 2.93 (s, 1H); **<sup>13</sup>C NMR** (CDCl<sub>3</sub>, 125 MHz) δ 159.0, 153.1, 140.0, 136.9, 136.1, 127.9, 113.8, 103.4, 75.7, 60.8, 56.0, 55.3; **IR** (neat) 3347, 2936, 2838, 1508, 1235, 1124 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>17</sub>H<sub>20</sub>O<sub>5</sub> (M + Na)<sup>+</sup> 327.1208, found 327.1208.



**1-((2-Methoxyethoxy)(3,4,5-trimethoxyphenyl)methyl)-4-methoxybenzene (2.15).** To a slurry of NaH (0.22 g, 9.0 mmol, 3.0 equiv) in DMF (2.0 mL) was added a solution of alcohol **2.14** (0.91 g, 3.0 mmol, 1.0 equiv) in DMF (3.0 mL). The mixture was stirred at room temperature for 20 min before addition of DMF (10 mL) and one equivalent of 2-bromoethyl methyl ether (0.28 mL, 3.0 mmol). After stirring for 10 min, a second equivalent of 2-bromoethyl methyl ether

(0.28 mL, 3.0 mmol) was added and the reaction stirred for 30 min. Excess NaH was quenched with saturated aqueous ammonium chloride (5 mL). H<sub>2</sub>O (10 mL) was added and the mixture was extracted with EtOAc (4 × 10 mL). The combined organics were washed with H<sub>2</sub>O (10 mL) and brine (10 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated in vacuo. The product was purified by flash column chromatography using 25–50% EtOAc/hexanes to afford the title compound as a yellow oil (1.0 g, 2.8 mmol, 92%). TLC R<sub>f</sub> = 0.2 (20% EtOAC/hexanes); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz) δ 7.27 (d, *J* = 8.7, 2H), 6.85 (d, *J* = 8.7, 2H), 6.60 (s, 2H), 5.31 (s, 1H), 3.81 (s, 9H), 3.76 (s, 3H), 3.64–3.59 (m, 4H), 3.38 (s, 3H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz) δ 159.1, 153.2, 138.0, 137.0, 134.2, 128.4, 113.8, 103.9, 83.6, 72.1, 68.4, 60.8, 59.0, 56.0, 55.2; IR (neat) 2935, 1506, 1232, 1123, 1088 cm<sup>-1</sup>; HRMS (TOF MS ES+) *m/z* calcd for C<sub>20</sub>H<sub>26</sub>O<sub>6</sub> (M + Na)<sup>+</sup> 385.1627, found 385.1620.

#### *Synthesis of Starting Materials for 2.4*

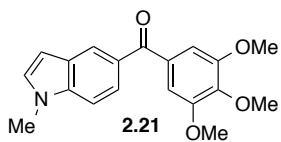


**(1-Methyl-1*H*-indol-5-yl)(3,4,5-trimethoxyphenyl)methanol (2.19).** Prepared according to a modified procedure reported by Rapoport and co-workers.<sup>23</sup> To a flame-dried flask charged with a stir bar was carefully added *tert*-butyllithium (10.4 mL, 17.6 mmol, 1.70 M in pentanes, 2.20 equiv), and the flask was cooled to –78 °C. A solution of 5-bromo-1-methylindole (1.7 g, 8.0 mmol, 1.0 equiv) in anhydrous 1:1 Et<sub>2</sub>O/THF (20 mL) was slowly added to the reaction flask at –78 °C, and the reaction was stirred at –78 °C for 15 min then used directly in the next step.

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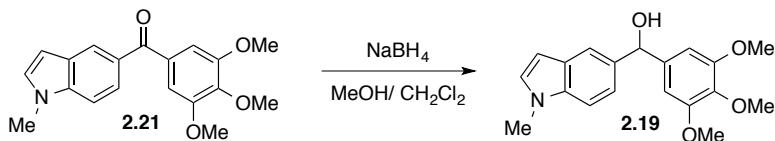
<sup>23</sup> Moyer, M. P.; Shiurba, J. F.; Rapoport, H. *J. Org. Chem.* **1986**, *51*, 5106.

To this mixture was slowly added a solution of 3,4,5-trimethoxybenzaldehyde (3.92 g, 20.0 mmol, 2.50 equiv) in 1:1 Et<sub>2</sub>O/THF (15 mL) via syringe at -78 °C. The reaction was allowed to warm to -20 °C slowly over several hours, then stirred at -20 °C overnight. The reaction was quenched with saturated aqueous ammonium chloride (30 mL) and extracted with EtOAc (3 x 30 mL). The combined organic layers were washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated in vacuo. The product was purified by flash column chromatography using 20–50% EtOAc/hexanes to afford the title compound as a white solid (2.0 g, 6.2 mmol, 78%). Analytical data are consistent with literature values.<sup>17</sup> TLC R<sub>f</sub> = 0.4 (40% EtOAc/hexanes); **m.p.** 125–126 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz) δ 7.68 (s, 1H), 7.36–7.28 (m, 2H), 7.13 (d, *J* = 3.2 Hz, 1H), 6.73 (s, 2H), 6.54 (d, *J* = 3.2 Hz, 1H), 5.94 (d, *J* = 3.3 Hz, 1H), 3.90 (s, 3H), 3.88 (s, 6H), 3.84 (s, 3H), 2.44 (d, *J* = 3.3 Hz, 1H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz) δ 153.2, 140.3, 136.9, 136.4, 135.1, 129.5, 128.4, 120.7, 119.2, 109.5, 103.4, 101.2, 76.9, 60.9, 56.1, 33.0; IR (neat) 3478 (br), 2938, 1123, 907, 724 cm<sup>-1</sup>; HRMS (TOF MS ES+) *m/z* calcd for C<sub>19</sub>H<sub>21</sub>NO<sub>4</sub> (M + Na)<sup>+</sup> 350.1368, found 350.1358.



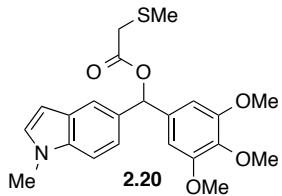
**(1-Methyl-1*H*-indol-5-yl)(3,4,5-trimethoxyphenyl)methanone (2.21).** Isolated as a byproduct from the lithiation reaction intended to form alcohol **2.19**, if the reaction was warmed to room temperature before quenching. The following amounts of reagents were used: *tert*-butyllithium (10.4 mL, 17.6 mmol, 1.70 M in pentanes, 2.20 equiv), 5-bromo-1-methylindole (1.7 g, 8.0 mmol, 1.0 equiv), 3,4,5-trimethoxybenzaldehyde (3.92 g, 20.0 mmol, 2.50 equiv), and anhydrous 1:1 Et<sub>2</sub>O/THF (35 mL). The product was purified by flash column chromatography using 20–30–

40% EtOAc/hexanes to afford the title compound as a white solid (1.9 g, 5.9 mmol, 74%). Analytical data are consistent with literature values.<sup>17</sup> **TLC**  $R_f = 0.5$  (40% EtOAc/hexanes); **m.p.** 115–116 °C; **<sup>1</sup>H NMR** ( $\text{CDCl}_3$ , 400 MHz)  $\delta$  8.14 (d,  $J = 1.6$  Hz, 1H), 7.80 (dd,  $J = 8.5, 1.7$  Hz, 1H), 7.40 (d,  $J = 8.8$  Hz, 1H), 7.15 (d,  $J = 3.2$  Hz, 1H), 7.08 (s, 2H), 6.61 (d,  $J = 3.1$  Hz, 1H), 3.95 (s, 3H), 3.88 (s, 6H), 3.86 (s, 3H); **<sup>13</sup>C NMR** ( $\text{CDCl}_3$ , 125 MHz)  $\delta$  196.5, 152.8, 141.3, 139.0, 134.3, 130.6, 129.4, 127.8, 125.2, 123.9, 109.1, 107.7, 103.0, 61.1, 56.4, 33.2; **IR** (neat) 2941, 1652, 1581, 1324, 1119  $\text{cm}^{-1}$ ; **HRMS** (TOF MS ES+)  $m/z$  calcd for  $\text{C}_{19}\text{H}_{19}\text{NO}_4$  (M + Na)<sup>+</sup> 326.1392, found 326.1382.



**(1-Methyl-1*H*-indol-5-yl)(3,4,5-trimethoxyphenyl)methanol (2.19)** could alternatively be prepared by reduction of ketone **2.21** according to a modified procedure reported by Franzén and co-workers.<sup>24</sup> Ketone **2.21** (0.98 g, 3.0 mmol, 1.0 equiv) was dissolved in 1:1 MeOH/CH<sub>2</sub>Cl<sub>2</sub> (20 mL) and the reaction cooled to 0 °C. NaBH<sub>4</sub> (0.18 g, 4.8 mmol, 1.6 equiv) was added in one portion and the reaction stirred 30 min at 0 °C, then 3 h at room temperature. The reaction was quenched with water (20 mL) and extracted with CH<sub>2</sub>Cl<sub>2</sub> (3 × 20 mL). The combined organic layers were washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated in vacuo. The product was purified by flash column chromatography using 20–30–40% EtOAc/hexanes to afford the title compound as a white solid (0.80 g, 2.5 mmol, 82%). Analytical data are consistent with the values listed for **2.19** (vide supra).

<sup>24</sup> Wang, Y.; Franzén, R. *Synlett*. **2012**, 23, 925.



**(1-Methyl-1*H*-indol-5-yl)(3,4,5-trimethoxyphenyl)methyl 2-(methylthio)acetate (2.20).**

Prepared according to a modified procedure reported by Meyer and co-workers.<sup>19</sup> To a stirring solution of alcohol **2.19** (0.46 g, 1.4 mmol, 1.0 equiv), N,N'-dicyclohexylcarbodiimide (DCC, 0.32 g, 1.5 mmol, 1.1 equiv), and 4-dimethylaminopyridine (DMAP, 94 mg, 0.77 mmol, 0.55 equiv) in anhydrous CH<sub>2</sub>Cl<sub>2</sub> (10 mL) was added 2-(methylthio)acetic acid (0.11 mL, 1.5 mmol, 1.1 equiv). The opaque reaction mixture was allowed to stir at room temperature for 20 h. The mixture was passed through a plug of Celite, rinsed with CH<sub>2</sub>Cl<sub>2</sub>, and concentrated in vacuo. The product was purified by flash column chromatography using 20–30% EtOAc/hexanes to afford the title compound a white solid (0.33 g, 0.79 mmol, 57%). **TLC R<sub>f</sub>** = 0.7 (40% EtOAc/hexanes); **m.p.** 92–93 °C; **<sup>1</sup>H NMR** (CDCl<sub>3</sub>, 400 MHz) δ 7.62 (s, 1H), 7.27 (d, *J* = 8.5 Hz, 1H), 7.21 (dd, *J* = 8.5, 1.7 Hz, 1H), 7.03 (d, *J* = 3.2 Hz, 1H), 6.99 (s, 1H), 6.66 (s, 2H), 6.45 (d, *J* = 3.1 Hz, 1H), 3.83 (s, 3H), 3.80 (s, 6H), 3.72 (s, 3H), 3.30 (d, *J* = 14.1 Hz, 1H), 3.24 (d, *J* = 14.1 Hz, 1H), 2.15 (s, 3H); **<sup>13</sup>C NMR** (CDCl<sub>3</sub>, 125 MHz) δ 169.2, 153.1, 137.1, 136.3 (2C), 130.5, 129.6, 128.2, 121.1, 120.1, 109.3, 103.8, 101.2, 78.4, 60.7, 56.0, 35.9, 32.8, 16.2; **IR** (neat) 2935, 1729, 1243, 1116, 727 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>22</sub>H<sub>25</sub>NO<sub>5</sub>S (M + Na)<sup>+</sup> 438.1351, found 438.1344.

## ***General Procedures for Biological Experiments***

### *Materials*

DMEM/high glucose media (containing 10% FBS and supplemented with L-glutamine and gentamycin) was purchased from HyClone.

### *Tubulin Polymerization Disruption Experiments*

Fluorescence imaging was performed on glass-bottom 3 mL plates suitable for microscopy that were seeded with near-confluent monolayers of an LLCPK cell line containing GFP-tagged tubulin. The cells were cultured in 10% DMEM media and imaged prior to substrate exposure. The compounds (*R*)-**2.7**, (*S*)-**2.7**, **2.10**, (*R*)-**2.13**, (*S*)-**2.13**, **2.15**, and colchicine control (**1.1**) were dissolved in molecular biology grade DMSO to achieve 10 mM stock solutions. The 10 mM DMSO stock solutions were subsequently diluted to a final concentration of 10 µM in 3 mL of 10% DMEM media. The media was aspirated from the plates containing the LLCPK cells and the media containing the compounds or DMSO vehicle controls were added. After culturing for 15 min, tubulin disruption was measured using a deconvolution fluorescence microscope. The cells were cultured for another 15 min and imaged again. Tubulin disruption was qualitatively determined by the degree of tubulin depolymerization relative to the colchicine and DMSO vehicle controls. This assay was repeated with all compounds to confirm the results.

Fluorescence imaging was performed on a Zeiss Axiovert 200 M using the Axiovision camera and software, and images were processed in Photoshop 8.0.

## *Chapter Three*

### **Stereospecific Nickel-Catalyzed Cross-Coupling Reactions of Aryl-Substituted Tetrahydropyrans for the Synthesis and Biological Evaluation of 3,5-Diaryl Alcohols**

#### **3.1 Introduction**

The structural complexity of unnatural polyketide analogues provides an intriguing challenge with respect to their synthesis.<sup>1,2,3,4</sup> A major obstacle is controlling the relative configuration of substituents during the synthesis of acyclic polyketides. In the total synthesis of erythromycin A, Woodward and co-workers presented an innovative solution: implementing a dithiadecalin system to control the relative stereochemistry of the cyclic molecule, followed by subsequent ring-opening to reveal the highly functionalized, acyclic natural product.<sup>5</sup> Similarly, others have utilized this approach to synthesize substituted polyketides.<sup>6</sup>

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<sup>1</sup> Portions of this Chapter were originally published as: Tollefson, E. J.; Dawson, D. D.; Osborne, C. A.; Jarvo, E. R. *J. Am. Chem. Soc.* **2014**, *136*, 14951.

<sup>2</sup> (a) Rohr, J. *Angew. Chem., Int. Ed.* **2000**, *39*, 2847; (b) Newman, D. J.; Cragg, G. M. *J. Nat. Prod.* **2007**, *70*, 461; (c) Cragg, G. M.; Grothaus, P. G.; Newman, D. J. *Chem. Rev.* **2009**, *109*, 3012.

<sup>3</sup> For selected reviews, see: (a) ter Horst, B.; Feringa, B. L.; Minnaard, A. J. *Chem. Commun.* **2010**, *46*, 2535; (b) Hanessian, S.; Giroux, S.; Mascitti, V. *Synthesis* **2006**, *7*, 1057; (c) Schetter, B.; Mahrwald, R. *Angew. Chem., Int. Ed.* **2006**, *45*, 7506; (d) Dechert-Schmitt, A.-M. R.; Schmitt, D. C.; Gao, X.; Itoh, T.; Krische, M. J. *Nat. Prod. Rep.* **2014**, *31*, 504.

<sup>4</sup> For biosynthetic strategies for synthesis of unnatural polyketides, see: (a) Tang, Y.; Khosla, C. Biosynthesis of “Unnatural” Natural Products. In *Exploiting Chemical Diversity for Drug Discovery*; Bartlett, P. A.; Entzeroth, M., Eds.; Royal Society of Chemistry: Dorset, U.K., 2006; (b) Zhang, W.; Tang, Y. *J. Med. Chem.* **2008**, *51*, 2629.

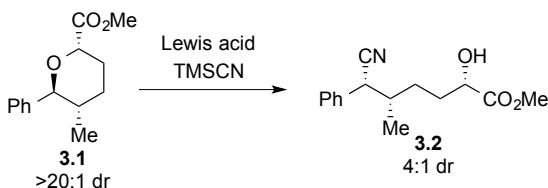
<sup>5</sup> Woodward, R. B.; Logusch, E.; Nambiar, K. P.; Sakan, K.; Ward, D. E.; Au-Yeung, B.-W.; Balaran, P.; Browne, L. J.; Card, P. J.; Chen, C. H.; Chenevert, R. B.; Fliri, A.; Frobel, K.; Gais, H.-J.; Garratt, D. G.; Hayakawa, K.; Heggie, W.; Hesson, D. P.; Hoppe, D.; Hoppe, I.; Hyatt, J. A.; Ikeda, D.; Jacobi, P. A.; Kim, K. S.; Kobuke, Y.; Kojima, K.; Krowicki, K.; Lee, V. J.; Leutert, T.; Malchenko, S.; Martens, J.; Matthews, R. S.; Ong, B. S.; Press, J. B.; Rajan Babu, T. V.; Rousseau, G.; Sauter, H. M.; Suzuki, M.; Tatsuta, K.; Tolbert, L. M.; Truesdale, E. A.; Uchida, I.; Ueda, Y.; Uyehara, A. T.; Vasella, W. C.; Vladuchick, W. C.; Wade, P. A.; Williams, R. M.; Wong, H. N.-C. *J. Am. Chem. Soc.* **1981**, *103*, 3210.

<sup>6</sup> Ward, D. E. *Chem. Commun.* **2011**, *47*, 11375.

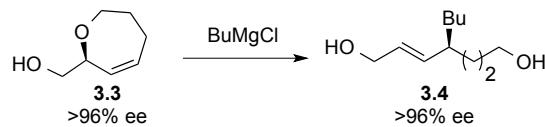
To access substituted unnatural polyketides, the ring-opening reactions of tetrahydropyrans have been developed,<sup>7</sup> although few examples of ring-opening by forming a new C<sub>sp</sub><sup>3</sup>–C<sub>sp</sub><sup>3</sup> bond exist.<sup>8,9</sup> Notably, Panek and co-workers disclosed the diastereoselective, Lewis acid-catalyzed ring-opening reaction of tetrahydropyrans using cyanide, where the stereochemical outcome of the reaction is consistent with carbocation formation (Scheme 3.1a).<sup>8a</sup>

**Scheme 3.1.** Stereoselective ring-opening and C–C bond formation strategy.

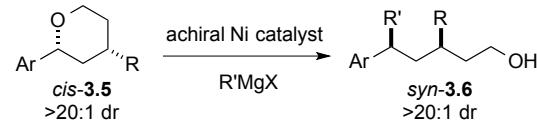
a) stereoab ablative via carbocation (Panek, 2007):



b) stereospecific S<sub>N</sub>2' (Hoveyda, 1997):



c) stereospecific via organonickel intermediate (this work):



While diastereoselective ring-opening reactions constitute one method of controlling relative stereochemistry, a complementary approach involves stereospecific transformations. Here, stereochemical information is conserved throughout the reaction. For example, Hoveyda and co-workers showed that unsaturated cyclic ethers activated by pendant alcohols could

<sup>7</sup> (a) Burwell, R. L., Jr. *Chem. Rev.* **1954**, *54*, 615; (b) Maercker, A. *Angew. Chem., Int. Ed.* **1987**, *26*, 972; (c) For a recent example, see: Mack, D. J.; Guo, B.; Njarðarson, J. T. *Chem. Commun.* **2012**, *48*, 7844.

<sup>8</sup> (a) Qin, H.-L.; Lowe, J. T.; Panek, J. S. *J. Am. Chem. Soc.* **2007**, *129*, 38; (b) Sawama, Y.; Shibata, K.; Sawama, Y.; Takubo, M.; Monguchi, Y.; Krause, N.; Sajiki, H. *Org. Lett.* **2013**, *15*, 5282; (c) Oku, A.; Homoto, Y.; Harada, T. *Chem. Lett.* **1986**, 1495; (d) Christensen, S. H.; Holm, T.; Madsen, R. *Tetrahedron* **2014**, *70*, 4942.

<sup>9</sup> For examples of allylic substitution reactions that open dihydropyrans and lactones, see: (a) Sawama, Y.; Sawama, Y.; Krause, N. *Org. Lett.* **2009**, *11*, 5034; (b) Matsushita, H.; Negishi, E.-i. *J. Chem. Soc., Chem. Commun.* **1982**, 160.

undergo stereospecific  $S_N2'$  displacements by Grignard reagents to provide enantioenriched acyclic products (Scheme 3.1b).<sup>10</sup>

Our goal was to expand the scope of stereospecific ring-opening reactions to include *saturated* cyclic ethers, such as tetrahydropyrans, that are not activated by ring strain.<sup>11,12</sup> Based on the Jarvo laboratory's enantiospecific Kumada cross-coupling reaction of ethers,<sup>13,14</sup> we anticipated that we could develop the stereospecific nickel-catalyzed ring-opening reactions of cyclic ethers (Scheme 3.1c). We envisioned that this stereospecific methodology would provide complex acyclic fragments containing both a medicinally relevant benzylic methyl substituent<sup>15</sup> and a pendant alcohol for further derivatization.

In this Chapter, we report the stereospecific Kumada cross-coupling reaction of tetrahydropyrans with a range of Grignard reagents. We also report the derivatization of an enantioenriched furan-containing carboxylic acid formed via the stereospecific Negishi cross-coupling reaction of a benzylic lactone with dimethyl zinc. This methodology has generated a range of unnatural polyketide analogues for biological testing. We report the discovery of several

<sup>10</sup> (a) Heron, N. M.; Adams, J. A.; Hoveyda, A. H. *J. Am. Chem. Soc.* **1997**, *119*, 6205; (b) Adams, J. A.; Heron, N. M.; Koss, A.-M.; Hoveyda, A. H. *J. Org. Chem.* **1999**, *64*, 854; (c) The Hoveyda group has also reported enantioselective, catalyst controlled ring-opening of cyclic unsaturated ethers with chiral zirconium-based catalysts. For a lead reference, see: Didiuk, M. T.; Johannes, C. W.; Morken, J. P.; Hoveyda, A. H. *J. Am. Chem. Soc.* **1995**, *117*, 7097.

<sup>11</sup> For nickel-catalyzed cross-coupling reaction of dihydrofurans with Grignard reagents, see: Cornell, J.; Martin, R. *Org. Lett.* **2013**, *24*, 6298.

<sup>12</sup> For examples of nickel-catalyzed addition to strained heterocycles, e.g., epoxides and aziridines, see: (a) Molinaro, C.; Jamison, T. F. *J. Am. Chem. Soc.* **2003**, *125*, 8076; (b) Lin, B. L.; Clough, C. R.; Hillhouse, G. L. *J. Am. Chem. Soc.* **2002**, *124*, 2890; (c) Nielsen, D. K.; Doyle, A. G. *Angew. Chem., Int. Ed.* **2011**, *50*, 6056; (d) Nielsen, D. K.; Huang, C.-Y.; Doyle, A. G. *J. Am. Chem. Soc.* **2013**, *135*, 13605; (e) Jensen, K. L.; Standley, E. A.; Jamison, T. F. *J. Am. Chem. Soc.* **2014**, *136*, 11145; (f) Takeda, Y.; Ikeda, Y.; Kuroda, A.; Tanaka, S.; Minakata, S. *J. Am. Chem. Soc.* **2014**, *136*, 8544.

<sup>13</sup> (a) Taylor, B. L. H.; Swift, E. C.; Waetzig, J. D.; Jarvo, E. R. *J. Am. Chem. Soc.* **2011**, *133*, 389; (b) Taylor, B. L. H.; Harris, M. R.; Jarvo, E. R. *Angew. Chem., Int. Ed.* **2012**, *51*, 7790; (c) Greene, M. A.; Yonova, I. M.; Williams, F. J.; Jarvo, E. R. *Org. Lett.* **2012**, *14*, 4293; (d) Yonova, I. M.; Johnson, A. G.; Osborne, C. A.; Moore, C. E.; Morrisette, N. S.; Jarvo, E. R. *Angew. Chem., Int. Ed.* **2014**, *53*, 2422.

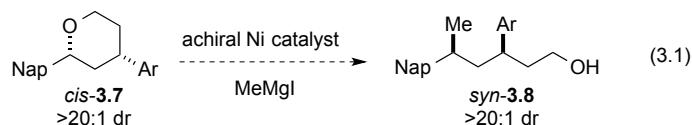
<sup>14</sup> For a recent review of nickel-catalyzed reactions, see: Tasker, S. Z.; Standley, E. A.; Jamison, T. F. *Nature* **2014**, *509*, 299.

<sup>15</sup> For a review, see: Barreiro, E. J.; Kümmerle, A. E.; Fraga, C. A. M. *Chem. Rev.* **2011**, *111*, 5215.

compounds that exhibit selective anti-breast cancer activity against the MCF-7 (ER+ breast cancer) and MDA-MB-468 (triple-negative breast cancer) cell lines.

### 3.2 Stereospecific Kumada Cross-Coupling Reactions of Tetrahydropyrans

We sought to develop this methodology as a powerful strategy for the cross-coupling reactions of heterocycles containing multiple stereogenic centers. Toward this end, we applied our reaction to the opening of *cis*-( $\pm$ )-2,4-disubstituted tetrahydropyrans, subunits of the calyxin family of natural products.<sup>16</sup> These cross-coupling reactions would provide synthetic access to *syn*-3,5-disubstituted alcohols (eq. 3.1).



There are several elegant methods for the diastereoselective synthesis of highly substituted tetrahydropyrans,<sup>17</sup> including diastereoselective Prins cyclization reactions.<sup>18</sup> We developed the two-step diastereoselective strategy outlined in Scheme 3.2 to easily obtain tetrahydropyrans containing a broad range of aryl substituents at the C4 position. First, a MgBr<sub>2</sub> and *p*-TsOH-promoted Prins cyclization afforded 4-bromotetrahydropyran **3.9** as a 2:1 mixture of diastereomers.<sup>19</sup> Then, we employed a diastereoselective nickel-catalyzed Suzuki cross-coupling reaction to install an array of aryl substituents.<sup>20</sup> This approach takes advantage of Fu

<sup>16</sup> Prasain, J. K.; Li, J.-X.; Tezuka, Y.; Tanaka, K.; Basnet, P.; Dong, H.; Namba, T.; Kadota, S. *J. Nat. Prod.* **1998**, 61, 212.

<sup>17</sup> (a) Nicolas, L.; Butkevich, A. N.; Guérinot, A.; Corbu, A.; Reymond, S.; Cossy, J. *Pure Appl. Chem.* **2013**, 85, 1203; (b) Pastor, I. M.; Yus, M. *Curr. Org. Chem.* **2012**, 16, 1277.

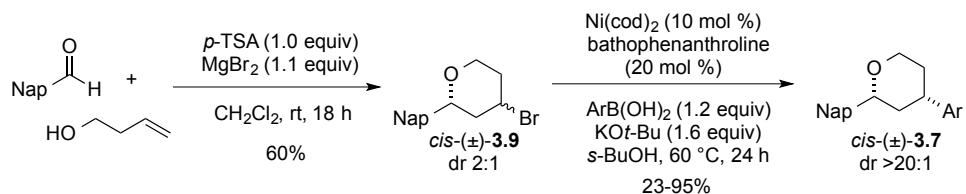
<sup>18</sup> (a) For a recent review, see reference 17b; (b) Jasti, R.; Rychnovsky, S. D. *J. Am. Chem. Soc.* **2006**, 128, 13640; (c) Alder, R. W.; Harvey, J. N.; Oakley, M. T. *J. Am. Chem. Soc.* **2002**, 124, 4960.

<sup>19</sup> Borkar, P.; van de Weghe, P.; Subba Reddy, B. V.; Yadav, J. S.; Grée, R. *Chem. Commun.* **2012**, 48, 9316.

<sup>20</sup> Zultanski, S. L.; Fu, G. C. *J. Am. Chem. Soc.* **2013**, 135, 624.

and co-workers' seminal stereoconvergent Suzuki reaction.<sup>21</sup> In this transformation, diastereoselectivity stems from the preferred conformation of the radical intermediate.<sup>22</sup> The thermodynamically-favored *cis* diastereomer is predicted as the major product; indeed, the cross-coupling reactions of **3.9** with a range of commercially available aryl boronic acids afforded a variety of 4-aryltetrahydropyrans in high diastereoselectivity. The relative configurations of these *cis*-2,4-diaryl tetrahydropyrans were assigned by NOE NMR experiments.<sup>23</sup>

**Scheme 3.2.** Diastereoselective synthesis of tetrahydropyrans.



Development of the nickel-catalyzed ring-opening reactions of cyclic ethers was guided by our prior experience developing Kumada cross-coupling reactions of benzylic ethers. We observed that 15 mol % catalyst loading of Ni(cod)<sub>2</sub> and *rac*-BINAP in the presence of methylmagnesium iodide resulted in good to excellent yields of the desired 3,5-diaryl alcohols (Table 3.1). We also examined the transfer of stereochemical information in the cross-coupling reaction by comparing the diastereomeric ratios of the starting materials to those of the acyclic products. For example, tetrahydropyran **3.10** (dr >20:1) afforded *syn*-**3.11** in 76% yield and >20:1 dr (entry 1), indicating the complete transfer of stereochemical information in the cross-coupling reaction.

<sup>21</sup> Zhou, J.; Fu, G. C. *J. Am. Chem. Soc.* **2004**, *126*, 1340.

<sup>22</sup> (a) Zultanski, S. L.; Fu, G. C. *J. Am. Chem. Soc.* **2011**, *133*, 15362; (b) Stille, J. K.; Cowell, A. B. *J. Organomet. Chem.* **1977**, *124*, 253.

<sup>23</sup> See Experimental Section for details.

**Table 3.1.** Scope of cross-coupling reaction of tetrahydropyrans.

Nap  
>20:1 dr

$\xrightarrow[\substack{\text{PhMe, rt, 24 h}}]{\substack{\text{Ni}(\text{cod})_2 \text{ (15 mol \%)} \\ \text{rac-BINAP (15 mol \%)} \\ \text{MeMgI (2.5 equiv)}}}$

Product

Yield (%)<sup>a</sup> Prod. dr<sup>b</sup>

---

Entry	Starting Material	Product	Yield (%) <sup>a</sup>	Prod. dr <sup>b</sup>
1	Nap <i>cis</i> -(±)-3.10	Nap <i>syn</i> -(±)-3.11	76 <sup>c</sup>	>20:1
2	Nap <i>cis</i> -(±)-3.12	Nap <i>syn</i> -(±)-3.13	72	>20:1
3	Nap <i>cis</i> -(±)-3.14	Nap <i>syn</i> -(±)-3.15	81 <sup>c</sup>	>20:1
4	Nap <i>cis</i> -(±)-3.16	Nap <i>syn</i> -(±)-3.17	42 <sup>c</sup>	>20:1
5	Nap <i>cis</i> -(±)-3.18	Nap <i>syn</i> -(±)-3.19	74	>20:1
6	Nap <i>cis</i> -(±)-3.20	Nap <i>syn</i> -(±)-3.21	63	>20:1

---

<sup>a</sup>Isolated yield after column chromatography. <sup>b</sup>Determined by <sup>1</sup>H NMR. <sup>c</sup>Calculated

yield; see Experimental Section for details. Nap = 2-naphthyl.

We evaluated the scope of the cross-coupling reaction and found that both electron-rich and electron-poor aryl substituents at the C4 position of the tetrahydropyran are well tolerated (Table 3.1, entries 1 and 2). To challenge this methodology, we sought to incorporate biologically-relevant moieties in our substrates. For example, the cross-coupling reaction of tetrahydropyran *cis*-**3.14** proceeded in 81% yield and >20:1 dr to form benzodioxane-substituted product *syn*-**3.15** (entry 3). 1,4-Benzodioxanes are present in a range of pharmaceutical agents such as piperoxan and idazoxan.<sup>24</sup> The reaction of tetrahydropyran **3.16** afforded product **3.17** in modest yield (entry 4); the 3,4,5-trimethoxyphenyl functionality is commonly found in compounds that inhibit tubulin polymerization, such as colchicine and combretastatin A-4.<sup>25</sup> We were also gratified to see that 3-furan-substituted tetrahydropyran *cis*-**3.18** was well tolerated in the reaction. Product *syn*-**3.19** was formed in high yield and dr, and contains a furan substituent that can be readily derivatized by oxidation or cycloaddition reactions (entry 5).<sup>26</sup>

To further challenge this method, we synthesized a stereotriad by the Kumada coupling of 2,4,6-trisubstituted tetrahydropyran *cis*-**3.20**. Subjecting *cis*-**3.20** to the reaction conditions afforded the secondary alcohol *syn*-**3.21**, containing three stereogenic centers, as a single diastereomer and with good yield (Table 3.1, entry 6). This strategy introduces a modular three-step synthesis of polyketide analogues where substituents in the C2, C4, and C6 positions can be easily altered by the use of commercially available aldehydes, arylboronic acids, and homoallylic alcohols, respectively.

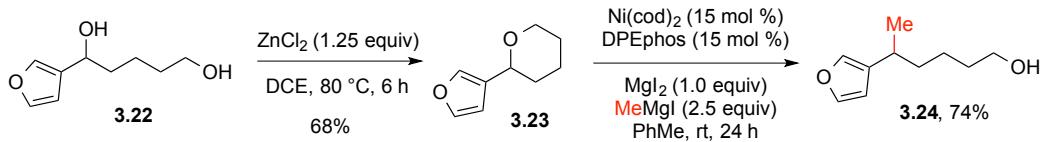
<sup>24</sup> (a) Piperoxan: Fourneau, E.; Bovet, D. *Arch. Int. Pharmacodyn. Thér.* **1933**, *46*, 178; (b) Idazoxan synthesis and pharmacology: Chapleo, C. B.; Myers, P. L.; Butler, R. C. M.; Doxey, J. C.; Roach, A. G.; Smith, C. F. C. *J. Med. Chem.* **1983**, *26*, 823.

<sup>25</sup> (a) Jordan, M. A.; Wilson, L. *Nat. Rev. Cancer* **2004**, *4*, 253; (b) Lin, C. M.; Ho, H. H.; Pettit, G. R.; Hamel, E. *Biochemistry* **1989**, *28*, 6984; (c) Tron, G. C.; Pirali, T.; Sorba, G.; Pagliai, F.; Busacca, S.; Genazzani, A. A. *J. Med. Chem.* **2006**, *49*, 3033.

<sup>26</sup> (a) Kobayashi, Y.; Kumar, G. B.; Kurachi, T.; Acharya, H. P.; Yamazaki, T.; Kitazume, T. *J. Org. Chem.* **2001**, *66*, 2011; (b) Diels, O.; Alder, K. *Ber. Dtsch. Chem. Ges.* **1929**, *62*, 554.

We were interested in determining whether this cross-coupling reaction could be performed using substrates containing aryl substituents other than naphthyl. We hypothesized that tetrahydropyran **3.23** would undergo a nickel-catalyzed Kumada cross-coupling reaction due to the lower aromatic stabilization energy of furan.<sup>27</sup> Our laboratory has observed a strong dependence of cross-coupling rates on the identity of the aryl substituent, and we predict that arenes possessing lower aromatic stabilization energy provide better ligation of the nickel catalyst and greater stabilization of the transition state for oxidative addition. Benzylic ethers and esters activated by extended aromatic rings such as naphthalene and benzofuran are sufficiently reactive, as are those activated by furan.<sup>28</sup> The synthesis of compound **3.23** is easily achieved by a Lewis acid-promoted cyclization of the corresponding 1,5-diol (**3.22**, Scheme 3.3).<sup>29</sup> Subjecting 3-furan-substituted tetrahydropyran **3.23** to the nickel-catalyzed Kumada reaction formed the desired product **3.24** in 74% yield. We found DPEphos to be a more effective ligand than *rac*-BINAP for this transformation, consistent with our observation that DPEphos provides the highest yields for heteroaromatic-containing substrates.<sup>30</sup> In addition, using one equivalent of MgI<sub>2</sub> in the reaction slightly increased the yield from 70% to 74%.<sup>31</sup>

**Scheme 3.3.** Synthesis and Kumada cross-coupling reaction of 3-furan-substituted tetrahydropyran **3.23**.



<sup>27</sup> (a) For thermodynamic parameters of bonding of arenes to nickel complexes, see: Brauer, D. J.; Krüger, C. *Inorg. Chem.* **1977**, *16*, 884; (b) For resonance energies of arenes, see: Smith, M. B.; March, J. *Advanced Organic Chemistry: Reactions, Mechanisms, and Structure*; John Wiley & Sons, Inc.: New York, 2007; pp 60–62.

<sup>28</sup> Harris, M. R.; Konev, M. O.; Jarvo, E. R. *J. Am. Chem. Soc.* **2014**, *136*, 7825.

<sup>29</sup> Kim, S.; Chung, K. N.; Yang, S. *J. Org. Chem.* **1987**, *52*, 3917.

<sup>30</sup> Tollefson, E. J.; Dawson, D. D.; Osborne, C. A.; Jarvo, E. R. *J. Am. Chem. Soc.* **2014**, *136*, 14951.

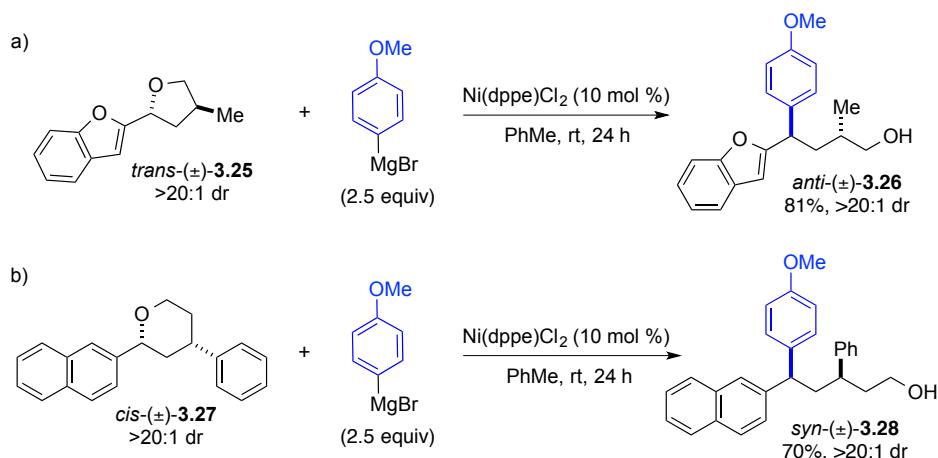
<sup>31</sup> Greene, M. A. Diastereoselective Synthesis of Seven Membered Ring *trans*-Alkenes and Development of Stereospecific Nickel-Catalyzed Cross-Coupling Reactions. Ph.D. Thesis, The University of California, Irvine, May 2013.

### 3.3 Grignard Reagent Scope

It is critical for synthetic methodology to provide access to analogues with a range of substituent patterns to achieve the long-term goal of determining structure-activity relationships (SAR). To this end, we were interested in examining the scope of our reaction with respect to the transmetallating agent. Successful incorporation of other Grignard reagents would provide modular access to acyclic products containing a variety of benzylic substituents.

The Jarvo laboratory has recently demonstrated that Ni(dppe)Cl<sub>2</sub> is a broadly applicable catalyst for cross-coupling reactions of alkyl and aryl Grignard reagents with benzylic ethers.<sup>13d</sup> We applied this catalyst system to the ring-opening reactions of a representative tetrahydrofuran and tetrahydropyran and found that aryl Grignard reagents were well tolerated in the synthesis of complex diarylalkanes (Scheme 3.4). For example, 4-methoxyphenylmagnesium bromide underwent smooth cross-coupling reactions with both tetrahydrofuran **3.25** and tetrahydropyran **3.27**. Products *anti*-**3.26** and *syn*-**3.28** were formed in high yield and dr (Scheme 3.4).

**Scheme 3.4.** Use of aryl Grignard reagent in the stereospecific Kumada cross-coupling reaction of (a) tetrahydrofuran **3.25** and (b) tetrahydropyran **3.27**.



### 3.4 Derivatization of a Furan-Containing Product Synthesized via the Negishi Cross-Coupling Reaction of Lactones

Based on our recently reported nickel-catalyzed Negishi cross-coupling reaction of benzylic esters,<sup>32</sup> we hypothesized that a similar reaction could be applied to the ring-opening of benzylic lactones. Utilizing alkyl zinc reagents in the transformation would allow for greater functional group tolerance as compared to Grignard reagents. Furthermore, the ring-opening reaction of enantioenriched lactones would provide products containing a benzylic stereocenter and a pendant carboxylic acid available for derivatization. Recently, Sawama and co-workers have disclosed the Lewis acid-catalyzed ring-opening reaction of aryl-substituted lactones with allylsilane.<sup>8b</sup> However, stereospecific ring-opening of lactones with other carbon-based nucleophiles has not been reported.

The required starting lactone (*R*)-**3.31** can be readily accessed as a single enantiomer (Scheme 3.5).<sup>33</sup> Corey–Bakshi–Shibata (CBS) reduction of benzylic ketone **3.29**<sup>34</sup> followed by hydroboration/ oxidation and cyclization of 1,5-diol (*R*)-**3.22** provided (*R*)-**3.31** in 90% ee.<sup>35</sup> The absolute configuration of the lactone was assigned as *R* based on the accepted CBS model of selectivity for intermediate alcohol (*R*)-**3.30** (see Experimental Section for details).<sup>34</sup>

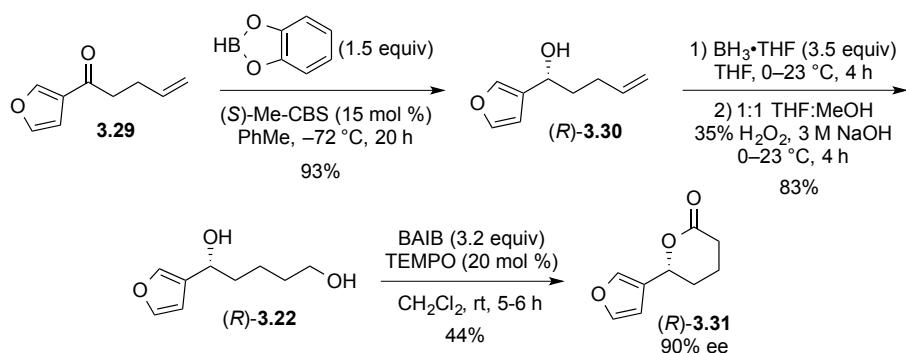
<sup>32</sup> Wisniewska, H. M.; Swift, E. C.; Jarvo, E. R. *J. Am. Chem. Soc.* **2013**, *135*, 9083.

<sup>33</sup> (a) For a review, see: Boucard, V.; Broustal, G.; Campagne, J. M. *Eur. J. Org. Chem.* **2007**, 225. For recent examples, see: (b) Murphy, S. K.; Dong, V. M. *J. Am. Chem. Soc.* **2013**, *135*, 5553; (c) Moran, J.; Smith, A. G.; Carris, R. M.; Johnson, J. S.; Krische, M. J. *J. Am. Chem. Soc.* **2011**, *133*, 18618.

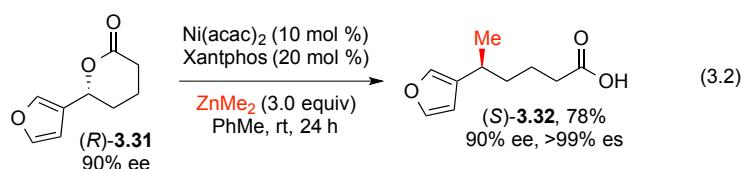
<sup>34</sup> Corey, E. J.; Helal, C. J. *Angew. Chem., Int. Ed.* **1998**, *37*, 1986.

<sup>35</sup> (a) Hansen, T. M.; Florence, G. J.; Lugo-Mas, P.; Chen, J.; Abrams, J. N.; Forsyth, C. J. *Tetrahedron Letters* **2003**, *44*, 57; (b) This type of oxidative cyclization is known to proceed with retention of stereochemical information; see: Kamal, A.; Sandbhor, M.; Shaik, A. A. *Tetrahedron: Asymmetry* **2003**, *14*, 1575.

**Scheme 3.5.** Synthesis of valerolactone (*R*)-3.31.<sup>36</sup>



Natural products including ricciocarpin A and salvinorin B contain  $\delta$ -valerolactones with furan substituents, such as **3.31**.<sup>37</sup> Methods to open such aryl-substituted valerolactones would provide a strategic synthesis of analogues for biological evaluation.<sup>38</sup> We predicted that, as with tetrahydropyran **3.23**, lactone **3.31** would undergo a stereospecific nickel-catalyzed cross-coupling reaction. Therefore, we evaluated 3-furan-substituted lactone (*R*)-**3.31** and found that Ni(acac)<sub>2</sub> and Xantphos afforded the cross-coupled carboxylic acid in 78% yield and >99% es (eq. 3.2).<sup>39</sup>



<sup>36</sup> (*R*)-**3.30** was synthesized by Emily Tollefson; see reference 30.

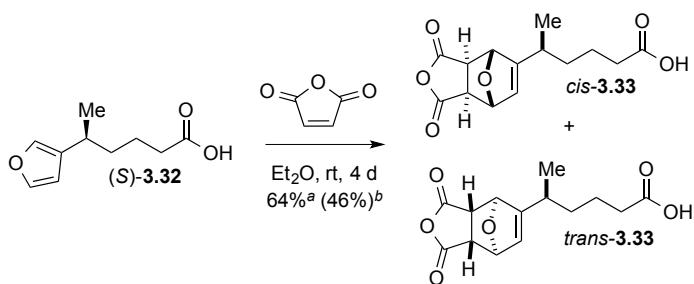
<sup>37</sup> (a) Wurzel, G.; Becker, H. *Phytochemistry* **1990**, *29*, 2565; (b) Ortega, A.; Blount, J. F.; Manchand, P. S. *J. Chem. Soc., Perkin Trans. 1* **1982**, *10*, 2505.

<sup>38</sup> (a) Clardy, J.; Walsh, C. *Nature* **2004**, *432*, 829; (b) Boldi, A. M. *Curr. Opin. Chem. Biol.* **2004**, *8*, 281; (c) Li, J.; Cisar, J. S.; Zhou, C.-Y.; Vera, B.; Williams, H.; Rodríguez, A. D.; Cravatt, B. F.; Romo, D. *Nat. Chem.* **2013**, *5*, 510; (d) Huigens, R. W.; Morrison, K. C.; Hicklin, R. W.; Flood, T. A., Jr.; Richter, M. F.; Hergenrother, P. J. *Nat. Chem.* **2013**, *5*, 195.

<sup>39</sup> es = ee<sub>product</sub>/ee<sub>starting material</sub>; see: Denmark, S. E.; Vogler, T. *Chem.–Eur. J.* **2009**, *15*, 11737.

To highlight the furan's functional utility, we derivatized product (*S*)-**3.32** by a Diels–Alder reaction with maleic anhydride (Scheme 3.6).<sup>26b,40</sup> The cycloaddition reaction furnished the enantioenriched bicyclic **3.33** in 64% yield as a 1:1 mixture of diastereomers. Based on Woodward's analysis of the thermodynamic product of the reaction, the Diels–Alder reaction is anticipated to be highly exo-selective.<sup>40</sup> Notably, this derivatization has generated an unnatural polyketide analogue containing no aromatic substituents.

**Scheme 3.6.** Diels–Alder reaction of furan (*S*)-**3.32**.



<sup>a</sup>Yield determined by <sup>1</sup>H NMR based on comparison to PhTMS as internal standard.

<sup>b</sup>Isolated yield after column chromatography.

### 3.5 Biological Evaluation of Cross-Coupling Products

We hypothesized that the 3,5-diaryl alcohol moiety would exhibit anti-cancer activity because of its structural similarity to the 1,1-diarylalkane pharmacophore. The latter scaffold is found in a range of bioactive molecules,<sup>41, 42</sup> and we have previously demonstrated in Chapter 1

<sup>40</sup> Woodward, R. B.; Baer, H. *J. Am. Chem. Soc.* **1948**, *70*, 1161.

<sup>41</sup> Representative examples: (a) as ligands for nuclear receptors, see: Kainuma, M; Kasuga, J.-i.; Hosoda, S; Wakabayashi, K.-i.; Tanatani, A; Nagasawa, K.; Miyachi, H.; Makishima, M; Hashimoto, Y. *Bioorg. Med. Chem. Lett.* **2006**, *16*, 3213; (b) as combretastatin analogs for colon cancer, see: Messaoudi, S.; Hamze, A.; Provot, O.; Tréguier, B.; Rodrigo De Losada, J.; Bignon, J.; Liu, J.-M.; Wdzieczak-Bakala, J.; Thoret, S.; Dubois, J.; Brion, J.-D.; Alami, M. *ChemMedChem.* **2011**, *6*, 488; (c) prostate cancer: Hu, Q. Z.; Yin, L. N.; Jagusch, C.; Hille, U. E.; Hartmann, R. W. *J. Med. Chem.* **2010**, *53*, 5049; (d) diabetes: Kim, R. M.; Parmee, E. R.; Tan, Q.; Yang, C.; Lins, A. R. U. S. Patent 12/227,030, May 11, 2007.

that enantioenriched diarylalkanes provide lead compounds with selective anti-breast-cancer activity.<sup>13d</sup> We evaluated several 3,5-diaryl alcohols, synthesized using our nickel-catalyzed cross-coupling reaction, against a range of cancer cell lines using an MTT assay.<sup>43</sup> Preliminary biological studies indicated that *syn*-**3.34** was selective for anti-proliferation of the MCF-7 breast cancer line compared to the MDA-MB-468 breast cancer line and prostate, kidney, and ovarian cancer lines (Figure 3.1a).

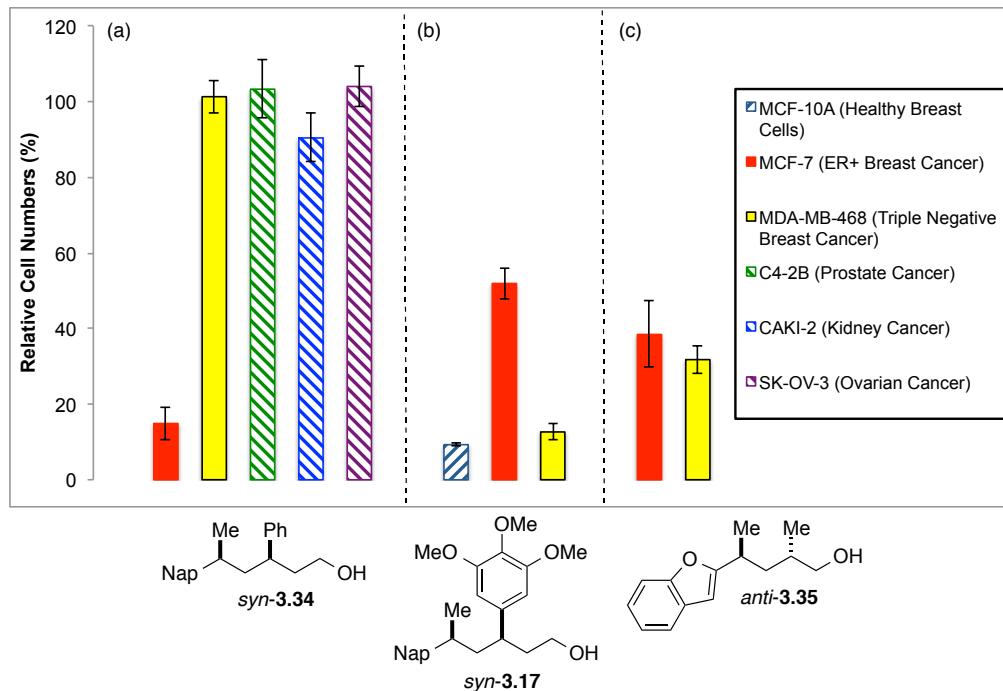
To generate analogues of this lead compound, we biased our library of tetrahydropyrans to include different aryl derivatives at the C4 position (Table 3.1). Upon evaluating analogues for biological activity, we determined that *syn*-**3.11**, *syn*-**3.13**, *syn*-**3.15**, *syn*-**3.17**, and *syn*-**3.19** did not exhibit significant anti-cancer activity. However, **3.17** proved extremely potent toward all cell lines, including the healthy MCF-10A breast cell line (Figure 3.1b). We hypothesize that the 3,4,5-trimethoxyphenyl motif of **3.17** interacts with the colchicine binding site of  $\beta$ -tubulin and thus demonstrates general cytotoxic activity.<sup>44</sup>

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<sup>42</sup> For breast cancer: (a) Pathak, T. P.; Gligorich, K. M.; Welm, B. E.; Sigman, M. S. *J. Am. Chem. Soc.* **2010**, *132*, 7870; (b) Pathak, T. P.; Osiak, J. G.; Vaden, R. M.; Welm, B. E.; Sigman, M. S. *Tetrahedron* **2012**, *68*, 5203.

<sup>43</sup> MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) assay: Alley, M. C.; Scudiero, D. A.; Monks, A.; Hursey, M. L.; Czerwinski, M. J.; Fine, D. L.; Abbott, B. J.; Mayo, J. G.; Shoemaker, R. H.; Boyd, M. R. *Cancer Res.* **1988**, *48*, 589.

<sup>44</sup> (a) see reference 25a; (b) Álvarez, R.; Álvarez, C.; Mollinedo, F.; Sierra, B. G.; Medarde, M.; Peláez, R. *Bioorg. Med. Chem.* **2009**, *17*, 6422.



**Figure 3.1.** Evaluation of compounds for anti-cancer activity at 10  $\mu\text{M}$ .<sup>45</sup> Cell proliferation is represented as relative cell numbers after treatment, where a low percentage indicates potent anti-proliferative activity. All data are normalized to the DMSO vehicle control.

Examination of additional compounds isolated from our cross-coupling reactions identified *anti*-3.35 as displaying potent activity toward both ER+ breast cancer (MCF-7) and triple-negative breast cancer (MDA-MB-468) cell lines (Figure 3.1c). Establishing activity against the latter cell line is a significant result because it is a particularly challenging cancer to treat.<sup>46</sup> These results have set the stage for future SAR studies of unnatural polyketide analogues.

<sup>45</sup> Products *syn*-3.34 and *anti*-3.35 were synthesized by Emily Tollefson; see reference 30.

<sup>46</sup> (a) Irvin, W. J., Jr.; Carey, L. A. *Eur. J. Cancer* **2008**, *44*, 2799; (b) Carey, L. A. *The Oncologist* **2011**, *16*, 71.

### **3.6 Conclusions**

We have developed the nickel-catalyzed, stereospecific cross-coupling reactions of aryl-substituted tetrahydropyrans. Through judicious choice of starting materials, cyclic ether intermediates have been utilized to set the desired relative stereochemical relationships. We have demonstrated the high stereospecificity of the reaction, where the dr of the product matches the dr of the starting tetrahydropyrans and tetrahydrofurans. The Negishi cross-coupling reaction of a benzylic lactone has also been established to synthesize an enantioenriched carboxylic acid. Evaluation of products for activity against a range of cancer cell lines identified several compounds that were selectively potent toward the MCF-7 ER+ breast cancer line and the MDA-MB-468 triple-negative breast cancer line. We are currently investigating the application of these methods toward the implementation of natural product editing to generate unnatural polyketides.

### **3.7 Experimental Details**

#### ***General Procedures***

All reactions were carried out under a N<sub>2</sub> atmosphere, unless otherwise stated. All glassware was either oven dried or flame-dried prior to use. Toluene (PhMe), diethyl ether (Et<sub>2</sub>O), dichloromethane (DCM), benzene (C<sub>6</sub>H<sub>6</sub>), and tetrahydrofuran (THF) were degassed with argon and then passed through two 4 x 36 inch columns of anhydrous neutral A-2 alumina (8 x 14 mesh; LaRoche Chemicals; activated under a flow of argon at 350 °C for 12 hours) to remove H<sub>2</sub>O. Other solvents were purchased “anhydrous” commercially, or were purified as described. <sup>1</sup>H NMR were recorded on Bruker DRX-400 (400 MHz <sup>1</sup>H, 100 MHz <sup>13</sup>C), GN-500 (500 MHz <sup>1</sup>H, 125.7 MHz <sup>13</sup>C), or CRYO-500 (500 MHz <sup>1</sup>H, 125.7 MHz <sup>13</sup>C) spectrometers. Proton chemical shifts are reported in ppm ( $\delta$ ) relative to internal trimethylsilane (TMS,  $\delta$  0.00). Data

are reported as follows: chemical shift (multiplicity [singlet (s), broad singlet (br s), doublet (d), doublet of doublets (dd), doublet of doublet of doublets (ddd), triplet (t), doublet of triplets (dt), triplet of doublets (td), doublet of doublet of triplets (ddt), quartet (q), quintet (quint), quintet of triplets (quintt), quintet of doublets (quintd), sextet (sext), septet (sept), multiplet (m), apparent doublet (ad), apparent triplet (at), apparent quartet (aq), apparent quintet (aquint)], coupling constants [Hz], integration). Carbon chemical shifts are reported in ppm ( $\delta$ ) relative to TMS with the solvent resonance as the internal standard ( $\text{CDCl}_3$ ,  $\delta$  77.16 ppm). NMR data were collected at 25 °C. Infrared spectra were obtained on a Thermo Scientific Nicolet iS5 spectrometer with an iD5 ATR tip (neat) and are reported in terms of frequency of absorption ( $\text{cm}^{-1}$ ). Analytical thin-layer chromatography (TLC) was performed using Silica Gel 60Å F254 precoated plates (0.25 mm thickness). Visualization was accomplished by irradiation with a UV lamp and/or staining with *p*-anisaldehyde (PAA), cerium ammonium molybdate (CAM), or potassium permanganate ( $\text{KMnO}_4$ ) solutions. Flash chromatography was performed using Silica Gel 60 (170-400 mesh) from Fisher Scientific or silver impregnated silica gel.<sup>47</sup> Melting points (m.p.) were obtained using a Mel-Temp melting point apparatus and are uncorrected. Optical rotations were measured with a Rudolph Research Analytical Autopol III Automatic Polarimeter. SFC determinations of enantiopurity were performed on a Berger Analytical instrument and an Aurora A5 Fusion instrument using a Daicel<sup>TM</sup> Chiralpak® column (OD-H, AD-H, OJ-H, or (*R,R*)-Whelk-O; 100 bar, 215 nm, 50 °C). High resolution mass spectrometry was performed by the University of California, Irvine Mass Spectrometry Center.

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<sup>47</sup> Shaghafi, M. B.; Kohn, B. L.; Jarvo, E. R. *Org. Lett.* **2008**, *10*, 4743.

Bis(1,5-cyclooctadiene)nickel was purchased from Strem, stored in a glove box freezer ( $-20\text{ }^{\circ}\text{C}$ ) under an atmosphere of  $\text{N}_2$  and used as received. Zinc (II) chloride was purchased from Strem and stored under an atmosphere of  $\text{N}_2$ . 1,2-Dichloroethane (DCE) was purchased from EMD Chemicals and distilled from  $\text{CaH}_2$  through a short-path distillation head ( $80\text{ }^{\circ}\text{C}$ , 40 torr). All ligands were purchased from Strem or Sigma Aldrich and were stored under  $\text{N}_2$  atmosphere and used as received. Dimethyl zinc ( $\text{ZnMe}_2$ ) was purchased from Sigma Aldrich and stored under  $\text{N}_2$  at  $4\text{ }^{\circ}\text{C}$ . All Grignard reagents and  $\text{ZnMe}_2$  were titrated with iodine prior to use.<sup>48</sup> Activated manganese oxide was prepared according to a procedure reported by Attenburrow.<sup>49</sup> All other chemicals were purchased commercially and used as received, unless otherwise noted.

### ***Stereochemical Proofs***

The relative or absolute configurations of all starting materials and cross-coupling reaction products were assigned by analogy based on the assumption that the reactions proceed with inversion of stereochemistry at the benzylic position. These stereochemical assignments are summarized in Tables 3.2, 3.3, and 3.4. These tables also summarize how the configuration of each compound was assigned.

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<sup>48</sup> Krasovskiy, A.; Knochel, P. *Synthesis* **2006**, 5, 890.

<sup>49</sup> Attenburrow, J.; Cameron, A. F. B.; Chapman, J. H.; Evans, R. M.; Hems, B. A.; Jansen, A. B. A.; Walker, T. *J. Chem. Soc.* **1952**, 1094.

**Table 3.2.** Configuration of starting materials and products for Table 3.1.

Tetrahydropyran	Configuration <sup>a</sup> assigned by:	Product	Configuration <sup>b</sup> assigned by:
Table 3.1:			
	3.10 <i>cis</i> NOE		3.11 <i>syn</i> by analogy
	3.12 <i>cis</i> NOE		3.13 <i>syn</i> by analogy
	3.14 <i>cis</i> NOE		3.15 <i>syn</i> by analogy
	3.17 <i>cis</i> by analogy		3.18 <i>syn</i> by analogy
	3.19 <i>cis</i> by analogy		3.20 <i>syn</i> by analogy
	3.21 <i>cis,cis</i> NOE		3.22 <i>syn,syn</i> by analogy

<sup>a</sup>For NOE values for each compound, see the characterization data. For comparison to literature values of derivatives, see the characterization data.

<sup>b</sup>In the absence of known relative or absolute configurations, the product configurations were assigned based on the assumption that the cross-coupling reaction proceeds with inversion.

**Table 3.3.** Configuration of starting materials and products for Scheme 3.4.

Tetrahydrofuran/ Tetrahydropyran	Configuration <sup>a</sup> assigned by:	Product	Configuration <sup>b</sup> assigned by:
Scheme 3.4:			
	<b>3.25</b> <i>trans</i> by analogy		<b>3.26</b> <i>anti</i> by analogy
	<b>3.27</b> <i>cis</i> NOE lit <sup>1</sup> H NMR X-ray		<b>3.28</b> <i>syn</i> by analogy

<sup>a</sup>For NOE values for each compound, see the characterization data. <sup>b</sup>For configuration assignment for products derived from tetrahydrofurans, see the Breit model.<sup>50</sup> In the absence of known relative or absolute configurations, the product configurations were assigned based on the assumption that the cross-coupling reaction proceeds with inversion.

**Table 3.4.** Configuration of starting materials and products for Scheme 3.5.

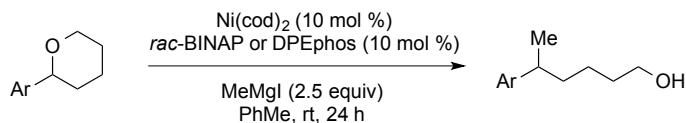
Alcohol	Corresponding Lactone:	Configuration <sup>a</sup> assigned by:	Product	Configuration <sup>b</sup> assigned by:
Scheme 3.5:				
		<i>R</i> (+) CBS model		<i>S</i> (+) by analogy

<sup>a</sup>For the optical rotation of each compound, see the characterization data. For the CBS model, see reference 34. For retention of stereochemistry of diol cyclizations to the corresponding lactones, see reference 35b. <sup>b</sup>In the absence of known optical rotations, the product configurations were assigned based on the assumption that the cross-coupling reaction proceeds with inversion.

<sup>50</sup> Schmidt, Y.; Lehr, K.; Colas, L.; Breit, B. *Chem. Eur. J.* **2012**, *18*, 7071.

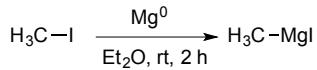
## **General Cross-Coupling Procedures**

### *Method A: Cross-Coupling with Methyl Grignard*



In a glovebox, a flame-dried 7 mL vial equipped with a stir bar was charged with substrate (1.0 equiv), Ni(cod)<sub>2</sub> (0.10 equiv), *rac*-BINAP or DPEphos (0.10 equiv), and PhMe. MeMgI (2.5 equiv) was then added dropwise. After 24 h the reaction was removed from the glovebox, quenched with isopropyl alcohol, filtered through a plug of silica gel (neat Et<sub>2</sub>O), and concentrated in vacuo. Phenyltrimethylsilane (PhTMS) was added as internal standard and a <sup>1</sup>H NMR yield was obtained before purification by flash column chromatography.

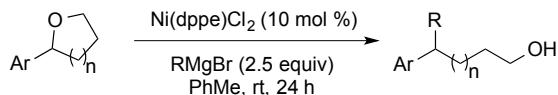
### *Preparation of Methyl Grignard Reagent*



Under a N<sub>2</sub> atmosphere, a 3-necked flask equipped with a stir bar, reflux condenser, and Schlenk filtration apparatus was charged with magnesium turnings (1.1 g, 45 mmol, 1.5 equiv). The flask and magnesium turnings were then flame-dried under vacuum and the flask was back-filled with N<sub>2</sub>. Anhydrous Et<sub>2</sub>O (7 mL) and a crystal of iodine (ca. 2 mg) were added to the flask. Freshly distilled iodomethane (1.9 mL, 31 mmol, 1.0 equiv) was slowly added over 30 min to maintain a gentle reflux. The mixture was stirred for 2 h at room temperature then filtered through the fritted Schlenk filter into the Schlenk bomb under N<sub>2</sub> atmosphere. The magnesium turnings were washed with Et<sub>2</sub>O (2 x 1.0 mL) then the Schlenk bomb was sealed, removed, and placed under an argon atmosphere. The resulting methyl Grignard reagent was typically between 2.4 and 3.0

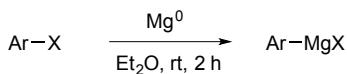
M as titrated by Knochel's method<sup>48</sup> and could be stored (sealed under argon atmosphere or in a glovebox) for up to 4 weeks.

*Method B: Cross-Coupling with Aryl Grignard*



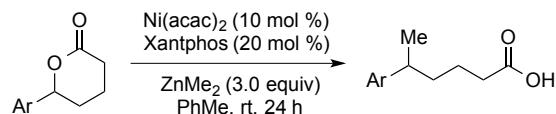
On the benchtop, a flame-dried 7 mL vial was charged with substrate (1.0 equiv) and Ni(dppe)Cl<sub>2</sub> (0.10 equiv), flushed with N<sub>2</sub>, and capped with a Teflon-lined septum. PhMe was added, followed by the alkyl or aryl Grignard reagent (2.5 equiv). The reaction was allowed to stir at room temperature for 24 h, then it was quenched with isopropyl alcohol, filtered through a plug of silica gel (neat Et<sub>2</sub>O), and concentrated in vacuo. Phenyltrimethylsilane (PhTMS) was added as internal standard and a <sup>1</sup>H NMR yield was obtained before purification by flash column chromatography.

*Preparation of Aryl Grignard Reagent*



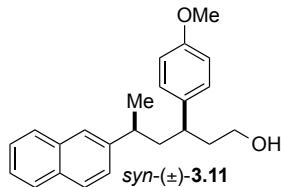
A 2-necked flask equipped with a stir bar and reflux condenser was charged with magnesium turnings (3.0 equiv). The reaction apparatus was flame-dried under vacuum and cooled under N<sub>2</sub>. Anhydrous Et<sub>2</sub>O and a crystal of I<sub>2</sub> (ca. 2 mg) were added to the flask. The organohalide (1.0 equiv) was added slowly over 30 min to maintain a gentle reflux. The mixture was stirred for 2 h at room temperature. The resulting Grignard reagent was typically between 1.5 and 2.5 M as titrated by Knochel's method.<sup>48</sup>

*Method C: Cross-Coupling with Dimethylzinc*

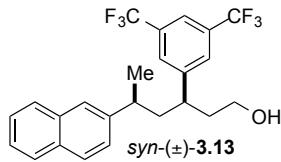


In a glovebox, a flame-dried 7 mL vial equipped with a stir bar was charged with substrate (1.0 equiv). To a separate flame-dried 7 mL vial equipped with a stir bar was added Ni(acac)<sub>2</sub> (0.10 equiv), Xantphos (0.20 equiv), and PhMe in the glovebox. After a 5 min prestir, half of the catalyst mixture was transferred to the vial containing substrate, which was then capped with a screw-cap fitted with a septum and removed from the glovebox. The reaction vial was equipped with a N<sub>2</sub> line and ZnMe<sub>2</sub> (3.0 equiv) was added, which resulted in an immediate color change from green to dark orange. The reaction was allowed to stir at room temperature for 2 h under N<sub>2</sub> before being sealed with Teflon tape and taken back into the glovebox. In the glovebox, the remaining catalyst mixture was added to the reaction vial. The reaction was allowed to stir at room temperature for an additional 22 h, then it was removed from the glovebox and quenched with 1M HCl (2 mL). The aqueous layer was extracted with E<sub>2</sub>O (3 x 2 mL), the combined organic layers were washed with brine (5 mL), dried over MgSO<sub>4</sub>, filtered, and concentrated in vacuo. Phenyltrimethylsilane (PhTMS) was added as internal standard and a <sup>1</sup>H NMR yield was obtained before purification by flash column chromatography.

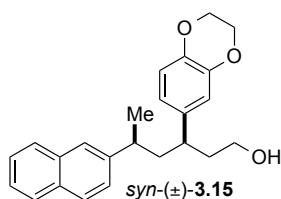
### Characterization Data for Products



***syn*-(±)-3-(4-Methoxyphenyl)-5-(naphthalen-2-yl)hexan-1-ol (*syn*-3.11)** was prepared according to Method A. The following amounts of reagents were used: Ni(cod)<sub>2</sub> (8.6 mg, 0.030 mmol, 0.15 equiv), *rac*-BINAP (19 mg, 0.030 mmol, 0.15 equiv), substrate *cis*-(±)-3.10 (63.7 mg, 0.200 mmol, 1.00 equiv, dr >20:1), PhMe (2.0 mL), and MeMgI (0.19 mL, 0.50 mmol, 2.7 M in Et<sub>2</sub>O, 2.5 equiv). The compound was purified by flash column chromatography using silver-impregnated silica gel (20% EtOAc/hexanes) to afford a colorless oil (53.8 mg) containing the title compound (76% calculated yield, dr >20:1). The dr was determined based on integration of the benzylic methines in the <sup>1</sup>H NMR spectrum. A small amount of analytically pure material was obtained for characterization. **TLC R<sub>f</sub>** = 0.3 (20% EtOAc/hexanes); **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>) δ 7.82–7.76 (m, 3H), 7.47–7.41 (m, 3H), 7.29 (dd, *J* = 8.5, 1.6, 1H), 7.24 (t, *J* = 7.7, 1H), 6.78 (dd, *J* = 8.1, 2.6, 1H), 6.68 (d, *J* = 7.7, 1H), 6.61, (m, 1H), 3.79 (s, 3H), 3.41–3.30 (m, 2H), 2.66–2.59 (m, 1H), 2.41 (sept, *J* = 5.0, 1H), 2.08–2.03 (m, 1H), 1.99–1.93 (m, 1H), 1.86–1.72 (m, 2H), 1.23 (d, *J* = 7.2, 3H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 159.8, 146.5, 144.4, 133.7, 132.4, 129.6, 128.2, 127.8, 127.6, 126.0, 125.8, 125.3, 120.4, 113.8, 111.6, 61.2, 55.3, 45.2, 40.6, 40.4, 37.7, 23.9; **IR** (neat) 3361 (br), 3051, 2925, 1599, 1486, 1257 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>23</sub>H<sub>26</sub>O<sub>2</sub>Na (M + Na)<sup>+</sup> 357.1830, found 357.1831.

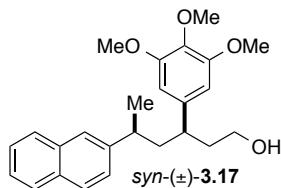


***syn*-( $\pm$ )-3-(3,5-Bis(trifluoromethyl)phenyl)-5-(naphthalen-2-yl)hexan-1-ol (syn-3.13)** was prepared according to Method A. The following amounts of reagents were used: Ni(cod)<sub>2</sub> (8.3 mg, 0.030 mmol, 0.15 equiv), *rac*-BINAP (19 mg, 0.030 mmol, 0.15 equiv), substrate *cis*-( $\pm$ )-3.12 (85 mg, 0.20 mmol, 1.0 equiv, dr >20:1), PhMe (2.0 mL), and MeMgI (0.21 mL, 0.50 mmol, 2.4 M in Et<sub>2</sub>O, 2.5 equiv). The compound was purified by flash column chromatography using silver-impregnated silica gel (5–10% EtOAc/hexanes) to afford the title compound as a colorless oil (63 mg, 0.14 mmol, 72%, dr >20:1). The dr was determined based on integration of the benzylic methines in the <sup>1</sup>H NMR spectrum. **TLC R<sub>f</sub>** = 0.3 (10% EtOAc/hexanes); **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>) δ 7.83 (d, *J* = 8.3, 2H), 7.76–7.74 (m, 2H), 7.50–7.44 (m, 4H), 7.35 (s, 1H), 7.24 (dd, *J* = 8.5, 1.5, 1H), 3.45–3.40 (m, 1H), 3.33–3.28 (m, 1H), 2.65–2.59 (m, 1H), 2.50–2.43 (m, 1H), 2.18 (ddd, *J* = 14.2, 11.4, 4.2, 1H), 1.99–1.89 (m, 2H), 1.82–1.75 (m, 1H), 1.25 (d, *J* = 6.9, 3H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 147.5, 143.1, 133.7, 132.5, 131.7 (q, *J* = 32.7), 128.7, 128.3 (q, *J* = 2.8), 127.8, 127.7, 126.3, 126.1, 125.6, 125.1, 123.6 (q, *J* = 272.8), 120.6 (septet, *J* = 3.9), 60.4, 44.6, 40.1, 39.7, 37.9, 23.8; **IR** (neat) 3330 (br), 2927, 1277, 1171 cm<sup>-1</sup>; **HRMS** (TOF MS ES-) *m/z* calcd for C<sub>24</sub>H<sub>21</sub>F<sub>6</sub>O (M – H)<sup>–</sup> 439.1497, found 439.1489.



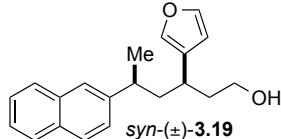
***syn*-( $\pm$ )-3-(2,3-Dihydrobenzo[b][1,4]dioxin-6-yl)-5-(naphthalen-2-yl)hexan-1-ol (syn-3.15)** was prepared according to Method A. The following amounts of reagents were used: Ni(cod)<sub>2</sub>

(8.3 mg, 0.030 mmol, 0.15 equiv), *rac*-BINAP (19 mg, 0.030 mmol, 0.15 equiv), substrate *cis*-( $\pm$ )-**3.14** (69.3 mg, 0.200 mmol, 1.00 equiv, dr >20:1), PhMe (2.0 mL), and MeMgI (0.21 mL, 0.50 mmol, 2.4 M in Et<sub>2</sub>O, 2.5 equiv). The compound was purified by flash column chromatography (20% EtOAc/hexanes) to afford a white foam (60.0 mg) containing a mixture of the title compound (81% calculated yield, dr >20:1) and  $\beta$ -H elimination (1% calculated yield). The dr was determined based on integration of the benzylic methines in the <sup>1</sup>H NMR spectrum. A small amount of analytically pure material was obtained for characterization. **TLC R<sub>f</sub>** = 0.2 (20% EtOAc/hexanes); **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.81–7.76 (m, 3H), 7.48 (s, 1H), 7.46–7.40 (m, 2H), 7.28 (dd, *J* = 8.5, 1.3, 1H), 6.79 (d, *J* = 8.3, 1H), 6.62 (d, *J* = 2.0, 1H), 6.50 (dd, *J* = 8.3, 2.0, 1H), 4.24 (s, 4H), 3.37–3.26 (m, 2H), 2.67–2.60 (m, 1H), 2.33–2.27 (m, 1H), 2.00 (ddd, *J* = 14.5, 10.9, 4.2, 1H), 1.89 (ddd, *J* = 14.8, 11.1, 4.3, 1H), 1.79–1.72 (m, 1H), 1.70–1.63 (m, 1H), 1.22 (d, *J* = 7.0, 3H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>)  $\delta$  144.4, 143.6, 142.0, 138.0, 133.7, 132.3, 128.1, 127.69, 127.65, 125.92, 125.88, 125.8, 125.2, 121.2, 117.1, 116.1, 64.5, 64.4, 61.2, 45.4, 40.4, 39.7, 37.6, 23.8; **IR** (neat) 3427 (br), 2926, 1453, 906, 729 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>24</sub>H<sub>26</sub>ONa (M + Na)<sup>+</sup> 341.1881, found 341.1886.



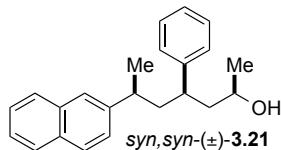
***syn*-( $\pm$ )-4-(1-Hydroxy-5-(naphthalen-2-yl)hexan-3-yl)-2,6-dimethoxyphenol (*syn*-**3.17**) was prepared according to Method A. The following amounts of reagents were used: Ni(cod)<sub>2</sub> (8.3 mg, 0.030 mmol, 0.15 equiv), *rac*-BINAP (18.7 mg, 0.030 mmol, 0.15 equiv), substrate *cis*-( $\pm$ )-**3.16** (75.7 mg, 0.20 mmol, 1.0 equiv, dr >20:1), PhMe (2.0 mL), and MeMgI (0.21 mL, 0.50**

mmol, 2.4 M in Et<sub>2</sub>O, 2.5 equiv). The compound was purified by flash column chromatography using silver-impregnated silica gel (15–40% EtOAc in hexanes) to afford a colorless oil (43.7 mg) containing a mixture of the title compound (42% calculated yield, dr >20:1) and β-H elimination (14% calculated yield). A small amount of analytically pure material was obtained for characterization. **TLC**  $R_f$  = 0.2 (40% EtOAc/hexanes); **<sup>1</sup>H NMR** (400 MHz, CDCl<sub>3</sub>) δ 7.83–7.79 (m, 2H), 7.75–7.73 (m, 1H), 7.45–7.41 (m, 3H), 7.30 (dd, *J* = 8.4, 1.8, 1H), 6.23 (s, 2H), 3.86 (s, 3H), 3.81 (s, 6H), 3.45–3.33 (m, 2H), 2.68–2.59 (m, 1H), 2.39–2.31 (m, 1H), 2.07 (ddd, *J* = 13.9, 10.8, 4.2, 1H), 1.95–1.71 (m, 3H), 1.24 (d, *J* = 6.9, 3H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 153.3, 144.3, 140.4, 136.3, 133.7, 132.4, 128.2, 127.8, 127.5, 126.14, 126.09, 125.7, 125.3, 104.6, 61.3, 61.0, 56.2, 45.0, 40.9, 40.3, 37.7, 24.0; **IR** (neat) 3504 (br), 2931, 1590, 1456, 1128 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>25</sub>H<sub>30</sub>O<sub>4</sub> (M + Na)<sup>+</sup> 417.2042, found 417.2049.



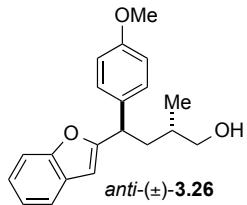
**syn-(±)-3-(Furan-3-yl)-5-(naphthalen-2-yl)hexan-1-ol (*syn*-3.19)** was prepared according to Method A. The following amounts of reagents were used: Ni(cod)<sub>2</sub> (7.9 mg, 0.029 mmol, 0.15 equiv), *rac*-BINAP (18 mg, 0.029 mmol, 0.15 equiv), substrate *cis*-(±)-3.18 (53 mg, 0.19 mmol, 1.0 equiv, dr >20:1), PhMe (1.9 mL), and MeMgI (0.18 mL, 0.48 mmol, 2.7 M in Et<sub>2</sub>O, 2.5 equiv). The compound was purified by flash column chromatography using silver-impregnated silica gel (20% EtOAc/hexanes) to afford the title compound as a white oil (42 mg, 0.14 mmol, 74%, dr >20:1). The dr was determined based on integration of the benzylic methines in the <sup>1</sup>H NMR spectrum. **TLC**  $R_f$  = 0.3 (20% EtOAc/hexanes); **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>) δ 7.81–7.76

(m, 3H), 7.51 (s, 1H), 7.46–7.40 (m, 3H), 7.29 (dd,  $J$  = 8.5, 1.6, 1H), 7.10 (s,  $J$  = 7.2, 2H), 6.27 (s, 1H), 3.45–3.41 (m, 1H), 3.38–3.33 (m, 1H), 2.78–2.71 (m, 1H), 2.38 (septet,  $J$  = 5.0, 1H), 2.00–1.95 (m, 1H), 1.80 (ddd,  $J$  = 13.8, 10.9, 4.3, 1H), 1.76–1.71 (m, 1H), 1.67–1.59 (m, 1H), 1.25 (d,  $J$  = 7.0, 3H);  **$^{13}\text{C}$  NMR** (125.7 MHz,  $\text{CDCl}_3$ )  $\delta$  144.3, 143.4, 139.6, 133.7, 132.4, 128.3, 127.7, 127.63, 127.60, 126.0, 125.8, 125.7, 125.3, 109.0, 61.1, 44.4, 39.5, 37.7, 30.4, 23.8; **IR** (neat) 3422 (br), 2927, 1455, 906, 728  $\text{cm}^{-1}$ ; **HRMS** (TOF MS ES+)  $m/z$  calcd for  $\text{C}_{20}\text{H}_{22}\text{O}_2\text{Na}$  ( $\text{M} + \text{Na}$ )<sup>+</sup> 317.1518, found 317.1517.

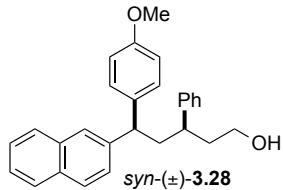


***syn,syn-(±)-6-(Naphthalen-2-yl)-4-phenylheptan-2-ol (syn-3.21)*** was prepared according to Method A. The following amounts of reagents were used:  $\text{Ni}(\text{cod})_2$  (8.3 mg, 0.030 mmol, 0.15 equiv), *rac*-BINAP (19 mg, 0.030 mmol, 0.15 equiv), substrate *cis*-(±)-**3.20** (61 mg, 0.20 mmol, 1.0 equiv, dr >20:1), PhMe (2.0 mL), and  $\text{MeMgI}$  (0.21 mL, 0.50 mmol, 2.4 M in  $\text{Et}_2\text{O}$ , 2.5 equiv). The compound was purified by flash column chromatography using silver-impregnated silica gel (10% EtOAc/hexanes) to afford the title compound as a colorless oil (40. mg, 0.13 mmol, 63%, dr >20:1). The dr was determined based on integration of the benzylic methines in the  **$^1\text{H}$  NMR** spectrum. **TLC**  $\text{R}_f$  = 0.3 (10% EtOAc/hexanes);  **$^1\text{H}$  NMR** (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.82–7.75 (m, 3H), 7.47–7.40 (m, 3H), 7.33–7.21 (m, 4H), 7.08 (dd,  $J$  = 8.4, 1.4, 2H), 3.41–3.33 (m, 1H), 2.65–2.54 (m, 2H), 2.05 (ddd,  $J$  = 13.9, 10.4, 4.6, 1H), 1.95 (ddd,  $J$  = 13.8, 10.6, 4.7, 1H), 1.72–1.59 (m, 2H), 1.22 (d,  $J$  = 7.0, 3H), 0.99 (d,  $J$  = 6.3, 3H);  **$^{13}\text{C}$  NMR** (125.7 MHz,  $\text{CDCl}_3$ )  $\delta$  144.9, 144.4, 133.7, 132.4, 128.6, 128.20, 128.15, 127.74, 127.68, 126.4, 125.94, 125.93, 125.8, 125.2, 65.7, 47.1, 45.5, 40.3, 37.8, 24.2, 23.9; **IR** (neat) 3427 (br), 2926, 1453,

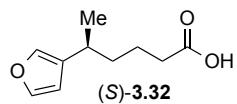
906, 729 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>23</sub>H<sub>26</sub>ONa (M + Na)<sup>+</sup> 341.1881, found 341.1886.



**anti-(±)-4-(Benzofuran-2-yl)-4-(4-methoxyphenyl)-2-methylbutan-1-ol (anti-3.26)** was prepared according to Method B. The following amounts of reagents were used: Ni(dppe)Cl<sub>2</sub> (9.1 mg, 0.017 mmol, 0.10 equiv), *trans*-(±)-**3.25** (35 mg, 0.17 mmol, 1.0 equiv, dr >20:1), phenylmagnesium bromide (0.31 mL, 0.43 mmol, 1.4 M in Et<sub>2</sub>O, 2.5 equiv), PhMe (1.6 mL). The crude product was purified via flash column chromatography (10–20% EtOAc/hexanes) to afford the title compound as a pale yellow oil (43 mg, 0.14 mmol, 81%, dr >20:1). The dr was determined based on integration of the benzylic methines in the <sup>1</sup>H NMR spectrum. **TLC R<sub>f</sub>** = 0.3 (20% EtOAc/hexanes); **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>) δ 7.45 (d, *J* = 7.2 Hz, 1H), 7.38 (d, *J* = 8.0 Hz, 1H), 7.24–7.23 (m, 2H), 7.20–7.13 (m, 2H), 6.85 (d, *J* = 8.7 Hz, 2H), 6.38 (s, 1H), 4.15 (dd, *J* = 10.0, 6.0 Hz, 1H), 3.77 (s, 3H), 3.46 (d, *J* = 6.0 Hz, 2H), 2.18–2.12 (m, 1H), 1.98–1.92 (m, 1H), 1.58–1.49 (m, 1H), 1.40 (br s, 1H), 0.97 (d, *J* = 6.7 Hz, 3H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 162.0, 158.6, 154.8, 133.5, 129.1, 128.8, 123.5, 122.6, 120.6, 114.1, 111.0, 102.1, 68.4, 55.4, 42.3, 37.9, 33.3, 16.3; **IR** (neat) 3367 (br), 2929, 1510, 1247, 750 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>20</sub>H<sub>22</sub>O<sub>3</sub>Na (M + Na)<sup>+</sup> 333.1467, found 333.1473.

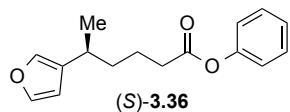


**syn-(±)-5-(4-Methoxyphenyl)-5-(naphthalen-2-yl)-3-phenylpentan-1-ol (syn-3.28)** was prepared according to Method B. The following amounts of reagents were used: Ni(dppe)Cl<sub>2</sub> (11 mg, 0.020 mmol, 0.10 equiv), *cis*-(±)-**3.27** (58 mg, 0.20 mmol, 1.0 equiv, dr >20:1), phenylmagnesium bromide (0.36 mL, 0.50 mmol, 1.4 M in Et<sub>2</sub>O, 2.5 equiv), PhMe (2.0 mL). The crude product was purified via flash column chromatography (10–20% EtOAc/hexanes) to afford the title compound as a white foam (55 mg, 0.14 mmol, 70%, dr >20:1). The dr was determined based on integration of the benzylic methines in the <sup>1</sup>H NMR spectrum. **TLC R<sub>f</sub>** = 0.3 (20% EtOAc/hexanes); **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>) δ 7.77 (d, *J* = 8.4 Hz, 2H), 7.74 (d, *J* = 8.6 Hz, 1H), 7.57 (s, 1H), 7.45–7.39 (m, 2H), 7.30 (t, *J* = 7.5 Hz, 2H), 7.26–7.19 (m, 2H), 7.09–7.05 (m, 4H), 6.75 (d, *J* = 8.7 Hz, 2H), 3.80 (dd, *J* = 10.5, 4.9 Hz, 1H), 3.70 (s, 3H), 3.40–3.36 (m, 1H), 3.34–3.29 (m, 1H), 2.59–2.49 (m, 2H), 2.40–2.31 (m, 1H), 1.97–1.91 (m, 1H), 1.85–1.78 (m, 1H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 157.9, 144.4, 141.8, 137.6, 133.6, 132.3, 128.71, 128.65, 128.4, 128.0, 127.8, 127.7, 126.8, 126.6, 126.1, 125.5, 113.9, 61.1, 55.3, 47.6, 42.8, 40.2, 40.1; **IR** (neat) 3388 (br), 2930, 1509, 907, 728 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>28</sub>H<sub>28</sub>O<sub>2</sub>Na (M + Na)<sup>+</sup> 419.1987, found 419.1970.



**(S)-5-(Furan-3-yl)hexanoic acid ((S)-3.32)** was prepared according to Method C. The following amounts of reagent were used: substrate (*R*)-**3.31** (58.5 mg, 0.350 mmol, 1.00 equiv, 90% ee),

$\text{Ni}(\text{acac})_2$  (9.0 mg, 0.035 mmol, 0.10 equiv), Xantphos (41 mg, 0.070 mmol, 0.20 equiv),  $\text{ZnMe}_2$  (0.59 mL, 1.8 M in PhMe, 1.1 mmol, 3.0 equiv), and PhMe (5.6 mL). The product was purified by flash column chromatography (10–50%  $\text{Et}_2\text{O}$ /pentane) to afford the title compound as a clear, colorless oil (49.5 mg, 0.272 mmol, 78%). The enantiomers of **3.32** could not be separated by SFC. The ee of (*S*)-**3.32** was determined by derivatization to ester (*S*)-**3.36**, which indicated 90% ee and >99% es. **TLC**  $\mathbf{R}_f = 0.2$  (20%  $\text{Et}_2\text{O}$ /pentanes);  **$^1\text{H NMR}$**  (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.35 (t,  $J = 1.5$ , 1H), 7.21 (d,  $J = 0.8$ , 1H), 6.27 (d,  $J = 0.8$ , 1H), 2.65 (sext,  $J = 6.8$ , 1H), 2.33 (t,  $J = 7.2$ , 2H), 1.64–1.51 (m, 4H), 1.19 (d,  $J = 6.9$ , 3H);  **$^{13}\text{C NMR}$**  (125.7 MHz,  $\text{CDCl}_3$ )  $\delta$  180.3, 143.0, 138.2, 130.5, 109.5, 36.9, 34.2, 30.0, 22.6, 21.4; **IR** (neat) 2959, 2930, 2872, 1708, 909  $\text{cm}^{-1}$ ; **HRMS** (TOF MS ES $-$ )  $m/z$  calcd for  $\text{C}_{10}\text{H}_{12}\text{O}_3$  ( $\text{M} - \text{H}^-$ ) 181.0865, found 181.0868;  $[\alpha]_D^{26} +9.9$  ( $c$  1.5,  $\text{CDCl}_3$ ).



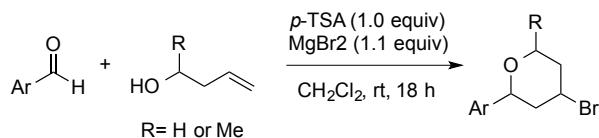
**Phenyl (*S*)-5-(furan-3-yl)hexanoate ((*S*)-**3.36**)** was prepared according to a modified procedure reported by Meyer.<sup>51</sup> To a stirring solution of substrate (*S*)-**3.32** (28 mg, 0.15 mmol, 1.0 equiv) in anhydrous DCM (2.0 mL) was added phenol (14 mg, 0.15 mmol, 1.0 equiv), N,N'-dicyclohexylcarbodiimide (DCC, 31 mg, 0.15 mmol, 1.0 equiv), and 4-dimethylaminopyridine (DMAP, 9.3 mg, 0.076 mmol, 0.50 equiv). The reaction mixture was allowed to stir at room temperature for 20 h. The mixture was passed through a plug of Celite, rinsed with DCM, and concentrated in vacuo. The product was purified by flash column chromatography (1%  $\text{Et}_2\text{O}$ /hexanes) to afford the title compound as a clear, colorless oil (26 mg, 0.10 mmol, 67%, 90% ee). **TLC**  $\mathbf{R}_f = 0.5$  (5%  $\text{EtOAc}$ /hexanes);  **$^1\text{H NMR}$**  (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.38–7.33 (m, 3H),

<sup>51</sup> Stayshich, R. M.; Meyer, T. Y. *J. Polym. Sci., Part A: Polym. Chem.* **2008**, *46*, 4704.

7.25–7.19 (m, 2H), 7.07–7.03 (m, 2H), 6.29 (d,  $J$  = 0.9, 1H), 2.69 (sext,  $J$  = 6.9, 1H), 2.54 (t,  $J$  = 7.3, 2H), 1.75–1.69 (m, 2H), 1.65–1.57 (m, 2H), 1.21 (d,  $J$  = 6.9, 3H);  **$^{13}\text{C}$  NMR** (125.7 MHz,  $\text{CDCl}_3$ )  $\delta$  172.2, 150.8, 143.0, 138.2, 130.5, 129.5, 125.9, 121.7, 109.5, 37.0, 34.5, 30.0, 22.8, 21.5; **IR** (neat) 2959, 2927, 1758, 1496, 1135, 873  $\text{cm}^{-1}$ ; **HRMS** (TOF MS ES+)  $m/z$  calcd for  $\text{C}_{16}\text{H}_{18}\text{O}_3\text{Na}$  ( $\text{M} + \text{Na}$ ) $^+$  281.1154, found 281.1152;  $[\alpha]_D^{26} + 8.2$  ( $c$  0.8,  $\text{CDCl}_3$ ); **SFC** analysis (AD-H, 10% MeOH, 3.0 mL/min, 215 nm) indicated 90% ee:  $t_R$  (major) = 2.0 minutes,  $t_R$  (minor) = 2.2 minutes.

### ***General Procedures for Starting Materials Synthesis***

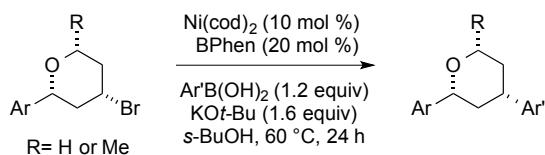
#### ***Method D: Prins Cyclization with $\text{MgBr}_2$***



Modified from a procedure reported by Grée.<sup>19</sup> Magnesium bromide (1.1 equiv) was added to a flame-dried flask equipped with a stir bar and then flame dried again under vacuum. *p*-Toluene sulfonic acid monohydrate (1.0 equiv) and anhydrous DCM (20 mL) were added and the reaction mixture was set to stir at ambient temperature. To a separate flame dried flask was added aldehyde (1.0 equiv). Anhydrous DCM (20 mL) and homoallylic alcohol (1.1 equiv) were added and the mixture was stirred for 5 min at room temperature. The aldehyde solution was added to the magnesium bromide solution and the reaction mixture was allowed to stir at room temperature for 18 h. The reaction was quenched with saturated aqueous  $\text{NaHCO}_3$  (20 mL) and was extracted with DCM (3 x 20 mL). The combined organic layers were washed with brine, dried over  $\text{MgSO}_4$ , filtered, and concentrated in vacuo.

To remove unreacted aldehyde that was difficult to separate from the desired product, the crude mixture was subjected to NaBH<sub>4</sub> reduction by a modified procedure reported by Franzén.<sup>52</sup> The crude mixture was dissolved in 1:1 MeOH/DCM and the reaction cooled to 0 °C. NaBH<sub>4</sub> (1.6 equiv relative to 1.0 equiv of aldehyde as determined by <sup>1</sup>H NMR integration) was added in one portion and the reaction stirred 30 min at 0 °C, then 30 min at room temperature. The reaction was quenched with water and extracted with DCM (x 3). The combined organic layers were washed with brine, dried over MgSO<sub>4</sub>, filtered, and concentrated in vacuo.

*Method E: Suzuki-Type Cross-Coupling of Alkyl Bromides*



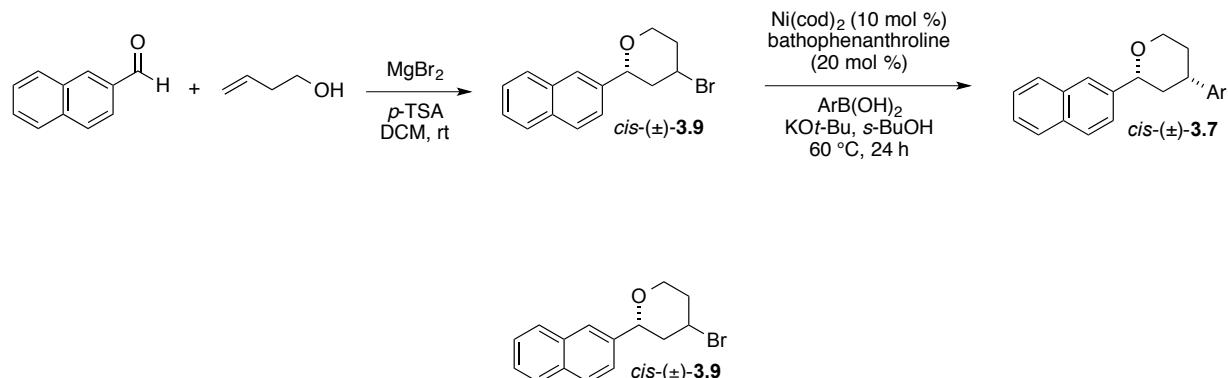
Modified from a procedure reported by Fu.<sup>21</sup> In a glove box, Ni(cod)<sub>2</sub> (0.10 equiv), bathophenanthroline (BPheN, 0.20 equiv), anhydrous KO<sup>t</sup>-Bu (1.6 equiv), arylboronic acid (1.2 equiv), and freshly distilled *s*-butanol (3.0 mL) were added to a flame dried 7 mL vial equipped with a stir bar and stirred at ambient temperature for 10 min. Substrate (1.0 equiv) was added to the reaction vial, which was then capped with a septum and removed from the glove box. The reaction vial was put under N<sub>2</sub> in a 60 °C oil bath. After stirring for 24 h, the reaction was allowed to cool to ambient temperature before it was passed through a silica gel plug (50% Et<sub>2</sub>O/hexanes). The filtrate was concentrated in vacuo.

<sup>52</sup> Wang, Y.; Franzén, R. *Synlett*. **2012**, 23, 925.

## Characterization Data for Starting Materials

### Synthesis of Tetrahydropyrans

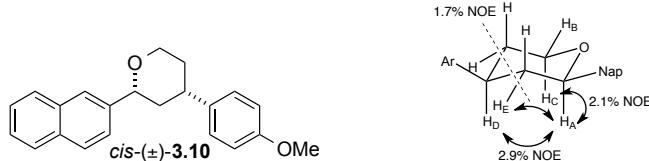
**Scheme 3.7.** Two-step synthesis of 2,4-disubstituted tetrahydropyrans.



**( $\pm$ )-4-Bromo-2-(2-naphthyl)-tetrahydropyran (*cis*-3.9)** was prepared according Method D.

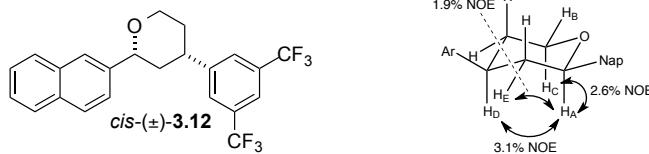
The following amounts of reagents were used: magnesium bromide (1.97 g, 10.7 mmol, 1.1 equiv), *p*-toluene sulfonic acid monohydrate (1.7 g, 9.7 mmol, 1.0 equiv), 2-naphthaldehyde (1.5 g, 9.7 mmol, 1.0 equiv), 3-buten-1-ol (0.92 mL, 10.7 mmol, 1.1 equiv), and anhydrous DCM (100 mL). The product was purified by flash column chromatography (2% EtOAc/hexanes) to afford the title compound as a white solid (1.84 g, 65%). The dr was determined based on integration of the benzylic methines in the <sup>1</sup>H NMR spectrum. <sup>1</sup>H NMR analysis indicated **3.9** was an inseparable 2:1 mixture of *cis* and *trans* diastereomers.<sup>19</sup> **m.p.** 55–57 °C; **TLC R<sub>f</sub>** = 0.6 (2% EtOAc/hexanes); **IR** (neat) 3059, 2956, 2927, 2853, 1601, 1505, 1445 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>15</sub>H<sub>15</sub>BrONH<sub>4</sub> (M + NH<sub>4</sub>)<sup>+</sup> 308.0650, found 308.0640. **cis-( $\pm$ )-3.9** (68% by <sup>1</sup>H NMR integration): **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>) δ 7.84–7.79 (m, 4H), 7.49–7.43 (m, 3H), 4.50 (d, *J* = 11.1, 1H), 4.37–4.30 (m, 1H), 4.24–4.19 (m, 1H), 3.66 (td, *J* = 12.1, 2.2, 1H), 2.56 (ddd, *J* = 13.0, 4.5, 2.2, 1H), 2.31–2.12 (m, 3H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 138.7, 133.4, 133.2, 128.4, 128.2, 127.82, 126.3, 126.1, 124.7, 124.0, 80.4, 68.5, 46.6, 45.7, 37.9. **trans-**

( $\pm$ )-**3.9** (32% by  $^1\text{H}$  NMR integration): **1H NMR** (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.84–7.79 (m, 4H), 7.49–7.43 (m, 3H), 5.08 (dd,  $J$  = 10.6, 1.8 1H), 4.84–4.82 (m, 1H), 4.24–4.19 (m, 1H), 4.08 (dd,  $J$  = 11.8, 5.0, 1H), 2.31–2.12 (m, 3H), 2.01 (app d,  $J$  = 15.1, 1H);  **$^{13}\text{C}$  NMR** (125.7 MHz,  $\text{CDCl}_3$ )  $\delta$  139.4, 133.5, 133.1, 128.3, 128.1, 127.79, 126.2, 126.0, 124.7, 124.19, 74.6, 63.7, 50.3, 42.0, 34.2.

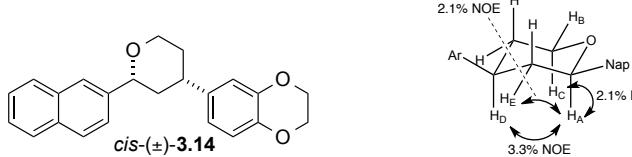


***cis*-( $\pm$ )-4-(4-Methoxyphenyl)-2-(2-naphthyl)-tetrahydropyran (*cis*-**3.10**)** was prepared according to Method E. The following amounts of reagents were used:  $\text{Ni}(\text{cod})_2$  (14 mg, 0.050 mmol, 0.10 equiv), bathophenanthroline (33 mg, 0.10 mmol, 0.20 equiv), anhydrous  $\text{KO}^\text{t-Bu}$  (90. mg, 0.80 mmol, 1.6 equiv), 4-methoxyphenylboronic acid (91 mg, 0.60 mmol, 1.2 equiv), *s*-butanol (3.0 mL) and substrate *cis*-( $\pm$ )-**3.9** (150 mg, 0.50 mmol, 1.0 equiv). The compound was purified by flash column chromatography (5%  $\text{Et}_2\text{O}$ /hexanes) to afford the title compound as a white solid (110 mg, 0.34 mmol, 68%, dr >20:1). The dr was determined based on integration of the benzylic methines in the  $^1\text{H}$  NMR spectrum. The relative configuration was assigned as *cis* by COSY and NOE NMR experiments. Irradiation of the benzylic proton (H<sub>A</sub>) gave an NOE enhancement of 2.1% of H<sub>C</sub>, an enhancement of 2.9% of H<sub>D</sub>, and an enhancement of 1.7% of H<sub>E</sub>. **m.p.** 69–70 °C; **TLC**  $R_f$  = 0.3 (5%  $\text{Et}_2\text{O}$ /hexanes);  **$^1\text{H}$  NMR** (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.84 (s, 1H), 7.83–7.79 (m, 3H), 7.49 (dd,  $J$  = 8.4, 1.6, 1H), 7.46–7.41 (m, 2H), 7.17 (d,  $J$  = 8.7, 2H), 6.85 (d,  $J$  = 8.8, 2H), 4.62 (dd,  $J$  = 11.2, 1.6, 1H), 4.33 (ddd,  $J$  = 11.5, 4.2, 1.6, 1H), 3.80 (td,  $J$  = 11.5, 2.9, 1H), 3.76 (s, 3H), 2.95 (tt,  $J$  = 11.9, 4.0, 1H), 2.14 (adt,  $J$  = 13.2, 1.6, 1H), 1.93–1.76 (m, 3H);  **$^{13}\text{C}$  NMR** (125.7 MHz,  $\text{CDCl}_3$ )  $\delta$  158.2, 140.4, 137.8, 133.5, 133.0, 128.14, 128.12, 127.75, 127.74, 126.1, 125.8, 124.4, 124.3, 114.0, 80.1, 68.9, 55.4, 41.8, 41.4, 33.8; **IR** (neat) 2937,

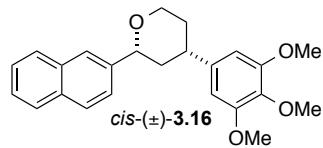
2836, 1600, 1258, 1083  $\text{cm}^{-1}$ ; **HRMS** (TOF MS ES+)  $m/z$  calcd for  $\text{C}_{22}\text{H}_{22}\text{O}_2\text{Na} (\text{M} + \text{Na})^+$  341.1518, found 341.1512.



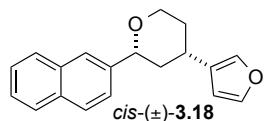
***cis*-( $\pm$ )-4-(3,5-Bis(trifluoromethyl)phenyl)-2-(2-naphthyl)-tetrahydropyran (cis-3.12)** was prepared according to Method E. The following amounts of reagents were used:  $\text{Ni}(\text{cod})_2$  (14 mg, 0.050 mmol, 0.10 equiv), bathophenanthroline (33 mg, 0.10 mmol, 0.20 equiv), anhydrous  $\text{KO}^\ddagger\text{-Bu}$  (90. mg, 0.80 mmol, 1.6 equiv), 3,5-bis(trifluoromethyl)phenylboronic acid (155 mg, 0.600 mmol, 1.2 equiv), *s*-butanol (3.0 mL) and substrate *cis*-( $\pm$ )-3.9 (150 mg, 0.50 mmol, 1.0 equiv). The compound was purified by flash column chromatography (2%  $\text{Et}_2\text{O}$ /hexanes) to afford the title compound as a white solid (130 mg, 0.31 mmol, 62%, dr >20:1). The dr was determined based on integration of the benzylic methines in the  $^1\text{H}$  NMR spectrum. The relative configuration was assigned as *cis* by COSY and NOE NMR experiments. Irradiation of the benzylic proton ( $\text{H}_\text{A}$ ) gave an NOE enhancement of 2.6% of  $\text{H}_\text{C}$ , an enhancement of 3.1% of  $\text{H}_\text{D}$ , and an enhancement of 1.9% of  $\text{H}_\text{E}$ . **m.p.** 73–74 °C; **TLC  $\text{R}_f$**  = 0.3 (5%  $\text{Et}_2\text{O}$ /hexanes);  **$^1\text{H NMR}$**  (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.86–7.81 (m, 4H), 7.75 (s, 1H), 7.71 (s, 2H), 7.50 (dd,  $J$  = 8.5, 1.5, 1H), 7.48–7.45 (m, 2H), 4.67 (dd,  $J$  = 10.9, 1.8, 1H), 4.38 (ddd,  $J$  = 11.5, 4.2, 1.8, 1H), 3.84 (td,  $J$  = 11.4, 3.3, 1H), 3.20–3.14 (m, 1H), 2.20 (adt,  $J$  = 13.2, 1.9, 1H), 2.00–1.91 (m, 2H), 1.85 (q,  $J$  = 12.3, 1H);  **$^{13}\text{C NMR}$**  (125.7 MHz,  $\text{CDCl}_3$ )  $\delta$  147.8, 139.7, 133.5, 133.1, 132.0 (q,  $J$  = 33.4), 128.4, 128.1, 127.8, 127.2 (q,  $J$  = 2.8), 126.3, 126.0, 124.5, 124.1, 123.5 (q,  $J$  = 272.8), 120.7, (septet,  $J$  = 3.9), 79.8, 68.4, 42.1, 41.2, 33.1; **IR** (neat) 2920, 2848, 1277, 1171  $\text{cm}^{-1}$ ; **HRMS** (TOF MS CI+)  $m/z$  calcd for  $\text{C}_{23}\text{H}_{18}\text{F}_6\text{ONH}_4 (\text{M} + \text{NH}_4)^+$  442.1606, found 442.1609.



**cis-( $\pm$ )-6-(2-(Naphthyl)-tetrahydropyranyl)-2,3-dihydrobenzo[*b*][1,4]dioxine (*cis*-3.14)** was prepared according to Method E. The following amounts of reagents were used: Ni(cod)<sub>2</sub> (14 mg, 0.050 mmol, 0.10 equiv), bathophenanthroline (33 mg, 0.10 mmol, 0.20 equiv), anhydrous KO<sup>t</sup>-Bu (90. mg, 0.80 mmol, 1.6 equiv), 1,4-benzodioxane-6-boronic acid (110 mg, 0.60 mmol, 1.2 equiv), *s*-butanol (3.0 mL) and substrate *cis*-( $\pm$ )-3.9 (150 mg, 0.50 mmol, 1.0 equiv). The compound was purified by flash column chromatography (5% Et<sub>2</sub>O/hexanes) to afford the title compound as a faintly pink oil (95 mg, 0.27 mmol, 55%, dr >20:1). The dr was determined based on integration of the benzylic methines in the <sup>1</sup>H NMR spectrum. The relative configuration was assigned as *cis* by COSY and NOE NMR experiments. Irradiation of the benzylic proton (H<sub>A</sub>) gave an NOE enhancement of 2.1% of H<sub>C</sub>, an enhancement of 3.3% of H<sub>D</sub>, and an enhancement of 2.1% of H<sub>E</sub>. **TLC** R<sub>f</sub> = 0.2 (5% Et<sub>2</sub>O/hexanes); **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.84–7.80 (m, 4H), 7.49 (dd, *J* = 8.6, 1.3, 1H), 7.47–7.42 (m, 2H), 6.82–6.77 (m, 2H), 6.73 (dd, *J* = 8.3, 1.9, 1H), 4.62 (dd, *J* = 11.1, 1.1, 1H), 4.32 (ddd, *J* = 11.5, 3.7, 2.0, 1H), 4.22 (s, 4H), 3.80 (td, *J* = 11.1, 3.6, 1H), 2.94–2.88 (m, 1H), 2.14 (add, *J* = 13.2, 2.0, 1H), 1.91–1.84 (m, 2H), 1.78 (q, *J* = 12.2, 1H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>)  $\delta$  143.6, 142.1, 140.3, 139.1, 133.5, 133.0, 128.1, 127.8, 126.1, 125.8, 124.5, 124.4, 119.8, 117.3, 115.6, 80.1, 68.9, 64.54, 64.45, 41.7, 41.5, 33.7; **IR** (neat) 2933, 1508, 1284, 1068 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>23</sub>H<sub>22</sub>O<sub>3</sub>Na (M + Na)<sup>+</sup> 369.1467, found 369.1469.



***cis*-(±)-2-(Naphthalen-2-yl)-4-(3,4,5-trimethoxyphenyl)tetrahydro-2*H*-pyran (*cis*-3.16)** was prepared according to Method E. The following amounts of reagents were used: Ni(cod)<sub>2</sub> (20.6 mg, 0.075 mmol, 0.15 equiv), bathophenanthroline (50.0 mg, 0.15 mmol, 0.30 equiv), anhydrous KO<sup>t</sup>-Bu (90 mg, 0.80 mmol, 1.6 equiv), 3,4,5-trimethoxyphenylboronic acid (127 mg, 0.60 mmol, 1.2 equiv), *s*-butanol (3.0 mL) and substrate *cis*-(±)-3.9 (146 mg, 0.5 mmol, 1.0 equiv). The compound was purified by flash column chromatography (20% EtOAc in hexanes) to afford the title compound as a colorless oil (133 mg, 0.35 mmol, 70%, dr >20:1). The dr was determined based on integration of the benzylic methines in the <sup>1</sup>H NMR spectrum. The relative configuration was assigned as *cis* by analogy to *cis*-3.10. **TLC R<sub>f</sub>** = 0.4 (20% EtOAc/hexanes); **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>) δ 7.86 (s, 1H), 7.83–7.80 (m, 3H), 7.50 (dd, *J* = 8.6, 1.3, 1H), 7.47–7.43 (m, 2H), 6.48 (s, 2H), 4.64 (dd, *J* = 11.1, 1.5, 1H), 4.36 (ddd, *J* = 11.3, 4.0, 1.6 1H), 3.85 (s, 6H), 3.83–3.79 (m, 4H), 3.00–2.94 (m, 1H), 2.18 (add, *J* = 13.4, 1.7, 1H), 1.96–1.87 (m, 2H), 1.81 (q, *J* = 12.5, 1H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 153.4, 141.4, 140.3, 136.5, 133.4, 133.0, 128.15, 128.07, 127.7, 126.1, 125.8, 124.3, 124.2, 103.7, 80.0, 68.8, 60.9, 56.2, 42.7, 41.8, 33.7; **IR** (neat) 2936, 2838, 1588, 1124 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>24</sub>H<sub>26</sub>O<sub>4</sub> (M + Na)<sup>+</sup> 401.1729, found 401.1717.

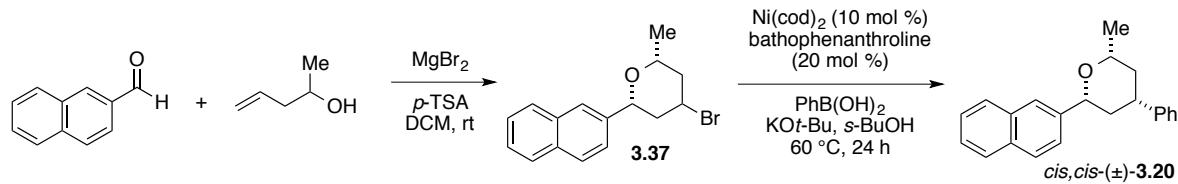


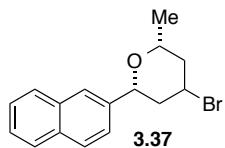
***cis*-(±)-4-(Furan-3-yl)-2-(naphthyl)-tetrahydropyran (*cis*-3.18)** was prepared according to Method E. The following amounts of reagents were used: Ni(cod)<sub>2</sub> (21 mg, 0.075 mmol, 0.15

equiv), bathophenanthroline (50. mg, 0.15 mmol, 0.30 equiv), anhydrous KO<sup>t</sup>-Bu (90. mg, 0.80 mmol, 1.6 equiv), 3-furanylboronic acid (67 mg, 0.60 mmol, 1.2 equiv), *s*-butanol (3.0 mL) and substrate *cis*-( $\pm$ )-**3.9** (150 mg, 0.50 mmol, 1.0 equiv). The compound was purified by flash column chromatography (5% Et<sub>2</sub>O/hexanes) to afford a mixture of desired product and dimer of the starting material; this mixture was further purified by flash column chromatography (40% benzene/hexanes) to afford the title compound as an opaque oil (31 mg, 0.11 mmol, 23%, dr >20:1). The dr was determined based on integration of the benzylic methines in the <sup>1</sup>H NMR spectrum. The relative configuration was assigned as *cis* by analogy to *cis*-**3.10**. TLC R<sub>f</sub> = 0.4 (5% Et<sub>2</sub>O/hexanes), 0.2 (40% benzene/hexanes), R<sub>f</sub> (dimer) = 0.3 (40% benzene/hexanes); <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.84–7.80 (m, 4H), 7.49 (dd, *J* = 8.4, 1.6, 1H), 7.48–7.43 (m, 2H), 7.36 (m, 1H), 7.24 (s, 1H), 6.33 (s, 1H), 4.61 (dd, *J* = 11.3, 1.7, 1H), 4.30 (ddd, *J* = 11.6, 4.5, 1.1, 1H), 3.79 (td, *J* = 12.0, 2.2, 1H), 2.94 (tt, *J* = 12.1, 3.8, 1H), 2.20 (adt, *J* = 13.2, 1.7, 1H), 1.92 (adq, *J* = 13.2, 2.0, 1H), 1.79 (qd, *J* = 12.6, 4.5, 1H), 1.71 (q, *J* = 12.3, 1H); <sup>13</sup>C NMR (125.7 MHz, CDCl<sub>3</sub>)  $\delta$  143.1, 140.3, 137.9, 133.5, 133.0, 129.6, 128.2, 128.1, 127.8, 126.1, 125.9, 124.5, 124.3, 109.4, 79.9, 68.6, 41.1, 33.0, 32.8; IR (neat) 2935, 2846, 1085, 907 cm<sup>-1</sup>; HRMS (TOF MS ES+) *m/z* calcd for C<sub>19</sub>H<sub>18</sub>O<sub>2</sub>Na (M + Na)<sup>+</sup> 301.1205, found 301.1206.

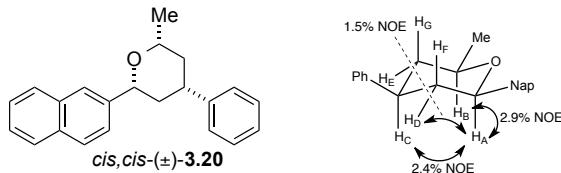
### Synthesis of *cis,cis*-( $\pm$ )-**3.20**

**Scheme 3.8.** Two-step synthesis of *cis,cis*-( $\pm$ )-**3.20**.





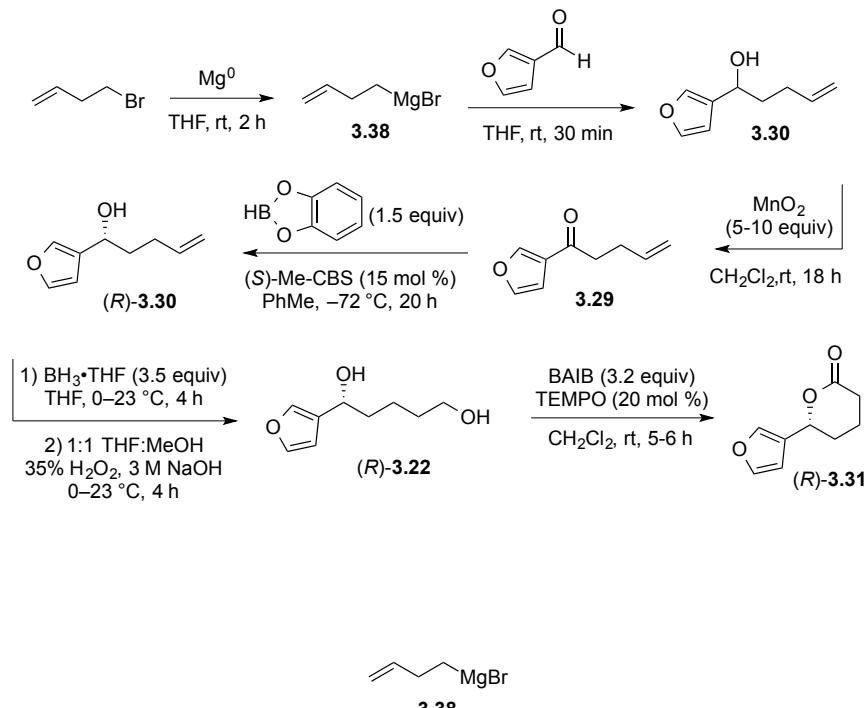
**( $\pm$ )-4-Bromo-6-methyl-2-(2-naphthyl)-tetrahydropyran (3.37)** was prepared according to Method D. The following amounts of reagents were used: magnesium bromide (1.6 g, 8.7 mmol, 1.1 equiv), *p*-toluene sulfonic acid monohydrate (1.4 g, 7.4 mmol, 1.0 equiv), 2-naphthaldehyde (1.2 g, 7.7 mmol, 1.0 equiv), 4-penten-2-ol (0.82 mL, 7.9 mmol, 1.0 equiv), and anhydrous DCM (80 mL). The compound was purified by flash column chromatography (0–1% EtOAc in hexanes) to afford the title compound as a white solid (0.91 g, 3.0 mmol, 39%). The dr was determined based on integration of the benzylic methines in the  $^1\text{H}$  NMR spectrum.  $^1\text{H}$  NMR analysis indicated the title compound was an inseparable mixture of *cis* and *trans* diastereomers by comparison to literature  $^1\text{H}$  NMR of the reported analogues.<sup>19</sup> The *cis:trans* ratios ranged from 1:1.1 to 1:1.7 for different batches. **m.p.** 78–80 °C; **TLC**  $\mathbf{R}_f = 0.6$  (2% EtOAc/hexanes, stains blue with PAA); **IR** (neat) 3060, 2966, 1312, 1069 cm<sup>−1</sup>; **HRMS** (TOF MS CI+) *m/z* calcd for C<sub>16</sub>H<sub>17</sub>BrONH<sub>4</sub> (M + NH<sub>4</sub>)<sup>+</sup> 322.0807, found 322.0815. **cis-( $\pm$ )-3.37** (37% by  $^1\text{H}$  NMR integration):  **$^1\text{H NMR}$**  (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.77–7.70 (m, 4H), 7.41–7.33 (m, 3H), 4.30 (dd, *J* = 11.4, 1.5, 1H), 4.15–4.07 (m, 1H), 3.48–3.40 (m, 1H), 2.37 (dt, *J* = 12.9, 2.2, 1H), 2.15 (dd, *J* = 14.4, 2.2, 1H), 2.02–1.86 (m, 1H), 1.79–1.64 (m, 1H), 1.21 (d, *J* = 6.3, 3H);  **$^{13}\text{C NMR}$**  (125.7 MHz, CDCl<sub>3</sub>)  $\delta$  138.8, 133.4, 133.1, 128.4, 128.13, 127.77, 126.2, 126.0, 124.76, 124.2, 79.7, 74.3, 46.6, 45.0, 41.0, 21.8. **trans-( $\pm$ )-3.37** (63% by  $^1\text{H}$  NMR integration):  **$^1\text{H NMR}$**  (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.77–7.70 (m, 4H), 7.41–7.33 (m, 3H), 5.06 (dd, *J* = 10.9, 1.4, 1H), 6.46 (t, *J* = 2.9, 1H), 4.24–4.16 (m, 1H), 2.15 (dd, *J* = 14.4, 2.2, 1H), 2.02–1.86 (m, 2H), 1.79–1.64 (m, 1H), 1.25 (d, *J* = 6.4, 3H);  **$^{13}\text{C NMR}$**  (125.7 MHz, CDCl<sub>3</sub>)  $\delta$  139.6, 133.5, 133.1, 128.3, 128.10, 127.75, 126.1, 125.9, 124.77, 124.4, 74.5, 69.0, 50.7, 44.8, 41.5, 21.6.



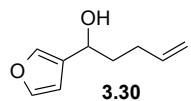
***cis,cis*-( $\pm$ )-6-Methyl-4-phenyl-2-(2-naphthyl)-tetrahydropyran (*cis,cis*-**3.20**)** was prepared according to Method E. The following amounts of reagents were used: Ni(cod)<sub>2</sub> (14 mg, 0.050 mmol, 0.10 equiv), bathophenanthroline (33 mg, 0.10 mmol, 0.20 equiv), anhydrous KO<sup>t</sup>-Bu (90. mg, 0.80 mmol, 1.6 equiv), phenylboronic acid (73 mg, 0.60 mmol, 1.2 equiv), *s*-butanol (3.0 mL) and substrate **3.37** (150 mg, 0.50 mmol, 1.0 equiv). The compound was purified by flash column chromatography (2% Et<sub>2</sub>O/hexanes) to afford the title compound as a white solid (99 mg, 0.33 mmol, 66%, dr >20:1). The dr was determined based on integration of the benzylic methines in the <sup>1</sup>H NMR spectrum. The relative configuration was assigned as *cis* by COSY and NOE NMR experiments. Irradiation of the benzylic proton (H<sub>A</sub>) gave an NOE enhancement of 2.9% of H<sub>B</sub>, an enhancement of 2.4% of H<sub>C</sub>, and an enhancement of 1.5% of H<sub>D</sub>. **m.p.** 67–69 °C; **TLC R<sub>f</sub>** = 0.5 (5% Et<sub>2</sub>O/hexanes); **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>) δ 7.86 (s, 1H), 7.83–7.79 (m, 3H), 7.52 (dd, *J* = 8.6, 1.5, 1H), 7.46–7.41 (m, 2H), 7.32–7.29 (m, 2H), 7.26–7.19 (m, 3H), 4.70 (dd, *J* = 11.1, 1.7, 1H), 3.89–3.83 (m, 1H), 3.03 (tt, *J* = 12.3, 3.7, 1H), 2.15 (dt, *J* = 13.2, 1.8, 1H), 1.93 (dt, *J* = 13.2, 1.8, 1H), 1.77 (q, *J* = 12.3, 1H), 1.56 (q, *J* = 11.8, 1H), 1.37 (d, *J* = 6.4, 3H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 145.6, 140.5, 133.5, 133.0, 128.7, 128.16, 128.15, 127.7, 126.9, 126.5, 126.0, 125.7, 124.59, 124.58, 79.8, 74.4, 42.3, 41.0, 40.8, 22.3; **IR** (neat) 3058, 2930, 1132, 757 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>22</sub>H<sub>22</sub>O (M + Na)<sup>+</sup> 325.1568, found 325.1563.

*Synthesis of Enantioenriched Valerolactone (R)-3.31.*

**Scheme 3.9.** Synthetic scheme for (R)-3.31.

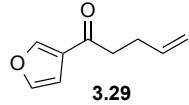


**3-Butenylmagnesium bromide (3.38).** Magnesium turnings (3.65 g, 150 mmol, 2.00 equiv.) were added to a round-bottom flask equipped with a stir bar. The reaction apparatus was flame-dried under vacuum and cooled under N<sub>2</sub>. A single crystal of I<sub>2</sub> (ca. 2 mg) was added to the flask, followed by anhydrous Et<sub>2</sub>O (30 mL). 4-Bromo-1-butene (7.5 mL, 74 mmol, 1.0 equiv.) was added portion-wise over 1 h at 0 °C. The reaction was stirred at 0 °C for 10 min, then at ambient temperature for 2 h, and titrated.<sup>48</sup>



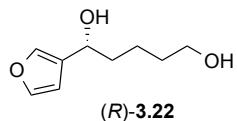
**1-(3-Furyl)-pent-4-en-1-ol (3.30).** A solution of freshly prepared Grignard reagent 3.38 (13 mL, 20 mmol, 1.5 M in Et<sub>2</sub>O, 1.1 equiv) at 0 °C was treated in a dropwise manner with 3-furaldehyde

(1.60 mL, 18.4 mmol, 1.00 equiv) to maintain a gentle reflux. The reaction mixture was stirred at room temperature 1 h, and then cooled to 0 °C and quenched with saturated aqueous NH<sub>4</sub>Cl (30 mL). The mixture was extracted with EtOAc (3 × 100 mL) and the combined organic extracts were washed with brine, dried over MgSO<sub>4</sub>, and concentrated in vacuo. The product was purified by flash column chromatography (5–10–20% Et<sub>2</sub>O/hexanes) to afford the title compound as a light yellow oil (2.78 g, 18.3 mmol, 99%). **TLC** R<sub>f</sub> = 0.4 (20% EtOAc/hexanes); **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>) δ 7.39–7.38 (m, 2H), 6.40 (s, 1H), 5.88–5.80 (m, 1H), 5.05 (dq, *J* = 17.1, 1.7, 1H), 4.99 (dq, *J* = 1.4, 10.2, 1H), 4.70–4.66 (m, 1H), 2.20–2.09 (m, 2H), 1.91–1.77 (m, 2H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 143.5, 139.2, 138.2, 129.1, 115.2, 108.5, 66.6, 36.9, 30.0; **IR** (neat) 3364 (br), 2936, 1502, 1159, 1022 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>9</sub>H<sub>12</sub>O<sub>2</sub> (M + Na)<sup>+</sup> 151.0759, found 151.0756.

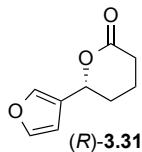


**1-(3-Furyl)-pent-4-en-1-one (3.29).** Activated MnO<sub>2</sub> (15.9 g, 183 mmol, 10.0 equiv) was added to a solution of substrate **3.30** (2.78 g, 18.3 mmol, 1.00 equiv) in wet DCM (200 mL). The reaction was stirred at ambient temperature for 18 h, then filtered through a bed of Celite with additional DCM and concentrated in vacuo. The product was purified by flash column chromatography (1% Et<sub>2</sub>O/hexanes) to afford the title compound as a clear, colorless oil (2.06 g, 13.7 mmol, 75%). **TLC** R<sub>f</sub> = 0.6 (10% Et<sub>2</sub>O/pentanes); **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>) δ 8.03 (as, 1H), 7.44 (at, *J* = 1.7, 1H), 6.77 (add, *J* = 1.9, 0.8, 1H), 5.91–5.82 (m, 1H), 5.09 (dq, *J* = 17.0, 1.6, 1H), 5.00 (dq, *J* = 10.2, 1.4, 1H), 2.85 (t, *J* = 7.5, 2H), 2.47 (qt, *J* = 7.1, 1.3, 2H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 194.3, 147.2, 144.3, 137.2, 127.8, 115.5, 108.7, 39.6, 28.2; **IR** (neat)

3137, 1676, 1562, 1511, 1155, 873 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>9</sub>H<sub>10</sub>O<sub>2</sub>H (M + H)<sup>+</sup> 151.0759, found 151.0765.



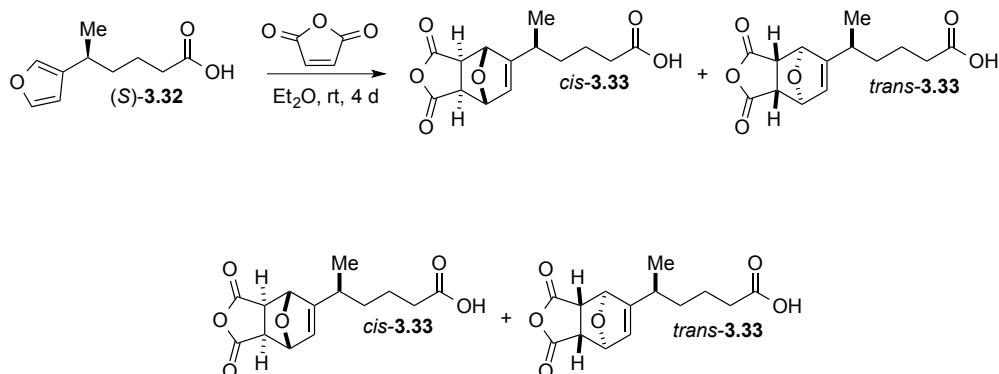
**(R)-1-(2-Furyl)-1,5-pentanediol ((R)-3.22).** Anhydrous THF (75 mL) was added to a flame-dried round-bottom flask equipped with a stir bar. Substrate **(R)-3.30** (1.22 g, 8.00 mmol, 1.00 equiv) was added and the reaction mixture cooled to 0 °C. BH<sub>3</sub>·THF (20 mL, 20 mmol, 1.0 M in THF, 2.5 equiv) was slowly added and the reaction mixture was stirred at ambient temperature for 4 h. The reaction was then cooled to 0 °C, diluted with 1:1 THF/MeOH (120 mL), and treated with 30% w/w H<sub>2</sub>O<sub>2</sub> (40 mL) and 3 M NaOH (75 mL) before stirring at ambient temperature another 4 h. The reaction mixture was partitioned between H<sub>2</sub>O and EtOAc and the aqueous layer was extracted with EtOAc (3 × 100 mL). The combined organic layers were washed with brine, dried over MgSO<sub>4</sub>, and concentrated in vacuo. The product was purified by flash column chromatography (20–70% EtOAc/hexanes) to afford the title compound as a pale yellow oil (1.13 g, 6.64 mmol, 83%). Enantiomeric excess could not be determined for the title compound using chiral SFC and GC instrumentation. **TLC** R<sub>f</sub> = 0.6 (10% EtOAc/hexanes); **<sup>1</sup>H NMR** (400 MHz, CDCl<sub>3</sub>) δ 7.39 (d, *J* = 1.5, 2H), 6.41 (s, 1H), 4.68 (at, *J* = 6.6, 1H), 3.65 (t, *J* = 6.4, 2H), 1.91–1.38 (m, 8H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 143.5, 139.2, 129.2, 108.5, 67.0, 62.9, 37.6, 32.5, 22.0; **IR** (neat) 3329, 2926, 2861, 1020, 874 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>9</sub>H<sub>14</sub>O<sub>3</sub>Na (M + Na)<sup>+</sup> 193.0841, found 193.0842; [α]<sub>D</sub><sup>28</sup> + 13.6 (*c* 0.9, CHCl<sub>3</sub>).



**(R)- $\delta$ -(3-Furyl)- $\delta$ -valerolactone ((R)-3.31)** was prepared according to a modified procedure reported by Forsyth.<sup>35a</sup> We found for this substrate in particular that the preceding diol **3.22** had to be pure for cyclization to proceed. To a stirring solution of diol **(R)-3.22** (0.16 g, 0.92 mmol, 1.0 equiv), in DCM (10 mL) was added bis-acetoxyiodobenzene (BAIB, 0.89 g, 2.8 mmol, 3.0 equiv) followed by 2,2,6,6-tetramethylpiperidinyloxy (TEMPO, 29 mg, 0.18 mmol, 0.20 equiv). After stirring for 5 h at ambient temperature, sat. aq. Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> (10 mL) and Et<sub>2</sub>O (10 x mL) were added and the organic phase was washed with sat. aq. NaHCO<sub>3</sub> (10 mL) and then H<sub>2</sub>O (10 mL). The combined aqueous layers were extracted with Et<sub>2</sub>O (3 x 10 mL). The combined organic layers were washed with brine, dried over MgSO<sub>4</sub>, filtered, and concentrated in vacuo. The product was purified by flash column chromatography (20–50% Et<sub>2</sub>O/pentanes) to afford the title compound as a pale yellow oil (0.67 g, 0.40 mmol, 44%, 90% ee). Absolute configuration was assigned as *R* from the preceding CBS reduction to **(R)-3.30**.<sup>34</sup> This type of oxidative cyclization is known to proceed with retention of stereochemical information from the requisite diol.<sup>35b</sup> **TLC** R<sub>f</sub> = 0.3 (50% Et<sub>2</sub>O/pentanes); **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.46 (as, 1H), 7.42 (as, 1H), 6.42 (as, 1H), 5.36 (dd, J = 9.5, 3.4, 1H), 2.70–2.63 (m, 1H), 2.58–2.51 (m, 1H), 2.19–2.14 (m, 1H), 2.03–1.87 (m, 3H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>)  $\delta$  171.2, 143.8, 139.7, 125.1, 108.5, 75.2, 29.6, 28.8, 18.5; **IR** (neat) 2922, 1727, 1504, 1346, 1227, 1156 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>9</sub>H<sub>10</sub>O<sub>3</sub>Na (M + Na)<sup>+</sup> 189.0528, found 189.0529; [α]<sub>D</sub><sup>29</sup> – 18.6 (*c* 1.1, CHCl<sub>3</sub>); **SFC** analysis (Whelk-(*R,R*), 5% IPA, 2.5 mL/min, 215 nm) indicated 90% ee: t<sub>R</sub> (major) = 16.8 minutes, t<sub>R</sub> (minor) = 13.9 minutes.

*Synthesis of cis- and trans-3.33*

**Scheme 3.10.** Diels–Alder cycloaddition of (*S*)-3.32.



**5-(1,3-Dioxo-1,3,3a,4,7,7a-hexahydro-4,7-epoxyisobenzofuran-5-yl)hexanoic acid (3.33)** was prepared according to a modified procedure by Woodward.<sup>40</sup> To a flame-dried 7 mL reaction vial equipped with a  $\text{N}_2$  line was added maleic anhydride (31 mg, 0.31 mmol, 3.0 equiv) and 0.5 mL  $\text{Et}_2\text{O}$ , and the mixture stirred vigorously until dissolution of maleic anhydride. Substrate (*S*)-3.32 (19 mg, 0.10 mmol, 1.0 equiv) was added as a solution in 0.5 mL  $\text{Et}_2\text{O}$ , and the reaction was stirred at ambient temperature for 4 d, whereupon the solvent was removed in vacuo. Phenyltrimethylsilane (PhTMS) was added as internal standard and a  $^1\text{H}$  NMR yield (64%) was obtained before purification. The crude mixture was purified by flash column chromatography in 50%  $\text{EtOAc}/\text{hexanes}$  to afford a colorless oil (13.3 mg) containing a mixture of the title compound (46% calculated yield) and maleic anhydride. A small amount of analytically pure sample was obtained for characterization. The dr (1:1.1) was determined based on integration of diastereotopic carbons in the  $^{13}\text{C}$  NMR spectrum. **TLC**  $\text{R}_f = 0.2$  (50%  $\text{EtOAc}/\text{hexanes}$ );  **$^1\text{H}$  NMR** (400 MHz,  $\text{CDCl}_3$ )  $\delta$  6.10 (t,  $J = 1.5$ , 1H), 5.38, (s, 1H), 5.29 (s, 1H), 3.25–3.19 (m, 2H), 2.49–2.43 (m, 1H), 2.41–2.35 (m, 2H), 1.66–1.51 (m, 3H), 1.49–1.39 (m, 1H), 1.12 (d,  $J = 7.0$ , 3H);  **$^{13}\text{C}$  NMR** (125.7 MHz,  $\text{CDCl}_3$ )  $\delta$  178.12, 178.10, 170.53, 170.49, 170.2, 157.0, 156.8, 128.68, 128.65, 83.5, 83.4, 83.23, 83.21, 50.2, 50.1, 49.1, 49.0, 34.2, 34.0, 33.54, 33.48, 32.8,

32.5, 22.1, 22.0, 18.8, 18.4; **IR** (neat) 2931, 1781, 905, 728 cm<sup>-1</sup>; **HRMS** (TOF MS ES-) *m/z* calcd for C<sub>14</sub>H<sub>15</sub>O<sub>6</sub> (M - H)<sup>-</sup> 279.0869, found 279.0870; [α]<sub>D</sub><sup>28</sup> + 7.1 (*c* 0.2, CDCl<sub>3</sub>).

### ***General Procedures for Biological Experiments***

Biological experiments were performed according to a modified procedure by Alley.<sup>43</sup>

### ***Materials***

The following reagents were obtained from commercial sources as indicated: RPMI 1640 medium (HyClone); McCoy's 5A medium (HyClone); fetal bovine serum (FBS), heat-inactivated (Omega Scientific); L-glutamine, 200 mM (Gibco); penicillin/streptomycin solution 50X (Mediatech); amphotericin B (HyClone); DMEM/Ham's Nutrient Mixture F12 containing 2.5 mM L-glutamine, 3151 mg/L dextrose, and 55 mg/L sodium pyruvate (Sigma-Aldrich); horse serum (Sigma-Aldrich); 50 μM hydrocortisone solution (Sigma-Aldrich); human insulin solution (Sigma-Aldrich); cholera toxin (Sigma-Aldrich); human Epidermal Growth Factor (EGF), recombinant (Sigma-Aldrich); 0.25% Trypsin-EDTA (Gibco); nuclease-free sterile water (Fisher Scientific); molecular biology grade DMSO (Sigma-Aldrich); MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) (Sigma-Aldrich).

### ***Cell Lines and Culture Conditions***

**MCF-10A** cells were maintained in standard medium according to a modified recipe by Brugge:<sup>53</sup> DMEM/F12 supplemented with 5% horse serum, 10 μg/mL human insulin, 0.5 μg/mL

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<sup>53</sup> Debnath, J.; Muthuswamy, S. K.; Brugge, J. S. *Methods* **2003**, 30, 256.

hydrocortisone, 10 ng/mL EGF, 100 ng/mL cholera toxin, and penicillin/streptomycin. Experiments with MCF-10A cells were performed in the same medium.

**MCF-7** cells were maintained in RPMI 1640 supplemented with 10% FBS, L-glutamine, and penicillin/streptomycin. Experiments with MCF-7 cells were performed in the same medium.

**MDA-MB-468** cells were maintained in RPMI 1640 supplemented with 10% FBS, L-glutamine, penicillin/streptomycin, and amphotericin B. Experiments with MDA-MB-468 cells were performed in the same medium.

**C4-2B** cells were maintained in RPMI 1640 supplemented with 10% FBS, L-glutamine, and penicillin/streptomycin. Experiments with C4-2B cells were performed in the same medium.

**CAKI-2** cells were maintained in McCoy's 5A supplemented with 10% FBS, L-glutamine, and penicillin/streptomycin. Experiments with CAKI-2 cells were performed in the same medium.

**SK-OV-3** cells were maintained in RPMI 1640 supplemented with 10% FBS, L-glutamine, and penicillin/streptomycin. Experiments with SK-OV-3 cells were performed in the same medium.

### ***Evaluation of Compounds Against Cell Lines***

#### ***Preparation of Cell Lines***

**MCF-10A** cells were centrifuged in DMEM/F12 at 1.0 rcf for 8 min, then the pellet was resuspended in DMEM/F12. The cells were seeded at 20,000 cells per well in 24-well flat bottom plates, using 500 µL per well DMEM/F12, and grown for 24 h in 5% CO<sub>2</sub> at 37 °C.

**MCF-7** cells were centrifuged in RPMI at 1.0 rcf for 8 min, then the pellet was resuspended in RPMI. The cells were seeded at 20,000 cells per well in 24-well flat bottom plates, using 500 µL per well RPMI, and grown for 24 h in 5% CO<sub>2</sub> at 37 °C.

**MDA-MB-468** cells were centrifuged in RPMI with amphotericin B at 1.0 rcf for 8 min, then the pellet was resuspended in RPMI with amphotericin B. The cells were seeded at 20,000 cells per well in 24-well flat bottom plates, using 500 µL per well RPMI with amphotericin B, and grown for 24 h in 5% CO<sub>2</sub> at 37 °C.

**C4-2B** cells were centrifuged in RPMI at 1.0 rcf for 8 min, then the pellet was resuspended in RPMI. The cells were seeded at 20,000 cells per well in 24-well flat bottom plates, using 500 µL per well RPMI, and grown for 24 h in 5% CO<sub>2</sub> at 37 °C.

**CAKI-2** cells were centrifuged in McCoy's 5A at 1.0 rcf for 8 min, then the pellet was resuspended in McCoy's 5A. The cells were seeded at 12,500 cells per well in 24-well flat bottom plates, using 500 µL per well McCoy's 5A, and grown for 24 h in 5% CO<sub>2</sub> at 37 °C.

**SK-OV-3** cells were centrifuged in RPMI at 1.0 rcf for 8 min, then the pellet was resuspended in RPMI. The cells were seeded at 20,000 cells per well in 24-well flat bottom plates, using 500 µL per well RPMI, and grown for 24 h in 5% CO<sub>2</sub> at 37 °C.

#### *Preparation of Compounds*

The compounds were dissolved in molecular biology grade DMSO to achieve a 100 mM stock solution, and then sterile filtered through a 0.45 µm PVDF syringe filter unit (Fisher Scientific). The 100 mM stock solutions were subsequently diluted to 20 mM stock solutions, then diluted to a final concentration of 10 µM in the corresponding medium for each cell line. Additionally, the corresponding DMSO vehicle control was diluted using the same medium.

After 24 h growth, the medium was carefully aspirated from each of the wells containing cells. The cells were treated by replacing the normal medium with 600 µL fresh medium per well containing the individual compounds or vehicle control (day 0). The plates were gently agitated,

then the cells were incubated with compound at 37 °C with 5% CO<sub>2</sub> for 72 h. On day 3, cell proliferation was measured using the MTT absorbance assay.

#### *MTT Absorbance Assay*

Quantitative analysis was performed according to a modified procedure by Alley.<sup>43</sup> After 72 h incubation with compound, 200 µL of MTT dye solution (3 mg MTT/ 1 mL PBS) was carefully added to each well containing cells. The plates were incubated at 37 °C for 1 h 15 min, after which the MTT dye was aspirated from each well. 300 µL of MTT dissolve solution (4% 1M HCl in IPA) was added to each well and the plates were agitated 15 min. From each well, 200 µL of cell lysate solution was carefully transferred to a new well in a clear-bottom 96-well plate suitable for UV-Vis spectroscopy. Cells were quantified by absorbance at 570 nm. The absorbance values were normalized to the DMSO vehicle control. The normalized values were plotted as an average ± standard deviation of 3 wells per compound.

#### *Raw Data for Cell Assays*

**Table 3.5.** Relative cell numbers for compound *syn-3.34*.

<b>MCF-7 cell line</b>	<b>Trial:</b>	<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>	<b>Average</b>	<b>SD</b>
Fluorescence counts (DMSO)		1177201	1525796	1288876	1409215	1350272	150561
Fluorescence counts ( <i>syn-3.34</i> )		138743	165322	255419	243509	200748	57495
Relative cell numbers ( <i>syn-3.34</i> ), normalized (%)		10.3	12.2	18.9	18.0	14.9	4.3
<b>MDA-MB-468 cell line</b>	<b>Trial:</b>	<b>1</b>	<b>2</b>	<b>3</b>		<b>Average</b>	<b>SD</b>
Absorbance values (DMSO)		1.6433	1.5081	1.4655		1.5390	0.0928
Absorbance values ( <i>syn-3.34</i> )		1.5390	1.5651	1.6679		1.5907	0.0681
Relative cell numbers ( <i>syn-3.34</i> ), normalized (%)		98.0	99.7	106.2		101.3	4.3

<b>C4-2B cell line</b>	<b>Trial:</b>	<b>1</b>	<b>2</b>	<b>3</b>	<b>Average</b>	<b>SD</b>
Absorbance values (DMSO)		0.9790	0.9650	0.9612	0.9684	0.0094
Absorbance values ( <i>syn-3.34</i> )		1.0003	0.9264	1.0744	1.0004	0.0740
Relative cell numbers ( <i>syn-3.34</i> ), normalized (%)		103.3	95.7	110.9	103.3	7.6

<b>CAKI-2 cell line</b>	<b>Trial:</b>	<b>1</b>	<b>2</b>	<b>3</b>	<b>Average</b>	<b>SD</b>
Absorbance values (DMSO)		1.2051	0.9685	0.9101	1.0279	0.1562
Absorbance values ( <i>syn-3.34</i> )		0.8728	0.9132	1.0017	0.9292	0.0659
Relative cell numbers ( <i>syn-3.34</i> ), normalized (%)		84.9	88.8	97.5	90.4	6.4

<b>SK-OV-3 cell line</b>	<b>Trial:</b>	<b>1</b>	<b>2</b>	<b>3</b>	<b>Average</b>	<b>SD</b>
Absorbance values (DMSO)		2.0732	1.8405	1.5620	1.8252	0.2559
Absorbance values ( <i>syn-3.34</i> )		1.9432	2.0110	2.1475	2.0339	0.1041
Relative cell numbers ( <i>syn-3.34</i> ), normalized (%)		99.4	102.8	109.8	104.0	5.3

**Table 3.6.** Relative cell numbers for compound *syn-3.17*.

<b>MCF-10A cell line</b>	<b>Trial:</b>	<b>1</b>	<b>2</b>	<b>3</b>	<b>Average</b>	<b>SD</b>
Absorbance values (DMSO)		2.7994	2.7073	2.7361	2.7476	0.0471
Absorbance values ( <i>syn-3.17</i> )		0.2520	0.2484	0.2635	0.2546	0.0079
Relative cell numbers ( <i>syn-3.17</i> ), normalized (%)		9.2	9.0	9.6	9.3	0.3

<b>MCF-7 cell line</b>	<b>Trial:</b>	<b>1</b>	<b>2</b>	<b>3</b>	<b>Average</b>	<b>SD</b>
Absorbance values (DMSO)		0.4127	0.3514	0.3336	0.3659	0.0415
Absorbance values ( <i>syn-3.17</i> )		0.1770	0.1858	0.2070	0.1899	0.0154
Relative cell numbers ( <i>syn-3.17</i> ), normalized (%)		48.4	50.8	56.6	51.9	4.2

<b>MDA-MB-468 cell line</b>	<b>Trial:</b>	<b>1</b>	<b>2</b>	<b>3</b>	<b>Average</b>	<b>SD</b>
Absorbance values (DMSO)		1.2696	1.1288	1.2005	1.1996	0.0704
Absorbance values ( <i>syn-3.17</i> )		0.1797	0.1338	0.1412	0.1516	0.0246
Relative cell numbers ( <i>syn-3.17</i> ), normalized (%)		15.0	11.2	11.8	12.6	2.1

**Table 3.7.** Relative cell numbers for compound *anti*-3.35.

<b>MCF-7 cell line</b>	<b>Trial:</b>	<b>1</b>	<b>2</b>	<b>3</b>	<b>Average</b>	<b>SD</b>
Absorbance values (DMSO)		0.9522	0.6718	0.7772	0.8004	0.1416
Absorbance values ( <i>anti</i> -3.35)		0.2498	0.2870	0.3857	0.3075	0.0702
Relative cell numbers ( <i>anti</i> -3.35), normalized (%)		31.2	35.9	48.2	38.4	8.8
<b>MDA-MB-468 cell line</b>	<b>Trial:</b>	<b>1</b>	<b>2</b>	<b>3</b>	<b>Average</b>	<b>SD</b>
Absorbance values (DMSO)		1.3253	1.2975	1.2052	1.2760	0.0629
Absorbance values ( <i>anti</i> -3.35)		0.3928	0.4578	0.3655	0.4054	0.0474
Relative cell numbers ( <i>anti</i> -3.35), normalized (%)		30.8	35.9	28.6	31.8	3.7

## *Chapter Four*

### **Silver-Catalyzed Enantioselective Propargylation Reactions of *N*-Sulfonyl Ketimines**

#### **4.1 Introduction**

Nearly half of the top 200 pharmaceuticals in 2012 contain functional groups that can be prepared from  $\alpha$ -chiral amines.<sup>1,2</sup> To access this moiety, numerous enantioselective methods for the synthesis of chiral amines have been developed, many of which involve addition of organometallic nucleophiles to aldimines.<sup>3</sup> Additions to ketimines pose specific challenges. For example, mixtures of E and Z isomers can lead to low levels of enantioinduction.<sup>4</sup> These obstacles have inspired creative approaches<sup>5</sup> including use of cyclic *N*-sulfonyl ketimines (e.g., 4.1), which do not undergo E/Z isomerization and are synthesized in one step from saccharin.<sup>6,7</sup> Hayashi and co-workers have pioneered the rhodium-catalyzed enantioselective arylation reactions of *N*-sulfonyl ketimines; other elegant examples of arylation, allylation, and

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<sup>1</sup> Portions of this Chapter were originally published as: Osborne, C. A.; Endean, T. B. D.; Jarvo, E. R. *Org. Lett.* **2015**, *17*, 5340.

<sup>2</sup> 94 of the top 200 pharmaceuticals by U.S. retail sales in 2012 contained functional groups that could be prepared from  $\alpha$ -chiral amines, while 25 of the top 200 pharmaceuticals contained  $\alpha$ -chiral amines. See: (a) Njarðarson Group. *Top Pharmaceuticals Poster*. <http://jon.oia.arizona.edu/top-pharmaceuticals-poster> (accessed May 30, 2015); (b) McGrath, N. A.; Brichacek, M.; Njarðarson, J. T. *J. Chem. Ed.* **2010**, *87*, 1348.

<sup>3</sup> For reviews on catalytic enantioselective methods for generating chiral amines, see: (a) Nugent, T. C.; El-Shazly, M. *Adv. Synth. Catal.* **2010**, *352*, 753; (b) Kobayashi, S.; Mori, Y.; Fossey, J. S.; Salter, M. M. *Chem. Rev.* **2011**, *111*, 2626.

<sup>4</sup> For a discussion and review, see: Riant, O.; Hannedouche, J. *Org. Biomol. Chem.* **2007**, *5*, 873.

<sup>5</sup> For a lead reference, see: Yin, L.; Otsuka, Y.; Takada, H.; Mouri, S.; Yazaki, R.; Kumagai, N.; Shibasaki, M. *Org. Lett.* **2013**, *15*, 698.

<sup>6</sup> (a) For enantioselective hydrogenation, see: Oppolzer, W.; Wills, M.; Starkemann, C.; Bernardinelli, G. *Tetrahedron Lett.* **1990**, *31*, 4117; (b) For homoenolate additions, see: Rommel, M.; Fukuzumi, T.; Bode, J. W. *J. Am. Chem. Soc.* **2008**, *130*, 17266; (c) For enantioselective arylation, see: Nishimura, T.; Noishiki, A.; Tsui, G. C.; Hayashi, T. *J. Am. Chem. Soc.* **2012**, *134*, 5056; (d) For formal [3+2] cycloadditions with TMM, see: Trost, B. M.; Silverman, S. M. *J. Am. Chem. Soc.* **2012**, *134*, 4941.

<sup>7</sup> Synthesis of *N*-sulfonyl ketimines from saccharin: Davis, F. A.; Towson, J. C.; Vashi, D. B.; ThimmaReddy, R.; McCauley, Jr., J. P.; Harakal, M. E.; Gosciniaik, D. J. *J. Org. Chem.* **1990**, *55*, 1254.

alkenylation reactions have also been reported.<sup>6c,8</sup> An enantioselective propargylation reaction would afford a chiral sultam with a pendant terminal alkyne, a valuable functional group handle that can be easily derivatized for further synthetic elaboration.<sup>9</sup> In this Chapter, we report the first enantioselective propargylation reaction of ketimines (eq. 4.1).



Building on early advances in enantioselective propargylation reactions of aldehydes, in the past five years there has been rapid development of enantioselective propargylation reactions of ketones and aldimines.<sup>10,11,12</sup> The Jarvo laboratory has reported the silver-catalyzed enantioselective propargylation reactions of aldimines and diarylketones.<sup>11b,13,14</sup> Using AgF and chiral phosphine ligands from the Walphos family provided a variety of homopropargylic amines and alcohols in good yield and high enantiomeric excess (ee). We hypothesized that a

<sup>8</sup> For enantioselective arylation, see: (a) Jiang, C.; Lu, Y.; Hayashi, T. *Angew. Chem., Int. Ed.* **2014**, *53*, 9936; (b) Yang, G.; Zhang, W. *Angew. Chem., Int. Ed.* **2013**, *52*, 7540; (c) Wang, H.; Jiang, T.; Xu, M.-H. *J. Am. Chem. Soc.* **2013**, *135*, 971; (d) Jiang, T.; Wang, Z.; Xu, M.-H. *Org. Lett.* **2015**, *17*, 528; (e) For enantioselective allylation, see: Luo, Y.; Hepburn, H. B.; Chotsang, N.; Lam, H. W. *Angew. Chem., Int. Ed.* **2012**, *51*, 8309; (f) For enantioselective alkenylation, see: Luo, Y.; Carnell, A. J.; Lam, H. W. *Angew. Chem., Int. Ed.* **2012**, *51*, 6762.

<sup>9</sup> For recent examples in synthesis of polyketides, see: (a) Mailhol, D.; Willwacher, J.; Kausch-Busies, N.; Rubitski, E. E.; Sobol, Z.; Schuler, M.; Lam, M.-H.; Musto, S.; Loganzo, F.; Maderna, A.; Fürstner, A. *J. Am. Chem. Soc.* **2014**, *136*, 15719; (b) Reznik, S. K.; Marcus, B. S.; Leighton, J. L. *Chem. Sci.* **2012**, *3*, 3326.

<sup>10</sup> (a) For a comprehensive review, see: Ding, C.-H.; Hou, X.-L. *Chem. Rev.* **2011**, *111*, 1914; (b) For a synopsis of catalytic enantioselective propargylation reactions of ketones and imines, see: Wisniewska, H. M.; Jarvo, E. R. *J. Org. Chem.* **2013**, *78*, 11629.

<sup>11</sup> For catalyst-controlled, enantioselective propargylation of imines, see: (a) Kagoshima, H.; Uzawa, T.; Akiyama, T. *Chem. Lett.* **2002**, *31*, 298; (b) Wisniewska, H. M.; Jarvo, E. R. *Chem. Sci.* **2011**, *2*, 807; (c) Viera, E. M.; Haeffner, F.; Snapper, M. L.; Hoveyda, A. H. *Angew. Chem., Int. Ed.* **2012**, *51*, 6618.

<sup>12</sup> For examples of diastereoselective propargylation of imines, see: (a) Gonzalez, A. Z.; Soderquist, J. A. *Org. Lett.* **2007**, *9*, 1081; (b) Fandrick, D. R.; Johnson, C. S.; Fandrick, K. R.; Reeves, J. T.; Tan, Z.; Lee, H.; Song, J. J.; Yee, N. K.; Senanayake, C. H. *Org. Lett.* **2010**, *12*, 748; (c) García-Muñoz, M. J.; Zacconi, F.; Foubelo, F.; Yus, M. *Eur. J. Org. Chem.* **2013**, 1287; (d) Guo, T.; Song, R.; Yuan, B.-H.; Chen, X.-Y.; Sun, X.-W.; Lin, G.-Q. *Chem. Commun.* **2013**, *49*, 5402; (e) Chen, D.; Xu, M.-H. *Chem. Commun.* **2013**, *49*, 1327.

<sup>13</sup> Kohn, B. L.; Ichiiishi, N.; Jarvo, E. R. *Angew. Chem., Int. Ed.* **2013**, *52*, 4414.

<sup>14</sup> For silver-catalyzed enantioselective allylation reactions of ketones, see: Wadamoto, M.; Yamamoto, H. *J. Am. Chem. Soc.* **2005**, *127*, 14556.

Ag/Walphos catalyst would be able to differentiate between the *Re* and *Si* faces of diaryl ketimines, based on their structural similarity to diarylketones.

## 4.2 Optimization of Reactions of Aryl and Alkyl Ketimines

Previous methods in the Jarvo laboratory for preparing the Ag/Walphos catalyst were time and labor intensive. Catalyst preparation involved formation of the silver phosphine complex in methanol followed by solvent replacement with THF.<sup>15</sup> We turned to the silver-catalyzed enantioselective propargylation of aldimine **4.4** as a well-developed model system to improve catalyst preparation. Heating AgF and Walphos-1 in DMF for a 30 minute pre-stir increased the solubility of AgF and simplified reaction set-up; these results (Table 4.1, entries 1 and 2) were comparable to previously published results obtained using a methanol pre-stir (entries 3 and 4).

**Table 4.1.** Optimization of catalyst formation.

entry	solvent	temp (°C)	yield 4.5 (%) <sup>a</sup>	ee 4.5 (%) <sup>b</sup>	note
1	DMF	22	75	91	pre-stir heated
2	DMF	-20	89	97	pre-stir heated
3 <sup>c</sup>	MeOH then THF	22	85	88	
4 <sup>c</sup>	MeOH then THF	-20	90	97	

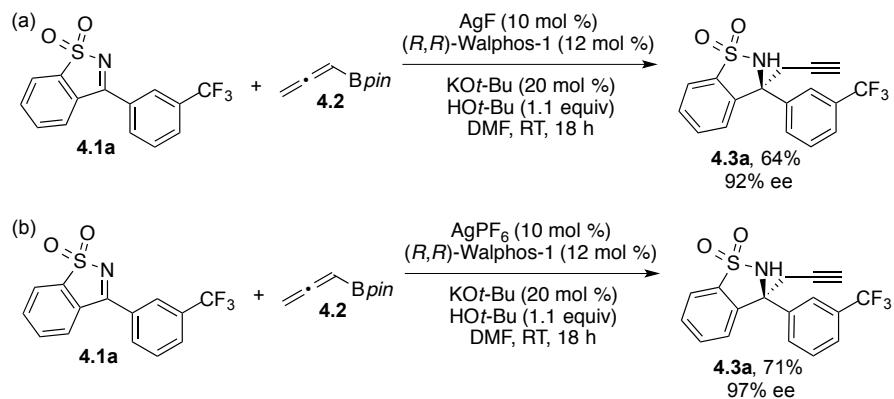
<sup>a</sup>Determined using <sup>1</sup>H NMR by comparison to PhTMS as internal standard. <sup>b</sup>Determined using chiral SFC chromatography. <sup>c</sup>See reference 11b.

With an improved method for catalyst preparation, we began to optimize reaction conditions for the enantioselective propargylation of diaryl ketimines. Employing AgF as the

<sup>15</sup> See references 11b, 13, and 14 for full details.

precatalyst and Walphos-1 as the ligand formed homopropargylic sultam **4.3a** in 64% yield and 92% ee (Scheme 4.1a). A survey of alternate silver sources found that AgPF<sub>6</sub> provided higher yield, likely due to increased solubility. Furthermore, AgPF<sub>6</sub> provided higher ee than AgF. Employing AgPF<sub>6</sub> as the precatalyst and Walphos-1 as the ligand provided **4.3a** in 71% yield and 97% ee (Scheme 4.1b).

**Scheme 4.1.** Enantioselective propargylation of ketimine **4.1a** with (a) AgF or (b) AgPF<sub>6</sub>.



We were interested in improving reaction conditions for the propargylation of substrate **4.1b** (Table 4.2). Initial reaction at –20 °C resulted in low conversion to sulfonamide **4.3b** (entry 1). We hypothesized that protodeborylation of allenylboronic acid pinacol ester **4.2** to allene (C<sub>3</sub>H<sub>4</sub>) was competitive with the desired addition reaction and thus resulted in modest yields.<sup>16</sup> Use of two additional equivalents of allenylboronic acid pinacol ester **4.2** via slow addition improved the yield and provided **4.3b** in 98% ee (entry 2). We increased the temperature and found that at ambient temperature, **4.3b** was formed in a modest yield; fortunately, the ee remained high (entry 3). Increasing the equivalents of **4.2** at room temperature further improved the yield, providing **4.3b** in 71% yield and 97% ee (entry 4).

<sup>16</sup> Kohn, B. L.; Jarvo, E. R. *Org. Lett.* **2011**, *13*, 4858.

**Table 4.2.** Optimization of silver-catalyzed propargylation reaction.

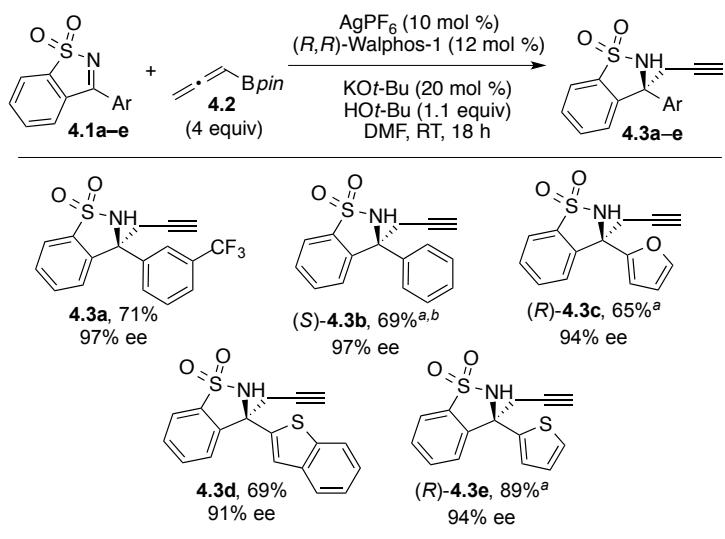
entry	temp (°C)	equiv 4.2	yield 4.1b (%) <sup>a</sup>	yield 4.3b (%) <sup>a</sup>	ee 4.3b (%) <sup>b</sup>
1	-20	2	57	12	nd
2	-20	4	49	52	98
3	22	2	43	52	96
4	22	4	34	71	97

<sup>a</sup>Determined using <sup>1</sup>H NMR by comparison to PhTMS as internal standard. <sup>b</sup>Determined using chiral SFC.

Having determined optimized conditions for this reaction, we proceeded to evaluate the substrate scope. A wide range of aryl ketimines underwent enantioselective propargylation in high yield and >91% ee (Scheme 4.2). Ketimines containing electron-withdrawing groups formed products in excellent ee (**4.3a**). We found that several heterocycles were also tolerated in the reaction. Ketimines containing furan, thiophene, and benzothiophene functionalities reacted smoothly to provide the corresponding homopropargylic sulfonamides (**4.3c–e**) in high ee. The absolute configurations of **4.3b**, **4.3c**, and **4.3e** were determined by X-ray crystallographic analysis.<sup>17</sup>

<sup>17</sup> For supplementary crystallographic data, see the Experimental Section and CCDC 1405841, 1405894 and 1405895.

**Scheme 4.2.** Scope of diaryl sultams.

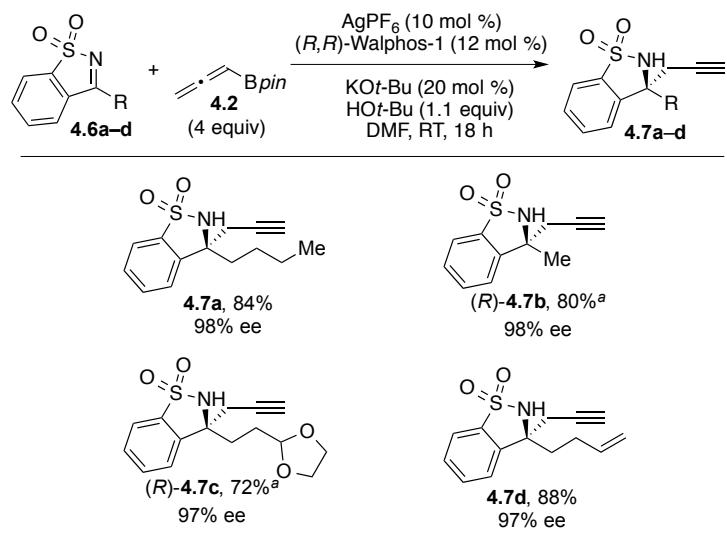


<sup>a</sup>Absolute configuration assigned by X-ray crystallographic analysis.<sup>17</sup> <sup>b</sup>Reaction performed for 6 hours.

We were gratified to find that alkyl ketimines were also well tolerated in the reaction (Scheme 4.3), since we were concerned that these substrates would tautomerize to enamines in the presence of potassium *tert*-butoxide. Several alkyl ketimines reacted to give homopropargylic products in excellent ee (**4.7a–c**). Furthermore, we found that other functional groups are compatible with this methodology: sultam **4.7c**, containing an acetal protecting group, was formed in high yield. The absolute configuration of **4.7c** was determined by X-ray crystallographic analysis.<sup>18</sup>

<sup>18</sup> For supplementary crystallographic data, see the Experimental Section and CCDC 1410049 and 1405843.

**Scheme 4.3.** Scope of alkyl sultams.



<sup>a</sup>Absolute configuration assigned by X-ray crystallographic analysis.<sup>18</sup>

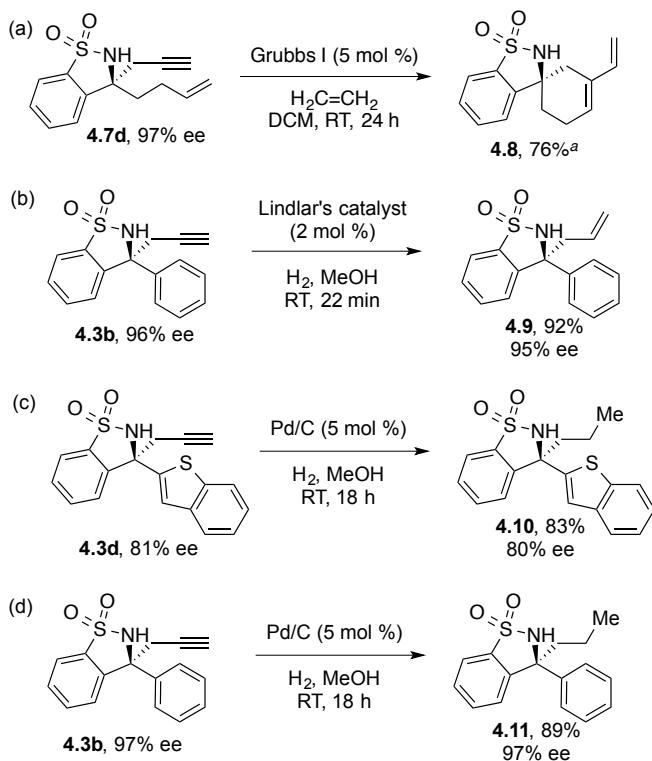
### 4.3 Synthetic Transformations of Homopropargylic Sultams

To emphasize the utility of the pendant terminal alkyne, we synthesized derivatives of several alkyl and aryl homopropargylic sulfonamides (Scheme 4.4). We prepared compound **4.7d** for an enyne ring-closing metathesis (Scheme 4.4a).<sup>19</sup> In the presence of 5 mol % Grubbs I catalyst and under an atmosphere of ethylene, the desired spirocycle **4.8** was obtained in 76% yield. Lindlar reduction of **4.3a** provided the corresponding enantioenriched diaryl allyl sultam **4.9**, a moiety that has not been previously reported (Scheme 4.4b).<sup>8e,20</sup>

<sup>19</sup> Diver, S. T.; Giessert, A. J. *Chem. Rev.* **2004**, *104*, 1317.

<sup>20</sup> For enantioselective allylation of alkyl cyclic N-sulfonyl ketimines, see: Hepburn, H. B.; Chotsaeng, N.; Luo, Y.; Lam, H. W. *Synthesis* **2013**, *45*, 2649.

**Scheme 4.4.** Synthetic transformations of homopropargylic sultams.



<sup>a</sup>Enantiomeric excess could not be determined using chiral SFC instrumentation.

We further highlighted the versatility of the alkyne moiety by reducing several alkynes to their corresponding alkanes. Alkyne **4.3d** was fully reduced to alkane **4.10** using palladium on carbon for 18 hours, without reduction of the cyclic benzylic sulfonamide (Scheme 4.4c). Likewise, reduction of **4.3b** afforded alkane **4.11** in 89% yield without racemization (Scheme 4.4d).

#### 4.4 Mechanism of Propargylation

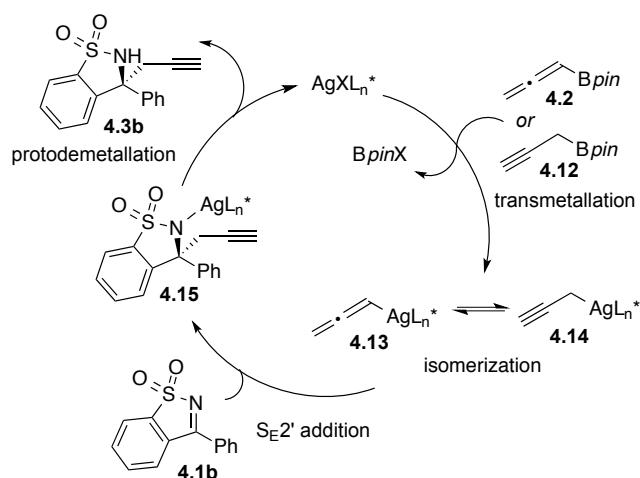
We sought to establish a reasonable mechanism for this propargylation reaction; two of the most likely possibilities are presented in Figure 4.1.<sup>21</sup> Our approach to distinguish between these mechanisms was to compare product distributions from reactions employing isomeric

<sup>21</sup> For a discussion of these mechanistic possibilities, including representative examples, see references 10a and 10b.

borolane reagents, allenyl borolane **4.2** and propargyl borolane **4.12**. Importantly, both mechanisms take into account our experimental observation that allenyl borolane **4.2** and propargyl borolane **4.12** are not in equilibrium under the reaction conditions during the time frame of the reaction (vide infra).

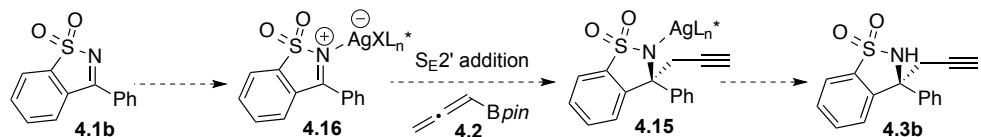
#### Mechanism A

Using either allenyl or propargyl borolane reagent:

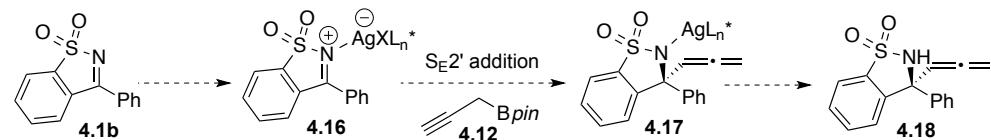


#### Mechanism B

1) Using allenyl borolane reagent:



2) Using propargyl borolane reagent:



**Figure 4.1.** Possible mechanisms for silver-catalyzed propargylation reaction. (a)

Proposed catalytic cycle involving transmetallation of silver catalyst with borolane reagent. (b) Lewis acid catalysis.

Mechanism A involves transmetallation and isomerization of the allenylmetal intermediates.<sup>22</sup> Transmetallation of the silver catalyst with the borolane reagent forms the key nucleophilic allenylsilver complex (**4.13**) *in situ*. Allenylsilver complex **4.13** is in equilibrium with propargylsilver complex **4.14**.<sup>23,24</sup> Addition of allenylsilver complex **4.13** to the ketimine via S<sub>E</sub>2' mechanism is favored to form alkyne **4.3b**. Therefore, if Mechanism A is operative, using either allenyl borolane **4.2** or propargyl borolane **4.12** would result in formation of alkyne **4.3b** via equilibration of **4.13** and **4.14** (Figure 4.1a).

An alternative pathway is Mechanism B, involving direct addition of the borolane reagent to the ketimine.<sup>10</sup> In this scenario, the silver catalyst acts as a chiral Lewis acid in the reaction (Figure 4.1b-1). Coordination of the silver catalyst to form intermediate **4.16** followed by S<sub>E</sub>2' addition of allenyl borolane **4.2** results in formation of alkyne **4.3b**. In this possible mechanism, isomerization of the allenyl- and propargylboron species is slower than attack on the activated electrophile (**4.16**). Therefore, using propargyl borolane **4.12** would provide a different product, allene **4.18** (Figure 4.1b-2).

<sup>22</sup> For selected examples of reactions that likely proceed through a similar mechanism, see: (a) Tamaru, Y.; Goto, S.; Tanaka, A.; Shimizu, N.; Kimura, M. *Angew. Chem., Int. Ed. Engl.* **1996**, *35*, 878; (b) Hameury, T.; Guillemont, J.; Van Hijfte, L.; Bellosta, V.; Cossy, J. *Org. Lett.* **2009**, *11*, 2397; (c) Fandrick, D. R.; Saha, J.; Fandrick, K. R.; Sanyal, S.; Ogikubo, J.; Lee, H.; Roschangar, F.; Song, J. J.; Senanayake, C. H. *Org. Lett.* **2011**, *13*, 5616; (d) Fandrick, K. R.; Ogikubo, J.; Fandrick, D. R.; Patel, N. D.; Saha, J.; Lee, H.; Ma, S.; Grinberg, N.; Busacca, C. A.; Senanayake, C. H. *Org. Lett.* **2013**, *15*, 1214; (e) Mszar, N. W.; Haeffner, F.; Hoveyda, A. H. *J. Am. Chem. Soc.* **2014**, *136*, 3362.

<sup>23</sup> For selected examples of isomerization of allenyl- and propargylmetal complexes, see: (a) Elsevier, C. J.; Kleijn, H.; Boersma, J.; Vermeer, P. *Organometallics* **1986**, *5*, 716; (b) Ogoshi, S.; Fukunishi, Y.; Tsutsumi, K.; Kurosawa, H. *J. Chem. Soc., Chem. Commun.* **1995**, 2485; (c) Ogoshi, S.; Nishida, T.; Fukunishi, Y.; Tsutsumi, K.; Kurosawa, H. *J. Organomet. Chem.* **2001**, *620*, 190.

<sup>24</sup> For examples in the context of palladium-catalyzed cross-coupling reactions, see: (a) Moriya, T.; Miyaura, N.; Suzuki, A. *Synlett* **1994**, 149; (b) Ma, S.; Zhang, A. *J. Org. Chem.* **2002**, *67*, 2287; (c) Marshall, J. A. *Chem. Rev.* **2000**, *100*, 3163.

To rule out one of these two mechanisms, we set out to examine reactions employing propargyl borolane **4.12**.<sup>25</sup> We synthesized **4.12** using a procedure recently published by Fandrick and co-workers, employing iodomethyl borolane **4.19** in a Matteson homologation in the presence of ethynylmagnesium bromide.<sup>22c</sup> This reaction proved extremely sensitive to trace moisture. All reagents and solvents were distilled over molecular sieves but additional attempts at the homologation reaction were plagued by isomerization to allenyl borolane **4.2**.<sup>26</sup> Excess Grignard reagent could catalyze bimolecular isomerization, so DMSO was added as a Lewis basic reagent to coordinate excess magnesium and prevent its chelation with the borolane.<sup>27</sup> To slow isomerization, the number of equivalents of Grignard reagent was scaled down to 0.95 equivalents; however, the reaction yielded primarily starting iodomethyl borolane **4.19**.<sup>28</sup> Further investigation of the effect of the number of equivalents of Grignard reagent on product formation showed a narrow range in which propargyl borolane **4.12** is the primary product of this reaction (Figure 4.2). The optimal number of equivalents of Grignard reagent was determined to be 1.02 equivalents ethynylmagnesium bromide, above which there is significant isomerization to allenyl borolane **4.2**, and below which almost no conversion was observed. The desired product was isolated in a 96:4 propargyl : allenyl ratio at 1.02 equivalents Grignard reagent.

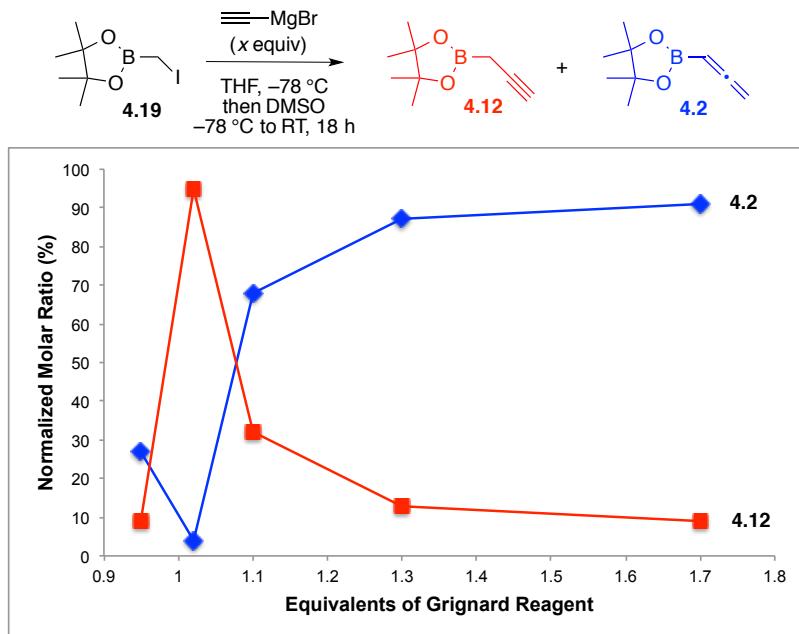
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<sup>25</sup> See reference 22c. A related strategy has been employed to elucidate the mechanistic details of transition-metal-catalyzed allylation reactions. See: (a) reference 14; (b) Shaghafi, M. B.; Kohn, B. L.; Jarvo, E. R. *Org. Lett.* **2008**, *10*, 4743.

<sup>26</sup> Soundararajan, R.; Li, G.; Brown, H. C. *Tetrahedron Lett.* **1994**, *35*, 8961.

<sup>27</sup> Fandrick, D. R.; *Personal communication*.

<sup>28</sup> Soundararajan, R.; Li, G.; Brown, H. C. *Tetrahedron Lett.* **1994**, *35*, 8957.

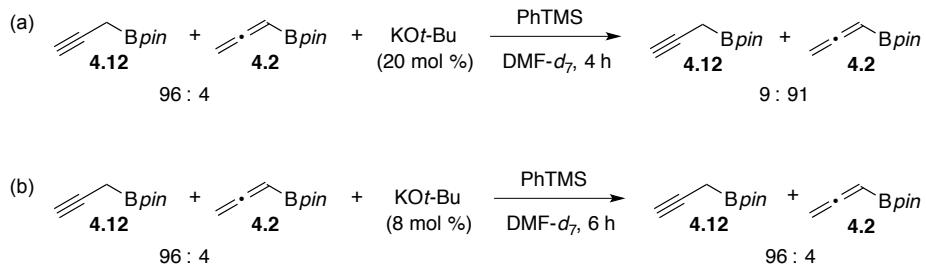


**Figure 4.2.** Isomerization of propargyl borolane **4.12** to allenyl borolane **4.2** in the presence of excess Grignard reagent.

Having established that excess Grignard reagent causes the isomerization of propargyl borolane **4.12** to allenyl borolane **4.2**, we reasoned that potassium *tert*-butoxide could also accelerate isomerization. We were concerned that equilibration might occur under the propargylation reaction conditions. To determine whether **4.12** could isomerize to **4.2** under the propargylation conditions, we performed control experiments employing different amounts of base. In the presence of 20 mol % potassium *tert*-butoxide, we observed the rapid isomerization of propargyl borolane **4.12** to allenyl borolane **4.2** (Scheme 4.5a). However, in the presence of 8 mol % potassium *tert*-butoxide, we observed no discernable isomerization after 6 hours (Scheme 4.5b).<sup>29</sup>

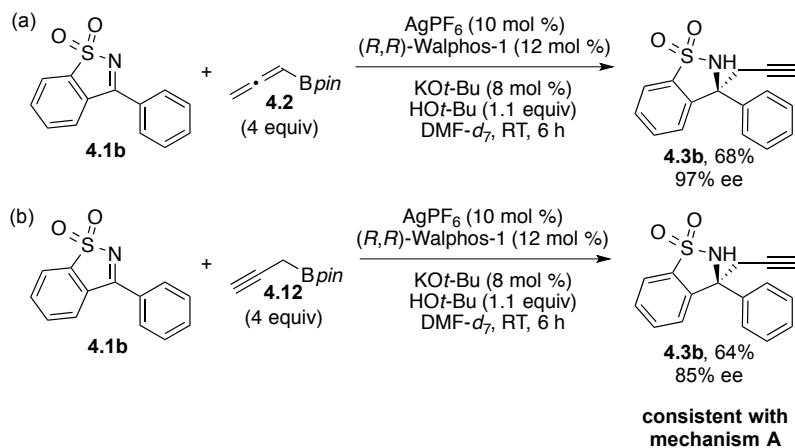
<sup>29</sup> See the Experimental Section for full details.

**Scheme 4.5.** Isomerization of propargyl borolane **4.12** to allenyl borolane **4.2** in the presence of (a) 20 mol % KOT-Bu, or (b) 8 mol % KOT-Bu.



With propargyl borolane **4.12** in hand, we performed the silver-catalyzed addition reaction to ketimine **4.1b**. Use of **4.12** in the reaction afforded alkyne **4.3b** in 64% yield (Scheme 4.6b). We found that the ee of the product remained high, with a slight decrease from 97% to 85% ee when using propargyl borolane **4.12**. Both allenyl borolane **4.2** and propargyl borolane **4.12** provide similar product distributions, most consistent with Mechanism A, where the silver catalyst undergoes transmetallation with the borolane reagent. The reaction was performed in deuterated DMF and the ratio of propargyl to allenyl borolane was analyzed by  $^1\text{H}$  NMR directly from the reaction mixture. The ratio of **4.12**:**4.2** before the reaction was 96:4, and after the reaction it was 94:6. This observation is most consistent with negligible equilibration of **4.12** and **4.2** over the time course of the propargylation reactions.

**Scheme 4.6.** Silver-catalyzed propargylation reaction using (a) allenylboronic acid pinacol ester **4.2**, or (b) propargylboronic acid pinacol ester **4.12**.



## 4.5 Conclusions

We have developed an enantioselective silver-catalyzed propargylation reaction of cyclic *N*-sulfonyl ketimines. Using a catalyst prepared from  $\text{AgPF}_6$  and Walphos-1, we found that many aryl and alkyl homopropargylic amines, including several heterocyclic products, were formed in high yield and excellent ee. Derivatization of the terminal alkyne yielded spirocyclic, alkenyl, or alkyl products. Mechanistic experiments employing propargyl borolane reagent are most consistent with a mechanism in which the silver catalyst undergoes transmetallation with the borolane reagent to generate a nucleophilic allenylboron reagent.

## 4.6 Experimental Details

### General Procedures

NMR spectra were recorded on Bruker DRX-400 (400 MHz  $^1\text{H}$ , 100 MHz  $^{13}\text{C}$ , 376.5 MHz  $^{19}\text{F}$ ), GN-500 (500 MHz  $^1\text{H}$ , 125.7 MHz  $^{13}\text{C}$ , 160.2 MHz  $^{11}\text{B}$ ), or CRYO-500 (500 MHz

<sup>1</sup>H, 125.7 MHz <sup>13</sup>C) spectrometers. Proton chemical shifts are reported in ppm ( $\delta$ ) relative to internal trimethylsilane (TMS,  $\delta$  0.00). Data are reported as follows: chemical shift (multiplicity [singlet (s), broad singlet (br s), doublet (d), apparent doublet (ad), doublet of doublets (dd), doublet of doublets of doublets (ddd), triplet (t), apparent triplet (at), doublet of triplets (dt), triplet of doublets (td), quartet (q), quintet (quint), apparent quintet of doublets (aquintd), sextet (sext), multiplet (m)], coupling constants [Hz], integration). Carbon chemical shifts are reported in ppm ( $\delta$ ) relative to TMS with the solvent resonance as the internal standard (CDCl<sub>3</sub>,  $\delta$  77.16 ppm or DMF-*d*<sub>7</sub>,  $\delta$  163.15 ppm). NMR data were collected at 25 °C. Infrared spectra were obtained on a Thermo Scientific Nicolet iS5 spectrometer with an iD5 ATR tip (neat) and are reported in terms of frequency of absorption (cm<sup>-1</sup>). Analytical thin-layer chromatography (TLC) was performed using Silica Gel 60 Å F254 precoated plates (0.25 mm thickness). Visualization was accomplished by irradiation with a UV lamp and/or staining with *p*-anisaldehyde (PAA) or potassium permanganate (KMnO<sub>4</sub>) solutions. Flash chromatography was performed using Silica Gel 60 (170-400 mesh) from Fisher Scientific. Melting points (m.p.) were obtained using a Mel-Temp melting point apparatus and are uncorrected. Optical rotations were measured with a Rudolph Research Analytical Autopol III Automatic Polarimeter. SFC determinations of enantiopurity were performed on a Berger Analytical instrument using a Daicel<sup>TM</sup> Chiralpak® column (OD-H, AD-H, AS-H, or (R,R)-Whelk-O); 100 bar, 215 nm, 50 °C). High resolution mass spectrometry was performed by the University of California, Irvine Mass Spectrometry Center.

All reactions were carried out under a N<sub>2</sub> atmosphere, unless otherwise stated. All glassware was either oven-dried or flame-dried prior to use. *N,N*-Dimethylformamide (DMF), tetrahydrofuran (THF), diethyl ether (Et<sub>2</sub>O), dichloromethane (DCM), triethylamine (TEA),

methanol (MeOH), and *N,N*-dimethylacetamide (DMA) were degassed with argon and then passed through two 4 x 36 inch columns of anhydrous neutral A-2 alumina (8 x 14 mesh; LaRoche Chemicals; activated under a flow of argon at 350 °C for 12 hours) to remove H<sub>2</sub>O. Other solvents were purchased “anhydrous” commercially, or were purified as described.

AgPF<sub>6</sub> was purchased as a white powder from Strem, stored in the dark in a glove box under an atmosphere of N<sub>2</sub>, and discarded upon turning to a brown powder.

(*R,R*)-Walphos W001-1 was purchased from Strem or Acros, stored in a glove box under an atmosphere of N<sub>2</sub>, and used as received. All other ligands were purchased from Strem or Sigma Aldrich and were stored under N<sub>2</sub> atmosphere and used as received.

Saccharin was purchased from Sigma Aldrich and used as received. All Grignard reagents were titrated with iodine prior to use.<sup>30</sup> *n*-Butyllithium and methyllithium solutions were purchased from Acros, stored at 4 °C, and titrated prior to use.<sup>31</sup>

*tert*-Butanol was purchased from Fisher and distilled every two weeks over CaH<sub>2</sub> through a short-path distillation head onto activated 4Å mol sieves.

Allenylboronic acid pinacol ester **4.2** was prepared according to Yoshida and co-workers<sup>32</sup> and distilled every month.

Propargylboronic acid pinacol ester **4.12** was prepared according to Fandrick and co-workers (*vide infra*).<sup>22c</sup> Ethynylmagnesium bromide was purchased from Sigma Aldrich, stored at 4 °C, and used within one week of opening the bottle.

*N,N*-Dimethylformamide-*d*<sub>7</sub> was purchased from Cambridge Isotope Laboratories and used as received.

All other chemicals were purchased commercially and used as received.

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<sup>30</sup> Krasovskiy, A.; Knochel, P. *Synthesis* **2006**, 5, 890.

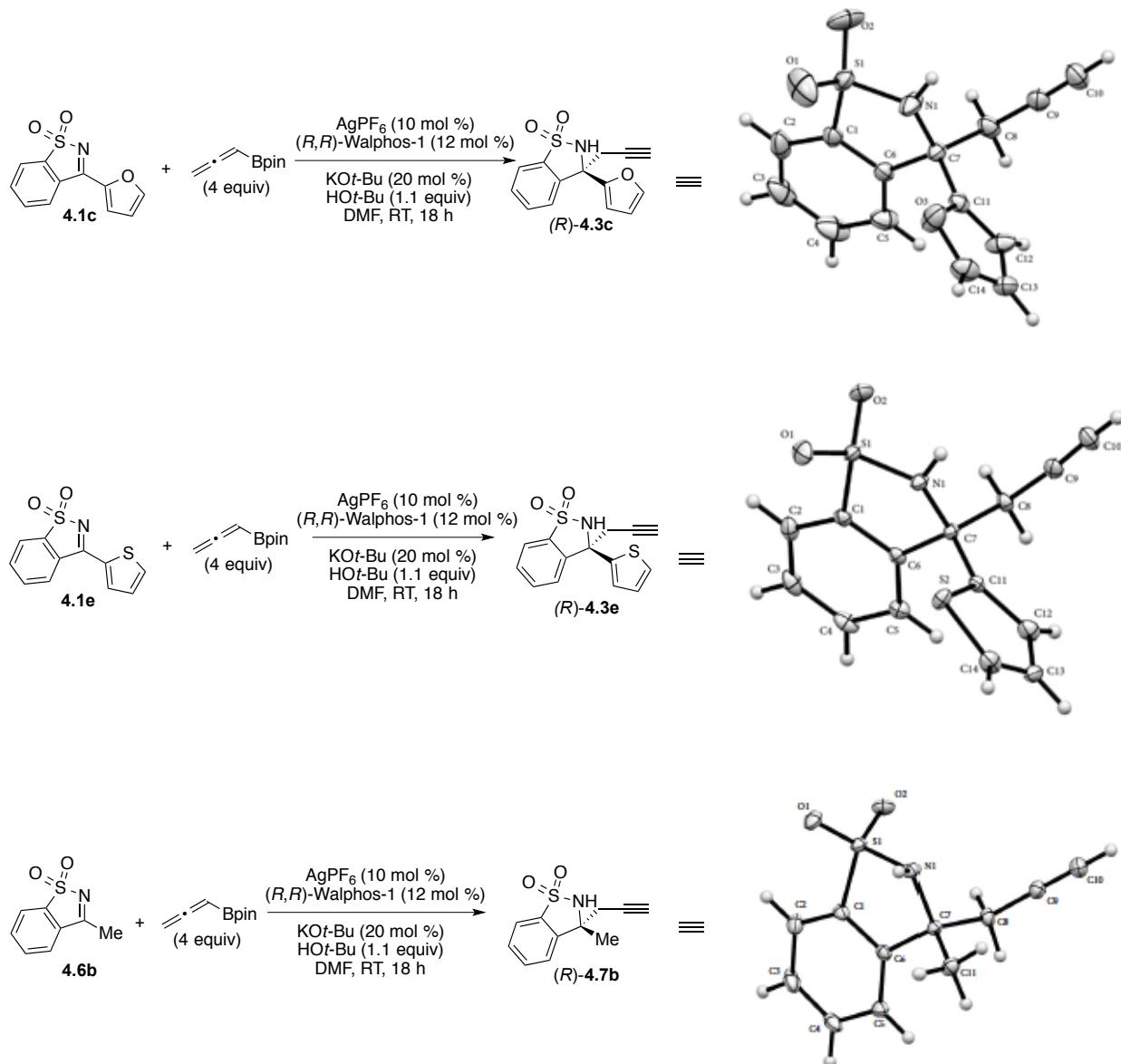
<sup>31</sup> Love, B. E.; Jones, E. G. *J. Org. Chem.* **1999**, 64, 3755.

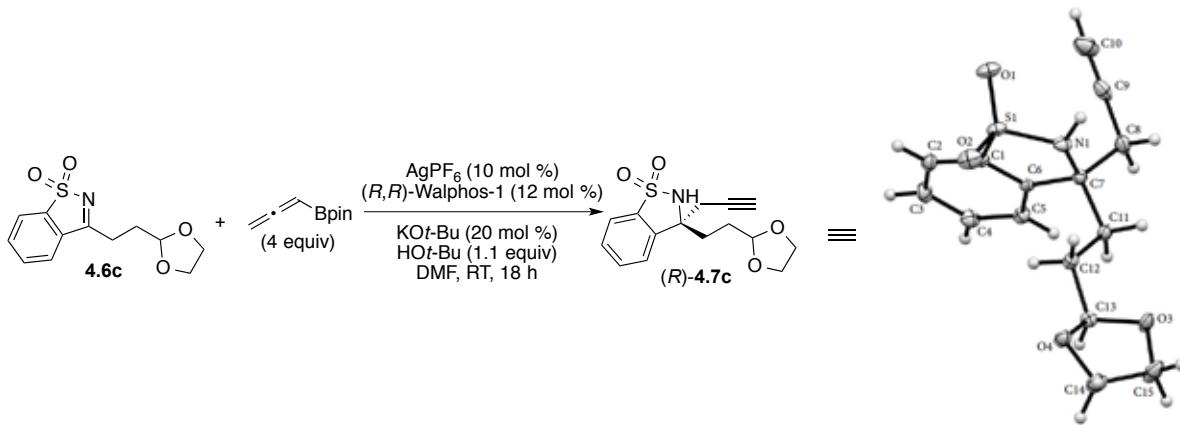
<sup>32</sup> Tonogaki, K.; Itami, K.; Yoshida, J.-i. *J. Am. Chem. Soc.* **2006**, 128, 1464.

## Stereochemical Proofs

The absolute configurations of products **4.3c**, **4.3e**, **4.7b**, and **4.7c** were assigned by X-ray crystallographic analysis (Scheme 4.7). The absolute configurations of all other products were assigned by analogy.

**Scheme 4.7.** Absolute configurations of products determined by X-ray crystallography.

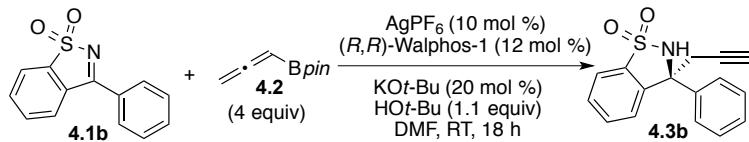




### **Representative Addition Procedures**

#### **Method A: Enantioselective Addition to Ketamines**

**Note:** All manipulations involving silver-catalyzed reactions were performed in the absence of direct light, using vials wrapped in aluminum foil.



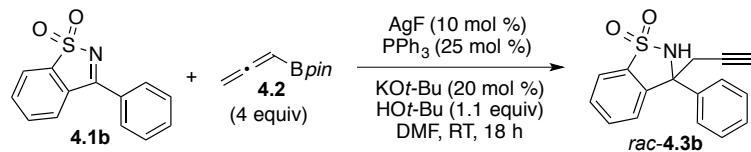
In a glovebox, an oven-dried 1.0 mL conical vial equipped with a triangular stir bar was charged with  $\text{AgPF}_6$  (5.0 mg, 0.020 mmol, 0.10 equiv) and Walphos W001-1 (22.3 mg, 0.0240 mmol, 0.120 equiv). The vial was sealed with a screw-top cap fit with a septum and removed from the glovebox. Anhydrous DMF (400  $\mu\text{L}$ ) was added and the solution was stirred for 5 min at RT. The  $\text{N}_2$  line was then removed and the solution was stirred for 30 min at 70 °C, then cooled to RT over 15 min.

To the catalyst solution was added *tert*-butanol (21  $\mu\text{L}$ , 0.22 mmol, 1.1 equiv), followed by potassium *tert*-butoxide (4.5 mg, 0.040 mmol, 0.20 equiv) and phenyl ketimine **4.1b** (48.6 mg, 0.200 mmol, 1.00 equiv) under a flow of  $\text{N}_2$ . The reaction was stirred at RT for 5 min to dissolve the ketimine. Allenylboronic acid pinacol ester **4.2** (72  $\mu\text{L}$ , 0.40 mmol, 2.0 equiv) was

added via syringe, followed by another portion of allenylboronic acid pincol ester (72  $\mu$ L, 0.40 mmol, 2.0 equiv) added via slow addition over 3 h using a syringe pump. The N<sub>2</sub> line was removed and the reaction was stirred at 22 °C for 18 h. The reaction mixture was filtered through a plug of silica gel eluting with 100% Et<sub>2</sub>O to remove the catalyst. Et<sub>2</sub>O was removed in vacuo and the resulting residue was purified by silica gel chromatography.

#### *Method B: Racemic Standards*

**Note:** All manipulations involving silver-catalyzed reactions were performed in the absence of direct light, using vials wrapped in aluminum foil.



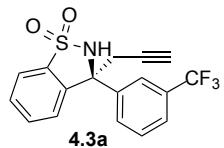
In a glovebox, a flame-dried 7 mL vial equipped with a stir bar was charged with AgF (2.5 mg, 0.020 mmol, 0.10 equiv) and  $\text{PPh}_3$  (13 mg, 0.050 mmol, 0.25 equiv). The vial was sealed with a screw-top cap fit with a septum and removed from the glovebox. Anhydrous DMF (800  $\mu$ L) was added and the solution was stirred for 5 min at RT. The N<sub>2</sub> line was then removed and the solution was stirred for 30 min at 70 °C, then cooled to RT over 15 min.

To the catalyst solution was added *tert*-butanol (21  $\mu$ L, 0.22 mmol, 1.1 equiv), followed by potassium *tert*-butoxide (4.5 mg, 0.040 mmol, 0.20 equiv) and phenyl ketimine **4.1b** (48.6 mg, 0.200 mmol, 1.00 equiv) under a flow of N<sub>2</sub>. The reaction was stirred at RT for 5 min to dissolve the ketimine. Allenylboronic acid pinacol ester **4.2** (72  $\mu$ L, 0.40 mmol, 2.0 equiv) was added via syringe. The N<sub>2</sub> line was removed and the reaction was stirred at 22 °C for 18 h. The reaction mixture was filtered through a plug of silica gel eluting with 100% Et<sub>2</sub>O to remove the

catalyst. Et<sub>2</sub>O was removed in vacuo and the resulting residue was purified by silica gel chromatography.

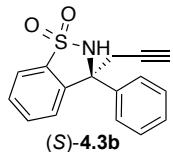
### **Characterization Data for Products**

**Note:** The yield of homoallenyllic sultam is typically less than 5–10% in these reactions and can be separated from the homopropargylic sultam using the column chromatography conditions specified below. The TLC R<sub>f</sub> of the homoallenyllic sultam is generally 0.1 higher than the R<sub>f</sub> of the homopropargylic sultam. The diagnostic peaks for the homoallenyllic sultam are found in the <sup>1</sup>H NMR range of δ 5.96 to 5.48 (t, J = 6.6 Hz, 1H) and δ 5.11 to 5.01 (d, J = 6.6 Hz, 2H).



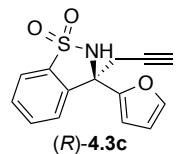
**Sultam 4.3a** was prepared according to Method A, using the following amounts of reagents: AgPF<sub>6</sub> (5.0 mg, 0.020 mmol, 0.10 equiv), Walphos W001-1 (22.3 mg, 0.0240 mmol, 0.120 equiv), *tert*-butanol (21 μL, 0.22 mmol, 1.1 equiv), potassium *tert*-butoxide (4.5 mg, 0.040 mmol, 0.20 equiv), substrate **4.1a** (62.3 mg, 0.200 mmol, 1.00 equiv), DMF (400 μL), and allenylboronic acid pincol ester (144 μL, 0.800 mmol, 4.00 equiv). The resulting mixture was purified by flash column chromatography using 0–1% TEA/benzene to separate product from unreacted starting material. The mixture was purified again by flash column chromatography using 5–10–20% EtOAc/hexanes (1% TEA) to separate product from excess ligand and afford the title compound as a beige solid (49.8 mg, 0.142 mmol, 71%, 97% ee). **TLC R<sub>f</sub>** = 0.1 (20% EtOAc/hexanes, stains pink with PAA); **m.p.** = 125–127 °C; **<sup>1</sup>H NMR** (400 MHz, CDCl<sub>3</sub>) δ 7.85–7.82 (m, 2H), 7.78 (d, J = 7.9 Hz, 1H), 7.70–7.58 (m, 3H), 7.52 (t, J = 7.9 Hz, 1H), 7.37 (d,

*J* = 7.9 Hz, 1H), 5.35 (br s, 1H), 3.32 (dd, *J* = 17.3, 2.7 Hz, 1H), 3.24 (dd, *J* = 17.3, 2.7 Hz, 1H), 2.10 (t, *J* = 2.7 Hz, 1H); **13C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 141.6, 141.5, 135.1, 133.8, 131.4 (q, *J* = 32.4 Hz), 130.4, 130.3, 129.8, 125.6 (q, *J* = 3.7 Hz), 124.6, 123.9 (q, *J* = 272.4 Hz), 123.3 (q, *J* = 4.0 Hz), 121.8, 77.7, 73.9, 66.7, 31.5; **19F NMR** (376.5 MHz, CDCl<sub>3</sub>) δ -62.6; **IR** (neat) 3302, 1329, 1164, 1125, 731 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>17</sub>H<sub>12</sub>F<sub>3</sub>NO<sub>2</sub>S (M + Na)<sup>+</sup> 374.0439, found 374.0433; [α]<sup>26</sup><sub>D</sub> +47 (*c* 1.2, CDCl<sub>3</sub>); **SFC** analysis (Whelk-O (*R,R*), 5% IPA, 3.0 mL/min, 215 nm) indicated 97% ee: t<sub>R</sub> (minor) = 10.8 min, t<sub>R</sub> (major) = 11.3 min.



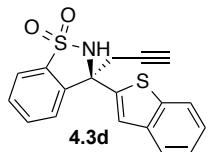
**Sultam (S)-4.3b** was prepared according to Method A, using the following amounts of reagents: AgPF<sub>6</sub> (5.0 mg, 0.020 mmol, 0.10 equiv), Walphos W001-1 (22.3 mg, 0.0240 mmol, 0.120 equiv), *tert*-butanol (21 μL, 0.22 mmol, 1.1 equiv), potassium *tert*-butoxide (4.5 mg, 0.040 mmol, 0.20 equiv), substrate **4.1b** (48.6 mg, 0.200 mmol, 1.00 equiv), DMF (400 μL), and allenylboronic acid pincol ester (144 μL, 0.800 mmol, 4.00 equiv). The resulting mixture was purified by flash column chromatography using 0–1% TEA/benzene to separate product from unreacted starting material. The mixture was purified again by flash column chromatography using 5–10–20% EtOAc/hexanes (1% TEA) to separate product from excess ligand and afford the title compound as a white solid (39.3 mg, 0.139 mmol, 69%, 97% ee). **TLC R<sub>f</sub>** = 0.2 (20% EtOAc/hexanes, stains pink with PAA); **m.p.** = 139–142 °C; **1H NMR** (500 MHz, CDCl<sub>3</sub>) δ 7.80 (ad, *J* = 7.3 Hz, 1H), 7.62–7.53 (m, 4H), 7.41–7.30 (m, 4H), 5.23 (br s, 1H), 3.30 (dd, *J* = 17.2, 2.6 Hz, 1H), 3.24 (dd, *J* = 17.2, 2.6 Hz, 1H), 2.06 (t, *J* = 2.6 Hz, 1H); **13C NMR** (125 MHz, CDCl<sub>3</sub>) δ 142.3, 140.4, 135.0, 133.5, 130.0, 129.1, 128.7, 126.6, 125.0, 121.5, 78.5, 73.3, 67.1,

31.3; **IR** (neat) 3286, 2923, 1713, 1293, 1165  $\text{cm}^{-1}$ ; **HRMS** (TOF MS ES+)  $m / z$  calcd for  $\text{C}_{16}\text{H}_{13}\text{NO}_2\text{S}$  ( $\text{M} + \text{Na}$ ) $^+$  306.0565, found 306.0564;  $[\alpha]^{24}_{\text{D}} +42$  ( $c$  0.7,  $\text{CHCl}_3$ ); **SFC** analysis (OD-H, 10% IPA, 3.0 mL/min, 215 nm) indicated 97% ee:  $t_{\text{R}}$  (minor) = 11.5 min,  $t_{\text{R}}$  (major) = 14.3 min.

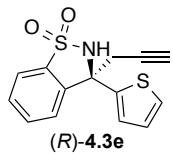


**Sultam (R)-4.3c** was prepared according to Method A, using the following amounts of reagents:  $\text{AgPF}_6$  (5.0 mg, 0.020 mmol, 0.10 equiv), Walphos W001-1 (22.3 mg, 0.0240 mmol, 0.120 equiv), *tert*-butanol (21  $\mu\text{L}$ , 0.22 mmol, 1.1 equiv), potassium *tert*-butoxide (4.5 mg, 0.040 mmol, 0.20 equiv), substrate **4.1c** (46.6 mg, 0.200 mmol, 1.00 equiv), DMF (400  $\mu\text{L}$ ), and allenylboronic acid pincol ester (144  $\mu\text{L}$ , 0.800 mmol, 4.00 equiv). The resulting mixture was purified by flash column chromatography using 0–1% TEA/benzene to separate product from unreacted starting material. The mixture was purified again by flash column chromatography using 5–20–40% EtOAc/hexanes (1% TEA) to separate product from excess ligand and afford the title compound as a white solid (35.4 mg, 0.130 mmol, 65%, 94% ee). **TLC**  $\mathbf{R}_f$  = 0.3 (20% EtOAc/hexanes, stains purple with PAA); **m.p.** = 126–127  $^{\circ}\text{C}$ ;  **$^1\text{H NMR}$**  (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.80 (d,  $J$  = 7.6 Hz, 1H), 7.66 (td,  $J$  = 7.6, 1.1 Hz, 1H), 7.60–7.58 (m, 2H), 7.40 (d,  $J$  = 1.1 Hz, 1H), 6.48 (d,  $J$  = 3.3 Hz, 1H), 6.35 (dd,  $J$  = 3.3, 1.8 Hz, 1H), 5.29 (br s, 1H), 3.25 (dd,  $J$  = 17.0, 2.6 Hz, 1H), 3.19 (dd,  $J$  = 17.0, 2.6 Hz, 1H), 2.07 (t,  $J$  = 2.6 Hz, 1H);  **$^{13}\text{C NMR}$**  (125.7 MHz,  $\text{CDCl}_3$ )  $\delta$  152.1, 143.6, 139.7, 135.2, 133.5, 130.4, 124.8, 121.5, 110.8, 108.8, 77.8, 73.1, 63.3, 30.6; **IR** (neat) 3283, 1295, 1166, 1132, 735  $\text{cm}^{-1}$ ; **HRMS** (TOF MS ES+)  $m / z$  calcd for  $\text{C}_{14}\text{H}_{11}\text{NO}_3\text{S}$  ( $\text{M} + \text{Na}$ ) $^+$  296.0357, found 296.0363;  $[\alpha]^{26}_{\text{D}} +7.0$  ( $c$  1.0,  $\text{CDCl}_3$ ); **SFC** analysis

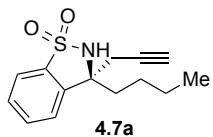
(OD-H, 10% IPA, 3.0 mL/min, 215 nm) indicated 94% ee:  $t_R$  (minor) = 7.2 min,  $t_R$  (major) = 8.6 min.



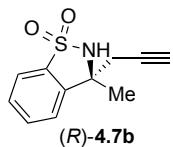
**Sultam 4.3d** was prepared according to Method A, using the following amounts of reagents: AgPF<sub>6</sub> (5.0 mg, 0.020 mmol, 0.10 equiv), Walphos W001-1 (22.3 mg, 0.0240 mmol, 0.120 equiv), *tert*-butanol (21  $\mu$ L, 0.22 mmol, 1.1 equiv), potassium *tert*-butoxide (4.5 mg, 0.040 mmol, 0.20 equiv), substrate **4.1d** (59.9 mg, 0.200 mmol, 1.00 equiv), DMF (400  $\mu$ L), and allenylboronic acid pincol ester (144  $\mu$ L, 0.800 mmol, 4.00 equiv). The resulting mixture was purified by flash column chromatography using 0–1% TEA/benzene to separate product from unreacted starting material. The mixture was purified again by flash column chromatography using 5–10–20% EtOAc/hexanes (1% TEA) to separate product from excess ligand and afford the title compound as a pale yellow oil (46.7 mg, 0.138 mmol, 69%, 91% ee). **TLC R<sub>f</sub>** = 0.3 (20% EtOAc/hexanes, stains purple with PAA); **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.83 (d,  $J$  = 7.8 Hz, 1H), 7.77–7.73 (m, 2H), 7.67 (td,  $J$  = 7.6, 1.1 Hz, 1H), 7.62–7.56 (m, 2H), 7.45 (s, 1H), 7.34 (quintd,  $J$  = 7.5, 1.4 Hz, 2H), 5.44 (s, 1H), 3.36 (dd,  $J$  = 17.1, 2.7 Hz, 1H), 3.29 (dd,  $J$  = 17.1, 2.7 Hz, 1H), 2.12 (t,  $J$  = 2.7 Hz, 1H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>)  $\delta$  145.5, 141.0, 139.9, 139.3, 134.9, 133.7, 130.5, 125.3, 124.9, 124.8, 124.3, 123.0, 122.5, 121.7, 77.8, 73.8, 65.4, 32.8; **IR** (neat) 3288, 1294, 1166, 1131, 726 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>18</sub>H<sub>13</sub>NO<sub>2</sub>S<sub>2</sub> (M + Na)<sup>+</sup> 362.0285, found 362.0279; **[ $\alpha$ ]<sup>28</sup><sub>D</sub>** +17 (*c* 1.1, CDCl<sub>3</sub>); **SFC** analysis (AS-H, 20% IPA, 3.0 mL/min, 215 nm) indicated 91% ee:  $t_R$  (major) = 11.0 min,  $t_R$  (minor) = 16.5 min.



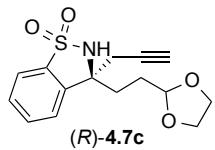
**Sultam (R)-4.3e** was prepared according to Method A, using the following amounts of reagents: AgPF<sub>6</sub> (5.0 mg, 0.020 mmol, 0.10 equiv), Walphos W001-1 (22.3 mg, 0.0240 mmol, 0.120 equiv), *tert*-butanol (21 μL, 0.22 mmol, 1.1 equiv), potassium *tert*-butoxide (4.5 mg, 0.040 mmol, 0.20 equiv), substrate **4.1e** (49.9 mg, 0.200 mmol, 1.00 equiv), DMF (400 μL), and allenylboronic acid pincol ester (144 μL, 0.800 mmol, 4.00 equiv). The resulting mixture was purified by flash column chromatography using 0–1% TEA/benzene to separate product from unreacted starting material. The mixture was purified again by flash column chromatography using 5–10–20% EtOAc/hexanes (1% TEA) to separate product from excess ligand and afford the title compound as a yellow solid (51.3 mg, 0.177 mmol, 89%, 94% ee). **TLC** **R<sub>f</sub>** = 0.3 (20% EtOAc/hexanes, stains purple with PAA); **m.p.** = 155–157 °C; **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>) δ 7.80 (d, *J* = 7.7 Hz, 1H), 7.66 (td, *J* = 7.6, 1.3 Hz, 1H), 7.59 (td, *J* = 7.5, 1.3 Hz, 1H), 7.54 (d, *J* = 7.8 Hz, 1H), 7.30 (dd, *J* = 5.1, 1.3 Hz, 1H), 7.18 (dd, *J* = 3.7, 1.3 Hz, 1H), 6.98 (dd, *J* = 5.1, 3.7 Hz, 1H), 5.34 (br s, 1H), 3.27 (d, *J* = 2.6 Hz, 2H), 2.09 (t, *J* = 2.6 Hz, 1H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 145.1, 141.6, 135.0, 133.6, 130.3, 127.4, 126.6, 126.2, 124.8, 121.5, 78.0, 73.5, 65.0, 33.2; **IR** (neat) 3305, 2925, 1302, 1168, 906, 728 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>14</sub>H<sub>11</sub>NO<sub>2</sub>S<sub>2</sub> (M + Na)<sup>+</sup> 312.0129, found 312.0142; [α]<sup>28</sup><sub>D</sub> -12 (*c* 0.9, CDCl<sub>3</sub>); **SFC** analysis (OD-H, 10% IPA, 3.0 mL/min, 215 nm) indicated 94% ee: t<sub>R</sub> (minor) = 12.5 min, t<sub>R</sub> (major) = 15.7 min.



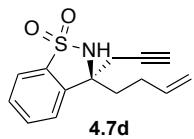
**Sultam 4.7a** was prepared according to Method A, using the following amounts of reagents: AgPF<sub>6</sub> (5.0 mg, 0.020 mmol, 0.10 equiv), Walphos W001-1 (22.3 mg, 0.0240 mmol, 0.120 equiv), *tert*-butanol (21  $\mu$ L, 0.22 mmol, 1.1 equiv), potassium *tert*-butoxide (4.5 mg, 0.040 mmol, 0.20 equiv), substrate **4.6a** (44.7 mg, 0.200 mmol, 1.00 equiv), DMF (400  $\mu$ L), and allenylboronic acid pincol ester (144  $\mu$ L, 0.800 mmol, 4.00 equiv). The resulting mixture was purified by flash column chromatography using 0–1% TEA/benzene to separate product from unreacted starting material. The mixture was purified again by flash column chromatography using 5–10% EtOAc/hexanes (1% TEA) to separate product from excess ligand and afford the title compound as a colorless oil (44.3 mg, 0.168 mmol, 84%, 98% ee). **TLC** R<sub>f</sub> = 0.6 (20% EtOAc/hexanes, stains pink with PAA); **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.77 (d, *J* = 7.8 Hz, 1H), 7.65 (t, *J* = 7.6, 1.0 Hz, 1H), 7.56 (t, *J* = 7.7, 0.9 Hz, 1H), 7.49 (d, *J* = 7.8 Hz, 1H), 4.87 (s, 1H), 2.82 (dd, *J* = 16.9, 2.7 Hz, 1H), 2.77 (dd, *J* = 16.9, 2.7 Hz, 1H), 2.13 (t, *J* = 2.7 Hz, 1H), 2.14–2.08 (m, 1H), 2.02–1.96 (m, 1H), 1.45–1.35 (m, 1H), 1.34–1.24 (m, 2H), 1.04–0.96 (m, 1H), 0.86 (t, *J* = 7.3 Hz, 3H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>)  $\delta$  141.9, 135.8, 133.4, 129.8, 123.6, 121.6, 78.8, 72.8, 65.2, 38.5, 31.8, 25.8, 22.7, 14.0; **IR** (neat) 3306, 2959, 1289, 1168, 907, 728 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>14</sub>H<sub>17</sub>NO<sub>2</sub>S (M + Na)<sup>+</sup> 286.0878, found 286.0884; [α]<sup>28</sup><sub>D</sub> -2.4 (*c* 1.1, CDCl<sub>3</sub>); **SFC** analysis (AS-H, 10% IPA, 3.0 mL/min, 215 nm) indicated 98% ee: t<sub>R</sub> (major) = 7.5 min, t<sub>R</sub> (minor) = 8.3 min.



**Sultam (R)-4.7b** was prepared according to Method A, using the following amounts of reagents: AgPF<sub>6</sub> (5.0 mg, 0.020 mmol, 0.10 equiv), Walphos W001-1 (22.3 mg, 0.0240 mmol, 0.120 equiv), *tert*-butanol (21  $\mu$ L, 0.22 mmol, 1.1 equiv), potassium *tert*-butoxide (4.5 mg, 0.040 mmol, 0.20 equiv), substrate **4.6b** (36.2 mg, 0.200 mmol, 1.00 equiv), DMF (400  $\mu$ L), and allenylboronic acid pincol ester (144  $\mu$ L, 0.800 mmol, 4.00 equiv). The resulting mixture was purified by flash column chromatography using 0–1% TEA/benzene to separate product from unreacted starting material. The mixture was purified again by flash column chromatography using 5–10–20% EtOAc/hexanes (1% TEA) to separate product from excess ligand and afford the title compound as a white solid (35.7 mg, 0.161 mmol, 80%, 98% ee). **TLC R<sub>f</sub>** = 0.3 (20% EtOAc/hexanes, stains pink with PAA); **m.p.** = 91–93 °C; **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.77 (d, *J* = 7.8 Hz, 1H), 7.65 (td, *J* = 7.8, 1.1 Hz, 1H), 7.56 (td, *J* = 7.7, 1.0 Hz, 1H), 7.51 (d, *J* = 7.8 Hz, 1H), 4.93 (s, 1H), 2.82 (dd, *J* = 17.0, 2.7 Hz, 1H), 2.77 (dd, *J* = 17.0, 2.7 Hz, 1H), 2.15 (t, *J* = 2.7 Hz, 1H), 1.76 (s, 3H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>)  $\delta$  143.3, 135.6, 133.5, 129.8, 123.4, 121.5, 78.8, 72.7, 62.0, 32.6, 26.9; **IR** (neat) 3274, 2980, 2342, 1281, 1156, 1132 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>11</sub>H<sub>11</sub>NO<sub>2</sub>S (M + Na)<sup>+</sup> 244.0408, found 244.0410; **[ $\alpha$ ]<sup>27</sup>D** +16 (*c* 0.9, CDCl<sub>3</sub>); **SFC** analysis (OD-H, 10% IPA, 3.0 mL/min, 215 nm) indicated 98% ee: t<sub>R</sub> (minor) = 5.8 min, t<sub>R</sub> (major) = 6.3 min.



**Sultam (R)-4.7c** was prepared according to Method A, using the following amounts of reagents: AgPF<sub>6</sub> (5.0 mg, 0.020 mmol, 0.10 equiv), Walphos W001-1 (22.3 mg, 0.0240 mmol, 0.120 equiv), *tert*-butanol (21  $\mu$ L, 0.22 mmol, 1.1 equiv), potassium *tert*-butoxide (4.5 mg, 0.040 mmol, 0.20 equiv), substrate **4.6c** (53.5 mg, 0.200 mmol, 1.00 equiv), DMF (400  $\mu$ L), and allenylboronic acid pincol ester (144  $\mu$ L, 0.800 mmol, 4.00 equiv). The resulting mixture was purified by flash column chromatography using 0–1% TEA/benzene to separate product from unreacted starting material. The mixture was purified again by flash column chromatography using 10–20–30% EtOAc/hexanes (1% TEA) to separate product from excess ligand and afford the title compound as a white solid (43.9 mg, 0.143 mmol, 72%, 97% ee). **TLC R<sub>f</sub>** = 0.2 (20% EtOAc/hexanes, stains yellow then pink with PAA); **m.p.** = 128–129 °C; **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.77 (d, *J* = 7.7 Hz, 1H), 7.64 (t, *J* = 7.7 Hz, 1H), 7.57–7.53 (m, 2H), 5.86 (s, 1H), 4.87 (t, *J* = 3.8 Hz, 1H), 4.03–3.95 (m, 2H), 3.91–3.83 (m, 2H), 2.83 (dd, *J* = 17.0, 2.7 Hz, 1H), 2.77 (dd, *J* = 17.0, 2.7 Hz, 1H), 2.33–2.20 (m, 2H), 2.14 (t, *J* = 2.7 Hz, 1H), 1.70–1.63 (m, 1H), 1.62–1.56 (m, 1H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>)  $\delta$  141.7, 136.4, 133.3, 129.9, 123.8, 121.5, 103.2, 79.0, 72.7, 65.3, 65.2, 64.6, 32.4, 31.1, 27.5; **IR** (neat) 3269, 2890, 1286, 1164, 1131, 729 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>15</sub>H<sub>17</sub>NO<sub>4</sub>S (M + Na)<sup>+</sup> 330.0776, found 330.0781; [α]<sup>27</sup><sub>D</sub> -26 (*c* 1.2, CDCl<sub>3</sub>); **SFC** analysis (AS-H, 10% IPA, 3.0 mL/min, 215 nm) indicated 97% ee: t<sub>R</sub> (major) = 12.4 min, t<sub>R</sub> (minor) = 13.8 min.

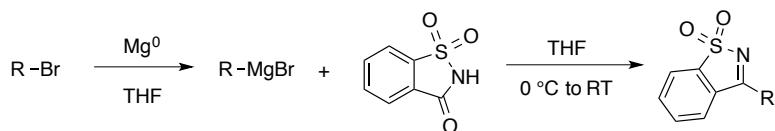


**Sultam 4.7d** was prepared according to Method A, using the following amounts of reagents: AgPF<sub>6</sub> (5.0 mg, 0.020 mmol, 0.10 equiv), Walphos W001-1 (22.3 mg, 0.0240 mmol, 0.120 equiv), *tert*-butanol (21 μL, 0.22 mmol, 1.1 equiv), potassium *tert*-butoxide (4.5 mg, 0.040 mmol, 0.20 equiv), substrate **4.6d** (44.3 mg, 0.200 mmol, 1.00 equiv), DMF (400 μL), and allenylboronic acid pincol ester (144 μL, 0.800 mmol, 4.00 equiv). The resulting mixture was purified by flash column chromatography using 0–1% TEA/benzene to separate product from unreacted starting material. The mixture was purified again by flash column chromatography using 5–10% EtOAc/hexanes (1% TEA) to separate product from excess ligand and afford the title compound as a white solid (46.2 mg, 0.177 mmol, 88%, 97% ee). **TLC** R<sub>f</sub> = 0.4 (20% EtOAc/hexanes, stains pink with PAA); **m.p.** = 80–81 °C; **<sup>1</sup>H NMR** (400 MHz, CDCl<sub>3</sub>) δ 7.78 (d, *J* = 7.8 Hz, 1H), 7.66 (td, *J* = 7.6, 1.3 Hz, 1H), 7.57 (td, *J* = 7.6, 1.0 Hz, 1H), 7.50 (d, *J* = 7.8 Hz, 1H), 5.79–5.69 (m, 1H), 5.02–4.93 (m, 3H), 2.86–2.76 (m, 2H), 2.27–2.16 (m, 2H), 2.15 (t, *J* = 2.6 Hz, 1H), 2.13–2.05 (m, 1H), 1.87–1.76 (m, 1H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 141.4, 137.0, 135.8, 133.4, 129.9, 123.6, 121.6, 115.7, 78.6, 72.9, 65.0, 37.8, 32.0, 28.0; **IR** (neat) 3273, 1285, 1165, 1131, 730 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>14</sub>H<sub>15</sub>NO<sub>2</sub>S (M + Na)<sup>+</sup> 284.0721, found 284.0728; **[α]<sub>D</sub><sup>24</sup>** +0.5 (*c* 1.3, CDCl<sub>3</sub>); **SFC** analysis (AS-H, 10% IPA, 3.0 mL/min, 215 nm) indicated 97% ee: t<sub>R</sub> (major) = 7.5 min, t<sub>R</sub> (minor) = 8.6 min.

### ***General Procedures for Starting Material Synthesis***

**Note:** When possible, Method D was preferentially used instead of Method C in order to minimize side reactions and obtain reaction mixtures that were easier to purify. The yields for starting material synthesis are unoptimized.

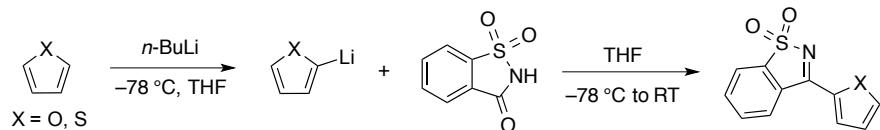
#### *Method C: Grignard Addition into Saccharin*



Prepared according to a modified procedure described by Hayashi and co-workers.<sup>6c</sup> The Grignard reagent was typically prepared using flame-dried magnesium turnings (2.0 equiv) with a few crystals of I<sub>2</sub> in anhydrous THF (10 mL). The aryl halide (1.0 equiv) was added to the solution until initiation of the Grignard reagent, after which the remaining aryl halide was added dropwise at 0 °C. The reaction was stirred 2 h at RT, then titrated.<sup>30</sup>

The Grignard reagent (2.0 equiv) was then slowly added to a solution of saccharin (1.0 equiv) in THF (6 mL) at 0 °C. The reaction was allowed to warm to RT and stirred at 22 °C overnight. The reaction was quenched at 0 °C with saturated aqueous NH<sub>4</sub>Cl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic layers were washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated in vacuo. The product was purified either by recrystallization (if the unpurified material was already crystalline) or by flash column chromatography using silica gel (generally, the unpurified material was first adsorbed onto silica).

*Method D: Organolithium Addition into Saccharin*

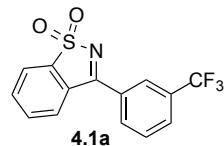


Prepared according to a modified procedure described by Bode and co-workers.<sup>6b</sup> The organolithium reagent was typically prepared by slow addition *n*-butyllithium (2.75 equiv) to a solution of heterocycle (2.5 equiv) in anhydrous THF (10 mL) at -78 °C. The reaction was stirred at -78 °C for 15 min then used directly in the next step.

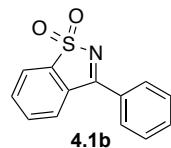
To this mixture was slowly added a solution of saccharin (1.0 equiv) in THF (6 mL) via syringe at -78 °C. The reaction was allowed to warm to RT slowly over several hours, then stirred at 22 °C overnight. The reaction was quenched at 0 °C with saturated aqueous NH<sub>4</sub>Cl (20 mL) and extracted with EtOAc (3 x 20 mL). The combined organic layers were washed with brine, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated in vacuo. The product was purified either by recrystallization (if the unpurified material was already crystalline) or by flash column chromatography using silica gel (generally, the unpurified material was first adsorbed onto silica).

**Note:** In the case of compounds **4.6a** or **4.6b**, *n*-butyllithium or methyl lithium (2.2 equiv) was slowly added directly to a solution of saccharin (1.0 equiv) in THF at -78 °C.

### **Characterization Data for Starting Materials**

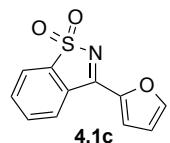


**N-Sulfonyl ketimine 4.1a** was prepared according Method C, using the following amounts of reagents: saccharin (0.46 g, 2.5 mmol, 1.0 equiv), (3-(trifluoromethyl)phenyl)magnesium iodide (4.00 mL, 5.00 mmol, 1.25 M in THF, 2.00 equiv) and THF (2.5 mL). The product was recrystallized from hot Et<sub>2</sub>O in CHCl<sub>3</sub> (1:1 Et<sub>2</sub>O/CHCl<sub>3</sub>) to afford the title compound as a pale yellow solid (0.21 g, 0.68 mmol, 27%). **TLC** R<sub>f</sub> = 0.4 (20% EtOAc/hexanes, UV active); **m.p.** = 154–156 °C; **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>) δ 8.22 (s, 1H), 8.16 (d, *J* = 7.8 Hz, 1H), 8.04 (d, *J* = 7.8 Hz, 1H), 7.96 (d, *J* = 8.0 Hz, 1H), 7.84–7.81 (m, 2H), 7.80–7.76 (m, 2H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 169.9, 141.3, 134.1, 133.9, 132.7, 132.2 (q, *J* = 33.3 Hz), 131.5, 130.14, 130.09, 129.9 (q, *J* = 3.7 Hz), 126.4 (q, *J* = 3.7 Hz), 126.3, 123.55, 123.54 (q, *J* = 272.8 Hz); **<sup>19</sup>F NMR** (376.5 MHz, CDCl<sub>3</sub>) δ –62.9; **IR** (neat) 1614, 1325, 1281, 1166, 1123 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>14</sub>H<sub>8</sub>F<sub>3</sub>NO<sub>2</sub>S (M + Na)<sup>+</sup> 334.0125, found 334.0131.

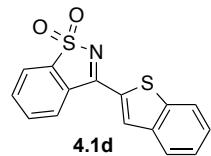


**N-Sulfonyl ketimine 4.1b** was prepared according Method C, using the following amounts of reagents: saccharin (1.10 g, 6.00 mmol, 1.00 equiv), phenylmagnesium bromide (6.0 mL, 12 mmol, 2.0 M in THF, 2.0 equiv) and THF (25 mL). The product was recrystallized from hot Et<sub>2</sub>O in CHCl<sub>3</sub> (1:1 Et<sub>2</sub>O/CHCl<sub>3</sub>) to afford the title compound as a white solid (0.715 g, 2.94

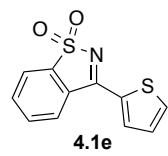
mmol, 49% yield). Analytical data are consistent with literature values.<sup>6c</sup> **TLC**  $R_f$  = 0.2 (20% EtOAc/hexanes, UV active); **m.p.** = 163–165 °C; **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>) δ 8.03 (d, *J* = 7.5 Hz, 1H), 7.98 (d, *J* = 7.3 Hz, 2H), 7.91 (d, *J* = 7.3 Hz, 1H), 7.80 (t, *J* = 7.3 Hz, 1H), 7.75 (t, *J* = 7.3 Hz, 1H), 7.71 (t, *J* = 7.5 Hz, 1H), 7.62 (t, *J* = 7.6 Hz, 2H); **<sup>13</sup>C NMR** (125 MHz, CDCl<sub>3</sub>) δ 171.2, 141.3, 133.8, 133.6, 133.5, 130.7, 130.6, 129.7, 129.4, 126.7, 123.3; **IR** (neat) 1599, 1531, 1332, 1171 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>13</sub>H<sub>9</sub>NO<sub>2</sub>S (M + Na)<sup>+</sup> 266.0252, found 266.0255.



**N-Sulfonyl ketimine 4.1c** was prepared according Method D, using the following amounts of reagents: furan (0.910 mL, 12.5 mmol, 2.50 equiv), *n*-BuLi (5.50 mL, 13.8 mmol, 2.50 M in hexanes, 2.75 equiv), saccharin (0.92 g, 5.0 mmol, 1.0 equiv), and THF (15 mL). The product was recrystallized from hot Et<sub>2</sub>O in CHCl<sub>3</sub> (1:1 Et<sub>2</sub>O/CHCl<sub>3</sub>) to afford the title compound as a yellow solid (0.16 g, 0.70 mmol, 14%). Analytical data are consistent with literature values.<sup>6c</sup> **TLC**  $R_f$  = 0.2 (20% EtOAc/hexanes, UV active); **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>) δ 8.49 (dd, *J* = 5.5, 3.0 Hz, 1H), 7.98 (dd, *J* = 5.5, 3.0 Hz, 1H), 7.91 (s, 1H), 7.79–7.76 (m, 3H), 6.79 (dd, *J* = 3.7, 1.6 Hz, 1H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 158.1, 149.4, 148.2, 141.0, 133.9, 133.6, 129.8, 127.6, 122.8, 122.2, 114.1; **IR** (neat) 1600, 1571, 1516, 1320, 1165 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>11</sub>H<sub>7</sub>NO<sub>3</sub>S (M + Na)<sup>+</sup> 256.0044, found 256.0043.

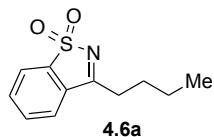


**N-Sulfonyl ketimine 4.1d** was prepared according Method D, using the following amounts of reagents: benzothiophene (1.46 mL, 12.5 mmol, 2.50 equiv), *n*-BuLi (5.50 mL, 13.8 mmol, 2.50 M in hexanes, 2.75 equiv), saccharin (0.92 g, 5.0 mmol, 1.0 equiv), and THF (16 mL). The product was purified by flash column chromatography using 20–30–50% EtOAc/hexanes to afford the title compound as a yellow solid (0.29 g, 0.96 mmol, 19%). **TLC**  $R_f$  = 0.2 (20% EtOAc/hexanes, UV active); **m.p.** = 266–268 °C; **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>) δ 8.46 (s, 1H), 8.30–8.28 (m, 1H), 8.05–8.03 (m, 1H), 7.99 (d, *J* = 8.1 Hz, 1H), 7.96 (d, *J* = 8.2 Hz, 1H), 7.83–7.81 (m, 2H), 7.56 (t, *J* = 7.6 Hz, 1H), 7.50 (t, *J* = 7.6 Hz, 1H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 164.0, 142.9, 141.4, 139.2, 133.9, 133.7, 132.7, 130.4, 128.6, 126.1, 126.0, 125.8, 123.3, 122.9; **IR** (neat) 1593, 1526, 1316, 1171, 744 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>15</sub>H<sub>9</sub>NO<sub>2</sub>S<sub>2</sub> (M + Na)<sup>+</sup> 321.9973, found 321.9986.



**N-Sulfonyl ketimine 4.1e** was prepared according Method C, using the following amounts of reagents: saccharin (0.92 g, 5.0 mmol, 1.0 equiv), 2-thienylmagnesium bromide (8.00 mL, 10.0 mmol, 1.25 M in THF, 2.00 equiv) and THF (5 mL). The product was recrystallized from hot Et<sub>2</sub>O in CHCl<sub>3</sub> (1:1 Et<sub>2</sub>O/CHCl<sub>3</sub>) to afford the title compound as an orange solid (0.22 g, 0.90 mmol, 18%). Analytical data are consistent with literature values.<sup>6b</sup> **TLC**  $R_f$  = 0.2 (20% EtOAc/hexanes, UV active); **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>) δ 8.23 (d, *J* = 3.9 Hz, 1H), 8.21–8.18

(m, 1H), 8.03–8.00 (m, 1H), 7.88 (d,  $J$  = 5.0 Hz, 1H), 7.82–7.77 (m, 2H), 7.34 (t,  $J$  = 4.4 Hz, 1H);  $^{13}\text{C}$  NMR (125.7 MHz,  $\text{CDCl}_3$ )  $\delta$  163.1, 141.4, 135.8, 135.3, 134.0, 133.8, 133.6, 130.5, 129.2, 125.8, 123.2; IR (neat) 1594, 1416, 1316, 1164, 725  $\text{cm}^{-1}$ ; HRMS (TOF MS ES+)  $m/z$  calcd for  $\text{C}_{11}\text{H}_7\text{NO}_2\text{S}_2$  ( $\text{M} + \text{Na}$ ) $^+$  271.9816, found 271.9816.

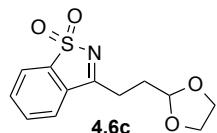


**N-Sulfonyl ketimine 4.6a** was prepared according Method D, using the following amounts of reagents: saccharin (0.92 g, 5.0 mmol, 1.0 equiv), *n*-BuLi (5.50 mL, 13.8 mmol, 2.50 M in hexanes, 2.75 equiv), and THF (18 mL). The product was purified by flash column chromatography using 20% EtOAc/hexanes to afford the title compound as a yellow solid (0.40 g, 1.8 mmol, 36%). Analytical data are consistent with literature values.<sup>6b</sup> TLC  $\mathbf{R}_f$  = 0.4 (20% EtOAc/hexanes, stains with KMnO<sub>4</sub>);  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.93–7.90 (m, 1H), 7.77–7.68 (m, 3H), 2.97 (t,  $J$  = 7.4 Hz, 2H), 1.88 (quint,  $J$  = 7.4 Hz, 2H), 1.51 (sext,  $J$  = 7.4 Hz, 2H), 0.99 (t,  $J$  = 7.4 Hz, 3H);  $^{13}\text{C}$  NMR (125.7 MHz,  $\text{CDCl}_3$ )  $\delta$  176.5, 140.0, 134.0, 133.6, 131.5, 124.0, 122.6, 31.0, 27.6, 22.5, 13.9; IR (neat) 2342, 1604, 1558, 1332, 1172  $\text{cm}^{-1}$ ; HRMS (TOF MS ES+)  $m/z$  calcd for  $\text{C}_{11}\text{H}_{13}\text{NO}_2\text{S}$  ( $\text{M} + \text{Na}$ ) $^+$  246.0565, found 246.0568.

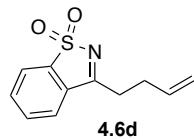


**N-Sulfonyl ketimine 4.6b** was prepared according Method D, using the following amounts of reagents: saccharin (1.83 g, 10.0 mmol, 1.00 equiv), MeLi (15.7 mL, 22.0 mmol, 1.40 M in  $\text{Et}_2\text{O}$ , 2.20 equiv) and THF (10 mL). The product was purified by flash column chromatography using

20% EtOAc/hexanes to afford the title compound as a white solid (1.13 g, 6.21 mmol, 62%). Analytical data are consistent with literature values.<sup>6b</sup> **TLC**  $R_f$  = 0.3 (20% EtOAc/hexanes, stains with KMnO<sub>4</sub>); **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>) δ 7.99–7.97 (m, 1H), 7.84–7.79 (m, 2H), 7.76–7.75 (m, 1H), 2.73 (s, 3H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 173.4, 139.7, 134.1, 133.7, 131.7, 124.3, 122.5, 17.7; **IR** (neat) 2341, 1558, 1316, 1168, 771 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>8</sub>H<sub>7</sub>NO<sub>2</sub>S (M + Na)<sup>+</sup> 204.0095, found 204.0098.



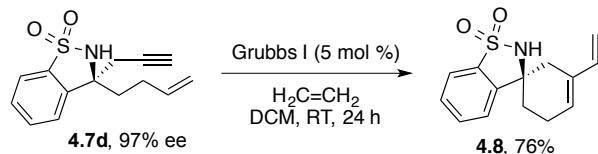
**N-Sulfonyl ketimine 4.6c** was prepared according Method C, using the following amounts of reagents: saccharin (1.2 g, 6.5 mmol, 1.0 equiv), (2-(1,3-dioxolan-2-yl)ethyl)magnesium bromide (14.0 mL, 13.0 mmol, 0.900 M in THF, 2.00 equiv) and THF (17 mL). The product was purified by flash column chromatography using 50% EtOAc/hexanes to afford the title compound as a white solid (0.80 g, 3.0 mmol, 46%). **TLC**  $R_f$  = 0.5 (50% EtOAc/hexanes, UV active); **m.p.** = 68–70 °C; **<sup>1</sup>H NMR** (400 MHz, CDCl<sub>3</sub>) δ 7.93–7.90 (m, 1H), 7.77–7.71 (m, 3H), 5.07 (t, *J* = 4.0 Hz, 1H), 4.01–3.95 (m, 2H), 3.93–3.87 (m, 2H), 3.12 (t, *J* = 7.4 Hz, 2H), 2.31 (td, *J* = 7.4, 4.0 Hz, 2H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 176.2, 139.8, 134.0, 133.6, 131.3, 124.1, 122.5, 102.6, 65.2, 29.0, 25.1; **IR** (neat) 1607, 1333, 1172, 909, 726 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>12</sub>H<sub>13</sub>NO<sub>4</sub>S (M + Na)<sup>+</sup> 290.0463, found 290.0465.



**N-Sulfonyl ketimine 4.6d** was prepared according Method C, using the following amounts of reagents: saccharin (0.82 g, 4.5 mmol, 1.0 equiv), 4-butenylmagnesium bromide (5.0 mL, 4.5 mmol, 0.90 M in THF, 1.0 equiv) and THF (6 mL). The product was purified by flash column chromatography using 10–20% EtOAc/hexanes to afford the title compound as a white solid (0.16 g, 0.71 mmol, 16%). Analytical data are consistent with literature values.<sup>33</sup> **TLC**  $R_f$  = 0.3 (20% EtOAc/hexanes, stains with KMnO<sub>4</sub>); **m.p.** = 81 °C; **<sup>1</sup>H NMR** (400 MHz, CDCl<sub>3</sub>) δ 7.93–7.88 (m, 1H), 7.78–7.68 (m, 3H), 5.99–5.89 (m, 1H), 5.15 (dd, *J* = 17.1, 1.5 Hz, 1H), 5.09 (dd, *J* = 10.2, 1.4 Hz, 1H), 3.07 (t, *J* = 7.5 Hz, 2H), 2.69–2.63 (m, 2H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 175.7, 139.9, 135.9, 134.1, 133.7, 131.3, 124.0, 122.6, 116.7, 30.6, 29.2; **IR** (neat) 2923, 2257, 1558, 1334, 907, 726 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>11</sub>H<sub>11</sub>NO<sub>2</sub>S (M + Na)<sup>+</sup> 244.0408, found 244.0419.

### *Synthetic Transformations of Homopropargylic Sultams (Scheme 4.4)*

#### *Enyne Ring-Closing Metathesis to Form 4.8*



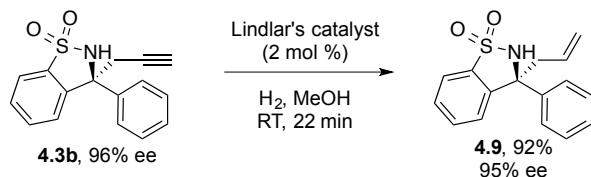
**Sultam 4.8** was prepared according to a modified procedure described by Mori and co-workers.<sup>34</sup> To a flame-dried 7 mL reaction vial equipped with a N<sub>2</sub> line and Grubbs 1<sup>st</sup> generation catalyst (6.0 mg, 0.0070 mmol, 0.050 equiv) was added substrate **4.7d** (36.8 mg, 0.140 mmol,

<sup>33</sup> Paderes M. C.; Chemler, S. R. *Org. Lett.* **2009**, *11*, 1915.

<sup>34</sup> Mori, M.; Sakakibara N.; Kinoshita, A. *J. Org. Chem.* **1998**, *63*, 6082.

1.00 equiv) in anhydrous DCM (5 mL). The N<sub>2</sub> atmosphere was exchanged with ethylene (1 atm, balloon), taking care to fully purge the vial of N<sub>2</sub>. After stirring 24 h at room temperature, the reaction mixture was concentrated in vacuo. The product was purified by flash column chromatography using 5–10% EtOAc/hexanes (1% TEA) to afford the title compound as a colorless oil (28.0 mg, 0.110 mmol, 76%). Enantiomeric excess could not be determined for the title compound using chiral SFC instrumentation due to lack of separation of the enantiomers. **TLC R<sub>f</sub>** = 0.4 (20% EtOAc/hexanes, stains blue with PAA); **<sup>1</sup>H NMR** (400 MHz, CDCl<sub>3</sub>) δ 7.78 (d, *J* = 7.7 Hz, 1H), 7.66 (t, *J* = 7.6 Hz, 1H), 7.56 (t, *J* = 7.6 Hz, 1H), 7.41 (d, *J* = 8.0 Hz, 1H), 6.42 (dd, *J* = 17.6, 10.9 Hz, 1H), 5.94 (s, 1H), 5.04–4.97 (m, 2H), 4.72 (s, 1H), 2.69 (d, *J* = 17.4 Hz, 1H), 2.58–2.49 (m, 2H), 2.45–2.37 (m, 1H), 2.07–1.94 (m, 2H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 144.6, 138.6, 135.5, 133.6, 133.2, 129.6, 128.7, 123.4, 121.6, 111.7, 61.9, 37.0, 33.3, 23.4; **IR** (neat) 3455, 3251, 2926, 1161, 1061, 744 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>14</sub>H<sub>15</sub>NO<sub>2</sub>S (M + Na)<sup>+</sup> 284.0721, found 284.0712; [α]<sup>24</sup><sub>D</sub> -14 (*c* 0.9, CDCl<sub>3</sub>).

#### Lindlar Reduction to Form **4.9**

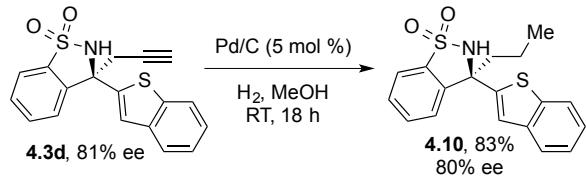


**Sultam 4.9** was prepared according to a modified procedure described by Jarvo and co-workers.<sup>35</sup> To a flame-dried 7 mL reaction vial equipped with a N<sub>2</sub> line, substrate **4.3b** (41.0 mg, 0.145 mmol, 1.00 equiv), and palladium, 5% on calcium carbonate, lead poisoned (6.4 mg, 2 mol % palladium relative to **4.3b**) was added anhydrous MeOH (2 mL). The N<sub>2</sub> atmosphere was exchanged with H<sub>2</sub> (1 atm, balloon) and the reaction was allowed to stir at room temperature.

<sup>35</sup> Harris, M. R.; Konev, M. O.; Jarvo, E. R. *J. Am. Chem. Soc.* **2014**, *136*, 7825.

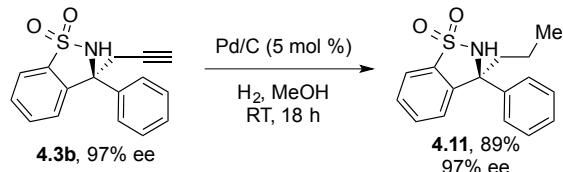
After 22 min, the H<sub>2</sub> atmosphere was exchanged with N<sub>2</sub> and the reaction mixture was filtered through a pad of Celite using 50% EtOAc/hexanes, and then concentrated in vacuo. The product was purified by flash column chromatography using 5–10% EtOAc/hexanes (1% TEA) to afford the title compound as a white solid (38.2 mg, 0.134 mmol, 92%, 95% ee). TLC R<sub>f</sub> = 0.1 (10% EtOAc/hexanes, stains blue with PAA); **m.p.** = 125–127 °C; **<sup>1</sup>H NMR** (500 MHz, CDCl<sub>3</sub>) δ 7.77 (d, J = 7.8 Hz, 1H), 7.62–7.58 (m, 3H), 7.52 (t, J = 7.7 Hz, 1H), 7.38 (t, J = 7.9 Hz, 3H), 7.30 (m, 1H), 5.66–5.57 (m, 1H), 5.28–5.21 (m, 2H), 5.00 (s, 1H), 3.25 (dd, J = 14.3, 6.6 Hz, 1H), 3.03 (dd, J = 14.3, 7.7 Hz, 1H); **<sup>13</sup>C NMR** (125.7 MHz, CDCl<sub>3</sub>) δ 143.2, 141.7, 134.8, 133.5, 131.7, 129.6, 129.1, 128.3, 126.3, 124.6, 122.0, 121.6, 67.7, 45.2; **IR** (neat) 3290, 3069, 1295, 1168, 906, 729 cm<sup>-1</sup>; **HRMS** (TOF MS ES+) *m/z* calcd for C<sub>16</sub>H<sub>15</sub>NO<sub>2</sub>S (M + Na)<sup>+</sup> 308.0721, found 308.0723; **[α]<sub>D</sub><sup>24</sup>** +72 (*c* 0.9, CDCl<sub>3</sub>); **SFC** analysis (OD-H, 10% IPA, 3.0 mL/min, 215 nm) indicated 95% ee: t<sub>R</sub> (minor) = 9.9 min, t<sub>R</sub> (major) = 12.6 min.

#### Pd/C Reduction to Form **4.10** and **4.11**



**Sultam 4.10.** To a flame-dried 7 mL reaction vial equipped with a N<sub>2</sub> line and palladium, 10% on carbon (5.6 mg, 5 mol % palladium relative to **4.3d**) was added substrate **4.3d** (35.7 mg, 0.105 mmol, 1.00 equiv) in anhydrous MeOH (2 mL). The vial was evacuated and refilled with N<sub>2</sub> three times. The N<sub>2</sub> atmosphere was exchanged with H<sub>2</sub> (1 atm, balloon) and the reaction was allowed to stir at room temperature. After 18 h, the H<sub>2</sub> atmosphere was exchanged with N<sub>2</sub> and the reaction mixture was filtered through a pad of Celite using MeOH, and then concentrated in

vacuo. The product was purified by flash column chromatography using 5–10–20% EtOAc/hexanes (1% TEA) to afford the title compound as a colorless oil (29.8 mg, 0.0870 mmol, 83%, 80% ee). **TLC**  $R_f$  = 0.4 (20% EtOAc/hexanes, UV active);  **$^1\text{H}$  NMR** (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.79 (d,  $J$  = 7.9 Hz, 1H), 7.74 (d,  $J$  = 7.9 Hz, 1H), 7.70 (dd,  $J$  = 6.8, 1.8 Hz, 1H), 7.63 (td,  $J$  = 7.6, 1.2 Hz, 1H), 7.55 (td,  $J$  = 7.5, 1.1 Hz, 1H), 7.50 (d,  $J$  = 7.9 Hz, 1H), 7.37 (s, 1H), 7.31 (aquitnd,  $J$  = 7.2, 1.5 Hz, 2H), 5.07 (br s, 1H), 2.47–2.31 (m, 2H), 1.64–1.51 (m, 1H), 1.23–1.10 (m, 1H), 0.96 (t,  $J$  = 7.4 Hz, 3H);  **$^{13}\text{C}$  NMR** (125.7 MHz,  $\text{CDCl}_3$ )  $\delta$  148.0, 142.1, 139.6, 139.4, 134.4, 133.7, 130.0, 125.0, 124.8, 124.5, 124.1, 122.4, 121.8, 121.7, 67.3, 44.1, 17.6, 14.0; **IR** (neat) 3258, 2961, 1456, 1287, 1157, 725  $\text{cm}^{-1}$ ; **HRMS** (TOF MS ES $^+$ )  $m/z$  calcd for  $\text{C}_{18}\text{H}_{17}\text{NO}_2\text{S}_2$  ( $\text{M} + \text{Na}$ ) $^+$  366.0598, found 366.0601;  $[\alpha]^{25}_{\text{D}} +47$  ( $c$  1.0,  $\text{CDCl}_3$ ); **SFC** analysis (AS-H, 20% IPA, 3.0 mL/min, 215 nm) indicated 80% ee:  $t_{\text{R}}$  (major) = 9.9 min,  $t_{\text{R}}$  (minor) = 15.7 min.



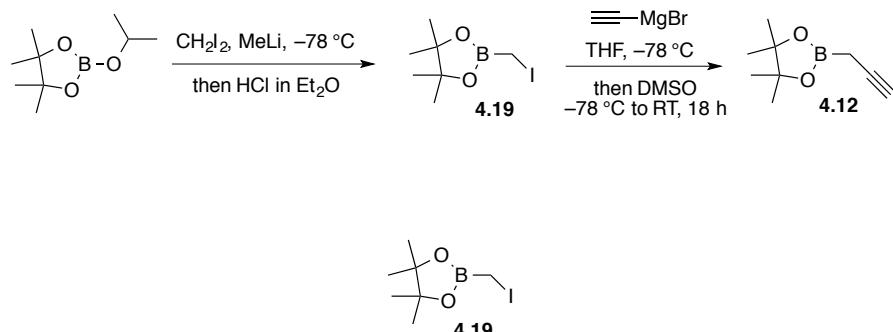
**Sultam 4.11.** To a flame-dried 7 mL reaction vial equipped with a  $\text{N}_2$  line and palladium, 10% on carbon (6.3 mg, 5 mol % palladium relative to **4.3b**) was added substrate **4.3b** (32.3 mg, 0.114 mmol, 1.00 equiv) in anhydrous MeOH (1.0 mL). The vial was evacuated and refilled with  $\text{N}_2$  three times. The  $\text{N}_2$  atmosphere was exchanged with  $\text{H}_2$  (1 atm, balloon) and the reaction was allowed to stir at room temperature. After 18 h, the  $\text{H}_2$  atmosphere was exchanged with  $\text{N}_2$  and the reaction mixture was filtered through a pad of Celite using MeOH, and then concentrated in vacuo. The product was purified by flash column chromatography using 5–10–20%

EtOAc/hexanes (1% TEA) to afford the title compound as a white solid (29.2 mg, 0.102 mmol, 89%, 97% ee). **TLC**  $R_f$  = 0.5 (20% EtOAc/hexanes, stains blue with PAA); **m.p.** = 160–162 °C;  **$^1\text{H NMR}$**  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.78 (d,  $J$  = 7.7 Hz, 1H), 7.60–7.49 (m, 4H), 7.36 (t,  $J$  = 7.7 Hz, 2H), 7.29 (t,  $J$  = 7.0 Hz, 2H), 4.86 (br s, 1H), 2.44–2.36 (m, 1H), 2.33–2.26 (m, 1H), 1.57–1.46 (m, 1H), 1.24–1.02 (m, 1H), 0.94 (t,  $J$  = 7.3 Hz, 3H);  **$^{13}\text{C NMR}$**  (125.7 MHz,  $\text{CDCl}_3$ )  $\delta$  143.6, 142.5, 134.5, 133.5, 129.4, 129.1, 128.2, 126.1, 124.4, 121.4, 68.9, 42.6, 17.5, 14.1; **IR** (neat) 3251, 2958, 1450, 1281, 1157, 765  $\text{cm}^{-1}$ ; **HRMS** (TOF MS ES+)  $m/z$  calcd for  $\text{C}_{16}\text{H}_{17}\text{NO}_2\text{S}$  (M + Na)<sup>+</sup> 310.0878, found 310.0883;  $[\alpha]^{25}_{\text{D}} +95$  (*c* 0.9,  $\text{CDCl}_3$ ); **SFC** analysis (OD-H, 10% IPA, 3.0 mL/min, 215 nm) indicated 97% ee:  $t_{\text{R}}$  (minor) = 11.0 min,  $t_{\text{R}}$  (major) = 13.9 min.

### Mechanistic Studies (Scheme 4.6)

#### Synthesis of Propargyl Borolane Reagent 4.12

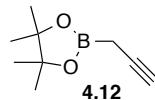
**Scheme 4.8.** Synthesis of propargyl borolane **4.12**.



**Iodomethyl borolane 4.19** was prepared according to a modified procedure by Brown and co-workers,<sup>36</sup> using the following amounts of reagents: 2-isopropoxy-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (17.7 mL, 86.9 mmol, 1.00 equiv), diiodomethane (7.00 mL, 86.9 mmol, 1.00

<sup>36</sup> Roy, C. D.; Soundararajan, R.; Brown, H. C. *Monatsch. Chem.* **2008**, 139, 241.

equiv), methylolithium (52.3 mL, 86.9 mmol, 1.66 M in Et<sub>2</sub>O, 1.00 equiv), THF (45 mL), and anhydrous HCl (93.0 mL, 93.0 mmol, 1.00 M in Et<sub>2</sub>O, 1.07 equiv). The resulting red solution was filtered through a plug of silica gel (40 mL) eluting with 10% Et<sub>2</sub>O/pentanes and concentrated in vacuo. Distillation twice through a short path distillation apparatus onto activated 4Å molecular sieves at 13.7 mmHg and T<sub>vap</sub> = 95 °C provided the title compound as a clear, colorless liquid (7.64 g, 28.5 mmol, 33%). Analytical data are consistent with literature values.<sup>36</sup> **TLC** R<sub>f</sub> = 0.7 (10% Et<sub>2</sub>O/pentanes, stains blue with PAA); **<sup>1</sup>H NMR** (CDCl<sub>3</sub>, 400 MHz) δ 2.16 (s, 2H), 1.28 (s, 12 H); **<sup>13</sup>C NMR** (CDCl<sub>3</sub>, 125.7 MHz) δ 84.1, 24.5; **<sup>11</sup>B NMR** (CDCl<sub>3</sub>, 160.2 MHz) δ 31.7; **IR** (neat) 2977, 1322, 1142, 844, 673, 577 cm<sup>-1</sup>; **HRMS** (TOF MS CI+) *m/z* calcd for C<sub>7</sub>H<sub>14</sub>BIO<sub>2</sub> (M)<sup>+</sup> 268.0133, found 268.0143.

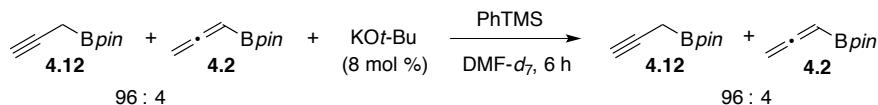


**Propargylboronic acid pinacol ester 4.12** was prepared according to a literature procedure by Fandrick and co-workers,<sup>22c</sup> using the following amounts of reagents: **4.19** (2.0 mL, 11 mmol, 1.0 equiv), ethynylmagnesium bromide (19.8 mL, 11.2 mmol, 0.566 M in THF, 1.02 equiv), THF (15 mL), and 1:1 THF/DMSO (24 mL). Distillation using a Kugelrohr distillation apparatus at 3–6 torr and T<sub>vap</sub> = 60 °C provided the title compound as a 96:4 mixture of propargyl borolane **4.12** and allenyl borolane **4.2** as a clear, colorless oil (0.37 g, 2.2 mmol, 20%). Analytical data for allenyl borolane **4.2** are consistent with literature values.<sup>32</sup> **TLC** R<sub>f</sub> = 0.9 (10% Et<sub>2</sub>O/hexanes, stains blue with PAA); **<sup>1</sup>H NMR** (400 MHz, DMF-*d*<sub>7</sub>) δ 4.90 (t, *J* = 7.0 Hz, 1H), 4.71 (d, *J* = 7.0 Hz, 2H), 1.26 (s, 12H). Analytical data for propargyl borolane **4.12** are consistent with literature values.<sup>22c</sup> **<sup>1</sup>H NMR** (400 MHz, DMF-*d*<sub>7</sub>) δ 2.46 (t, *J* = 2.9 Hz, 1H), 1.78 (d, *J* = 2.9 Hz, 2H), 1.26 (s, 12H); **<sup>13</sup>C NMR** (125.7 MHz, DMF-*d*<sub>7</sub>) δ 84.9, 81.9, 69.5, 25.3.

**Note:** We found that in order to obtain high ratios of propargyl borolane to allenyl borolane, it was necessary to use precisely 1.02 equivalents of ethynylmagnesium bromide relative to iodomethyl borolane **4.19**. Excess Grignard reagent (1.1 equivalents) causes isomerization to allenyl borolane, while fewer than 1.0 equivalents of Grignard reagent results in low conversion to product (Figure 4.2).

### Control Reactions with **4.12**

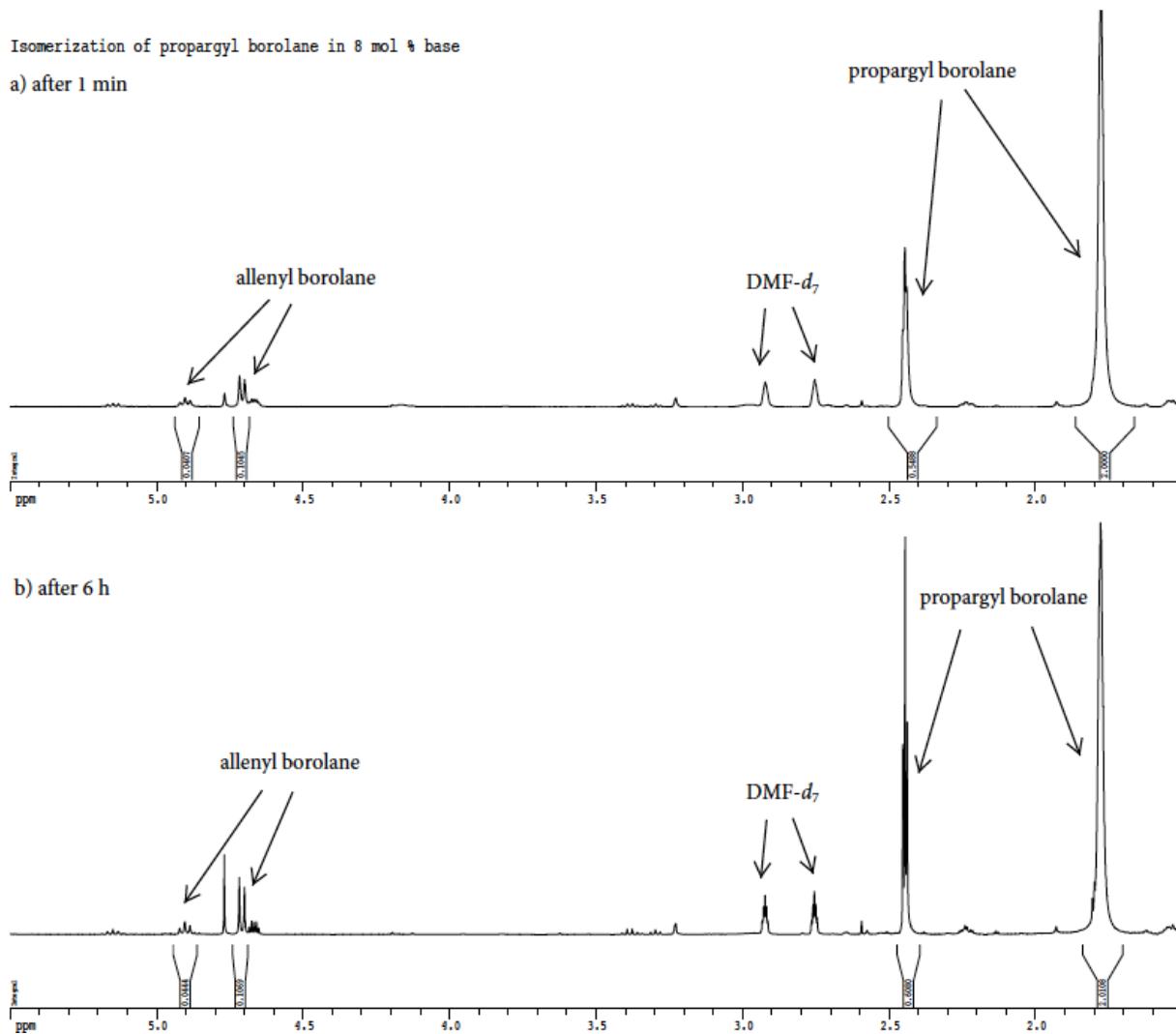
#### *Isomerization in Presence of 8 mol % Base*



In a glovebox, a flame-dried vial was charged with potassium *tert*-butoxide (1.7 mg, 0.015 mmol, 0.076 equiv). *N,N*-Dimethylformamide-*d*<sub>7</sub> (+0.05% V/V TMS, 0.6 mL) was added from an ampule opened in the glovebox and the solution was transferred into an oven-dried NMR tube. The NMR tube was capped with a rubber septum, sealed with parafilm, and removed from the glovebox. Phenyltrimethylsilane (PhTMS, internal standard) (17.2  $\mu$ L, 0.100 mmol, 0.500 equiv) was added via syringe to the NMR tube through the septum, and the NMR tube was inverted to mix. An initial <sup>1</sup>H NMR spectrum was collected of the solution, after which propargyl borolane **4.12** (72  $\mu$ L, 0.40 mmol, 2.0 equiv) was added via syringe to the NMR tube through the septum, and the NMR tube was inverted to mix. A <sup>1</sup>H NMR spectrum was collected (1 minute after adding **4.12**, Figure 4.3a), followed by sequential <sup>1</sup>H NMR spectra collected at the time points listed in Table 4.3.

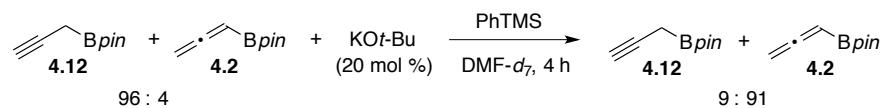
**Table 4.3.** Absence of isomerization of propargyl borolane **4.12** in the presence of 8 mol % KOt-Bu at RT.

time elapsed	ratio <b>4.12 : 4.2</b>
1 min	96:4
4 min	96:4
7 min	96:4
9 min	96:4
10 min	96:4
15 min	96:4
20 min	96:4
30 min	96:4
40 min	96:4
1 h	96:4
3 h	96:4
6 h	96:4



**Figure 4.3.** Absence of isomerization of propargyl borolane **4.12** in the presence of 8 mol %  $\text{KO}t\text{-Bu}$  at RT.

#### *Isomerization in Presence of 20 mol % Base*

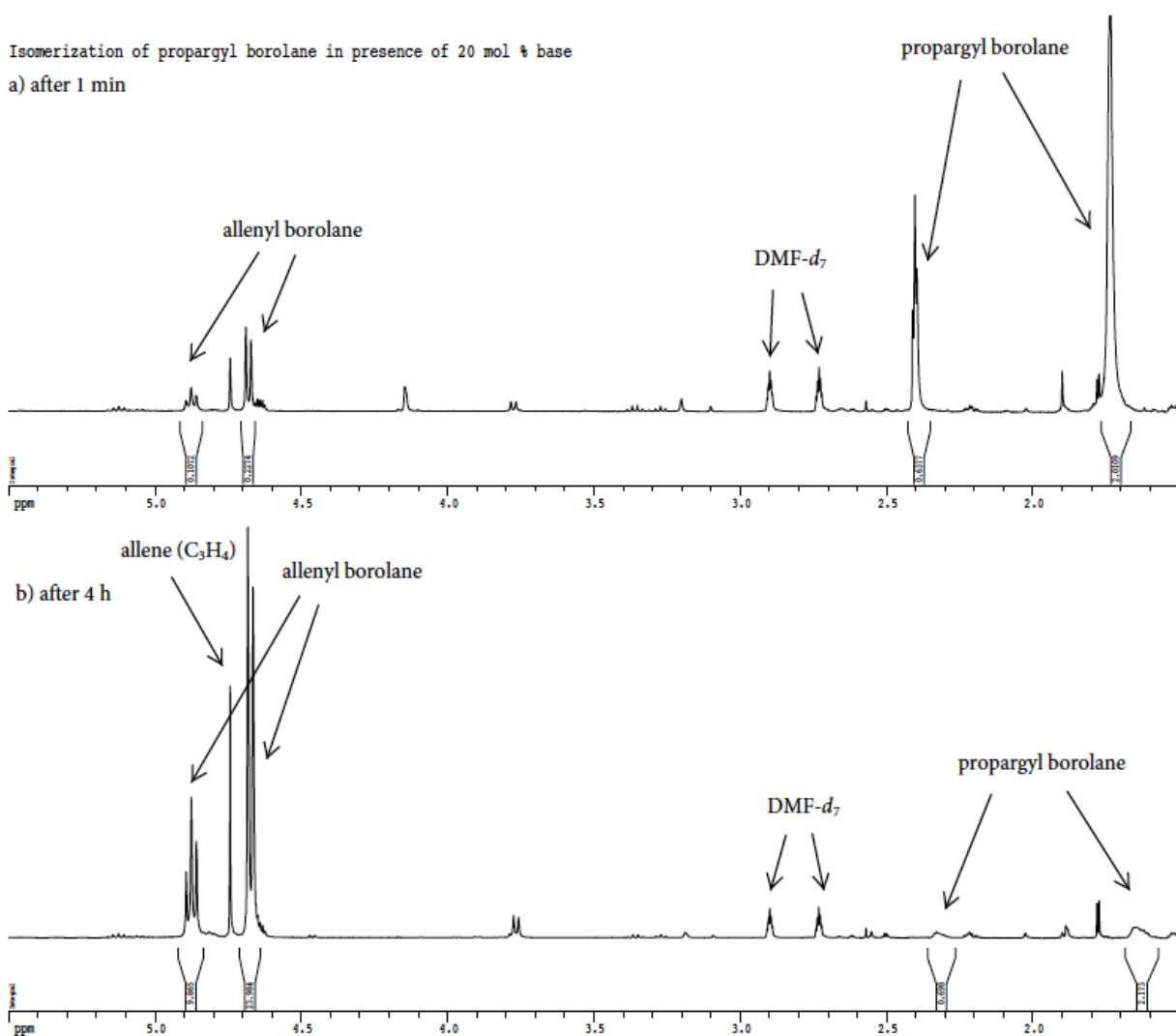


In a glovebox, a flame-dried vial was charged with potassium *tert*-butoxide (4.5 mg, 0.040 mmol, 0.20 equiv). *N,N*-Dimethylformamide- $d_7$  (+0.05% V/V TMS, 0.6 mL) was added

from an ampule opened in the glovebox and the solution was transferred into an oven-dried NMR tube. The NMR tube was capped with a rubber septum, sealed with parafilm, and removed from the glovebox. Phenyltrimethylsilane (PhTMS, internal standard) (17.2  $\mu$ L, 0.100 mmol, 0.500 equiv) was added via syringe to the NMR tube through the septum, and the NMR tube was inverted to mix. An initial  $^1\text{H}$  NMR spectrum was collected, after which propargyl borolane **4.12** (72  $\mu$ L, 0.40 mmol, 2.0 equiv) was added via syringe to the NMR tube through the septum, and the NMR tube was inverted to mix. A  $^1\text{H}$  NMR spectrum was collected (1 minute after adding **4.12**, Figure 4.4a), followed by sequential  $^1\text{H}$  NMR spectra collected at the time points listed in Table 4.4.

**Table 4.4.** Isomerization of propargyl borolane **4.12** to allenyl borolane **4.2** in the presence of 20 mol %  $\text{KO}t\text{-Bu}$  at RT.

time elapsed	ratio <b>4.12</b> : <b>4.2</b>
1 min	92:8
5 min	87:13
8 min	84:16
10 min	82:18
20 min	69:31
30 min	64:36
35 min	60:40
40 min	57:43
1 h	46:54
2 h	27:73
3 h	18:82
4 h	9:91



**Figure 4.4.** Isomerization of propargyl borolane **4.12** to allenyl borolane **4.2** in the presence of 20 mol % KOt-Bu at RT.

### Mechanistic Studies with Borolane Reagents

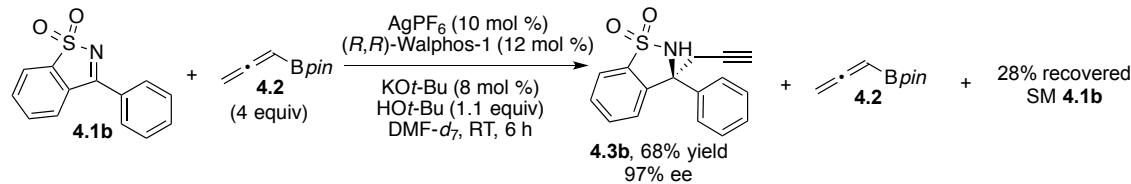
We were interested in distinguishing between two possible mechanisms for this propargylation reaction: transmetallation of the silver catalyst with the borolane reagent (Mechanism A), or Lewis acid catalysis (Mechanism B). To distinguish between these mechanisms, we examined reactions employing propargyl borolane reagent **4.12** while lowering

the base loading to 8 mol %. This experimental modification was performed to minimize isomerization of **4.12** to allenyl borolane **4.2** (vide supra). We also performed the reactions in deuterated solvent in order to determine the ratio of **4.12** to **4.2** by  $^1\text{H}$  NMR immediately after the reaction.

A control reaction using allenyl borolane **4.2** (Scheme 4.6a) demonstrated that under these conditions, alkyne **4.3b** was formed in 68% yield with 28% recovered starting material **4.1b** (vide infra). Using propargyl borolane **4.12** in the reaction (Scheme 4.6b) yielded alkyne **4.3b** in 64% yield with 32% recovered starting material **4.1b** (vide infra). This product distribution is consistent with Mechanism A.

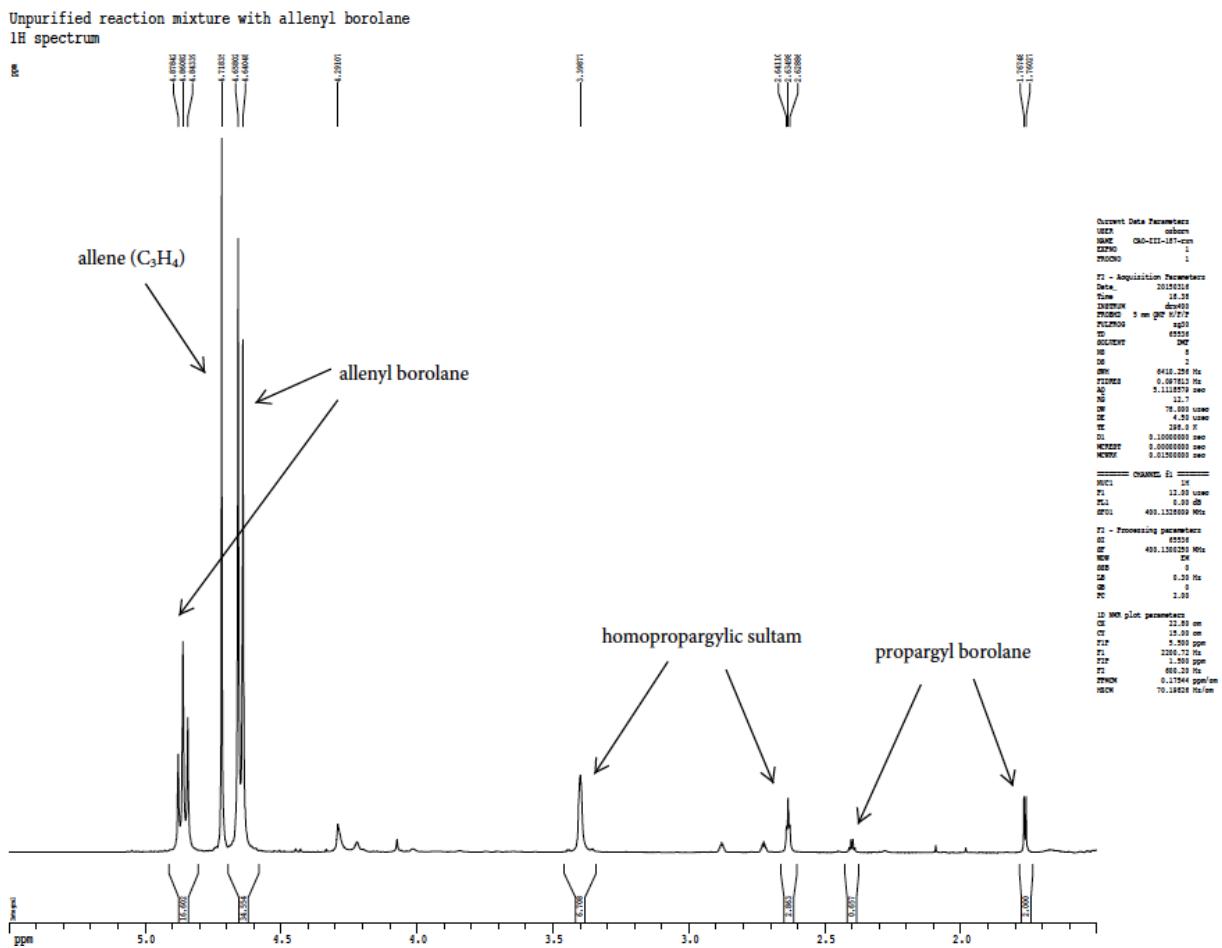
**Note:** All manipulations involving silver-catalyzed reactions were performed in the absence of direct light, using vials and NMR tubes wrapped in aluminum foil.

*Reaction using Allenyl Borolane **4.2** (Scheme 4.6a)*



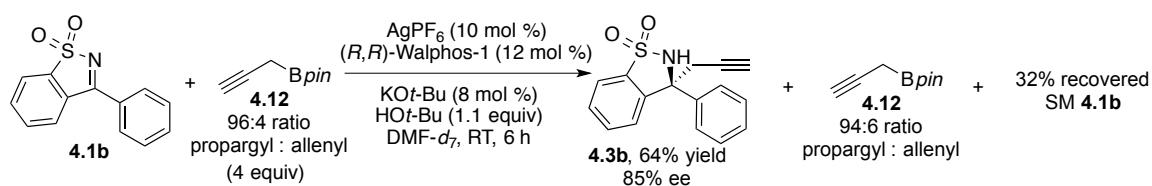
In a glovebox, an oven-dried 1.0 mL conical vial equipped with a triangular stir bar was charged with  $\text{AgPF}_6$  (5.0 mg, 0.020 mmol, 0.10 equiv) and Walphos W001-1 (22.3 mg, 0.0240 mmol, 0.102 equiv). The vial was sealed with a screw-top cap fit with a septum and  $\text{DMF}-d_7$  (+0.05% V/V TMS, 400  $\mu\text{L}$ ) was added from an ampule opened in the glovebox. The vial was removed from the glovebox and the solution was stirred for 5 min at RT. The  $\text{N}_2$  line was then removed and the solution was stirred for 30 min at 70 °C, then cooled to RT over 15 min.

To the catalyst solution was added *tert*-butanol (21  $\mu$ L, 0.22 mmol, 1.1 equiv), followed by potassium *tert*-butoxide (1.7 mg, 0.015 mmol, 0.076 equiv) and phenyl ketimine **4.1b** (48.6 mg, 0.200 mmol, 1.00 equiv) under a flow of N<sub>2</sub>. The reaction was stirred at RT for 5 min to dissolve the ketimine. Allenylboronic acid pinacol ester **4.2** (72  $\mu$ L, 0.40 mmol, 2.0 equiv) was added via syringe, followed by another portion of allenylboronic acid pincol ester (72  $\mu$ L, 0.40 mmol, 2.0 equiv) added via slow addition over 3 h using a syringe pump. The N<sub>2</sub> line was removed and the reaction was stirred at 22 °C for another 3 h. The reaction mixture in DMF-*d*<sub>7</sub> was transferred to an NMR tube and the ratio of allenyl borolane **4.2** to propargyl borolane **4.12** was determined to be 94:6 by <sup>1</sup>H NMR (Figure 4.5). The mixture was then filtered through a plug of silica gel eluting with 100% Et<sub>2</sub>O to remove the catalyst. Et<sub>2</sub>O was removed in vacuo and the resulting residue was purified by flash column chromatography using 0–1% TEA/benzene to separate product from unreacted starting material. The mixture was purified again by flash column chromatography using 5–10–20% EtOAc/hexanes (1% TEA) to separate product from excess ligand and afford alkyne **4.3b** as a white solid (38.5 mg, 0.136 mmol, 68%, 97% ee). Analytical data are consistent with the values listed for **4.3b** (vide supra). SFC analysis (OD-H, 10% IPA, 3.0 mL/min, 215 nm) indicated 97% ee: t<sub>R</sub> (minor) = 11.8 min, t<sub>R</sub> (major) = 13.7 min.



**Figure 4.5.** Unpurified reaction mixture in  $\text{DMF}-d_7$ .

#### *Reaction using Propargyl Borolane 4.12 (Scheme 4.6b)*

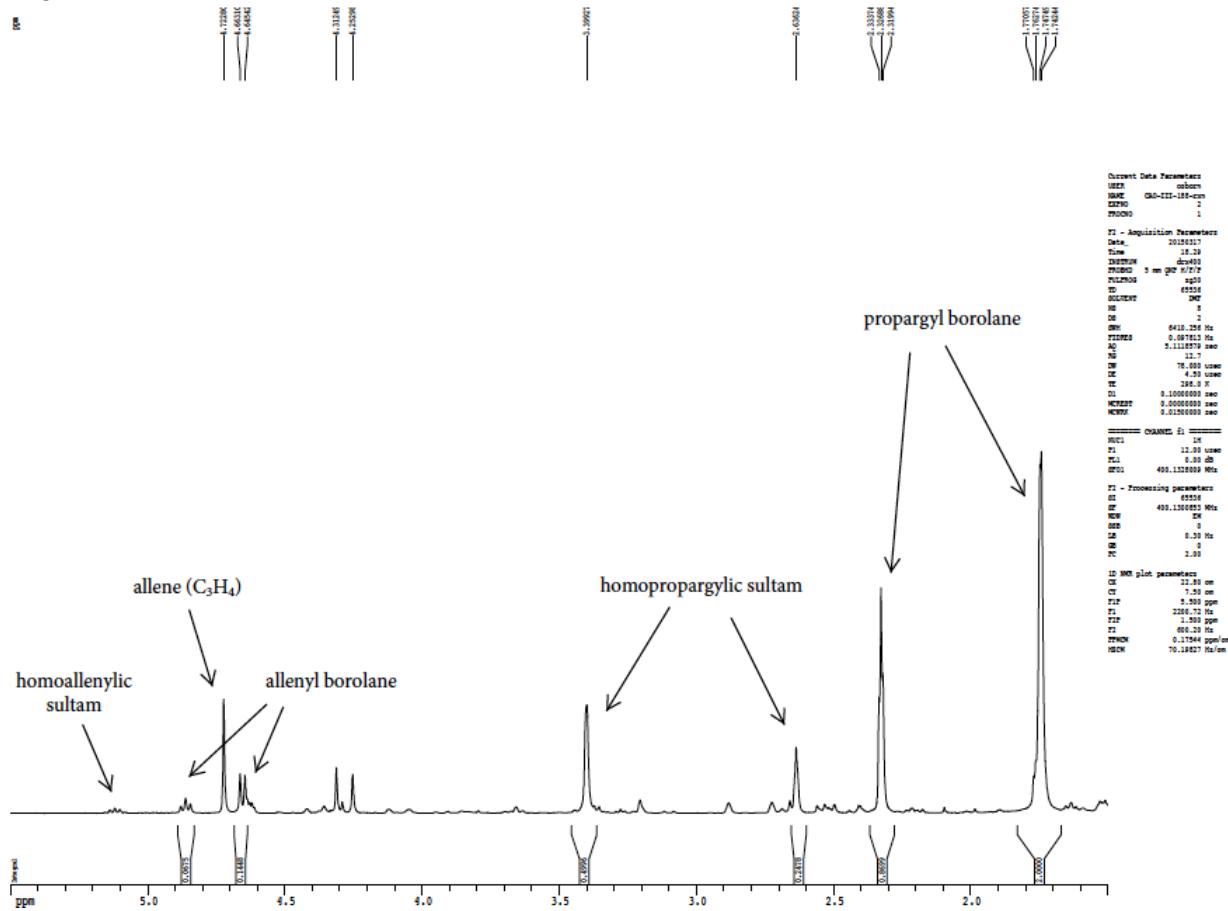


In a glovebox, an oven-dried 1.0 mL conical vial equipped with a triangular stir bar was charged with AgPF<sub>6</sub> (5.0 mg, 0.020 mmol, 0.10 equiv) and Walphos W001-1 (22.3 mg, 0.0240 mmol, 0.120 equiv). The vial was sealed with a screw-top cap fit with a septum and DMF-*d*<sub>7</sub>

(+0.05% V/V TMS, 400  $\mu$ L) was added from an ampule opened in the glovebox. The vial was removed from the glovebox and the solution was stirred for 5 min at RT. The N<sub>2</sub> line was then removed and the solution was stirred for 30 min at 70 °C, then cooled to RT over 15 min.

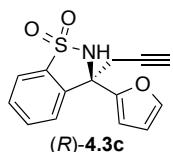
To the catalyst solution was added *tert*-butanol (21  $\mu$ L, 0.22 mmol, 1.1 equiv), followed by potassium *tert*-butoxide (1.7 mg, 0.015 mmol, 0.076 equiv) and phenyl ketimine **4.1b** (48.6 mg, 0.200 mmol, 1.00 equiv) under a flow of N<sub>2</sub>. The reaction was stirred at RT for 5 min to dissolve the ketimine. Propargylboronic acid pinacol ester **4.12** (72  $\mu$ L, 0.40 mmol, 2.0 equiv) was added via syringe, followed by another portion of propargylboronic acid pinacol ester (72  $\mu$ L, 0.40 mmol, 2.0 equiv) added via slow addition over 3 h using a syringe pump. The N<sub>2</sub> line was removed and the reaction was stirred at 22 °C for another 3 h. The reaction mixture in DMF-*d*<sub>7</sub> was transferred to an NMR tube and the ratio of propargyl borolane **4.12** to allenyl borolane **4.2** was determined to be 94:6 by <sup>1</sup>H NMR (Figure 4.6). The mixture was then filtered through a plug of silica gel eluting with 100% Et<sub>2</sub>O to remove the catalyst. Et<sub>2</sub>O was removed in vacuo and the resulting residue was purified by flash column chromatography using 0–1% TEA/benzene to separate product from unreacted starting material. The mixture was purified again by flash column chromatography using 5–10–20% EtOAc/hexanes (1% TEA) to separate product from excess ligand and afford alkyne **4.3b** as a white solid (36.5 mg, 0.129 mmol, 64%, 85% ee). Analytical data are consistent with the values listed for **4.3b** (*vide supra*). SFC analysis (OD-H, 10% IPA, 3.0 mL/min, 215 nm) indicated 85% ee: t<sub>R</sub> (minor) = 11.8 min, t<sub>R</sub> (major) = 13.7 min.

Unpurified reaction mixture with propargyl borolane  
1H spectrum



## Crystallographic Data

### A. X-ray Data Collection, Structure Solution and Refinement for (R)-4.3c:



A single crystal was grown from EtOAc with slow diffusion of pentanes at room temperature. A colorless crystal of approximate dimensions 0.202 x 0.333 x 0.426 mm was mounted on a glass fiber and transferred to a Bruker SMART APEX II diffractometer. The APEX2<sup>37</sup> program package was used to determine the unit-cell parameters and for data collection (15 sec/frame scan time for a sphere of diffraction data). The raw frame data was processed using SAINT<sup>38</sup> and SADABS<sup>39</sup> to yield the reflection data file. Subsequent calculations were carried out using the SHELXTL<sup>40</sup> program. The diffraction symmetry was *mmm* and the systematic absences were consistent with the orthorhombic space group *P*2<sub>1</sub>2<sub>1</sub>2<sub>1</sub> that was later determined to be correct.

The structure was solved by direct methods and refined on F<sup>2</sup> by full-matrix least-squares techniques. The analytical scattering factors<sup>41</sup> for neutral atoms were used throughout the analysis. Hydrogen atoms H(1) and H(10) were located from a difference-Fourier map and refined (x,y,z and U<sub>iso</sub>). The remaining hydrogen atoms were included using a riding model. O(3) and C(12) were disordered and included using partial site-occupancy-factors. The disorder was included to account for the approximate distribution of carbon (50%) and oxygen (50%) over the two sites.

At convergence, wR2 = 0.0909 and Goof = 1.070 for 180 variables refined against 3100 data (0.74 Å), R1 = 0.0363 for those 2870 data with I > 2.0σ(I). The absolute structure was assigned by refinement of the Flack parameter.<sup>42</sup>

<sup>37</sup> APEX2 Version 2014.11-0, Bruker AXS, Inc.; Madison, WI 2014.

<sup>38</sup> SAINT Version 8.34a, Bruker AXS, Inc.; Madison, WI 2013.

<sup>39</sup> Sheldrick, G. M. SADABS, Version 2014/5, Bruker AXS, Inc.; Madison, WI 2014.

<sup>40</sup> Sheldrick, G. M. SHELXTL, Version 2014/7, Bruker AXS, Inc.; Madison, WI 2014.

<sup>41</sup> International Tables for Crystallography 1992, Vol. C., Dordrecht: Kluwer Academic Publishers.

<sup>42</sup> Parsons, S., Flack, H. D., Wagner, T. Acta Cryst. B69, 249-259, 2013.

**Definitions:**

$$wR2 = [\sum[w(F_o^2 - F_c^2)^2] / \sum[w(F_o^2)^2]]^{1/2}$$

$$R1 = \sum||F_o - F_c|| / \sum|F_o|$$

Goof = S =  $[\sum[w(F_o^2 - F_c^2)^2] / (n-p)]^{1/2}$  where n is the number of reflections and p is the total number of parameters refined.

The thermal ellipsoid plot is shown at the 30% probability level.

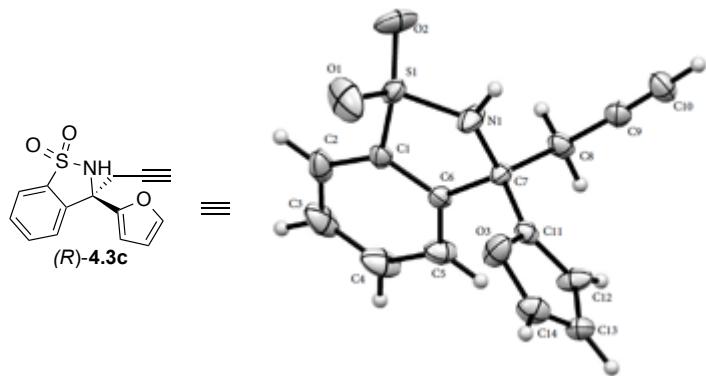


Table 4.5. Crystal data and structure refinement for (*R*)-**4.3c**.

Identification code	erj23 (Charlotte Osborne)	
Empirical formula	$C_{14} H_{11} N O_3 S$	
Formula weight	273.30	
Temperature	133(2) K	
Wavelength	0.71073 Å	
Crystal system	Orthorhombic	
Space group	$P2_1 2_1 2_1$	
Unit cell dimensions	$a = 7.5472(5)$ Å	$a = 90^\circ.$
	$b = 10.3052(7)$ Å	$b = 90^\circ.$
	$c = 16.2327(10)$ Å	$g = 90^\circ.$
Volume	1262.51(14) Å <sup>3</sup>	
Z	4	
Density (calculated)	1.438 Mg/m <sup>3</sup>	
Absorption coefficient	0.259 mm <sup>-1</sup>	
F(000)	568	
Crystal color	colorless	
Crystal size	0.426 x 0.333 x 0.202 mm <sup>3</sup>	
Theta range for data collection	2.341 to 28.839°	
Index ranges	$-10 \leq h \leq 10, -13 \leq k \leq 13, -21 \leq l \leq 20$	
Reflections collected	15206	
Independent reflections	3100 [R(int) = 0.0274]	
Completeness to theta = 25.500°	99.9 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.8621 and 0.8165	
Refinement method	Full-matrix least-squares on F <sup>2</sup>	
Data / restraints / parameters	3100 / 0 / 180	
Goodness-of-fit on F <sup>2</sup>	1.070	
Final R indices [I>2sigma(I) = 2870 data]	R1 = 0.0363, wR2 = 0.0881	
R indices (all data, 0.74Å)	R1 = 0.0403, wR2 = 0.0909	

Absolute structure parameter	0.04(2)
Largest diff. peak and hole	0.298 and -0.366 e. $\text{\AA}^{-3}$

Table 4.6. Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for (R)-4.3c. U(eq) is defined as one third of the trace of the orthogonalized  $U^{ij}$  tensor.

	x	y	z	U(eq)
S(1)	3053(1)	7602(1)	8953(1)	25(1)
N(1)	4347(4)	6393(2)	9154(2)	36(1)
O(1)	1714(3)	7709(3)	9570(1)	58(1)
O(2)	4026(4)	8766(2)	8781(1)	48(1)
O(3)	3159(3)	3886(2)	9539(2)	38(1)
C(1)	2206(3)	6903(2)	8057(2)	24(1)
C(2)	897(3)	7416(3)	7551(2)	37(1)
C(3)	430(4)	6713(3)	6868(2)	47(1)
C(4)	1240(4)	5532(3)	6696(2)	46(1)
C(5)	2551(4)	5035(3)	7203(2)	33(1)
C(6)	3040(3)	5741(2)	7897(1)	22(1)
C(7)	4499(3)	5402(2)	8508(1)	20(1)
C(8)	6309(3)	5461(3)	8061(2)	28(1)
C(9)	7799(3)	5308(2)	8617(2)	28(1)
C(10)	8994(4)	5223(3)	9089(2)	34(1)
C(11)	4214(3)	4100(2)	8891(2)	22(1)
C(12)	4765(3)	2981(2)	8580(1)	32(1)
C(13)	4119(3)	1987(3)	9100(2)	31(1)
C(14)	3160(4)	2544(3)	9680(2)	33(1)

Table 4.7. Bond lengths [ $\text{\AA}$ ] and angles [ $^\circ$ ] for (*R*)-**4.3c**.

---

S(1)-O(1)	1.426(2)
S(1)-O(2)	1.435(2)
S(1)-N(1)	1.616(2)
S(1)-C(1)	1.745(3)
N(1)-C(7)	1.468(3)
O(3)-C(11)	1.338(3)
O(3)-C(14)	1.402(4)
C(1)-C(6)	1.378(3)
C(1)-C(2)	1.389(4)
C(2)-C(3)	1.371(5)
C(3)-C(4)	1.390(5)
C(4)-C(5)	1.385(4)
C(5)-C(6)	1.391(4)
C(6)-C(7)	1.522(3)
C(7)-C(11)	1.494(3)
C(7)-C(8)	1.548(3)
C(8)-C(9)	1.451(4)
C(9)-C(10)	1.186(4)
C(11)-C(12)	1.326(3)
C(12)-C(13)	1.414(3)
C(13)-C(14)	1.319(4)
O(1)-S(1)-O(2)	115.75(16)
O(1)-S(1)-N(1)	110.28(16)
O(2)-S(1)-N(1)	112.00(15)
O(1)-S(1)-C(1)	110.97(13)
O(2)-S(1)-C(1)	111.74(12)
N(1)-S(1)-C(1)	94.07(12)

C(7)-N(1)-S(1)	116.09(18)
C(11)-O(3)-C(14)	106.9(2)
C(6)-C(1)-C(2)	123.0(3)
C(6)-C(1)-S(1)	110.40(18)
C(2)-C(1)-S(1)	126.6(2)
C(3)-C(2)-C(1)	117.4(3)
C(2)-C(3)-C(4)	120.8(3)
C(5)-C(4)-C(3)	121.3(3)
C(4)-C(5)-C(6)	118.5(3)
C(1)-C(6)-C(5)	119.1(2)
C(1)-C(6)-C(7)	114.0(2)
C(5)-C(6)-C(7)	126.8(2)
N(1)-C(7)-C(11)	108.4(2)
N(1)-C(7)-C(6)	104.41(19)
C(11)-C(7)-C(6)	111.9(2)
N(1)-C(7)-C(8)	112.1(2)
C(11)-C(7)-C(8)	111.0(2)
C(6)-C(7)-C(8)	108.93(19)
C(9)-C(8)-C(7)	112.8(2)
C(10)-C(9)-C(8)	177.4(3)
C(12)-C(11)-O(3)	110.0(2)
C(12)-C(11)-C(7)	125.2(2)
O(3)-C(11)-C(7)	124.2(2)
C(11)-C(12)-C(13)	107.2(2)
C(14)-C(13)-C(12)	107.5(2)
C(13)-C(14)-O(3)	108.3(2)

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Table 4.8. Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for (R)-**4.3c**. The anisotropic displacement factor exponent takes the form:  $-2p^2 [ h^2 a^{*2} U^{11} + \dots + 2 h k a^{*} b^{*} U^{12} ]$

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{23}$	$U^{13}$	$U^{12}$
S(1)	26(1)	19(1)	30(1)	-2(1)	8(1)	2(1)
N(1)	44(1)	26(1)	38(1)	-13(1)	-19(1)	10(1)
O(1)	38(1)	91(2)	45(1)	-20(1)	19(1)	1(1)
O(2)	74(2)	19(1)	50(1)	4(1)	-6(1)	-12(1)
O(3)	38(1)	28(1)	48(1)	4(1)	21(1)	3(1)
C(1)	19(1)	22(1)	32(1)	6(1)	2(1)	0(1)
C(2)	24(1)	35(2)	53(2)	18(1)	-2(1)	5(1)
C(3)	36(2)	52(2)	53(2)	23(2)	-24(2)	-9(1)
C(4)	50(2)	48(2)	39(2)	10(1)	-26(2)	-16(2)
C(5)	38(2)	30(1)	32(1)	2(1)	-11(1)	-7(1)
C(6)	21(1)	21(1)	25(1)	5(1)	-4(1)	-2(1)
C(7)	23(1)	18(1)	20(1)	-2(1)	-3(1)	2(1)
C(8)	24(1)	28(1)	31(1)	6(1)	0(1)	-4(1)
C(9)	25(1)	25(1)	35(1)	5(1)	3(1)	-2(1)
C(10)	24(1)	38(2)	40(2)	10(1)	-2(1)	-4(1)
C(11)	18(1)	25(1)	23(1)	4(1)	-5(1)	-2(1)
C(12)	44(1)	25(1)	27(1)	-6(1)	12(1)	-10(1)
C(13)	29(1)	22(1)	43(2)	-1(1)	-4(1)	-5(1)
C(14)	36(1)	33(1)	31(1)	9(1)	4(1)	-9(1)

Table 4.9. Hydrogen coordinates ( $\times 10^4$ ) and isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ )  
for (R)-4.3c.

	x	y	z	U(eq)
H(1)	5160(50)	6570(30)	9490(20)	44(10)
H(2B)	346	8222	7674	45
H(3A)	-457	7037	6507	57
H(4A)	887	5056	6223	55
H(5A)	3105	4230	7079	40
H(8A)	6353	4769	7639	33
H(8B)	6414	6306	7774	33
H(10)	9950(60)	5160(40)	9500(30)	69(13)
H(12A)	5464	2869	8098	38
H(13A)	4336	1084	9042	38
H(14A)	2571	2110	10117	40

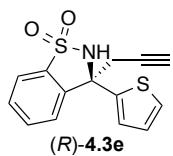
Table 4.10. Hydrogen bonds for (*R*)-**4.3c** [Å and °].

D-H...A	d(D-H)	d(H...A)	d(D...A)	$\angle$ (DHA)
N(1)-H(1)...O(1) <sup>#1</sup>	0.84(4)	2.06(4)	2.888(3)	167(3)

Symmetry transformations used to generate equivalent atoms:

#1 x+1/2,-y+3/2,-z+2

B. X-ray Data Collection, Structure Solution and Refinement for (R)-4.3e:



A single crystal was grown from EtOAc with slow diffusion of pentanes at room temperature. A colorless crystal of approximate dimensions 0.573 x 0.369 x 0.266 mm was mounted on a glass fiber and transferred to a Bruker SMART APEX II diffractometer. The APEX2<sup>37</sup> program package was used to determine the unit-cell parameters and for data collection (5 sec/frame scan time for a sphere of diffraction data). The raw frame data was processed using SAINT<sup>38</sup> and SADABS<sup>39</sup> to yield the reflection data file. Subsequent calculations were carried out using the SHELXTL<sup>40</sup> program. The diffraction symmetry was *mmm* and the systematic absences were consistent with the orthorhombic space group *P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>* that was later determined to be correct.

The structure was solved by direct methods and refined on F<sup>2</sup> by full-matrix least-squares techniques. The analytical scattering factors<sup>41</sup> for neutral atoms were used throughout the analysis. S(2) and C(12) were disordered and included using partial site-occupancy-factors. The disorder was included to account for the approximate distribution of carbon (25%) / sulfur (75%) at the position of S(2) and carbon (75%) / sulfur (25%) at the position of C(12). H(1) and H(10) were located from a difference-Fourier map and refined (x,y,z and U<sub>iso</sub>). All other hydrogen atoms were included using a riding model.

At convergence, wR2 = 0.0762 and Goof = 1.076 for 180 variables refined against 3338 data (0.73 Å), R1 = 0.0275 for those 3263 data with I > 2.0σ(I). The absolute structure was assigned by refinement of the Flack parameter.<sup>42</sup>

**Definitions:**

$$wR2 = [\sum[w(F_o^2 - F_c^2)^2] / \sum[w(F_o^2)^2]]^{1/2}$$

$$R1 = \sum||F_o| - |F_c|| / \sum|F_o|$$

Goof = S =  $[\sum[w(F_o^2 - F_c^2)^2] / (n-p)]^{1/2}$  where n is the number of reflections and p is the total number of parameters refined.

The thermal ellipsoid plot is shown at the 50% probability level.

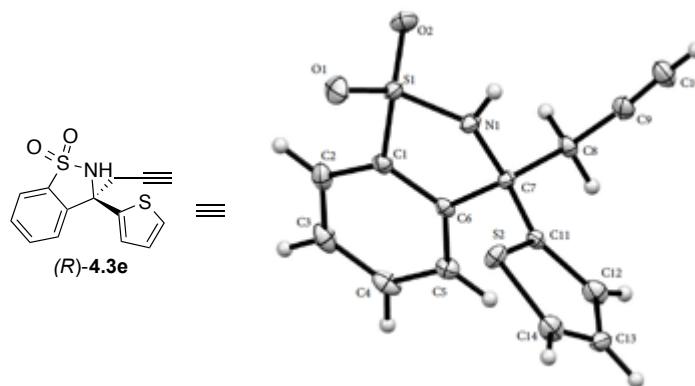


Table 4.11. Crystal data and structure refinement for (*R*)-**4.3e**.

Identification code	erj24 (Charlotte Osborne)	
Empirical formula	$C_{14} H_{11} N O_2 S_2$	
Formula weight	289.36	
Temperature	88(2) K	
Wavelength	0.71073 Å	
Crystal system	Orthorhombic	
Space group	$P2_1 2_1 2_1$	
Unit cell dimensions	$a = 7.5169(3)$ Å	$a = 90^\circ.$
	$b = 10.5263(5)$ Å	$b = 90^\circ.$
	$c = 16.5993(8)$ Å	$g = 90^\circ.$
Volume	1313.42(10) Å <sup>3</sup>	
Z	4	
Density (calculated)	1.463 Mg/m <sup>3</sup>	
Absorption coefficient	0.401 mm <sup>-1</sup>	
F(000)	600	
Crystal color	colorless	
Crystal size	0.573 x 0.369 x 0.266 mm <sup>3</sup>	
Theta range for data collection	2.291 to 29.140°	
Index ranges	$-9 \leq h \leq 10, -14 \leq k \leq 14, -22 \leq l \leq 22$	
Reflections collected	16368	
Independent reflections	3338 [R(int) = 0.0221]	
Completeness to theta = 25.500°	99.9 %	
Absorption correction	Numerical	
Max. and min. transmission	0.9277 and 0.8316	
Refinement method	Full-matrix least-squares on F <sup>2</sup>	
Data / restraints / parameters	3338 / 0 / 180	
Goodness-of-fit on F <sup>2</sup>	1.076	
Final R indices [I>2sigma(I) = 3263 data]	R1 = 0.0275, wR2 = 0.0754	
R indices (all data, 0.73Å)	R1 = 0.0283, wR2 = 0.0762	

Absolute structure parameter	0.040(17)
Largest diff. peak and hole	0.529 and -0.343 e. $\text{\AA}^{-3}$

Table 4.12. Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for (R)-4.3e. U(eq) is defined as one third of the trace of the orthogonalized  $U^{ij}$  tensor.

	x	y	z	U(eq)
S(1)	3023(1)	2396(1)	1085(1)	14(1)
S(2)	2903(1)	6156(1)	301(1)	14(1)
O(1)	1668(2)	2128(2)	496(1)	25(1)
O(2)	4157(2)	1338(1)	1302(1)	23(1)
N(1)	4166(2)	3658(2)	827(1)	15(1)
C(1)	2139(3)	3119(2)	1943(1)	14(1)
C(2)	841(3)	2609(2)	2448(1)	20(1)
C(3)	343(3)	3334(2)	3112(1)	23(1)
C(4)	1123(3)	4508(2)	3257(1)	21(1)
C(5)	2439(3)	4994(2)	2750(1)	17(1)
C(6)	2945(3)	4280(2)	2079(1)	13(1)
C(7)	4410(3)	4602(2)	1480(1)	12(1)
C(8)	6241(3)	4446(2)	1908(1)	15(1)
C(9)	7722(3)	4540(2)	1342(1)	18(1)
C(10)	8858(3)	4545(2)	845(2)	24(1)
C(11)	4195(2)	5919(2)	1128(1)	12(1)
C(12)	4882(2)	7120(2)	1456(1)	23(1)
C(13)	4238(3)	8128(2)	940(1)	21(1)
C(14)	3182(3)	7740(2)	320(1)	22(1)

Table 4.13. Bond lengths [ $\text{\AA}$ ] and angles [ $^\circ$ ] for (*R*)-**4.3e**.

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S(1)-O(1)	1.4387(15)
S(1)-O(2)	1.4478(16)
S(1)-N(1)	1.6393(18)
S(1)-C(1)	1.747(2)
S(2)-C(14)	1.682(2)
S(2)-C(11)	1.6990(19)
N(1)-C(7)	1.482(2)
C(1)-C(6)	1.382(3)
C(1)-C(2)	1.394(3)
C(2)-C(3)	1.392(3)
C(3)-C(4)	1.389(3)
C(4)-C(5)	1.396(3)
C(5)-C(6)	1.396(3)
C(6)-C(7)	1.522(3)
C(7)-C(11)	1.513(3)
C(7)-C(8)	1.557(3)
C(8)-C(9)	1.460(3)
C(9)-C(10)	1.187(3)
C(11)-C(12)	1.470(2)
C(12)-C(13)	1.447(3)
C(13)-C(14)	1.363(3)
O(1)-S(1)-O(2)	115.80(10)
O(1)-S(1)-N(1)	110.63(10)
O(2)-S(1)-N(1)	112.31(10)
O(1)-S(1)-C(1)	111.70(10)
O(2)-S(1)-C(1)	110.87(9)
N(1)-S(1)-C(1)	93.38(9)

C(14)-S(2)-C(11)	93.42(10)
C(7)-N(1)-S(1)	114.66(13)
C(6)-C(1)-C(2)	123.30(19)
C(6)-C(1)-S(1)	110.61(14)
C(2)-C(1)-S(1)	126.07(16)
C(3)-C(2)-C(1)	117.0(2)
C(4)-C(3)-C(2)	120.8(2)
C(3)-C(4)-C(5)	121.4(2)
C(6)-C(5)-C(4)	118.5(2)
C(1)-C(6)-C(5)	119.15(18)
C(1)-C(6)-C(7)	114.06(16)
C(5)-C(6)-C(7)	126.69(18)
N(1)-C(7)-C(11)	108.54(15)
N(1)-C(7)-C(6)	103.80(15)
C(11)-C(7)-C(6)	112.32(15)
N(1)-C(7)-C(8)	111.86(16)
C(11)-C(7)-C(8)	111.52(16)
C(6)-C(7)-C(8)	108.58(15)
C(9)-C(8)-C(7)	111.95(16)
C(10)-C(9)-C(8)	174.7(2)
C(12)-C(11)-C(7)	127.37(16)
C(12)-C(11)-S(2)	111.95(13)
C(7)-C(11)-S(2)	120.48(14)
C(13)-C(12)-C(11)	107.10(15)
C(14)-C(13)-C(12)	114.98(18)
C(13)-C(14)-S(2)	112.53(16)

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Table 4.14. Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for (R)-**4.3e**. The anisotropic displacement factor exponent takes the form:  $-2p^2[ h^2 a^{*2}U^{11} + \dots + 2 h k a^{*} b^{*} U^{12} ]$

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{23}$	$U^{13}$	$U^{12}$
S(1)	15(1)	12(1)	16(1)	-1(1)	-2(1)	0(1)
S(2)	20(1)	11(1)	12(1)	3(1)	-5(1)	-2(1)
O(1)	21(1)	31(1)	24(1)	-9(1)	-7(1)	-2(1)
O(2)	25(1)	14(1)	31(1)	2(1)	0(1)	4(1)
N(1)	18(1)	12(1)	14(1)	-2(1)	3(1)	-1(1)
C(1)	13(1)	15(1)	15(1)	2(1)	-1(1)	2(1)
C(2)	15(1)	23(1)	23(1)	7(1)	-1(1)	-4(1)
C(3)	17(1)	32(1)	21(1)	11(1)	4(1)	1(1)
C(4)	20(1)	30(1)	13(1)	3(1)	3(1)	7(1)
C(5)	17(1)	19(1)	14(1)	0(1)	0(1)	3(1)
C(6)	11(1)	15(1)	12(1)	3(1)	-2(1)	2(1)
C(7)	13(1)	11(1)	12(1)	-1(1)	1(1)	0(1)
C(8)	13(1)	16(1)	16(1)	3(1)	-1(1)	1(1)
C(9)	15(1)	17(1)	22(1)	5(1)	-5(1)	1(1)
C(10)	15(1)	27(1)	29(1)	7(1)	2(1)	2(1)
C(11)	11(1)	13(1)	13(1)	2(1)	1(1)	1(1)
C(12)	21(1)	24(1)	24(1)	7(1)	1(1)	2(1)
C(13)	22(1)	13(1)	28(1)	-2(1)	8(1)	-1(1)
C(14)	20(1)	22(1)	22(1)	10(1)	3(1)	2(1)

Table 4.15. Hydrogen coordinates ( $\times 10^4$ ) and isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ )  
for (R)-4.3e.

	x	y	z	U(eq)
H(1)	4980(40)	3440(30)	590(20)	34(9)
H(2B)	321	1803	2344	24
H(3A)	-542	3022	3470	28
H(4A)	754	4991	3711	25
H(5A)	2977	5793	2858	20
H(8A)	6368	5112	2325	18
H(8B)	6283	3610	2180	18
H(10)	9730(50)	4590(40)	510(20)	55(11)
H(12A)	5610	7216	1920	27
H(13A)	4530	8995	1027	25
H(14A)	2661	8303	-61	26

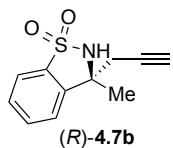
Table 4.16. Hydrogen bonds for (*R*)-**4.3e** [Å and °].

D-H...A	d(D-H)	d(H...A)	d(D...A)	$\angle$ (DHA)
N(1)-H(1)...O(1)#1	0.76(3)	2.28(3)	3.008(2)	159(3)
C(8)-H(8A)...O(2)#2	0.99	2.65	3.590(3)	158.9
C(13)-H(13A)...O(2)#3	0.95	2.52	3.432(3)	160.2

Symmetry transformations used to generate equivalent atoms:

#1 x+1/2,-y+1/2,-z #2 -x+1,y+1/2,-z+1/2 #3 x,y+1,z

C. X-ray Data Collection, Structure Solution and Refinement for (R)-4.7b:



A single crystal was grown from Et<sub>2</sub>O with slow diffusion of pentanes at room temperature. A colorless crystal of approximate dimensions 0.284 x 0.299 x 0.489 mm was mounted on a glass fiber and transferred to a Bruker SMART APEX II diffractometer. The APEX2<sup>37</sup> program package was used to determine the unit-cell parameters and for data collection (10 sec/frame scan time for a sphere of diffraction data). The raw frame data was processed using SAINT<sup>38</sup> and SADABS<sup>39</sup> to yield the reflection data file. Subsequent calculations were carried out using the SHELXTL<sup>40</sup> program. The diffraction symmetry was *mmm* and the systematic absences were consistent with the orthorhombic space group *P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>* that was later determined to be correct.

The structure was solved by direct methods and refined on F<sup>2</sup> by full-matrix least-squares techniques. The analytical scattering factors<sup>41</sup> for neutral atoms were used throughout the analysis. Hydrogen atoms were located from a difference-Fourier map and refined (x,y,z and U<sub>iso</sub>).

At convergence, wR2 = 0.0679 and Goof = 1.058 for 180 variables refined against 2571 data (0.74 Å), R1 = 0.0259 for those 2521 data with I > 2.0σ(I). The absolute structure was assigned by refinement of the Flack parameter.<sup>42</sup>

**Definitions:**

$$wR2 = [\sum[w(F_o^2 - F_c^2)^2] / \sum[w(F_o^2)^2]]^{1/2}$$

$$R1 = \sum||F_o - F_c|| / \sum|F_o|$$

Goof = S =  $[\sum[w(F_o^2 - F_c^2)^2] / (n-p)]^{1/2}$  where n is the number of reflections and p is the total number of parameters refined.

The thermal ellipsoid plot is shown at the 50% probability level.

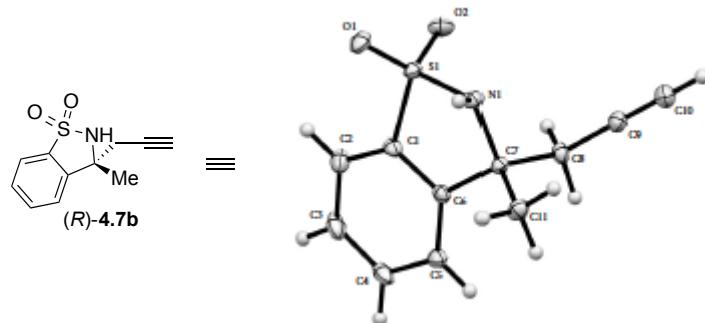


Table 4.17. Crystal data and structure refinement for (*R*)-**4.7b**.

Identification code	erj26 (Charlotte Osborne)	
Empirical formula	$C_{11} H_{11} N O_2 S$	
Formula weight	221.27	
Temperature	133(2) K	
Wavelength	0.71073 Å	
Crystal system	Orthorhombic	
Space group	$P2_1 2_1 2_1$	
Unit cell dimensions	$a = 8.0916(6)$ Å	$a = 90^\circ.$
	$b = 9.4218(7)$ Å	$b = 90^\circ.$
	$c = 13.8016(10)$ Å	$g = 90^\circ.$
Volume	1052.20(13) Å <sup>3</sup>	
Z	4	
Density (calculated)	1.397 Mg/m <sup>3</sup>	
Absorption coefficient	0.285 mm <sup>-1</sup>	
F(000)	464	
Crystal color	colorless	
Crystal size	0.489 x 0.299 x 0.284 mm <sup>3</sup>	
Theta range for data collection	2.617 to 28.724°	
Index ranges	$-10 \leq h \leq 10, -12 \leq k \leq 12, -18 \leq l \leq 18$	
Reflections collected	12648	
Independent reflections	2571 [R(int) = 0.0277]	
Completeness to theta = 25.500°	99.9 %	
Absorption correction	Numerical	
Max. and min. transmission	1.0000 and 0.8521	
Refinement method	Full-matrix least-squares on F <sup>2</sup>	
Data / restraints / parameters	2571 / 0 / 180	
Goodness-of-fit on F <sup>2</sup>	1.058	
Final R indices [I>2sigma(I) = 2521 data]	R1 = 0.0259, wR2 = 0.0672	

R indices (all data, 0.74Å) R1 = 0.0265, wR2 = 0.0679  
Absolute structure parameter 0.01(3)  
Largest diff. peak and hole 0.314 and -0.279 e.Å<sup>-3</sup>

Table 4.18. Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ )for (R)-4.7b. U(eq) is defined as one third of the trace of the orthogonalized  $U^{ij}$  tensor.

	x	y	z	U(eq)
S(1)	995(1)	3781(1)	7431(1)	14(1)
O(1)	639(2)	4308(2)	8386(1)	21(1)
O(2)	360(2)	2389(1)	7201(1)	22(1)
N(1)	370(2)	4924(2)	6611(1)	14(1)
C(1)	3091(2)	3848(2)	7115(1)	15(1)
C(2)	4403(2)	3209(2)	7592(1)	22(1)
C(3)	5950(3)	3368(2)	7180(1)	26(1)
C(4)	6157(2)	4107(2)	6320(1)	24(1)
C(5)	4813(2)	4700(2)	5835(1)	19(1)
C(6)	3250(2)	4566(2)	6245(1)	14(1)
C(7)	1612(2)	5056(2)	5812(1)	14(1)
C(8)	1157(2)	4018(2)	4982(1)	17(1)
C(9)	-448(2)	4300(2)	4531(1)	18(1)
C(10)	-1745(2)	4511(2)	4149(1)	23(1)
C(11)	1642(3)	6587(2)	5455(1)	19(1)

Table 4.19. Bond lengths [ $\text{\AA}$ ] and angles [ $^\circ$ ] for (*R*)-**4.7b**.

---

S(1)-O(1)	1.4377(13)
S(1)-O(2)	1.4438(14)
S(1)-N(1)	1.6418(16)
S(1)-C(1)	1.7524(17)
N(1)-C(7)	1.497(2)
N(1)-H(1)	0.77(3)
C(1)-C(6)	1.384(2)
C(1)-C(2)	1.388(3)
C(2)-C(3)	1.384(3)
C(2)-H(2)	0.93(3)
C(3)-C(4)	1.386(3)
C(3)-H(3)	0.91(3)
C(4)-C(5)	1.394(3)
C(4)-H(4)	0.95(3)
C(5)-C(6)	1.392(2)
C(5)-H(5)	0.94(3)
C(6)-C(7)	1.525(2)
C(7)-C(11)	1.525(2)
C(7)-C(8)	1.550(2)
C(8)-C(9)	1.464(3)
C(8)-H(8A)	1.01(2)
C(8)-H(8B)	0.95(2)
C(9)-C(10)	1.192(3)
C(10)-H(10)	0.92(3)
C(11)-H(11A)	0.98(3)
C(11)-H(11B)	0.93(3)
C(11)-H(11C)	0.97(3)

O(1)-S(1)-O(2)	116.34(8)
O(1)-S(1)-N(1)	110.12(8)
O(2)-S(1)-N(1)	109.58(8)
O(1)-S(1)-C(1)	114.17(8)
O(2)-S(1)-C(1)	108.82(9)
N(1)-S(1)-C(1)	95.88(8)
C(7)-N(1)-S(1)	110.82(12)
C(7)-N(1)-H(1)	113.8(19)
S(1)-N(1)-H(1)	107.4(18)
C(6)-C(1)-C(2)	123.60(16)
C(6)-C(1)-S(1)	108.88(13)
C(2)-C(1)-S(1)	127.35(14)
C(3)-C(2)-C(1)	116.72(17)
C(3)-C(2)-H(2)	120.5(15)
C(1)-C(2)-H(2)	122.6(15)
C(2)-C(3)-C(4)	121.09(18)
C(2)-C(3)-H(3)	121.4(17)
C(4)-C(3)-H(3)	117.5(17)
C(3)-C(4)-C(5)	121.24(18)
C(3)-C(4)-H(4)	120.9(17)
C(5)-C(4)-H(4)	117.9(17)
C(6)-C(5)-C(4)	118.52(17)
C(6)-C(5)-H(5)	121.0(16)
C(4)-C(5)-H(5)	120.5(16)
C(1)-C(6)-C(5)	118.79(16)
C(1)-C(6)-C(7)	114.00(15)
C(5)-C(6)-C(7)	127.10(15)
N(1)-C(7)-C(11)	109.09(14)
N(1)-C(7)-C(6)	105.65(13)
C(11)-C(7)-C(6)	113.48(15)

N(1)-C(7)-C(8)	109.46(14)
C(11)-C(7)-C(8)	111.19(14)
C(6)-C(7)-C(8)	107.79(14)
C(9)-C(8)-C(7)	114.22(15)
C(9)-C(8)-H(8A)	109.7(14)
C(7)-C(8)-H(8A)	108.6(14)
C(9)-C(8)-H(8B)	109.7(15)
C(7)-C(8)-H(8B)	109.1(14)
H(8A)-C(8)-H(8B)	105.1(19)
C(10)-C(9)-C(8)	178.6(2)
C(9)-C(10)-H(10)	177.4(18)
C(7)-C(11)-H(11A)	109.3(16)
C(7)-C(11)-H(11B)	114.0(15)
H(11A)-C(11)-H(11B)	109(2)
C(7)-C(11)-H(11C)	109.1(17)
H(11A)-C(11)-H(11C)	107(2)
H(11B)-C(11)-H(11C)	108(2)

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Table 4.20. Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for (R)-**4.7b**. The anisotropic displacement factor exponent takes the form:  $-2p^2 [ h^2 a^*{}^2 U^{11} + \dots + 2 h k a^* b^* U^{12} ]$

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{23}$	$U^{13}$	$U^{12}$
S(1)	13(1)	16(1)	14(1)	1(1)	3(1)	0(1)
O(1)	21(1)	27(1)	14(1)	1(1)	5(1)	4(1)
O(2)	22(1)	17(1)	26(1)	3(1)	4(1)	-4(1)
N(1)	14(1)	15(1)	14(1)	0(1)	2(1)	1(1)
C(1)	12(1)	19(1)	15(1)	-4(1)	1(1)	0(1)
C(2)	20(1)	30(1)	15(1)	-1(1)	-2(1)	5(1)
C(3)	15(1)	41(1)	22(1)	-6(1)	-6(1)	6(1)
C(4)	12(1)	36(1)	23(1)	-9(1)	1(1)	-2(1)
C(5)	16(1)	25(1)	16(1)	-3(1)	3(1)	-4(1)
C(6)	14(1)	16(1)	14(1)	-4(1)	-1(1)	-2(1)
C(7)	14(1)	16(1)	12(1)	0(1)	2(1)	-1(1)
C(8)	18(1)	19(1)	14(1)	-3(1)	-1(1)	0(1)
C(9)	20(1)	20(1)	14(1)	-1(1)	1(1)	-2(1)
C(10)	21(1)	28(1)	19(1)	1(1)	-3(1)	-3(1)
C(11)	23(1)	16(1)	18(1)	2(1)	2(1)	-1(1)

Table 4.21. Hydrogen coordinates ( $\times 10^4$ ) and isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for (R)-**4.7b**.

	x	y	z	U(eq)
H(1)	190(30)	5630(30)	6875(19)	20(6)
H(2)	4280(30)	2750(30)	8185(18)	23(6)
H(3)	6870(30)	2970(30)	7450(20)	28(6)
H(4)	7230(30)	4240(30)	6048(19)	32(7)
H(5)	4960(30)	5170(30)	5240(20)	34(7)
H(8A)	2060(30)	4060(30)	4474(18)	21(6)
H(8B)	1180(30)	3070(20)	5224(17)	18(5)
H(10)	-2740(40)	4630(30)	3840(20)	36(7)
H(11A)	2430(30)	6670(30)	4925(19)	28(6)
H(11B)	1900(30)	7240(30)	5932(18)	23(6)
H(11C)	570(40)	6830(30)	5200(20)	36(7)

Table 4.22. Torsion angles [°] for (*R*)-**4.7b**.

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O(1)-S(1)-N(1)-C(7)	138.66(12)
O(2)-S(1)-N(1)-C(7)	-92.16(13)
C(1)-S(1)-N(1)-C(7)	20.23(13)
O(1)-S(1)-C(1)-C(6)	-126.43(12)
O(2)-S(1)-C(1)-C(6)	101.75(14)
N(1)-S(1)-C(1)-C(6)	-11.27(14)
O(1)-S(1)-C(1)-C(2)	58.3(2)
O(2)-S(1)-C(1)-C(2)	-73.50(18)
N(1)-S(1)-C(1)-C(2)	173.48(17)
C(6)-C(1)-C(2)-C(3)	2.9(3)
S(1)-C(1)-C(2)-C(3)	177.46(15)
C(1)-C(2)-C(3)-C(4)	-1.4(3)
C(2)-C(3)-C(4)-C(5)	-0.7(3)
C(3)-C(4)-C(5)-C(6)	1.6(3)
C(2)-C(1)-C(6)-C(5)	-2.1(3)
S(1)-C(1)-C(6)-C(5)	-177.56(13)
C(2)-C(1)-C(6)-C(7)	174.41(17)
S(1)-C(1)-C(6)-C(7)	-1.05(18)
C(4)-C(5)-C(6)-C(1)	-0.2(3)
C(4)-C(5)-C(6)-C(7)	-176.18(17)
S(1)-N(1)-C(7)-C(11)	-144.84(13)
S(1)-N(1)-C(7)-C(6)	-22.51(16)
S(1)-N(1)-C(7)-C(8)	93.30(15)
C(1)-C(6)-C(7)-N(1)	14.5(2)
C(5)-C(6)-C(7)-N(1)	-169.29(17)
C(1)-C(6)-C(7)-C(11)	134.01(16)
C(5)-C(6)-C(7)-C(11)	-49.8(2)
C(1)-C(6)-C(7)-C(8)	-102.40(17)

C(5)-C(6)-C(7)-C(8)	73.8(2)
N(1)-C(7)-C(8)-C(9)	61.96(19)
C(11)-C(7)-C(8)-C(9)	-58.6(2)
C(6)-C(7)-C(8)-C(9)	176.40(14)

---

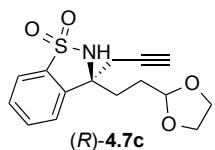
Table 4.23. Hydrogen bonds for (*R*)-**4.7b** [Å and °].

D-H...A	d(D-H)	d(H...A)	d(D...A)	$\angle$ (DHA)
N(1)-H(1)...O(2) <sup>#1</sup>	0.77(3)	2.14(3)	2.903(2)	171(3)

Symmetry transformations used to generate equivalent atoms:

#1 -x,y+1/2,-z+3/2

D. X-ray Data Collection, Structure Solution and Refinement for (R)-4.7c:



A single crystal was grown from EtOAc with slow diffusion of pentanes at room temperature. A colorless crystal of approximate dimensions 0.288 x 0.160 x 0.108 mm was mounted on a glass fiber and transferred to a Bruker SMART APEX II diffractometer. The APEX2<sup>37</sup> program package was used to determine the unit-cell parameters and for data collection (60 sec/frame scan time for a sphere of diffraction data). The raw frame data was processed using SAINT<sup>38</sup> and SADABS<sup>39</sup> to yield the reflection data file. Subsequent calculations were carried out using the SHELXTL<sup>40</sup> program. The diffraction symmetry was *mmm* and the systematic absences were consistent with the orthorhombic space group *P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>* that was later determined to be correct.

The structure was solved by direct methods and refined on F<sup>2</sup> by full-matrix least-squares techniques. The analytical scattering factors<sup>41</sup> for neutral atoms were used throughout the analysis. Hydrogen atoms were located from a difference-Fourier map and refined (x,y,z and U<sub>iso</sub>).

At convergence, wR2 = 0.0706 and Goof = 1.040 for 258 variables refined against 3562 data (0.75 Å), R1 = 0.0294 for those 3283 data with I > 2.0σ(I). The absolute structure was assigned by refinement of the Flack parameter.<sup>42</sup>

**Definitions:**

$$wR2 = [\sum[w(F_o^2 - F_c^2)^2] / \sum[w(F_o^2)^2]]^{1/2}$$

$$R1 = \sum|F_o - |F_c|| / \sum|F_o|$$

Goof = S =  $[\sum[w(F_o^2 - F_c^2)^2] / (n-p)]^{1/2}$  where n is the number of reflections and p is the total number of parameters refined.

The thermal ellipsoid plot is shown at the 50% probability level.

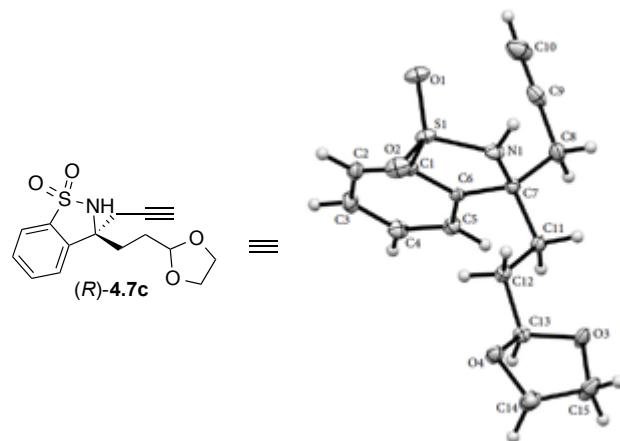


Table 4.24. Crystal data and structure refinement for (*R*)-4.7c.

Identification code	erj22 (Charlotte Osborne)	
Empirical formula	$C_{15}H_{17}NO_4S$	
Formula weight	307.35	
Temperature	133(2) K	
Wavelength	0.71073 Å	
Crystal system	Orthorhombic	
Space group	$P2_12_12_1$	
Unit cell dimensions	$a = 7.7171(5)$ Å	$a = 90^\circ$ .
	$b = 7.9486(5)$ Å	$b = 90^\circ$ .
	$c = 23.8160(14)$ Å	$g = 90^\circ$ .
Volume	1460.88(16) Å <sup>3</sup>	
Z	4	
Density (calculated)	1.397 Mg/m <sup>3</sup>	
Absorption coefficient	0.237 mm <sup>-1</sup>	
F(000)	648	
Crystal color	colorless	
Crystal size	0.288 x 0.160 x 0.108 mm <sup>3</sup>	
Theta range for data collection	1.710 to 28.288°	
Index ranges	$-10 \leq h \leq 10, -10 \leq k \leq 10, -31 \leq l \leq 31$	
Reflections collected	17581	
Independent reflections	3562 [R(int) = 0.0306]	
Completeness to theta = 25.500°	100.0 %	
Absorption correction	Numerical	
Max. and min. transmission	0.9980 and 0.9400	
Refinement method	Full-matrix least-squares on F <sup>2</sup>	
Data / restraints / parameters	3562 / 0 / 258	
Goodness-of-fit on F <sup>2</sup>	1.040	
Final R indices [I>2sigma(I) = 3283 data]	R1 = 0.0294, wR2 = 0.0678	
R indices (all data, 0.75 Å)	R1 = 0.0342, wR2 = 0.0706	

Absolute structure parameter	0.07(2)
Largest diff. peak and hole	0.316 and -0.247 e. $\text{\AA}^{-3}$

Table 4.25. Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for (R)-4.7c. U(eq) is defined as one third of the trace of the orthogonalized  $U^{ij}$  tensor.

	x	y	z	U(eq)
S(1)	9619(1)	2605(1)	9054(1)	17(1)
N(1)	10456(2)	3214(2)	8456(1)	18(1)
O(1)	10861(2)	1665(2)	9377(1)	25(1)
O(2)	7974(2)	1792(2)	8972(1)	25(1)
O(3)	7309(2)	5275(2)	6830(1)	22(1)
O(4)	5065(2)	4151(2)	7301(1)	20(1)
C(1)	9266(2)	4647(2)	9298(1)	14(1)
C(2)	8545(3)	5077(3)	9810(1)	19(1)
C(3)	8250(3)	6771(3)	9908(1)	22(1)
C(4)	8676(3)	7967(3)	9505(1)	21(1)
C(5)	9407(2)	7505(2)	8996(1)	17(1)
C(6)	9706(2)	5809(2)	8890(1)	13(1)
C(7)	10489(3)	5054(2)	8361(1)	15(1)
C(8)	12367(3)	5666(3)	8281(1)	23(1)
C(9)	13497(3)	5248(3)	8757(1)	24(1)
C(10)	14373(3)	4848(3)	9138(1)	34(1)
C(11)	9414(3)	5506(3)	7836(1)	16(1)
C(12)	7675(3)	4603(3)	7816(1)	16(1)
C(13)	6485(3)	5257(3)	7365(1)	16(1)
C(14)	4400(3)	4455(4)	6752(1)	30(1)
C(15)	5922(3)	5167(4)	6424(1)	30(1)

Table 4.26. Bond lengths [ $\text{\AA}$ ] and angles [ $^\circ$ ] for (*R*)-**4.7c**.

---

S(1)-O(2)	1.4379(16)
S(1)-O(1)	1.4383(15)
S(1)-N(1)	1.6361(18)
S(1)-C(1)	1.746(2)
N(1)-C(7)	1.480(3)
O(3)-C(13)	1.425(2)
O(3)-C(15)	1.445(3)
O(4)-C(13)	1.413(2)
O(4)-C(14)	1.425(3)
C(1)-C(2)	1.383(3)
C(1)-C(6)	1.384(3)
C(2)-C(3)	1.385(3)
C(3)-C(4)	1.390(3)
C(4)-C(5)	1.387(3)
C(5)-C(6)	1.391(3)
C(6)-C(7)	1.521(3)
C(7)-C(8)	1.541(3)
C(7)-C(11)	1.543(3)
C(8)-C(9)	1.468(3)
C(9)-C(10)	1.175(3)
C(11)-C(12)	1.523(3)
C(12)-C(13)	1.505(3)
C(14)-C(15)	1.519(3)
O(2)-S(1)-O(1)	115.30(10)
O(2)-S(1)-N(1)	111.35(9)
O(1)-S(1)-N(1)	110.84(10)
O(2)-S(1)-C(1)	108.98(9)
O(1)-S(1)-C(1)	114.11(9)

N(1)-S(1)-C(1)	94.39(9)
C(7)-N(1)-S(1)	115.63(13)
C(13)-O(3)-C(15)	105.46(17)
C(13)-O(4)-C(14)	105.85(16)
C(2)-C(1)-C(6)	123.63(18)
C(2)-C(1)-S(1)	125.90(16)
C(6)-C(1)-S(1)	110.34(14)
C(1)-C(2)-C(3)	117.03(19)
C(2)-C(3)-C(4)	120.6(2)
C(5)-C(4)-C(3)	121.3(2)
C(4)-C(5)-C(6)	118.85(18)
C(1)-C(6)-C(5)	118.58(17)
C(1)-C(6)-C(7)	114.62(16)
C(5)-C(6)-C(7)	126.80(17)
N(1)-C(7)-C(6)	104.84(15)
N(1)-C(7)-C(8)	110.31(17)
C(6)-C(7)-C(8)	110.57(16)
N(1)-C(7)-C(11)	110.20(16)
C(6)-C(7)-C(11)	111.43(15)
C(8)-C(7)-C(11)	109.42(16)
C(9)-C(8)-C(7)	113.10(18)
C(10)-C(9)-C(8)	177.2(3)
C(12)-C(11)-C(7)	112.91(16)
C(13)-C(12)-C(11)	113.39(17)
O(4)-C(13)-O(3)	104.82(15)
O(4)-C(13)-C(12)	109.61(16)
O(3)-C(13)-C(12)	111.66(17)
O(4)-C(14)-C(15)	104.84(18)
O(3)-C(15)-C(14)	104.59(18)

---

Table 4.27. Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for (R)-**4.7c**. The anisotropic displacement factor exponent takes the form:  $-2p^2[ h^2 a^{*2}U^{11} + \dots + 2 h k a^{*} b^{*} U^{12} ]$

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{23}$	$U^{13}$	$U^{12}$
S(1)	19(1)	12(1)	18(1)	1(1)	-6(1)	0(1)
N(1)	21(1)	17(1)	18(1)	-4(1)	-2(1)	6(1)
O(1)	28(1)	20(1)	27(1)	4(1)	-9(1)	5(1)
O(2)	26(1)	18(1)	32(1)	1(1)	-9(1)	-7(1)
O(3)	22(1)	31(1)	12(1)	5(1)	-3(1)	-6(1)
O(4)	17(1)	26(1)	17(1)	4(1)	-4(1)	-5(1)
C(1)	14(1)	13(1)	16(1)	0(1)	-3(1)	-1(1)
C(2)	17(1)	24(1)	15(1)	3(1)	1(1)	-5(1)
C(3)	22(1)	28(1)	15(1)	-6(1)	4(1)	0(1)
C(4)	24(1)	18(1)	21(1)	-4(1)	-2(1)	4(1)
C(5)	18(1)	16(1)	18(1)	1(1)	-2(1)	-2(1)
C(6)	11(1)	17(1)	11(1)	1(1)	-3(1)	-1(1)
C(7)	14(1)	16(1)	14(1)	-1(1)	0(1)	1(1)
C(8)	16(1)	34(1)	18(1)	-2(1)	3(1)	-3(1)
C(9)	14(1)	31(1)	28(1)	-8(1)	3(1)	-1(1)
C(10)	23(1)	43(1)	34(1)	-10(1)	-8(1)	4(1)
C(11)	17(1)	19(1)	12(1)	1(1)	0(1)	-1(1)
C(12)	19(1)	17(1)	14(1)	2(1)	0(1)	-1(1)
C(13)	17(1)	15(1)	16(1)	0(1)	0(1)	0(1)
C(14)	24(1)	45(2)	22(1)	9(1)	-8(1)	-5(1)
C(15)	33(1)	37(1)	19(1)	9(1)	-10(1)	-12(1)

Table 4.28. Hydrogen coordinates ( $\times 10^4$ ) and isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ )  
for (R)-4.7c.

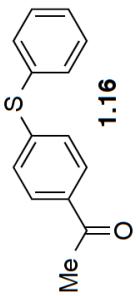
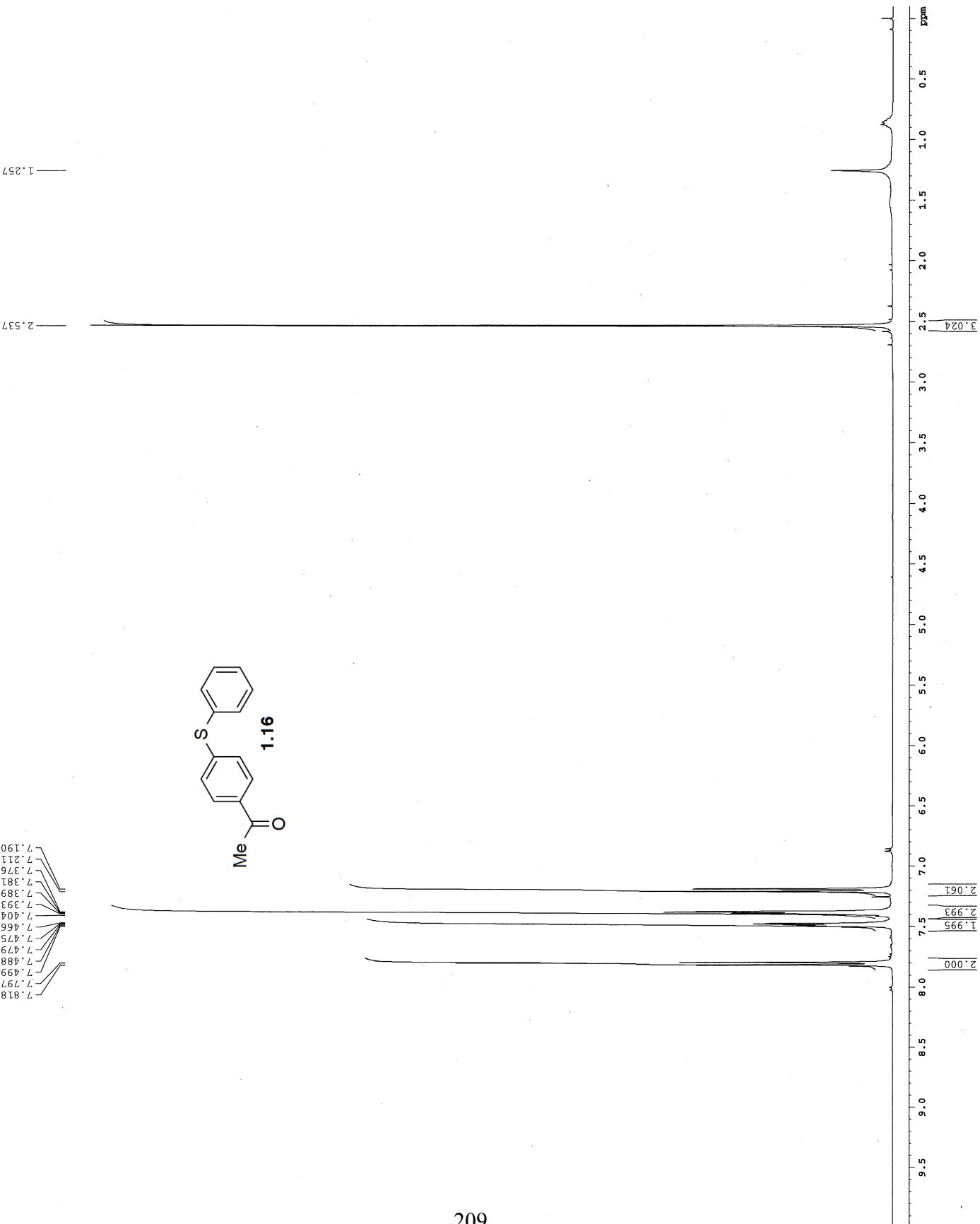
	x	y	z	U(eq)
H(1)	11280(40)	2610(40)	8381(11)	34(7)
H(2B)	8200(40)	4230(40)	10076(12)	37(8)
H(3A)	7770(30)	7090(30)	10250(10)	20(6)
H(4A)	8440(30)	9070(40)	9586(11)	28(7)
H(5A)	9700(30)	8340(30)	8715(10)	27(7)
H(8A)	12820(30)	5150(30)	7952(11)	23(6)
H(8B)	12330(30)	6880(40)	8216(10)	24(6)
H(10)	15060(40)	4580(40)	9411(13)	51(9)
H(11A)	9240(30)	6750(30)	7845(9)	21(6)
H(11B)	10110(30)	5230(30)	7509(10)	15(5)
H(12A)	7860(30)	3410(30)	7749(10)	20(6)
H(12B)	7070(30)	4720(30)	8154(11)	23(6)
H(13A)	6040(30)	6350(30)	7452(9)	12(5)
H(14A)	3410(40)	5190(40)	6780(12)	40(8)
H(14B)	4030(50)	3470(50)	6598(15)	67(11)
H(15A)	6290(40)	4470(40)	6122(14)	55(10)
H(15B)	5690(30)	6250(30)	6278(11)	27(7)

```

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  EXPNO     4
  PROCNO    1

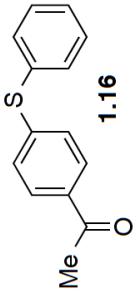
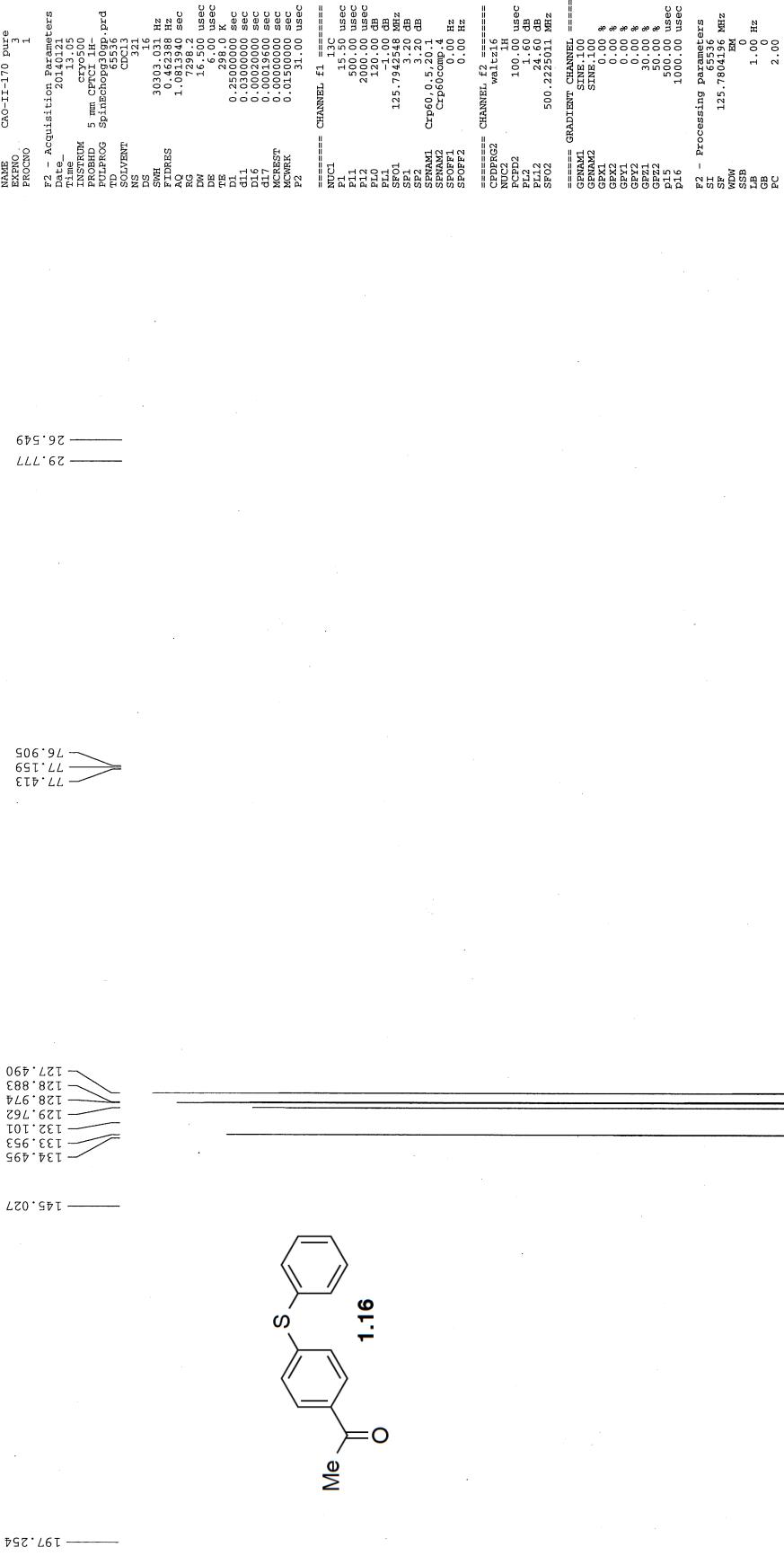
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Time: 14:26
INSTRUM: dxx400
PROBOD: 5 mm
TUPPERQ: 2530
DOLLENT: 65536
CD1: CD1
SOLENT: 8
NS: 2
DS: 2
SNR: 64.10/256
FIDRES: 0.037913
AQ: 5.1118739
RG: 78.64
DW: 74.50
DE: 4.50
TE: 290.00
T1: 0.10000000
D1: 0.00000000
MCREST: 0.01500000
MCWORK: ===== CHANNEL F1 =====
          NHC1      12.1H
          PL1      0.00.00
          SPL1     400.132809
          SFO1     400.130536
          SFI      400.130536
          SF1      400.130536
          SSB      400.130536
          SSB1     400.130536
          GB      0.30.00
          PC      2.00.00

```



1H spectrum

Z-restored spin-echo 13C spectrum with 1H decoupling

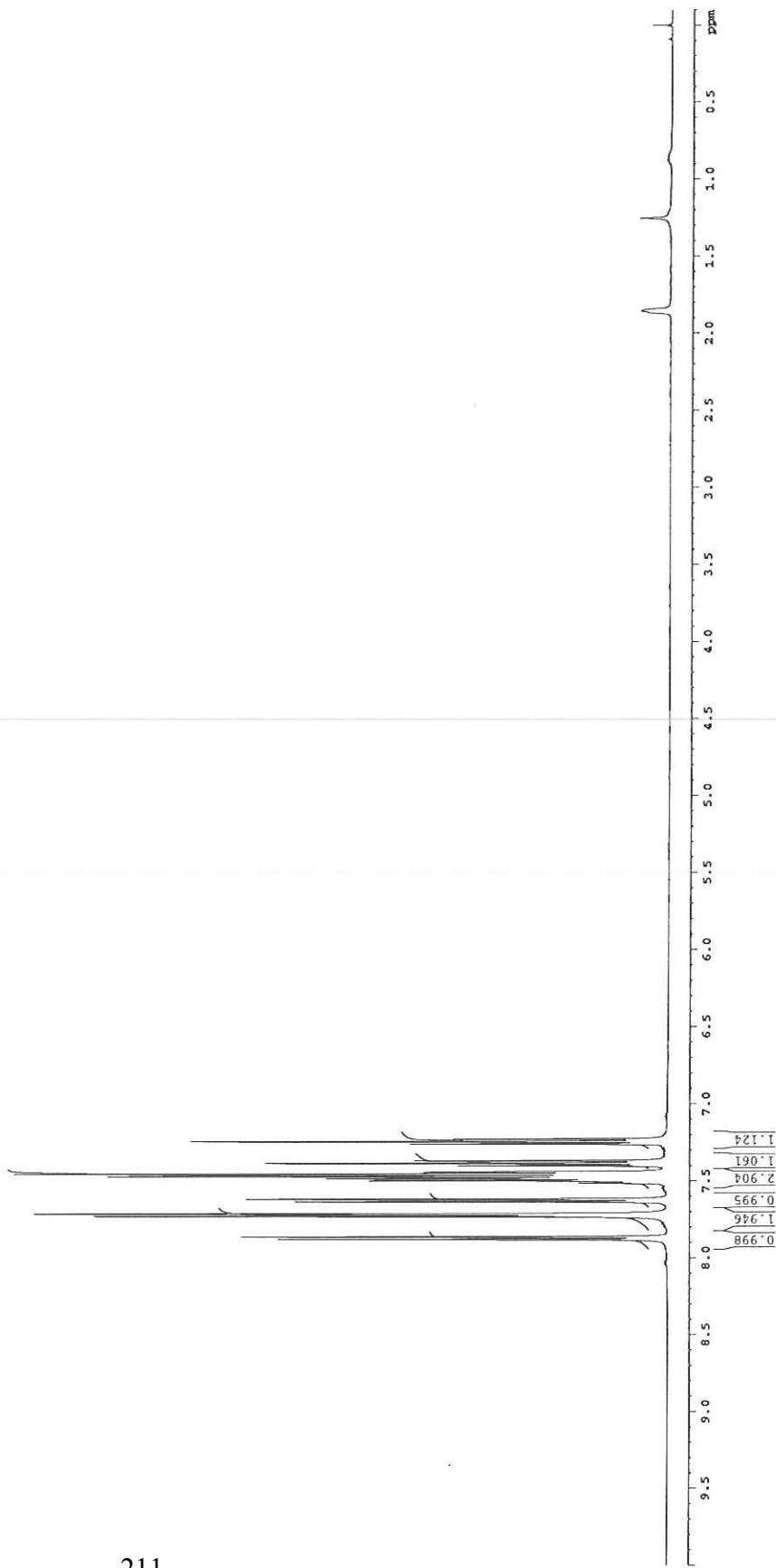
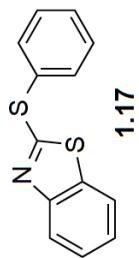


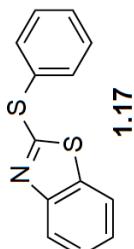
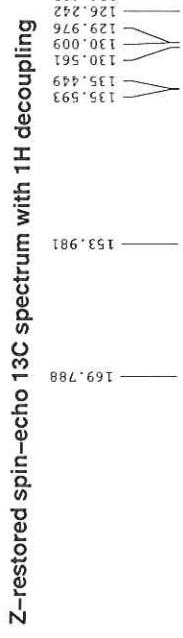
**1H spectrum**

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Current Data Parameters
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NAME   imy2 - Benzothiazole-1
PRONO
P2 - Acquisition Parameters
Date_ 20130916
Time_ 17.05
INSTRUM cryo500
PROBID 5 mm CPMR11
PULPROG 2930
TD 61728
SOLVENT D2O
NS 2
DS 8012.820 Hz
FIDRES 5.099674 sec
AQ 62.450 usec
RG 6.00 usec
TE 298.0 K
D1 0.1000000 sec
MCOUNT 65536
NCWDW 0.0000000 sec
NCWDW 0.0100000 sec
===== CHANNEL f1 =====
NUCL   1H
PL1    7.50 usec
PL11   1.00 dB
SP01   500.2235015 MHz
P2 - Processing parameters
SI 65536
SP 500.22350397 MHz
WDW SSB
SSB 0
LB 0.30 Hz
QBY 0.0
PC 4.00

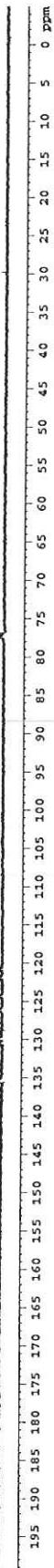
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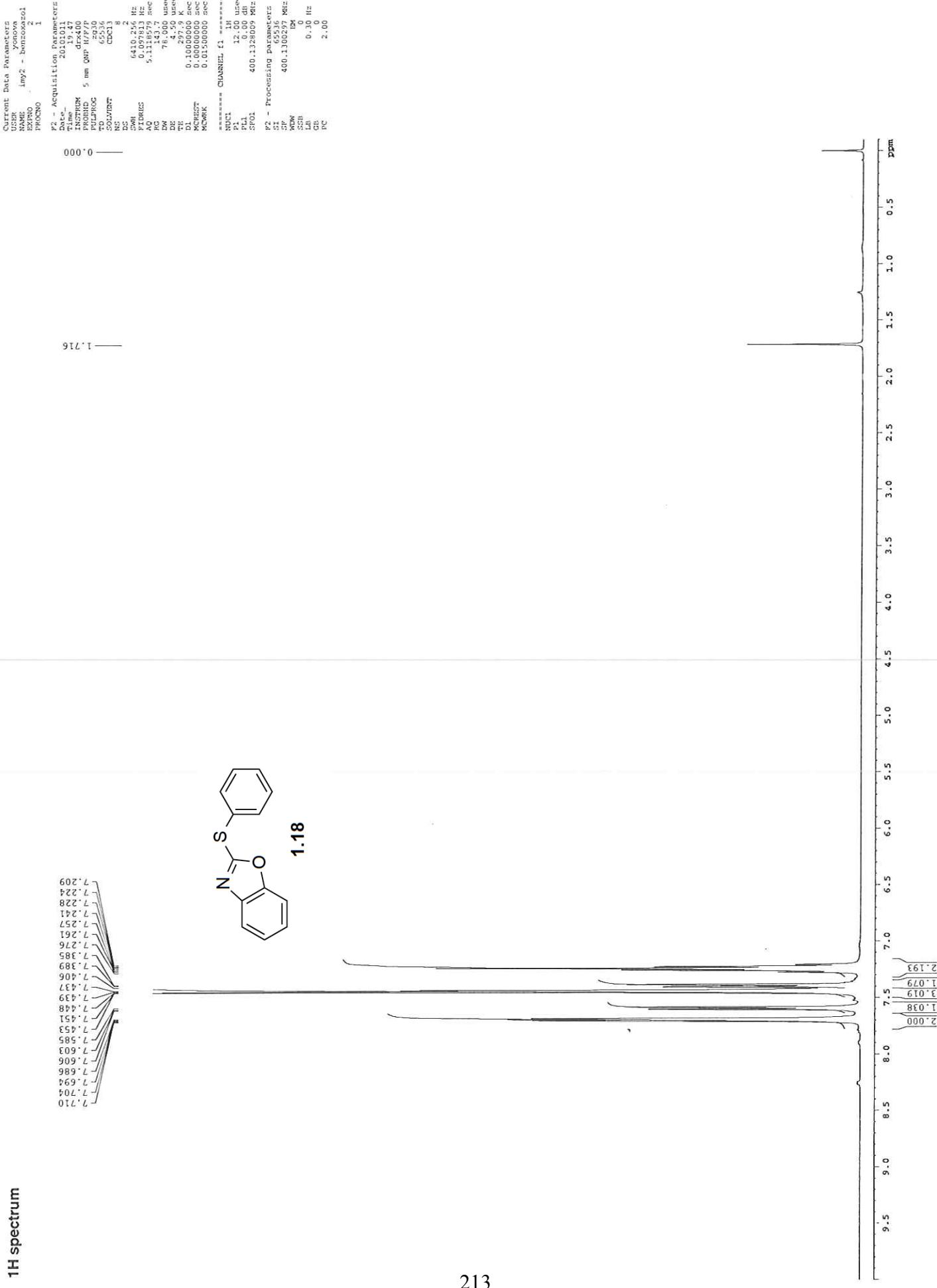


1.17

76.907  
77.161  
77.416

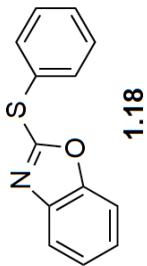


**1H spectrum**



<sup>13</sup>C spectrum with 1H decoupling

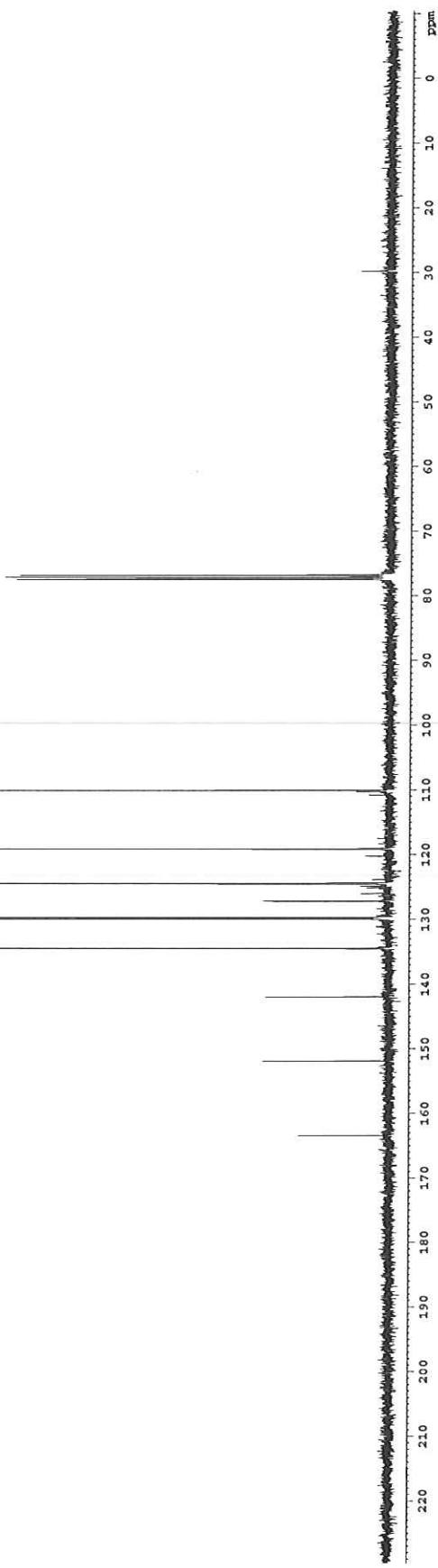
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 119.145  
 124.469  
 127.193  
 129.744  
 129.950  
 134.510  
 142.008  
 151.946  
 129.499  
 127.193  
 124.469  
 77.491  
 77.173  
 76.855  
 =====



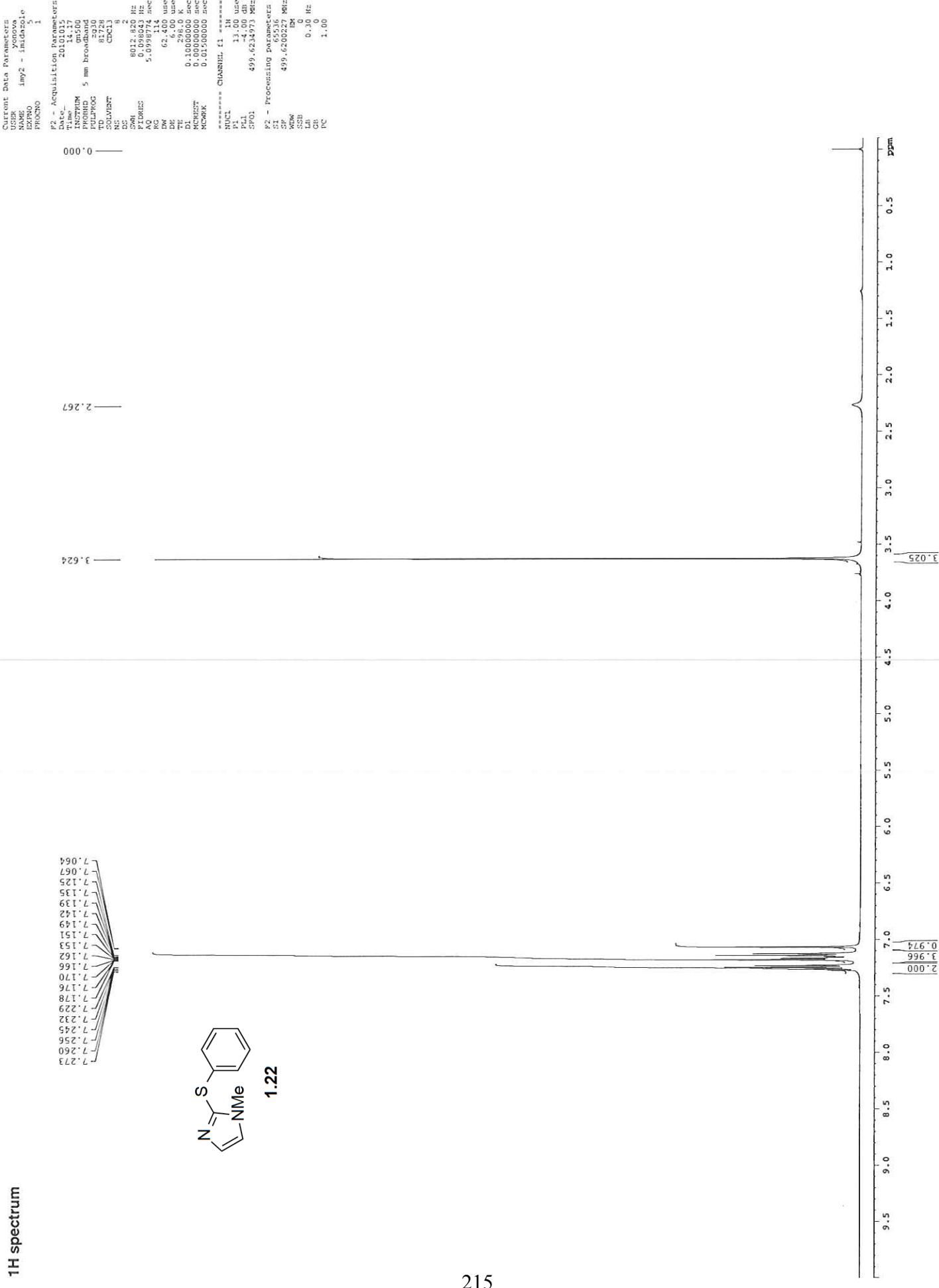
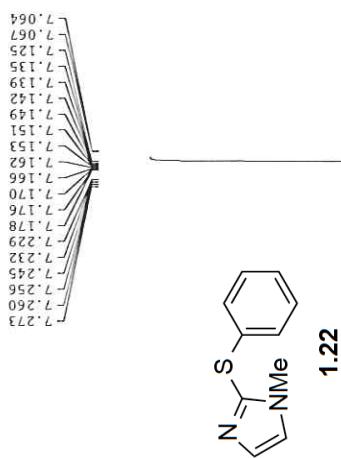
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NAME      imp2 = 217
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PROCNO    1
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Time_   3.20
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PROBODIM 5 mm QNP H7/TIP
TDPROG   20480
TD       65536
SOLVENT  CDCl3
NS       337
DS       4
SW0     24154.590 Hz
FIDRES  0.306719 Hz
AQ      1.000000 sec
RG      14590.5
DW      20.700 usec
DE      20.39 usec
TM      207.9 K
TWA    0.1000000 sec
D1      0.0000000 sec
MIXTET  0.0000000 sec
MIXNCR  0.0150000 sec
===== CHANNEL 1 =====
NUC1    13C
P1      10.75 usec
T1L    0.00 dB
SP01   100.6237954 Hz
===== CHANNEL 12 =====
CPDPG2
NUC2    1H
P1P2    90.00 usec
P1L2    0.00 dB
P2P2    1170.00
SP02   400.1328059 MHz
P2 - Processing parameters
SI      65536
SF      100.6127672 MHz
WDW    8000
SSB    0
LB     1.00
GAP   0
PC     1.00

```



**1H spectrum**



Z-restored spin-echo 13C spectrum with 1H decoupling

138,085  
134,975  
130,235  
129,321  
128,025  
126,625  
123,922

77,414  
76,906  
77,160

33,947

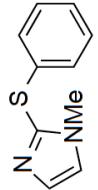
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PRDENO    1
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Date_   2010/01/05
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PROBODC 5-mm CRYP540
TD      65536
SOLVENT  CDCl3
NS      140
DW      30000.018 Hz
TR      1.0013940 sec
TE      16.500 usec
RG      6.000 usec
TEC     298.0 K
D1      0.250000 sec
P1      0.0300000 sec
D2      0.0020000 sec
D3      0.00019600 sec
MPCONT  0.01500000 sec
MCWRFK  0.01500000 sec
T2      31.00 usec
P2      31.00 usec

===== CHANNEL F1 =====
NUC1      13C
P1       15.50 usec
P2      200.00 usec
P3      120.00 dB
SP01    125.79451 MHz
SP1      3.20 dB
SP2      Crp0, 0.5, 20.1 dB
SPRM01  Crp4Comp, 4
SPRM02  Crp4Comp, 4
SP0P1   0.00 Hz
SP0P2   0.00 Hz

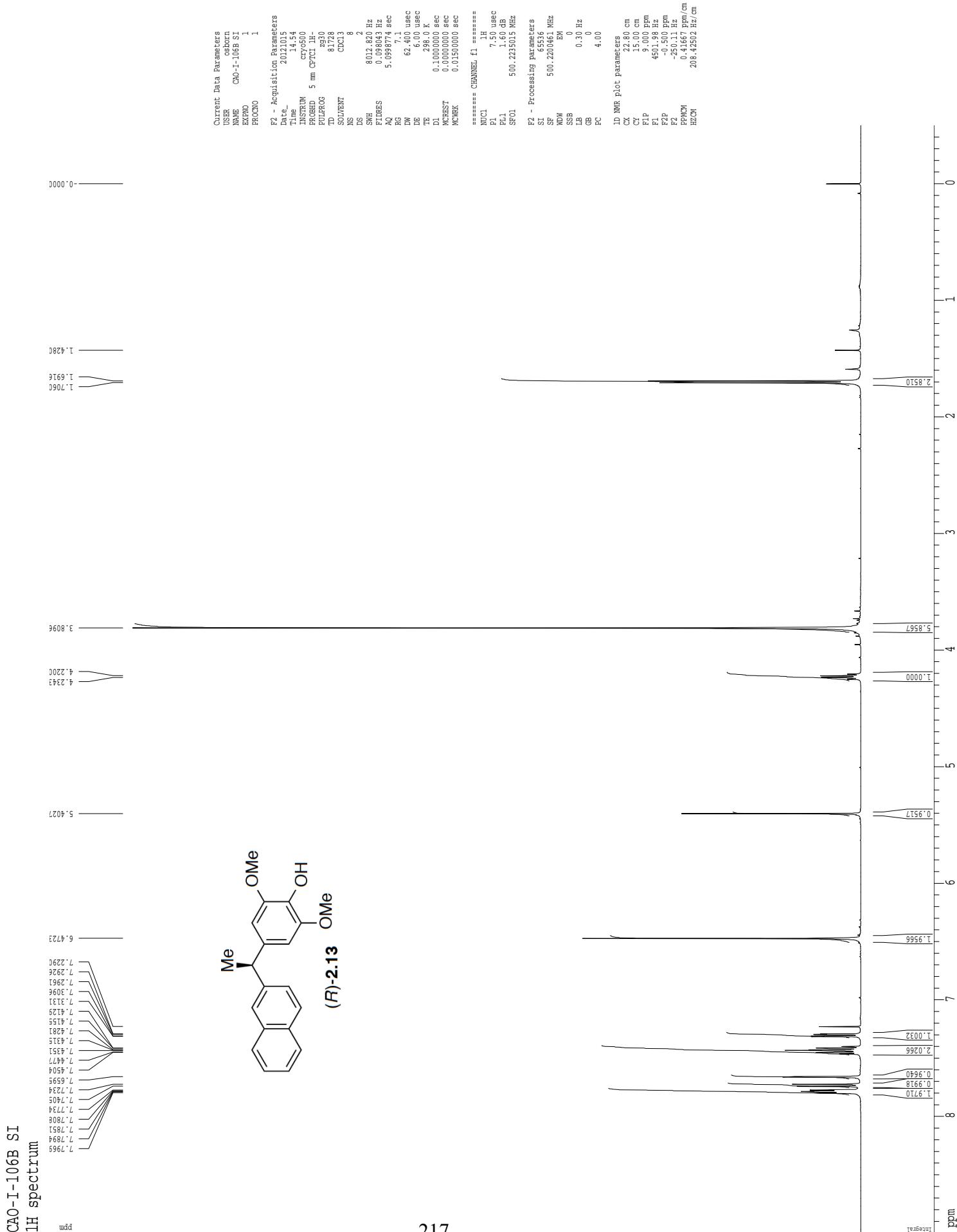
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NUC2      1H
PDPB2   100.00 usec
PDPB2   1.00 dB
PDPB2   24.00 dB
SP02    500.2225011 MHz

===== GRADIENT CHANNEL =====
GRADM1  SINE,1.00
GRADM2  SINE,1.00
GRV1   0.00 %
GRV2   0.00 %
GRV2   0.00 %
GRV2   30.00 %
GRV2   50.00 %
P15   500.00 usec
P16   1000.00 usec

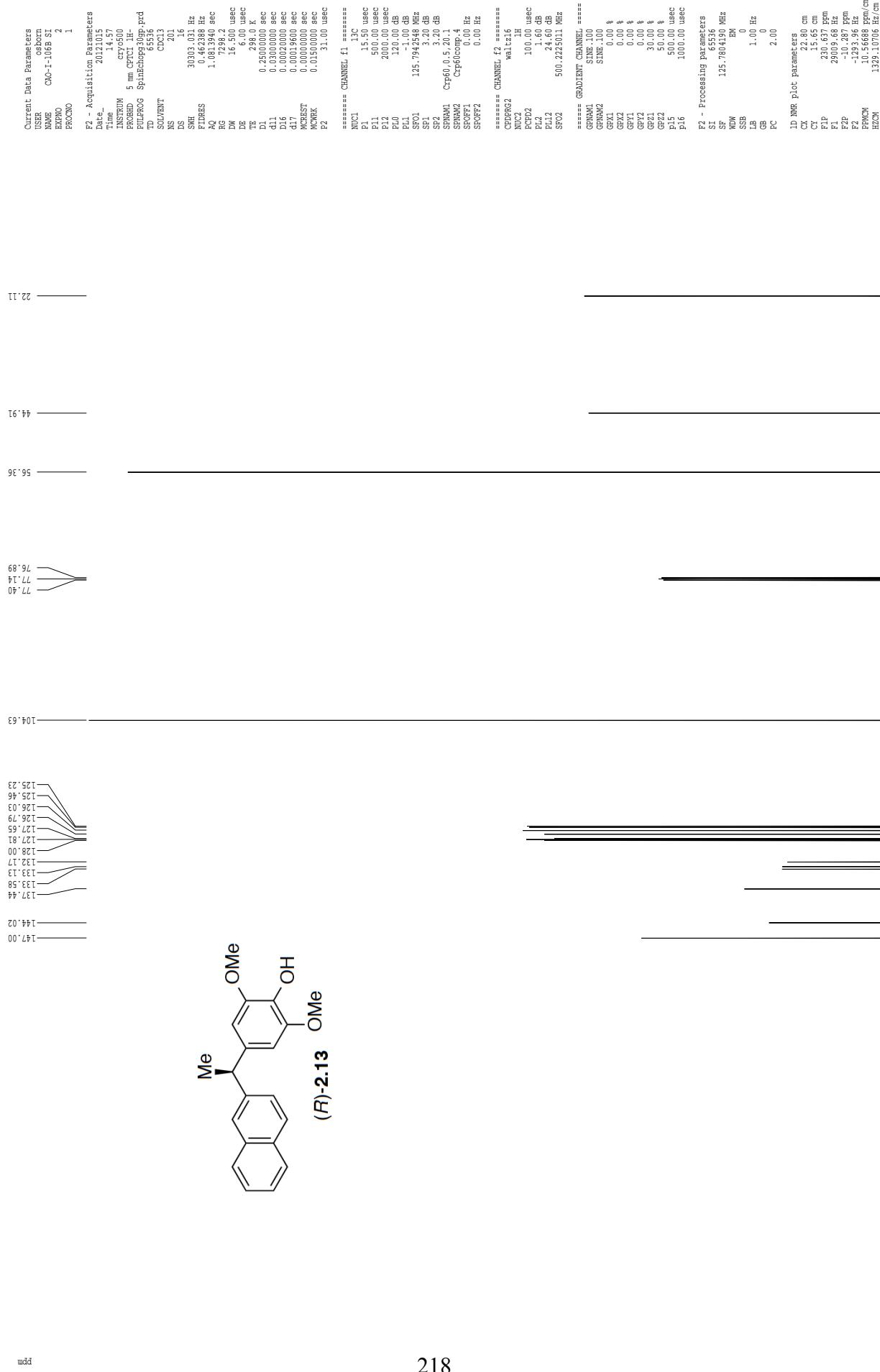
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SSB    0
LB      1.00 Hz
CD      0
PC      2.00
```



1.22

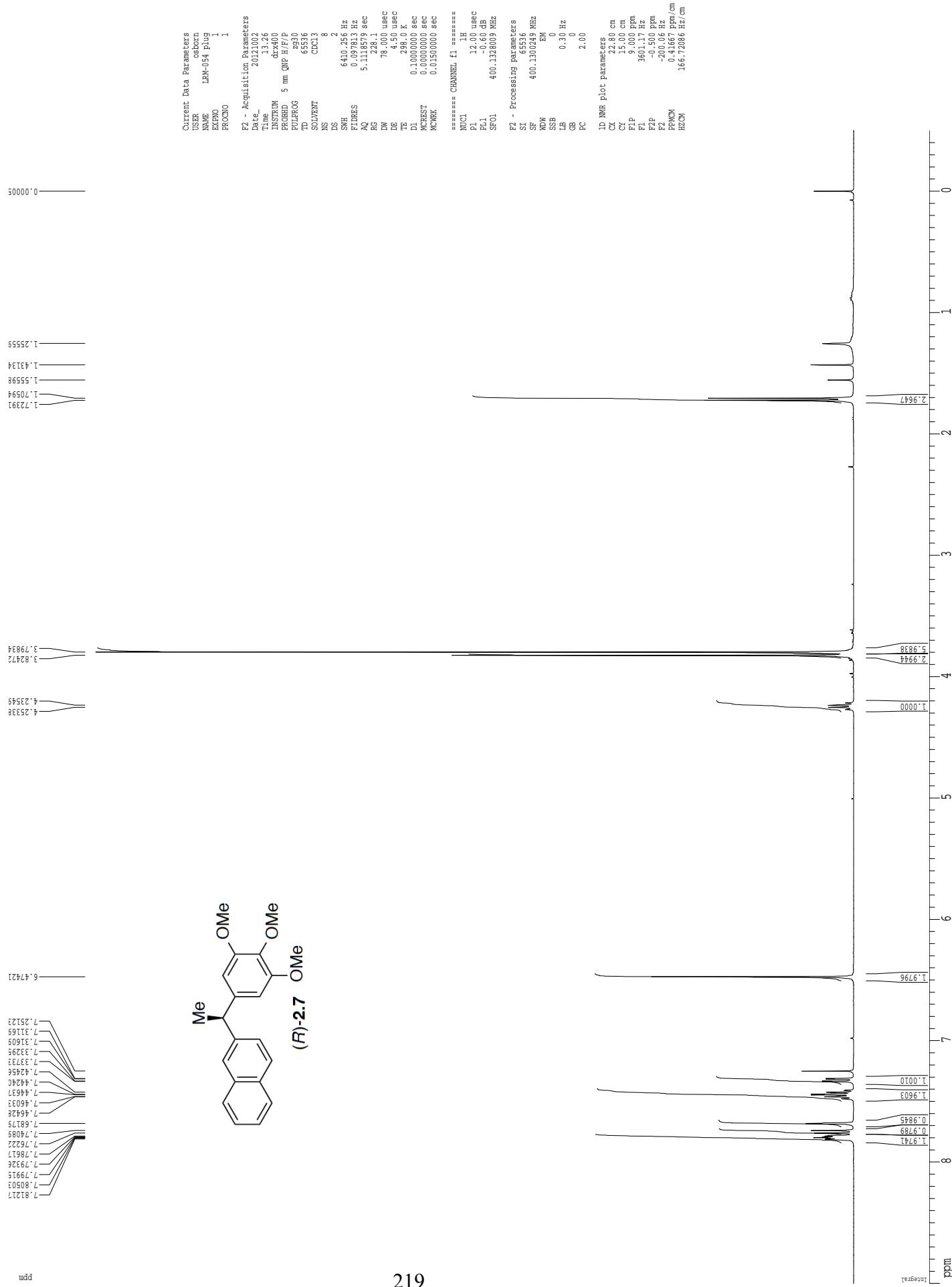


CAO-I-106B SI  
Z-restored spin-echo 13C spectrum with 1H decoupling

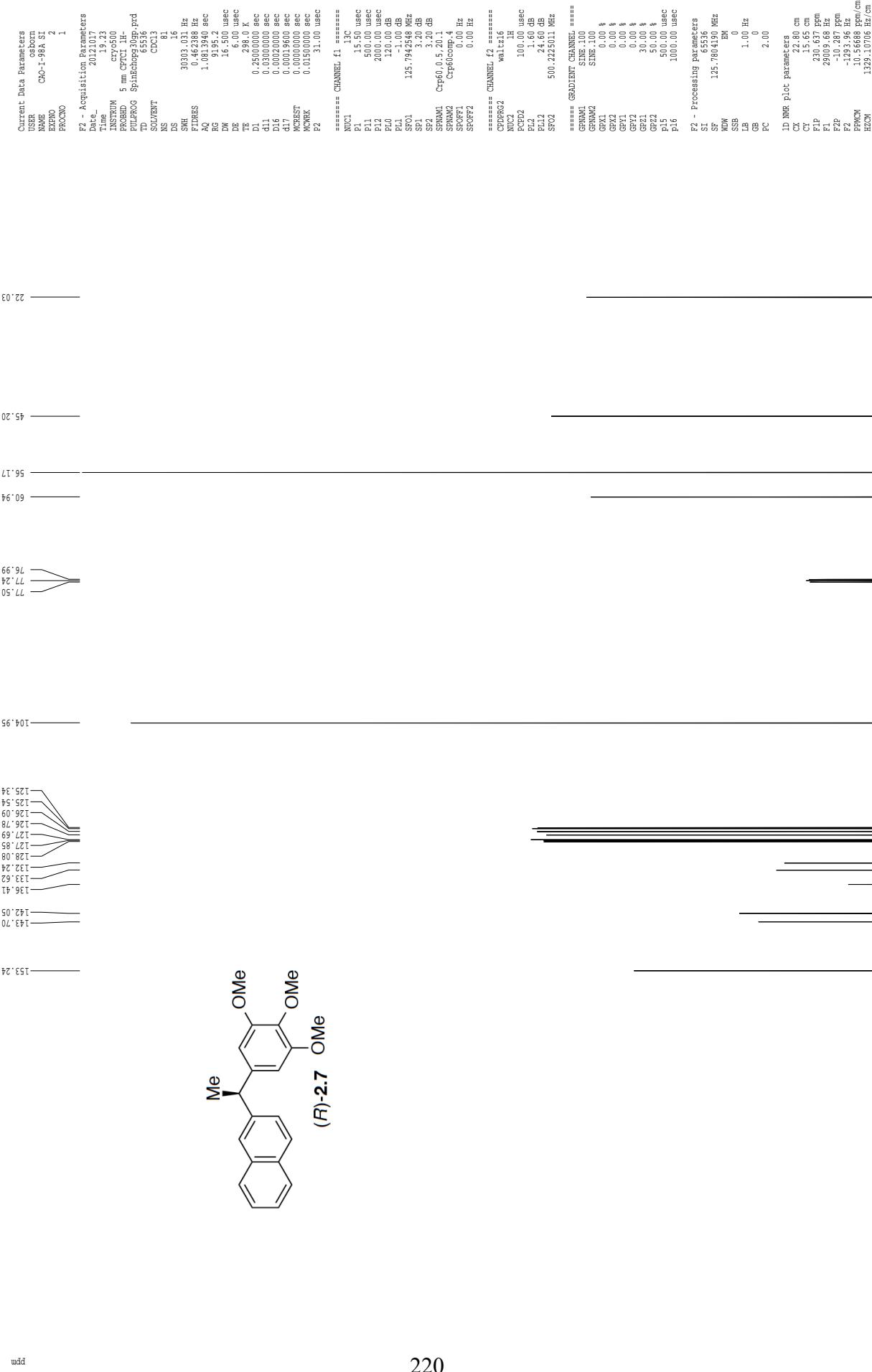


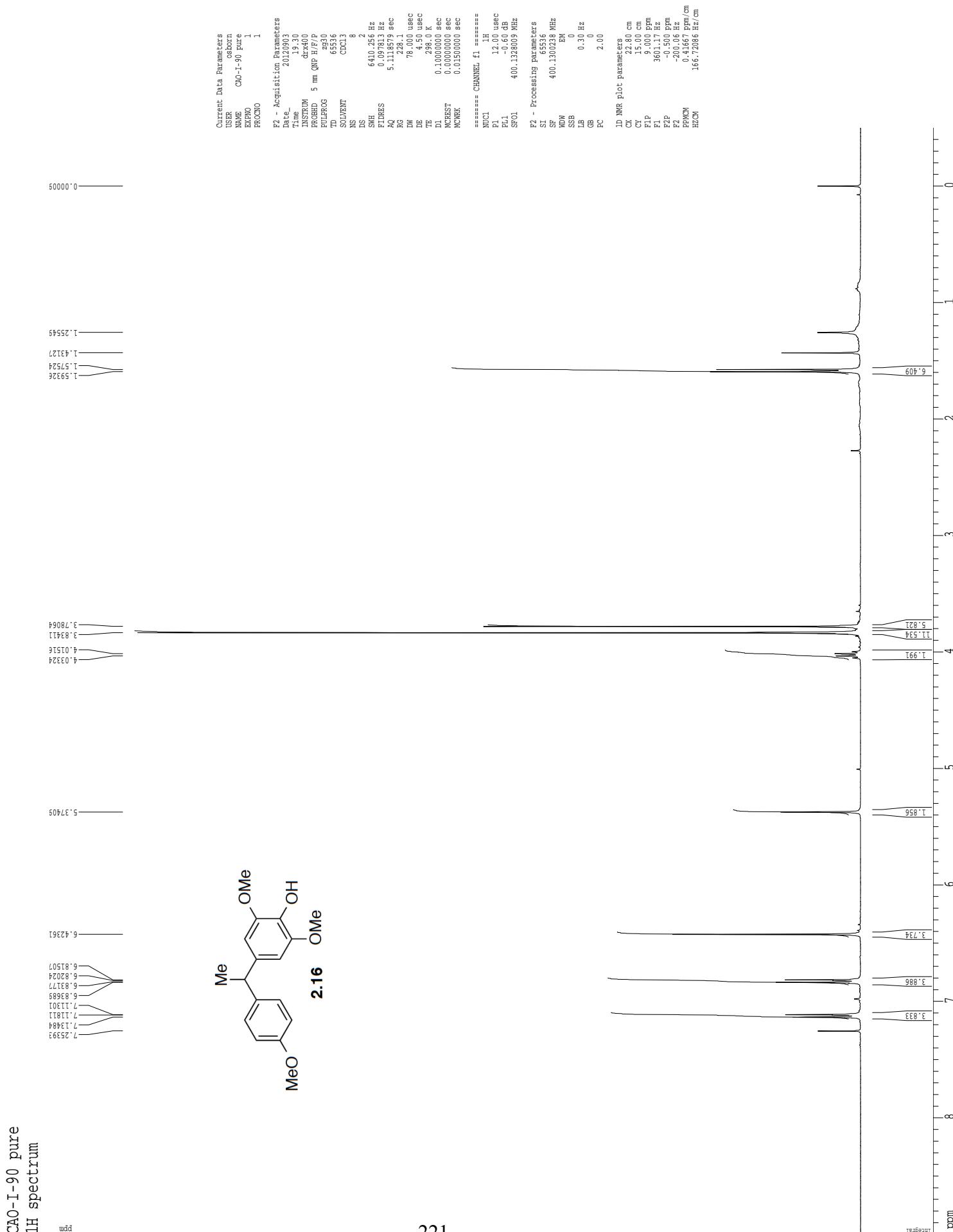
1H spectrum

ppm



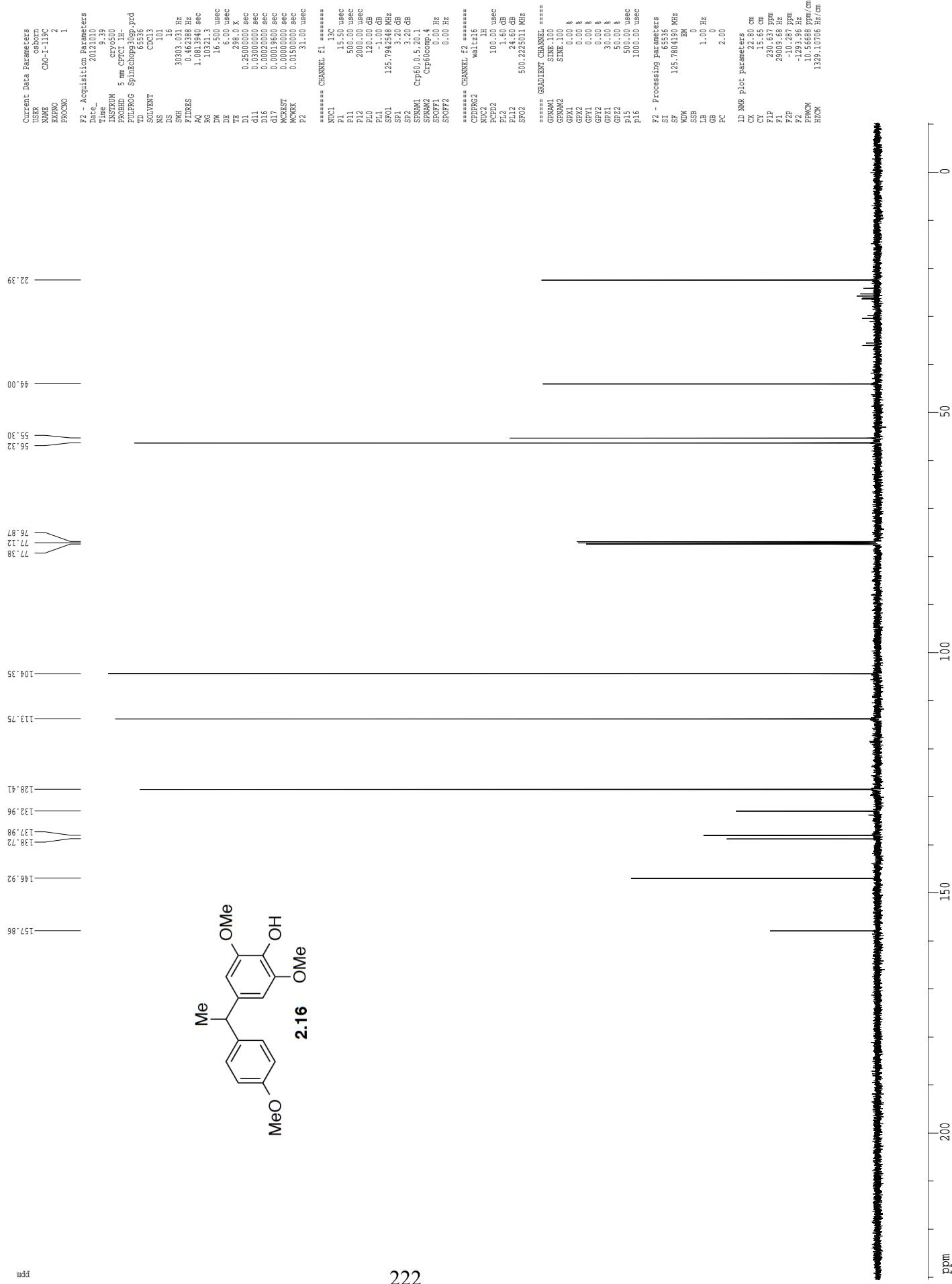
Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling





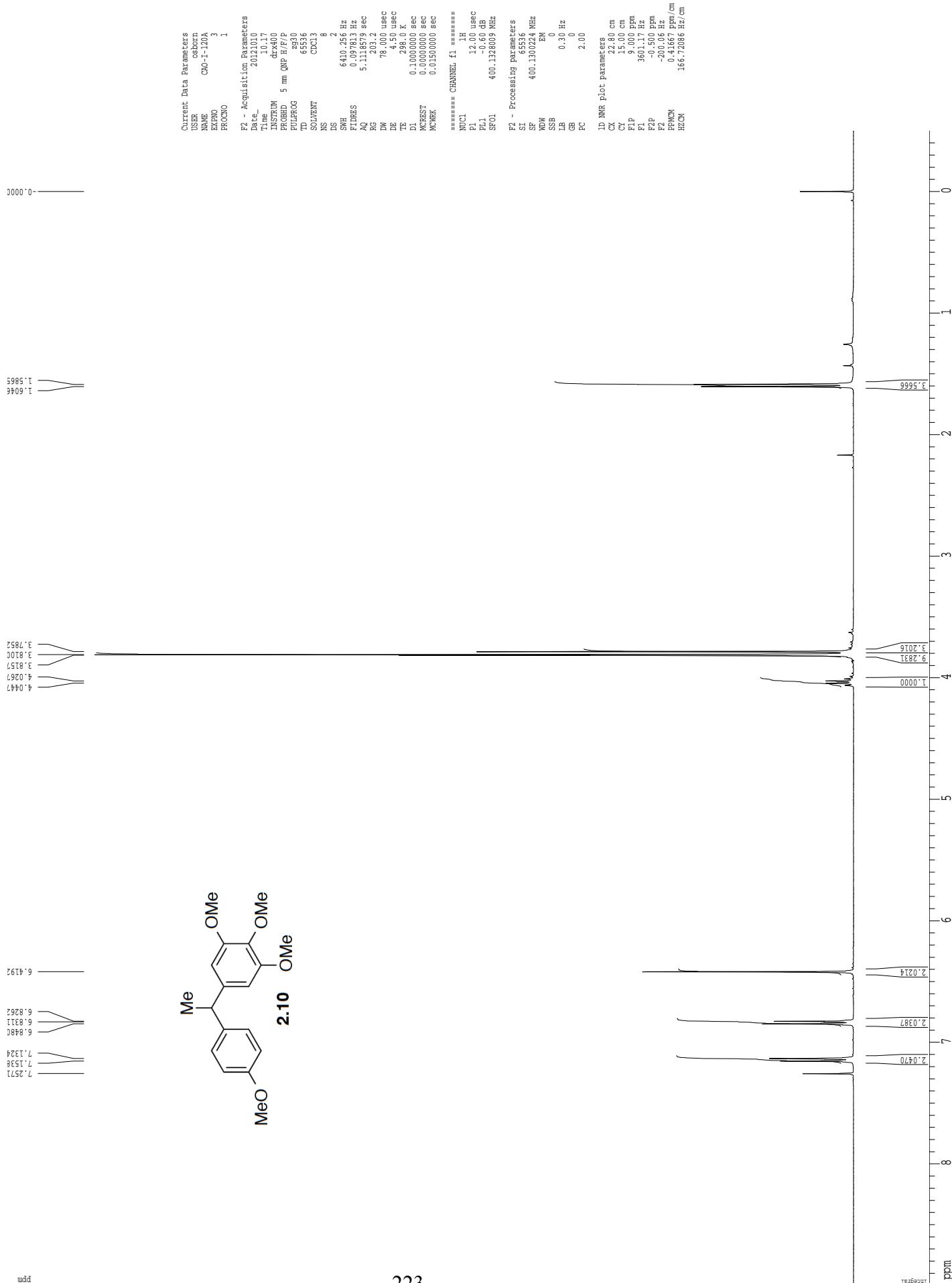
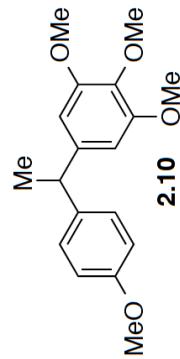
CAO-I-90 pure  
<sup>1</sup>H spectrum

CAO-I-119C  
Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling



CAO-I-102A  
1H spectrum

wdc



CAO-I-120A  
Z-restored spin-echo 13C spectrum with 1H decoupling

ppm

22.26

44.25

55.29

55.11

60.88

76.83

77.08

77.74

104.61

113.77

128.43

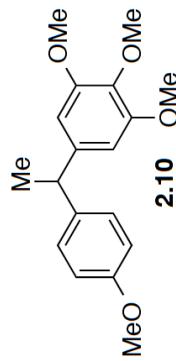
136.18

138.37

142.54

153.09

157.92



```

Current Data Parameters
USER   osborn
NAME   CAO-I-120A
EXNO   2
PROCNO 1

F2 - Acquisition Parameters
Date_  2012/10/0
Time_  9.49
INSTRUM  cryo500
PROBHD  5 mm CCP1 1H-
PULPROG SpinEchoes3D90P-prd
TD    65536
SOLVENT
NUC1  13C
PC1  211
DS    16
SWH  3033.031 Hz
ETRIM 0.46238 Hz
ETRIM_NQ 1.001940 sec
TE    11211.3
TM    16.500 usec
D1    3.00 usec
T1    296.0 K
TD0   0.260000 sec
D11   0.000000 sec
D16   0.002000 sec
D17   0.001960 sec
MC1   0.000000 sec
MC2T  0.000000 sec
MC3K  0.0150000 sec
P2K2  31.00 usec

=====
CHANNEL F1 =====
NUC1  13C
P1    15.30 usec
P11   50.00 usec
P12   200.00 usec
PL0   120.00 dB
PL1   -1.00 dB
SP01  125.794258 MHz
SP1   3.20 dB
SP2   3.20 dB
SPW01 Crp60.0,5,20.1
SPW02 Crp60.0,5,20.1
CPD16 Cpm30.0,4
SP0FF1 0.00 Hz
SP0FF2 0.00 Hz

=====
CHANNEL F2 =====
CPDPG2  SINE.100
NUC2  1H
P2D2  10.00 usec
P2L2  1.60 dB
PL12  24.60 dB
SF02  500.2225011 MHz

=====
GRADIENT CHANNEL =====
GP0M1  SINE.100
GP0M2  SINE.100
GPX1  0.00 %
GPX2  0.00 %
GPY1  0.00 %
GPY2  0.00 %
GPZ1  30.00 %
GPZ2  50.00 %
GP15  500.00 usec
GP16  1000.00 usec

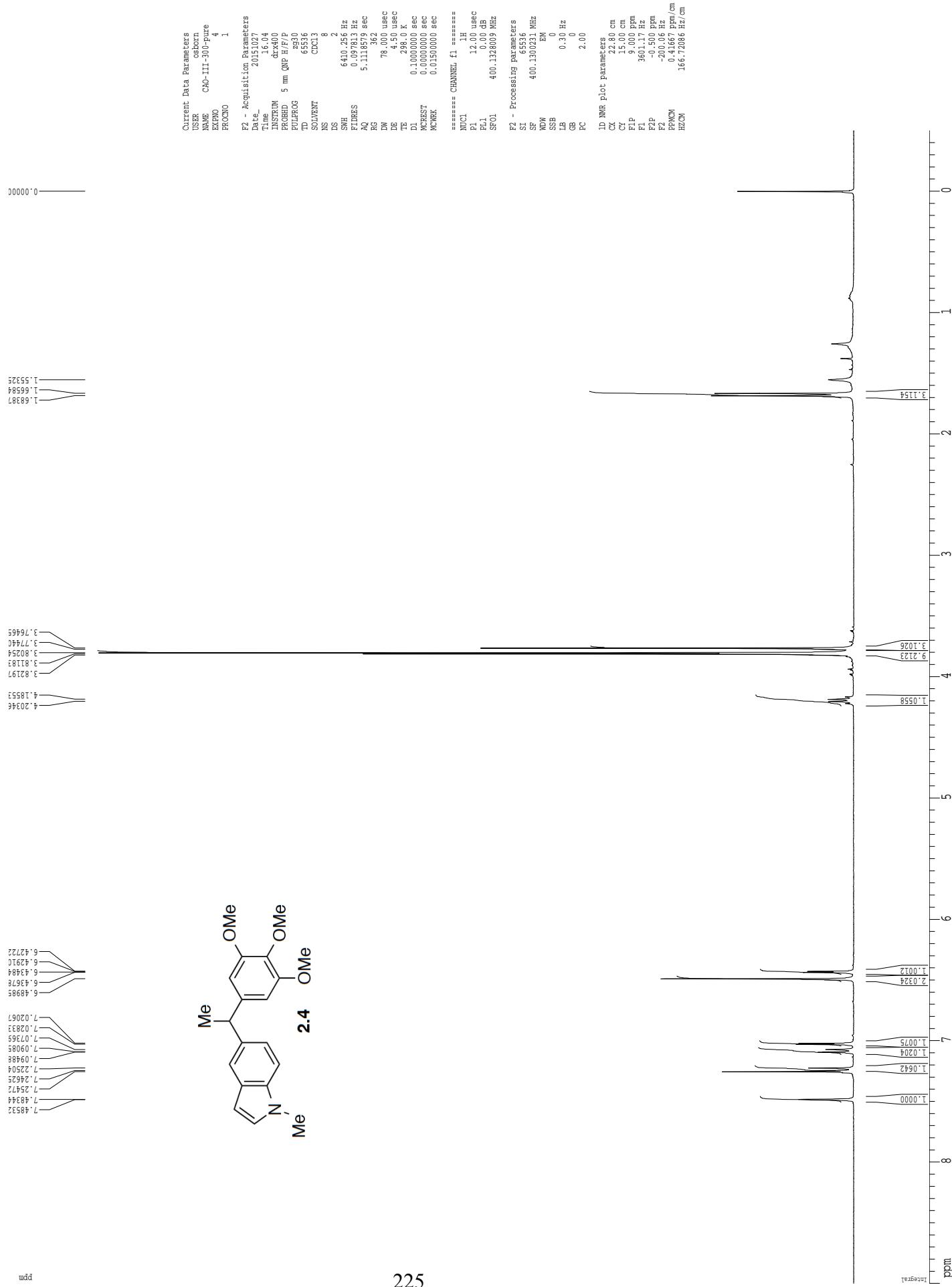
F2 - Processing Parameters
CX    22.80 cm
CY    15.65 cm
SF    125.7804190 MHz
WDW   EM
SSB   0
LB    1.00 Hz
GB    0
PC    2.00

1D NMR Plot parameters
T     ppm

```

1H spectrum

ppm



Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

Current Data Parameters  
 USER osborn  
 NAME Cb-111-30J-pure  
 EXNO 2  
 PRCNO 1  
 F2 - Acquisition Parameters  
 Date 2015/07/27  
 Time 14:10  
 INSTRUM cryo500  
 PROBHD 5 mm CCP1 1H-  
 PULPROG SpinEchoes30P.prd  
 TD 65536  
 SOLVENT  
 CR13  
 NS 116  
 DS 16  
 SWH 303.021 Hz  
 FIDRES 0.015390 Hz  
 AQ 1.00 sec  
 TS 128.2  
 TG 16.50 usec  
 FA 30.00  
 TE 296.0 K  
 D1 0.260000 sec  
 D11 0.000000 sec  
 D16 0.002000 sec  
 D17 0.001960 sec  
 MC 1  
 NEST 0.000000 sec  
 NCRR 0.015000 sec  
 PC2 33.10 usec

===== CHANNEL F1 =====  
 NUCL 13C  
 P1 16.55 usec  
 P11 50.00 usec  
 P12 200.00 usec  
 PL0 120.00  
 PLL 125.794258 MHz  
 SP1 2.70 dB  
 SP2 2.70 dB  
 SPW01 Crp60.0,5,20.1  
 SPW02 Crp60.0,5,20.1  
 CP60.0,4  
 SPFF1 0.00 Hz  
 SPFF2 0.00 Hz

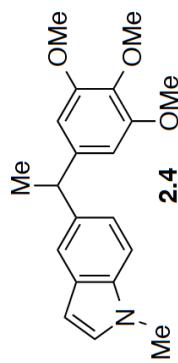
===== CHANNEL F2 =====  
 NUCL 1H  
 P1 16.55 usec  
 P11 50.00 usec  
 P12 200.00 usec  
 PL0 120.00  
 PLL 125.794258 MHz  
 SP1 2.70 dB  
 SP2 2.70 dB  
 SPW01 Crp60.0,5,20.1  
 SPW02 Crp60.0,5,20.1  
 CP60.0,4  
 SPFF1 0.00 Hz  
 SPFF2 0.00 Hz

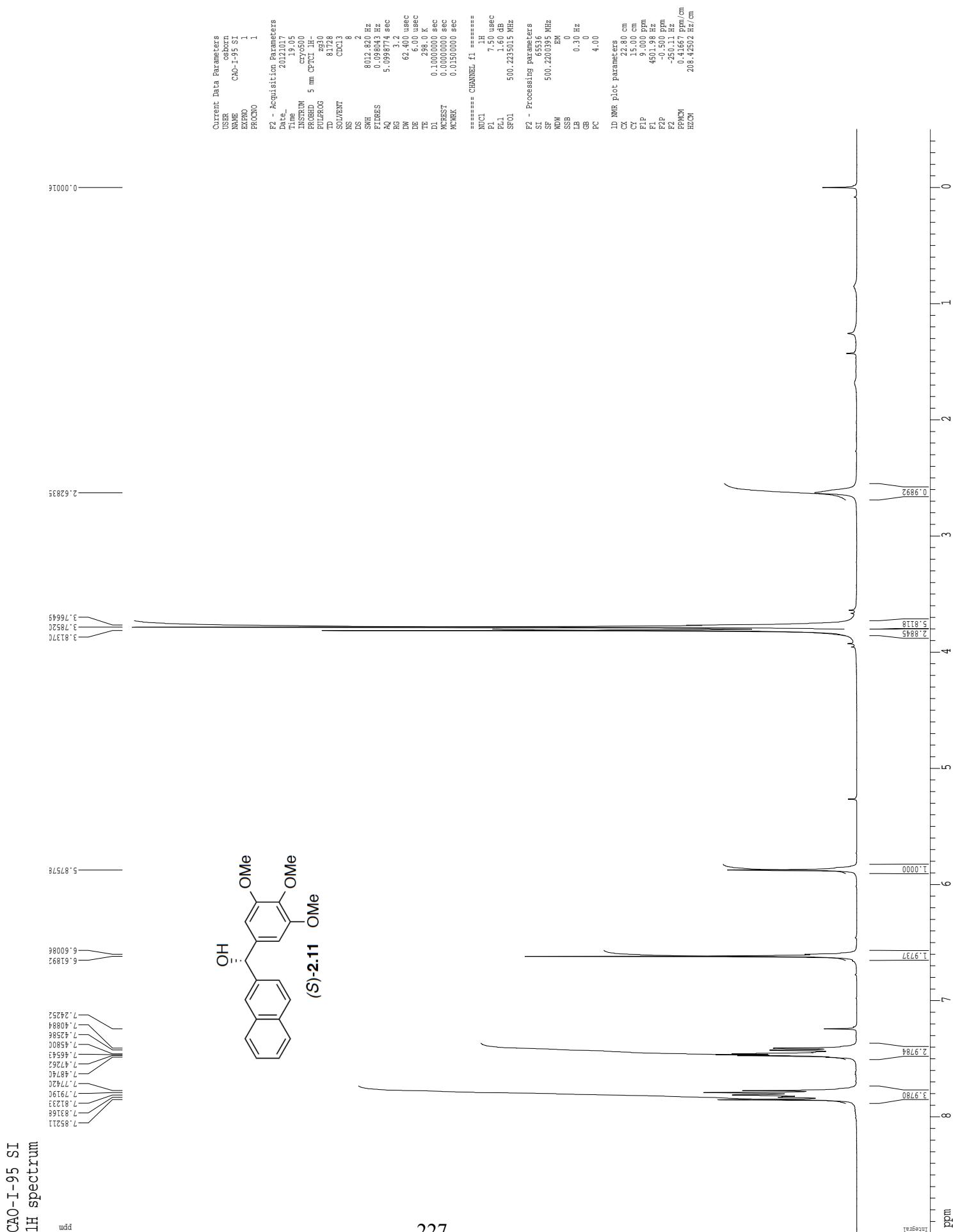
===== GRADIENT CHANNEL =====  
 GPRM01 SINE,100  
 GPRM02 SINE,100  
 GPEX1 0.00 %  
 GPEX2 0.00 %  
 GPY1 0.00 %  
 GPY2 0.00 %  
 GPZ1 30.00 %  
 GPZ2 50.00 %  
 P15 50.00 usec  
 P16 1000.00 usec

F2 - Processing Parameters  
 SI 65536  
 SF 125.7804090 MHz  
 SW 20.00 cm  
 EM 0  
 SSB 0  
 LB 1.00 Hz  
 GB 0  
 PC 2.00  
 1D NMR Plot Parameters  
 CX 22.80 cm  
 CY 20.00 cm  
 F1P 230.67 ppm  
 F1 290.968 Hz  
 F2P -10.287 ppm  
 F2 -129.346 Hz  
 PPMW 10.56688 ppm/cm  
 HZWM 1325.10693 Hz/cm

22.69  
 33.00  
 45.19  
 56.18  
 60.96  
 76.90  
 77.16  
 77.41

100.89  
 104.08  
 109.23  
 112.97  
 113.09  
 122.59  
 123.13  
 135.53  
 136.12  
 137.41  
 143.28  
 153.10

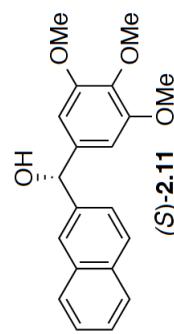




CAO-I-95 SI  
Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

ppm

153.32 —  
141.01 —  
139.44 —  
137.27 —  
132.38 —  
132.16 —  
126.90 —  
126.28 —  
127.75 —  
128.33 —  
128.11 —  
124.79 —  
77.34 —  
77.14 —  
76.49 —  
60.90 —  
56.14 —



```

Current Data Parameters
USER          usborn
NAME         CAO-I-95 SI
EXNO          2
PRCNO         1

F2 - Acquisition Parameters
Date        2012-07-17
Time       19.08
INSTRUM    INSPINW
PROBHD   5 mm PCD1 1H-
PULPROG  SpinEchoes30Pp.prd
TD        65536
SOLVENT    CS2
PCP1      CPM3
CPD1      121
DS        16
SWH      3033.021 Hz
ETRIM     0.063288 Hz
AQ        1.003940 sec
RG        128.2
TE        16.50 usec
TM        0.00 usec
FA        296.1 K
DW        0.260000 sec
D11      0.000000 sec
D16      0.002000 sec
D17      0.001960 sec
MC        1024
N1        0.000000 sec
W1CRK    0.0150000 sec
P2CRK    31.00 usec

===== CHANNEL F1 =====
NUCL1      13C
P1        15.30 usec
P11      50.00 usec
P12      200.00 usec
PL0      120.00 dB
PL1      -1.00 dB
SP01     125.794258 MHz
SP1      3.20 dB
SP2      3.20 dB
SPR001   Crp60.0,5,20.1
SPR002   Crp60.0,5,20.1
CPD1     0.00 Hz
SPCPFF2  0.00 Hz

===== CHANNEL F2 =====
NUCL2      1H
W1Z16     SINE.100
CPDPG2   10.00 usec
P2D2     1.60 dB
PL12     24.60 dB
SF02     500.2225011 MHz

===== GRADIENT CHANNEL =====
GP0001   SINE.100
GP0002   SINE.100
GPX1     0.00 %
GPX2     0.00 %
GPY1     0.00 %
GPY2     0.00 %
GPZ1     30.00 %
GPZ2     50.00 %
GP15    500.00 usec
GP16    1000.00 usec

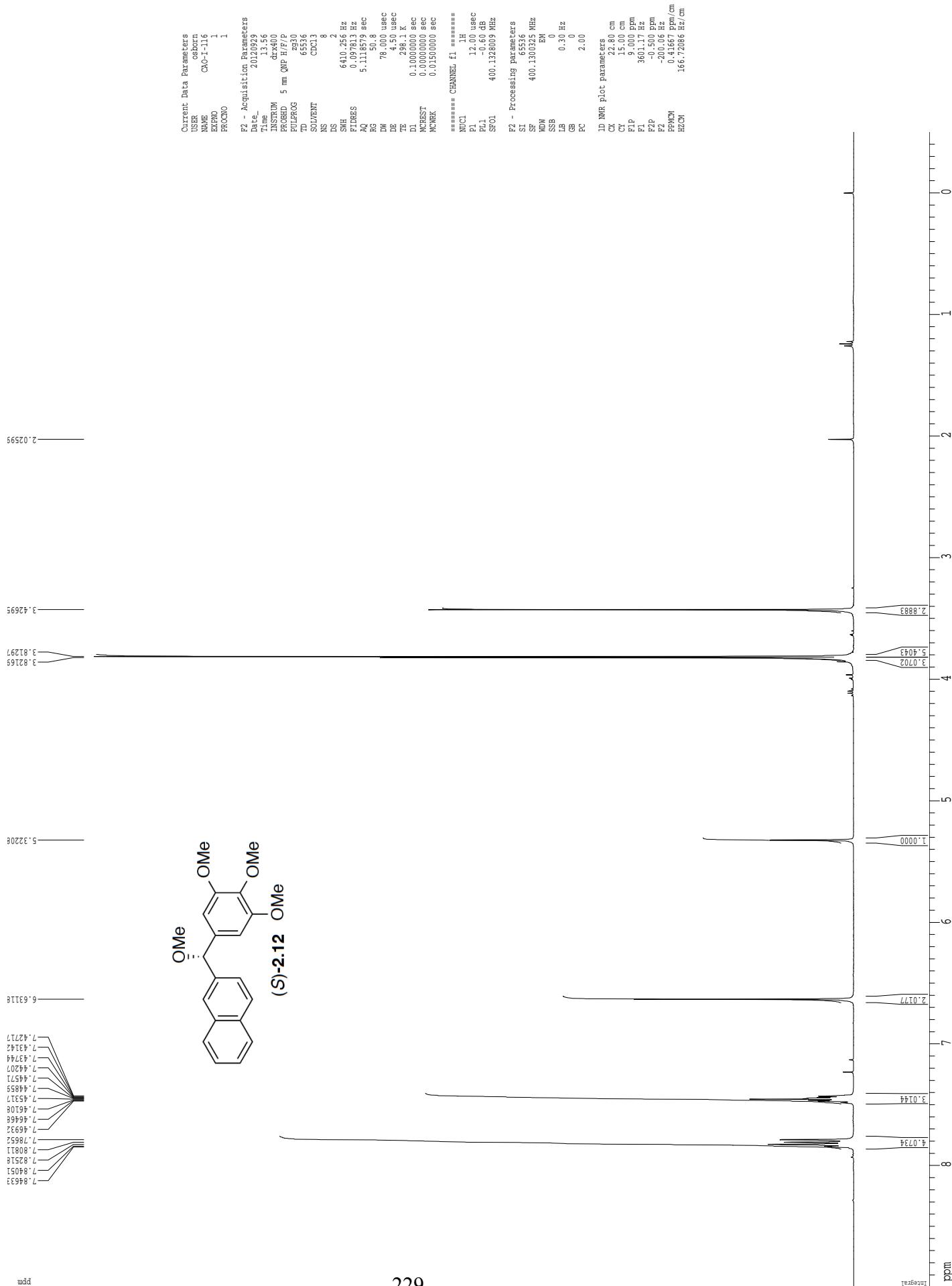
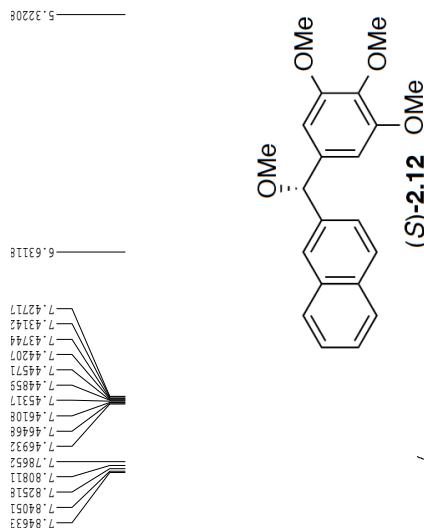
F2 - Processing Parameters
SI        65536
SF        125.7804190 MHz
WDW      EM
SSB      0
LB        1.00 Hz
GB      0
PC        2.00

1D NMR Plot parameters
CX        22.80 cm
CY        15.65 cm
F1P      20.67 ppm
F1       280.968 Hz
F2P      -10.287 ppm
F2      -129.346 Hz
PPCM    10.56688 ppm/cm
HZCM   1325.10706 Hz/cm

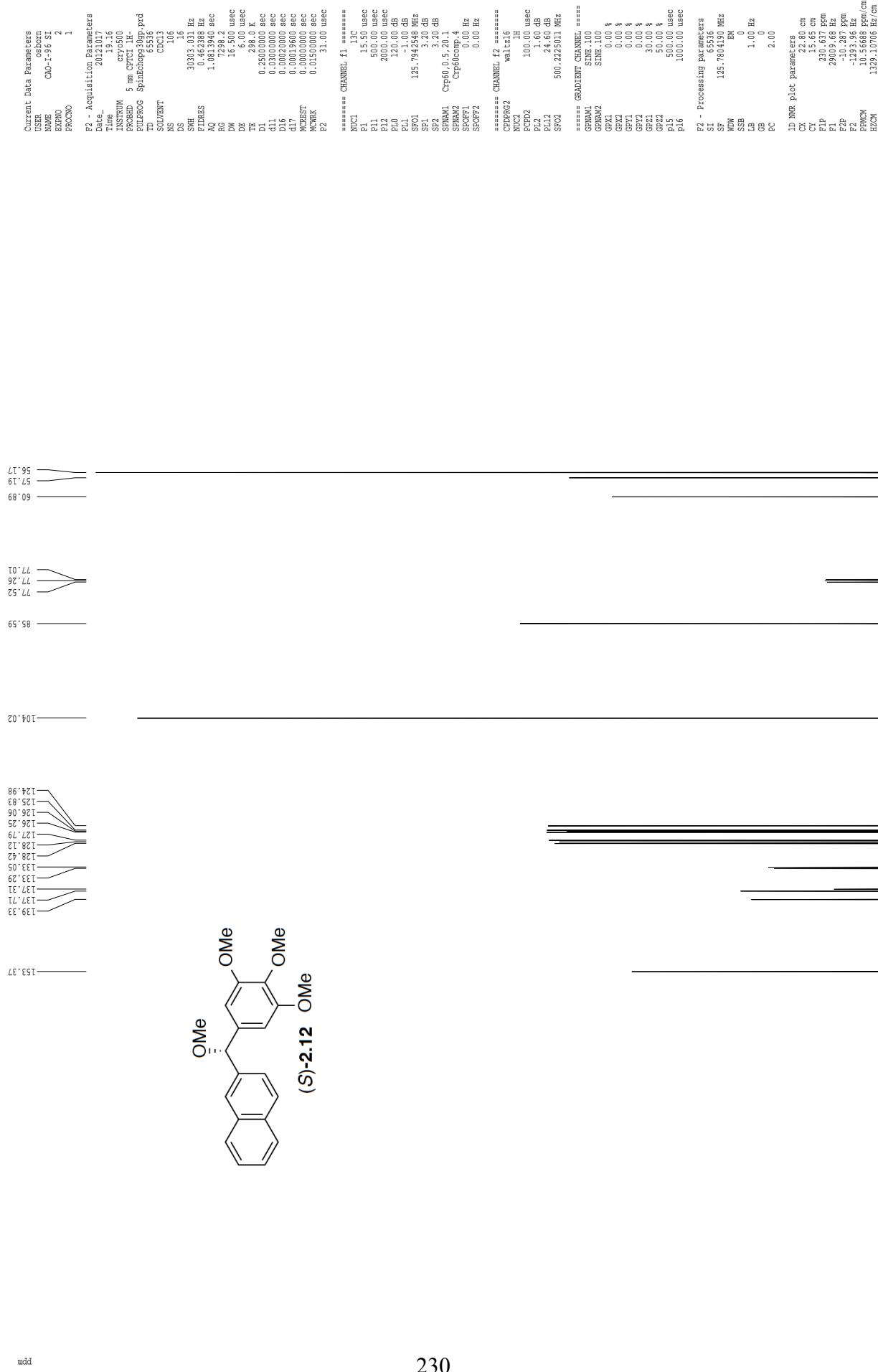
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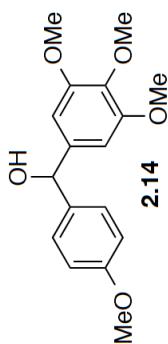
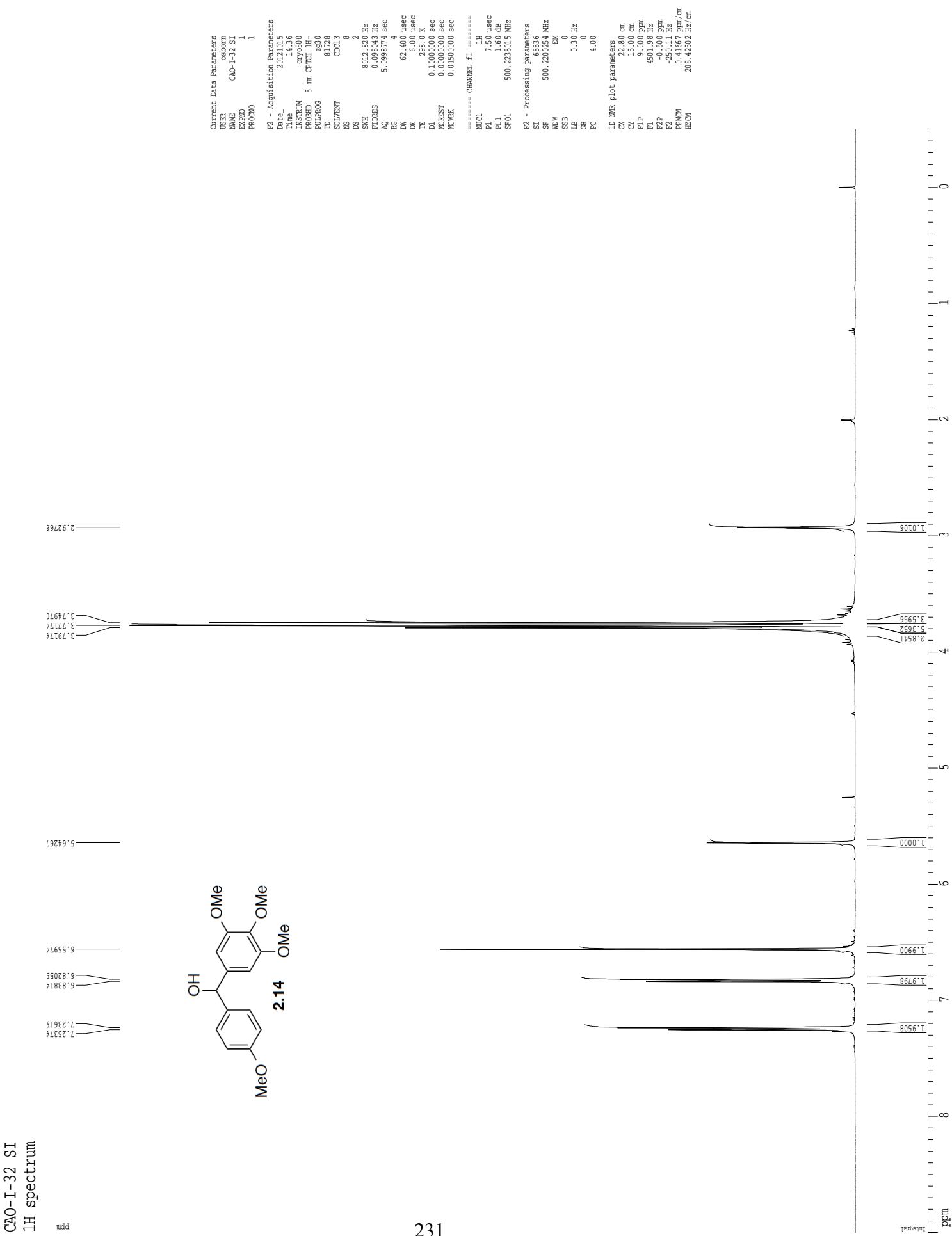
CAO-I-1116  
1H spectrum

wdci

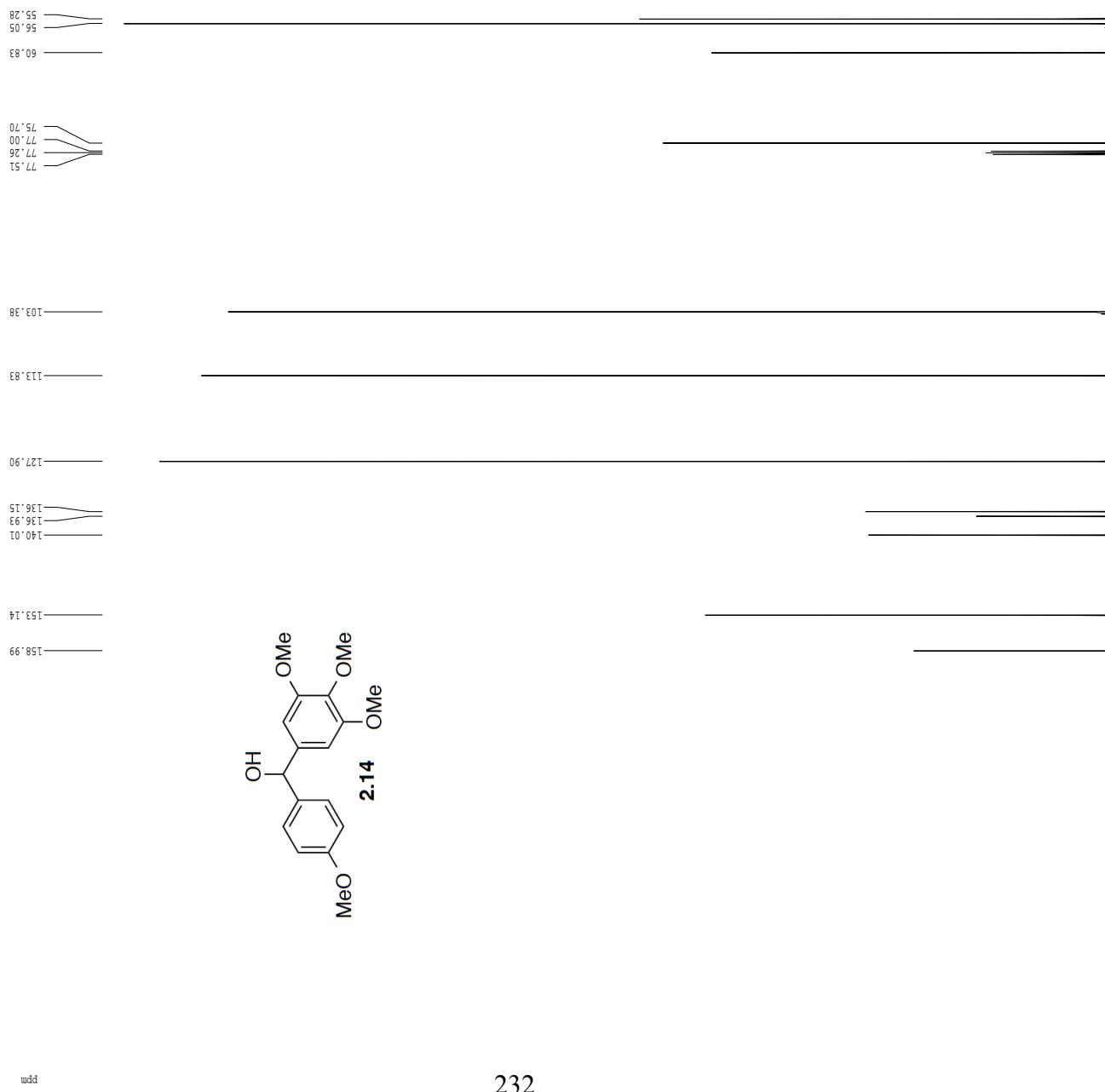


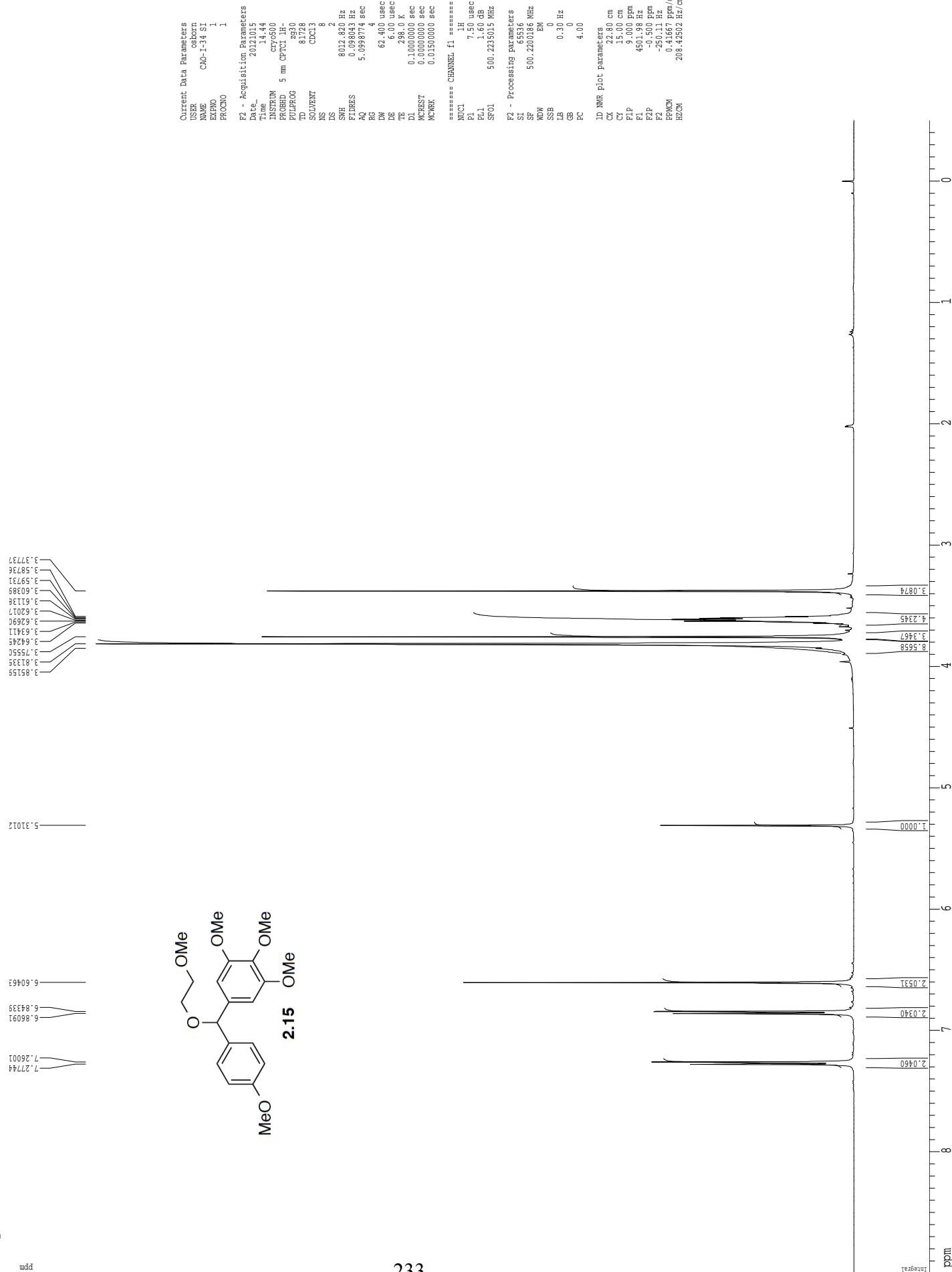
CAO-I-96 SI  
Z-restored spin-echo 13C spectrum with 1H decoupling



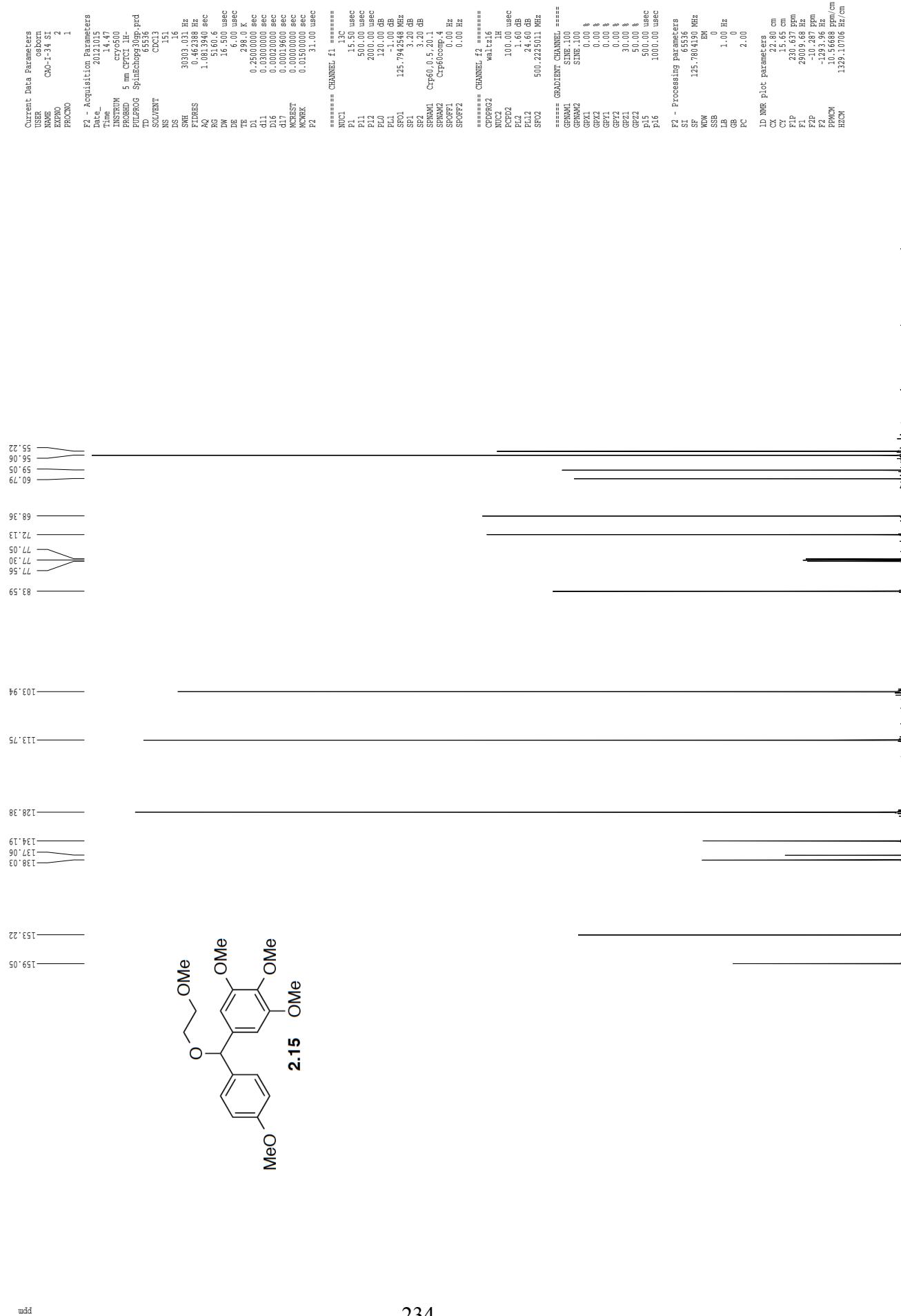


CAO-I-32 SI  
Z-restored spin-echo 13C spectrum with 1H decoupling



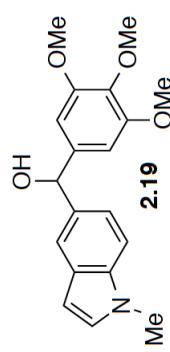


CAO-I-34 SI  
Z-restored spin-echo  $^{13}\text{C}$  spectrum with 1H decoupling



1H spectrum

7.60537  
7.60531  
7.60530  
7.28306  
7.28305  
7.20122  
7.20121  
7.21955  
7.21953  
7.03944  
7.03922  
6.65302  
6.41141  
6.41140  
6.41139  
6.41138  
6.41137  
6.41136  
5.85453  
5.85452  
3.84746  
3.83922  
3.83921  
3.75710  
3.75709  
2.33876  
2.33875  
2.34746  
2.01144  
1.00471  
1.00466  
2.28388  
1.00085  
2.0760  
1.0047  
1.0287  
3.30203  
6.34968  
3.30203  
0.00000



```

=====
Current Data Parameters
USER          osborn
NAME         C40-III-278-92
EXNO          1
PROCNO        1

P2 - Acquisition Parameters
Date       20130925
Time       12.39
INSTRUM   drx400
PROBHD   5 mm QNP H/F/P
PULPROG  TD
TD        65336
SOLVENT    CDCl3
NS           8
DS           2
SWH      6410.056 Hz
FIDRES   0.07713 Hz
AQ        5.11857 sec
RG        143.7
RG2       78.000 usec
DE        4.500 usec
TE        298.0 K
D1       0.1000000 sec
M1       0.0000000 sec
MCWRF  0.0150000 sec

=====
CHANNEL f1
=====
[NH]I
P1        12.00 usec
PL1       0.00 dB
SF01    400.138000 MHz

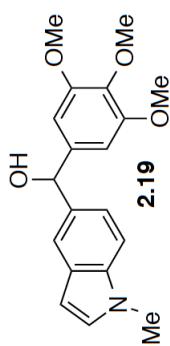
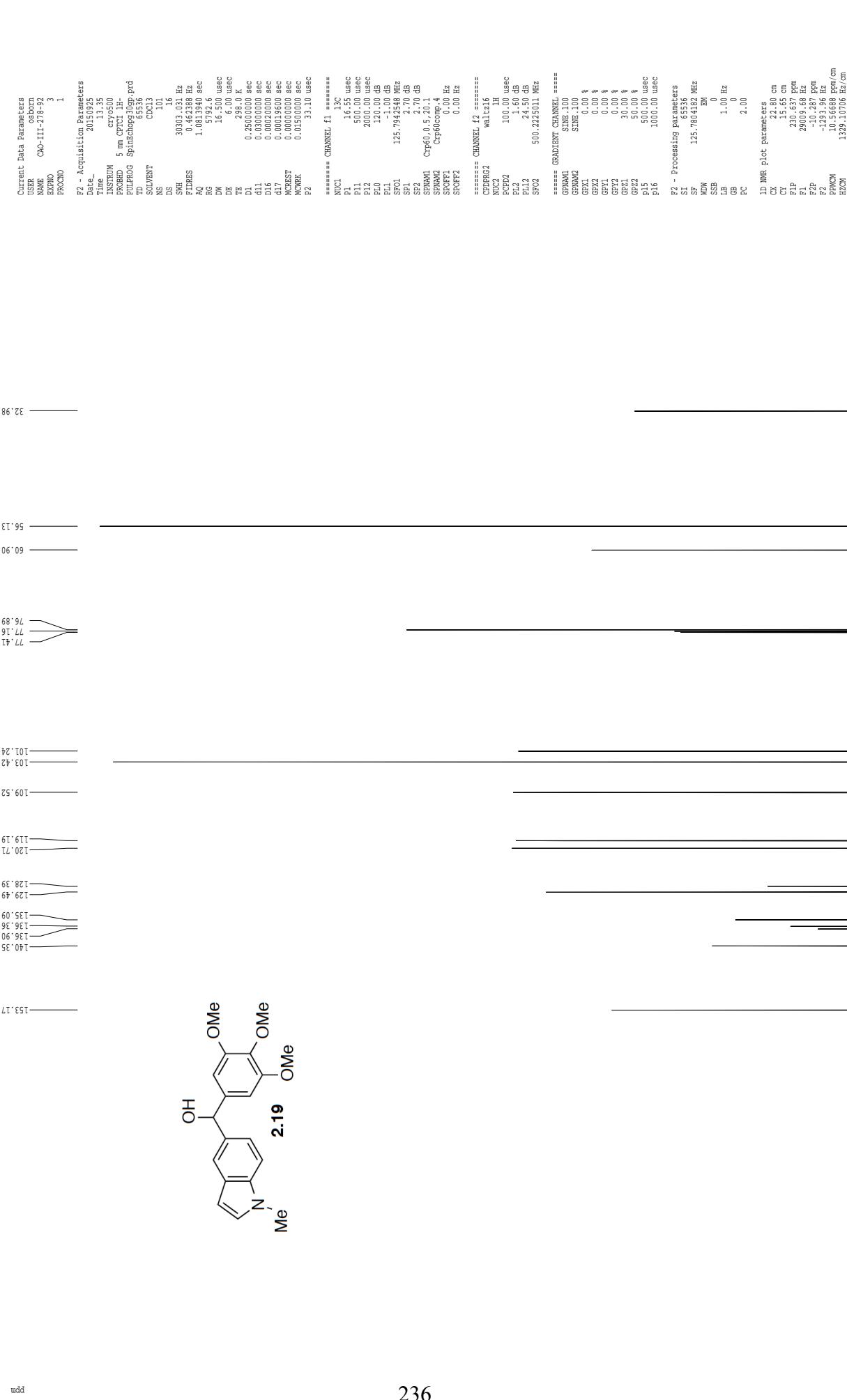
P2 - Processing parameters
SI       65536
SP      400.130089 kHz
WDW        E9
SSB        0
LB        0.30 Hz
GB        0
PC        2.00

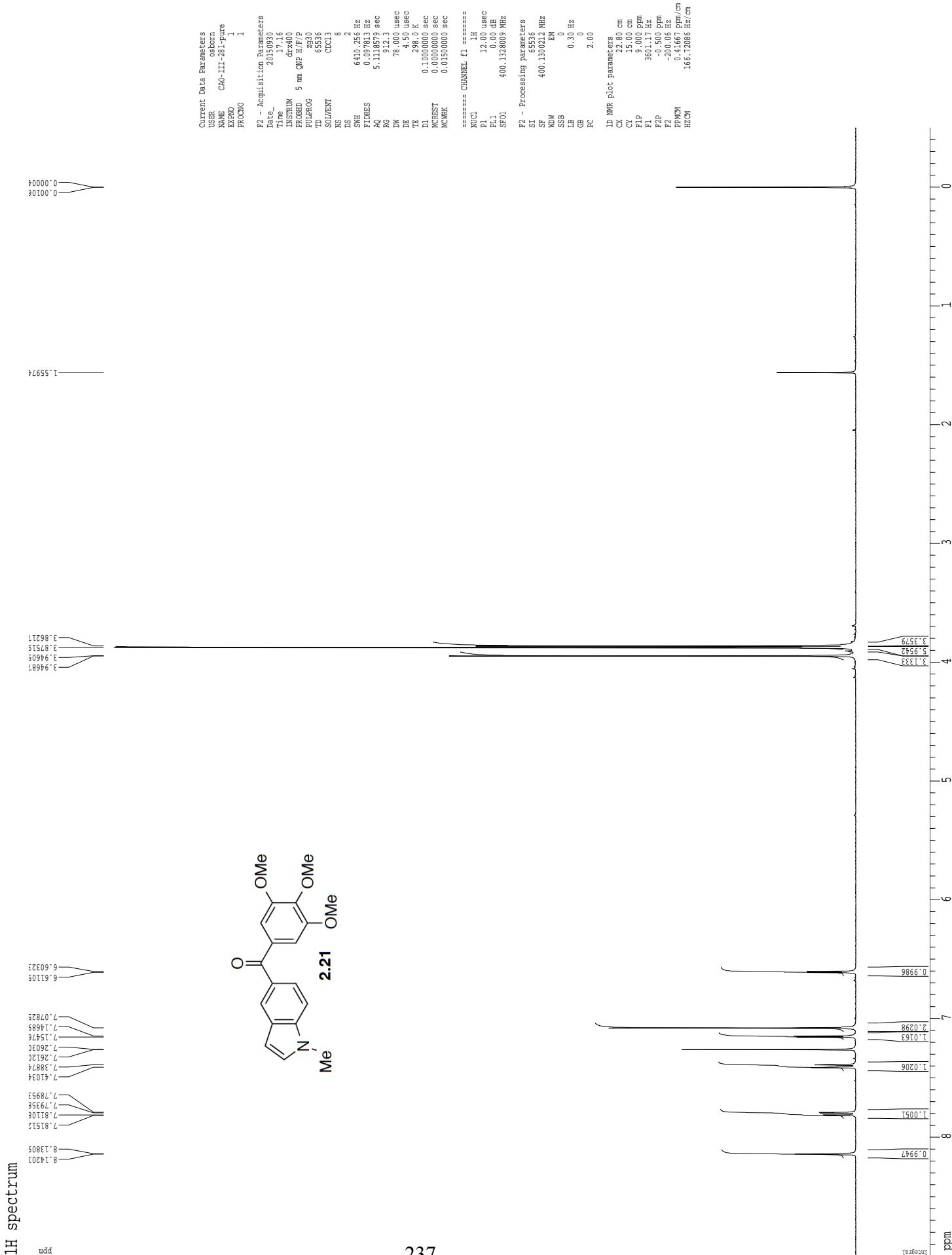
1D NMR plot parameters
CX        22.80 cm
CY        15.00 cm
F1P      9.000 ppm
F1       3601.17 Hz
F2P     -0.500 ppm
F2      -200.06 Hz
PPMCM  0.13167 ppm/cm
HZCM  166.12086 Hz/cm

```

Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

ppm

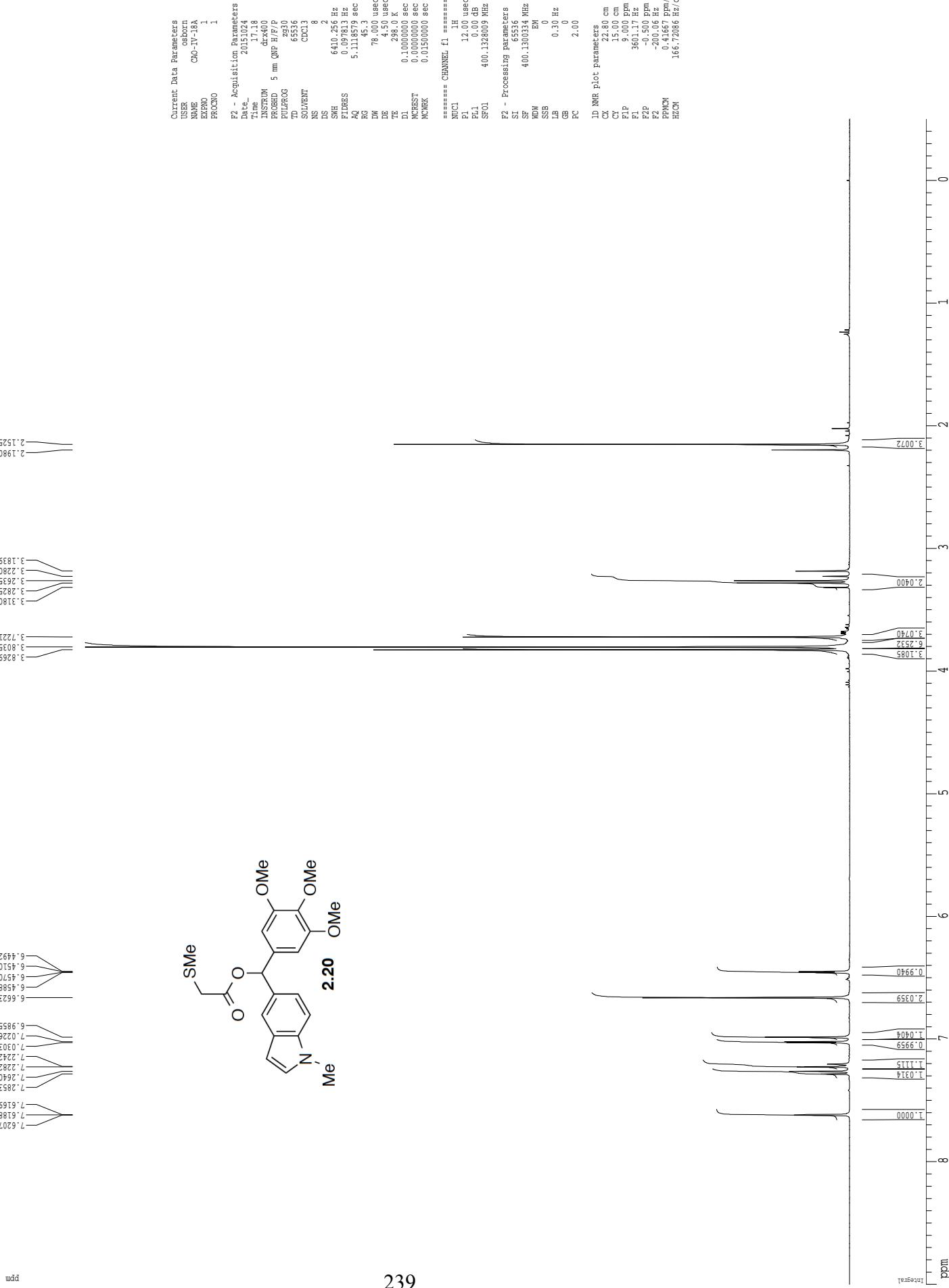




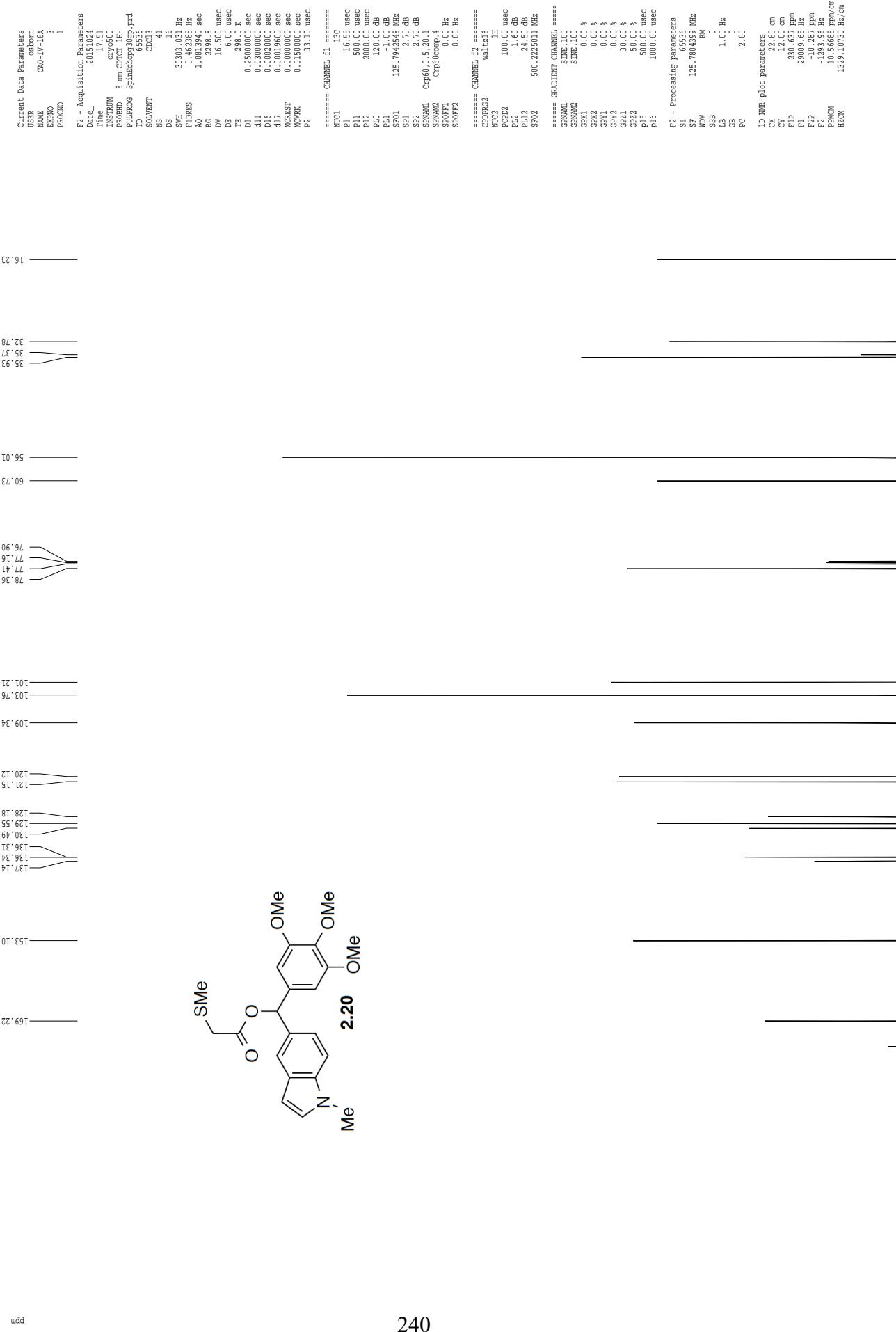
Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

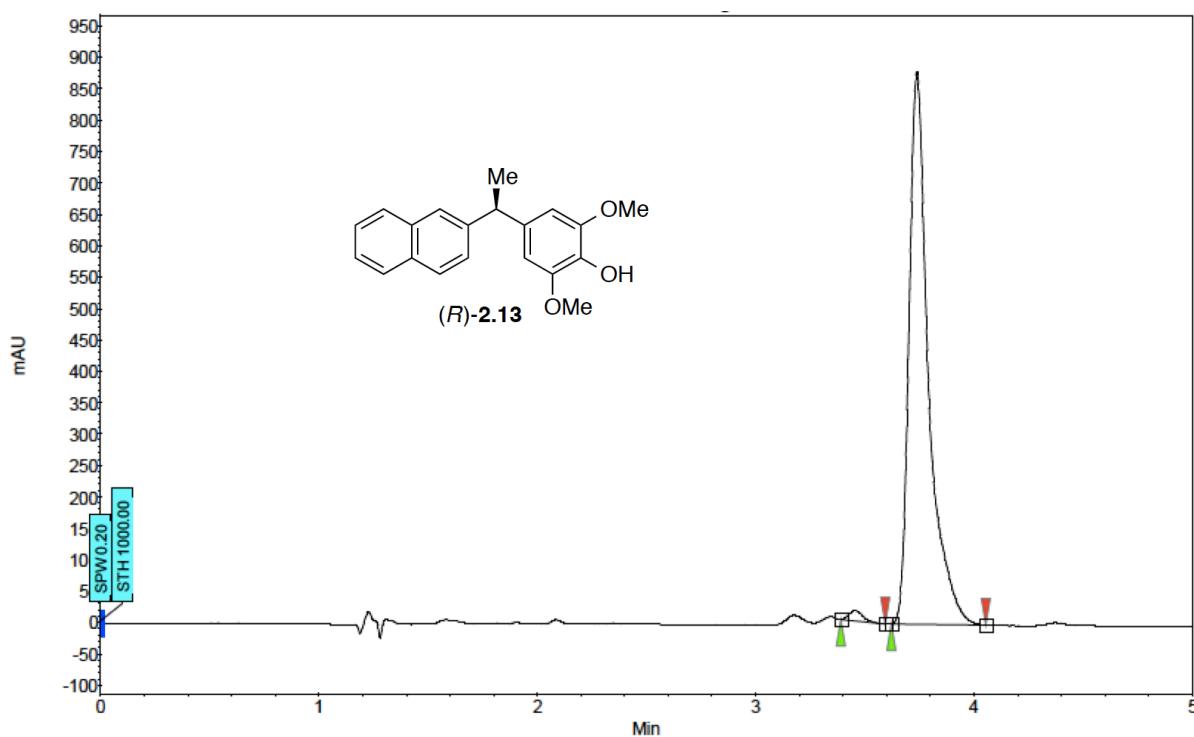
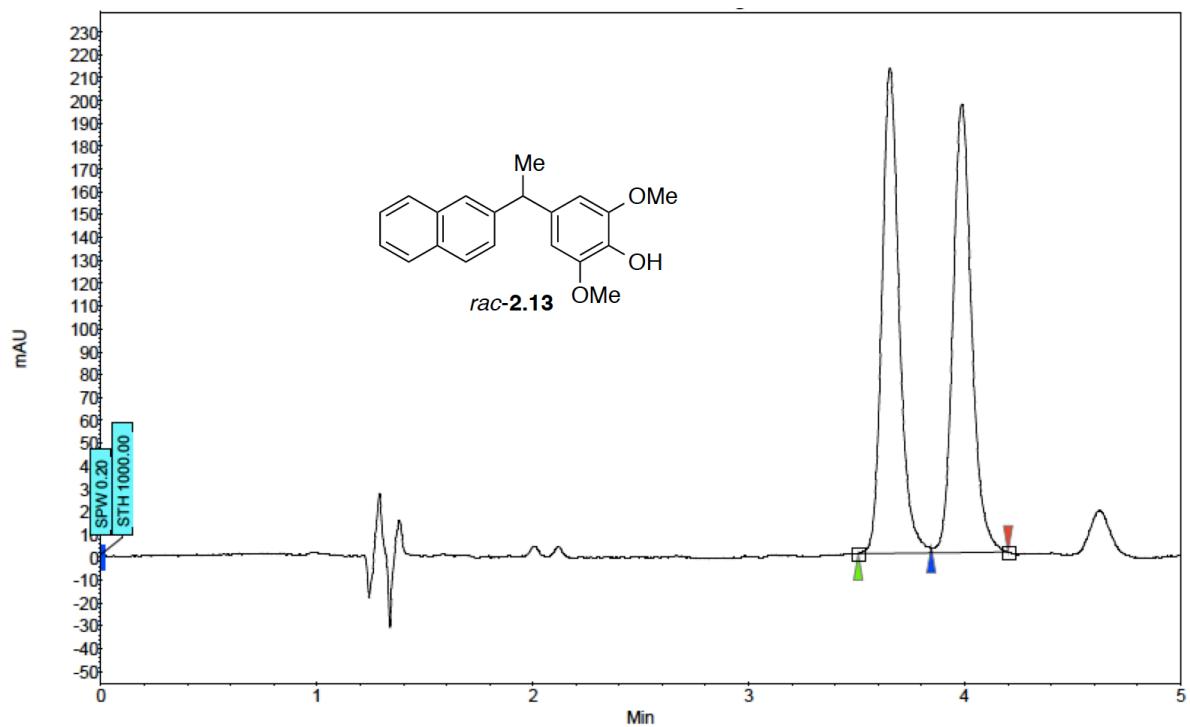


1H spectrum

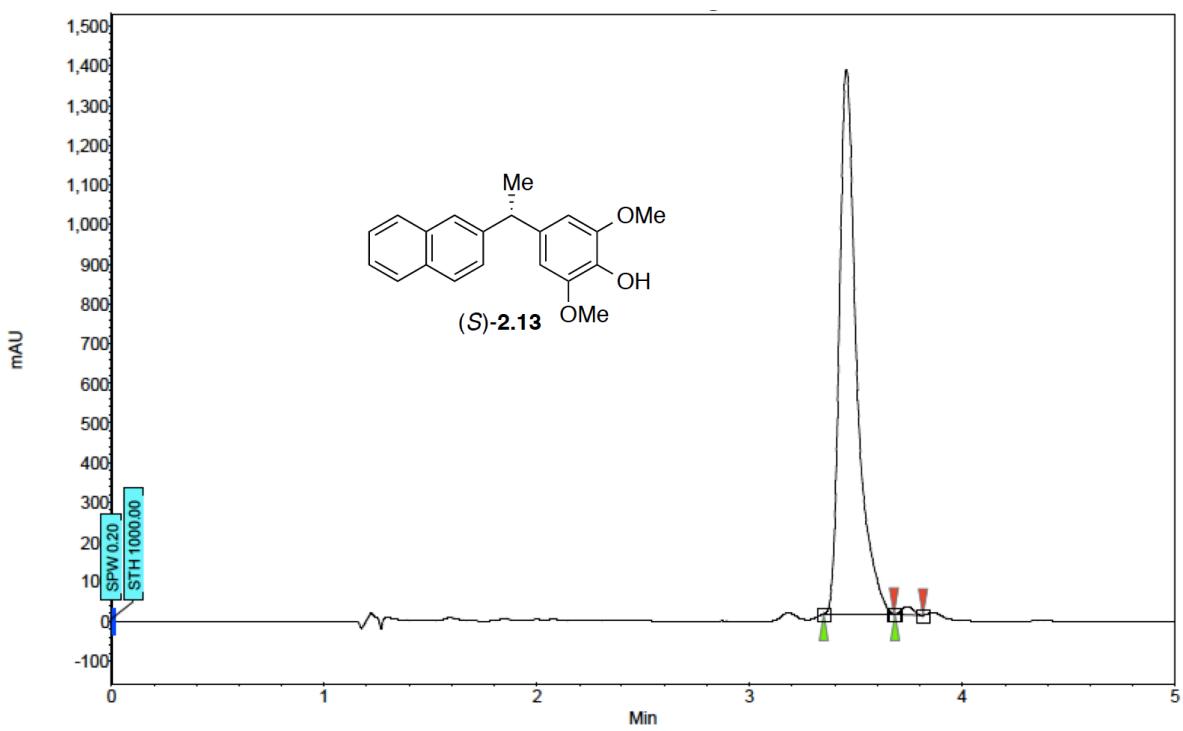
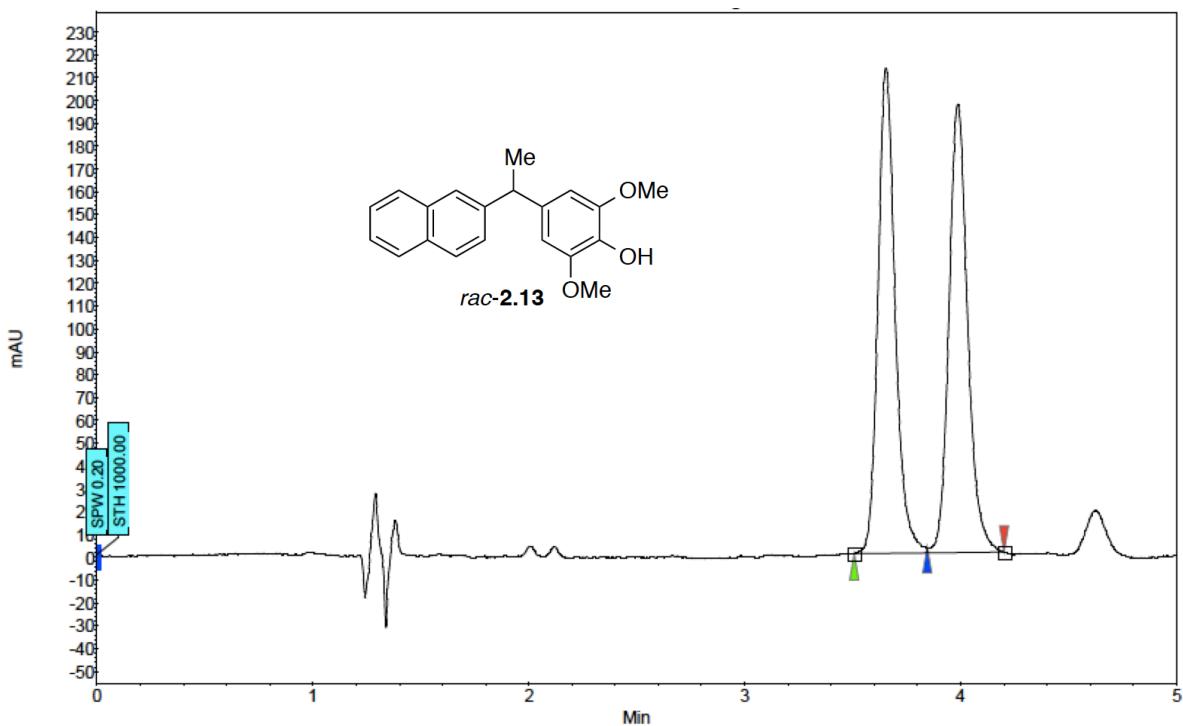


Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

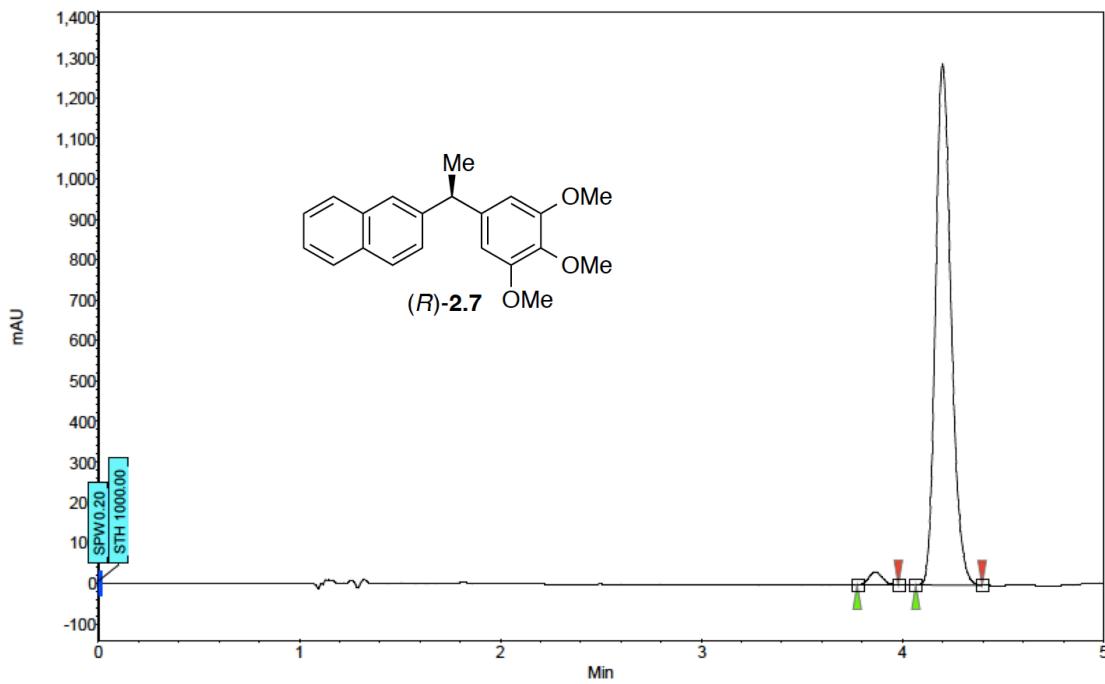
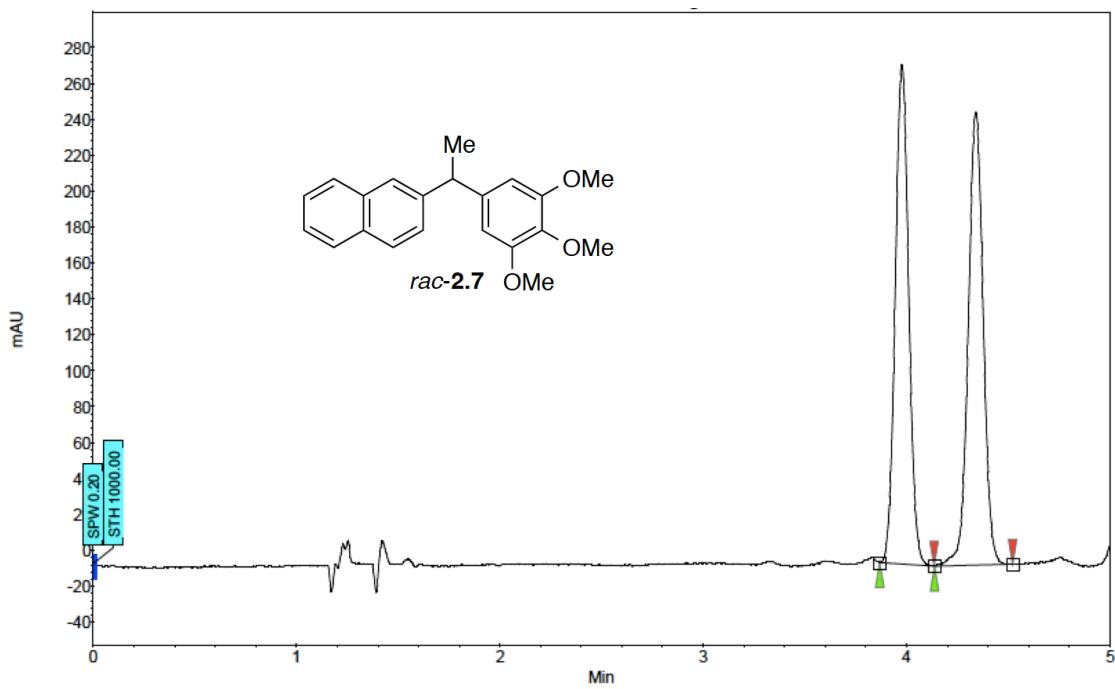




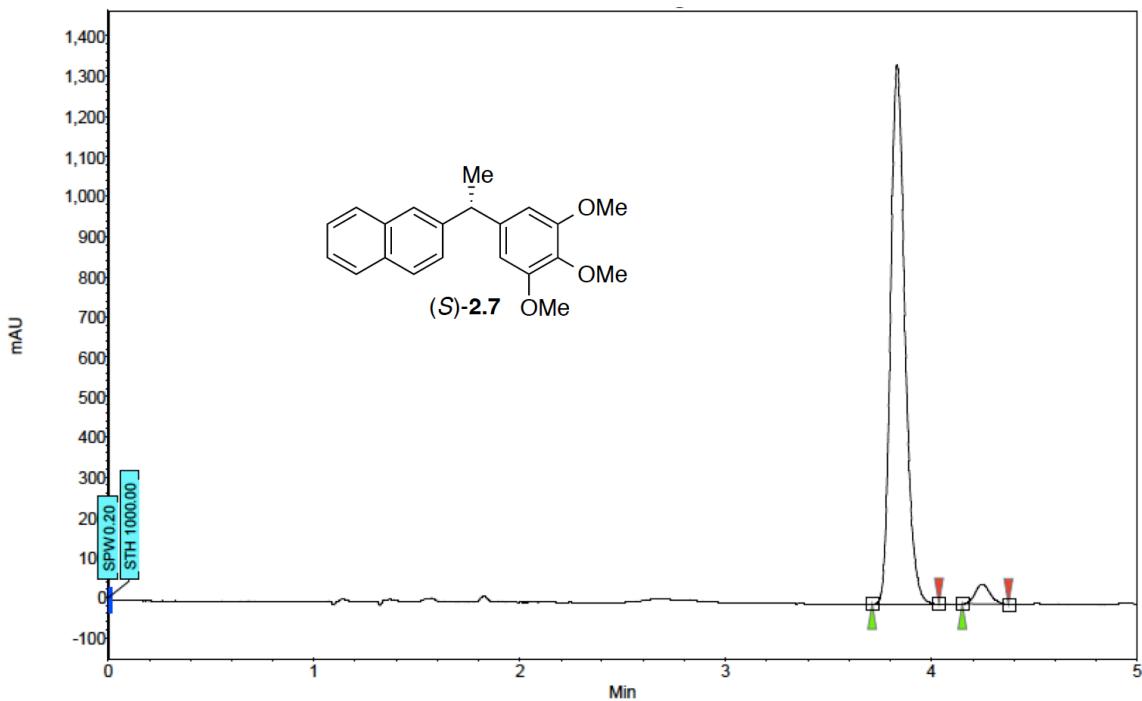
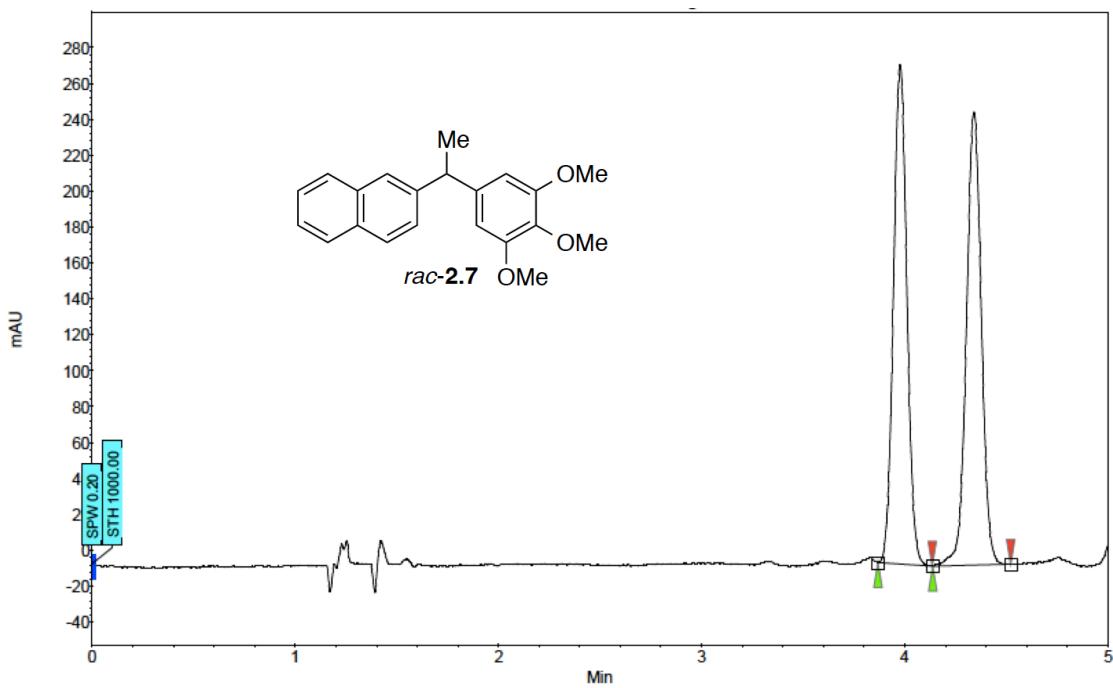
Index	Name	Start Time [Min]	Time [Min]	End [Min]	RT Offset [Min]	Quantity [% Area]	Height [µV]	Area [µV.Min]	Area [%]
1	UNKNOWN	3.39	3.46	3.59	0.00	1.29	16.4	1.2	1.292
2	UNKNOWN	3.62	3.74	4.05	0.00	98.71	879.9	90.7	98.708
Total						100.00	896.2	91.9	100.000



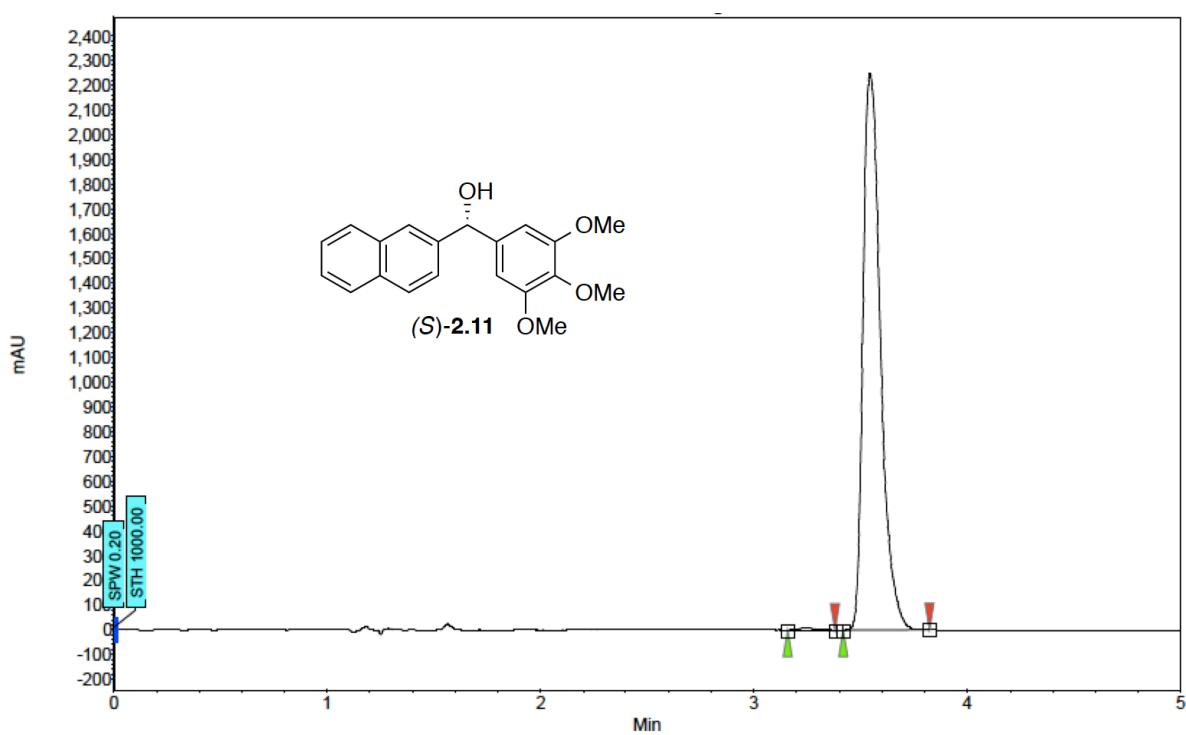
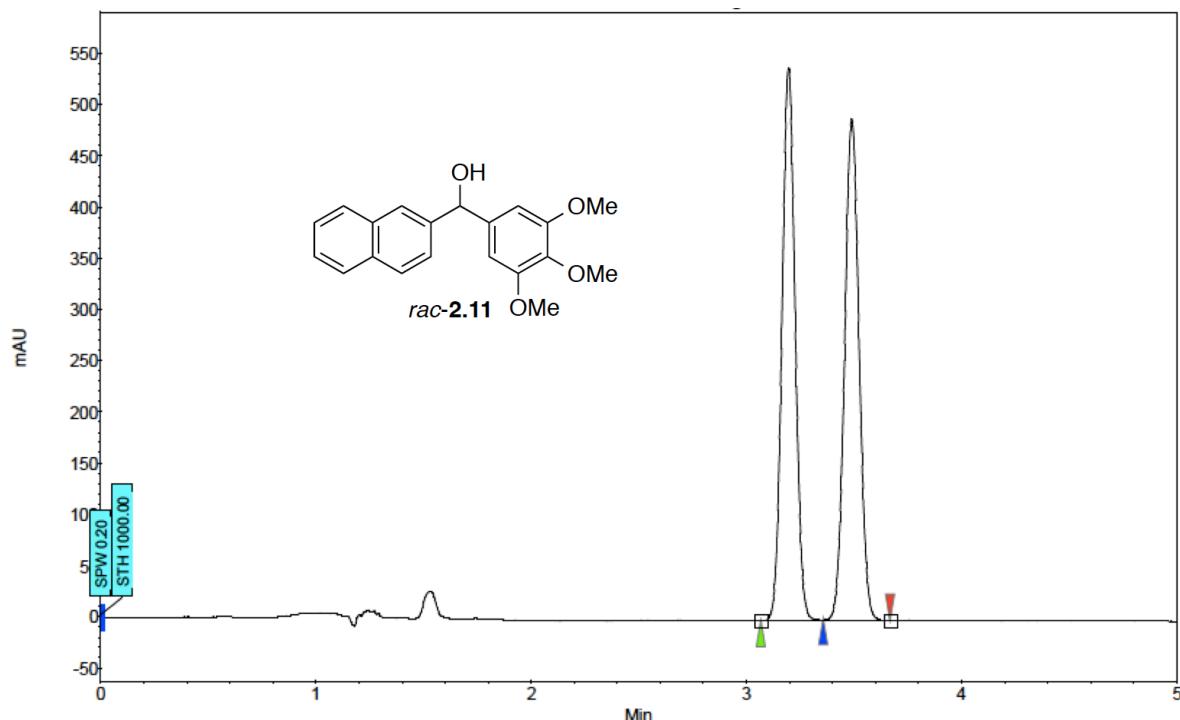
Index	Name	Start [Min]	Time [Min]	End [Min]	RT Offset [Min]	Quantity [% Area]	Height [ $\mu$ V]	Area [ $\mu$ V.Min]	Area [%]
1	UNKNOWN	3.35	3.46	3.68	0.00	99.06	1372.2	129.7	99.060
2	UNKNOWN	3.68	3.74	3.82	0.00	0.94	20.3	1.2	0.940



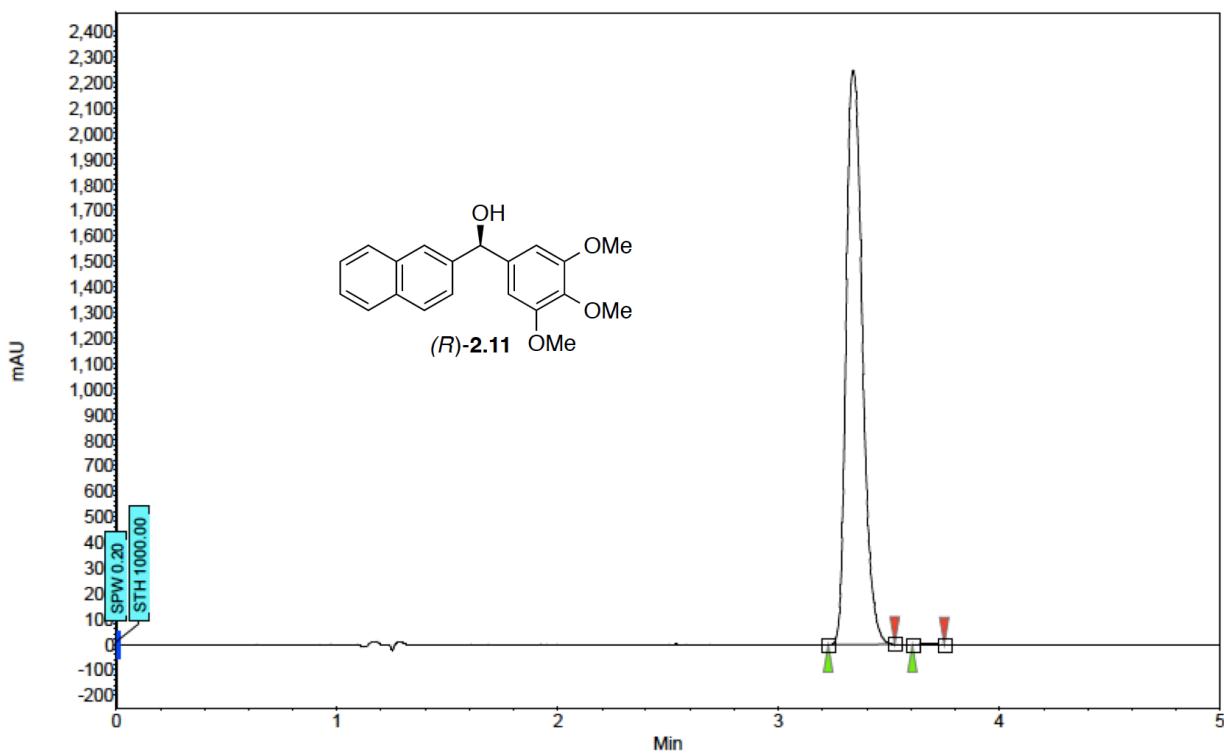
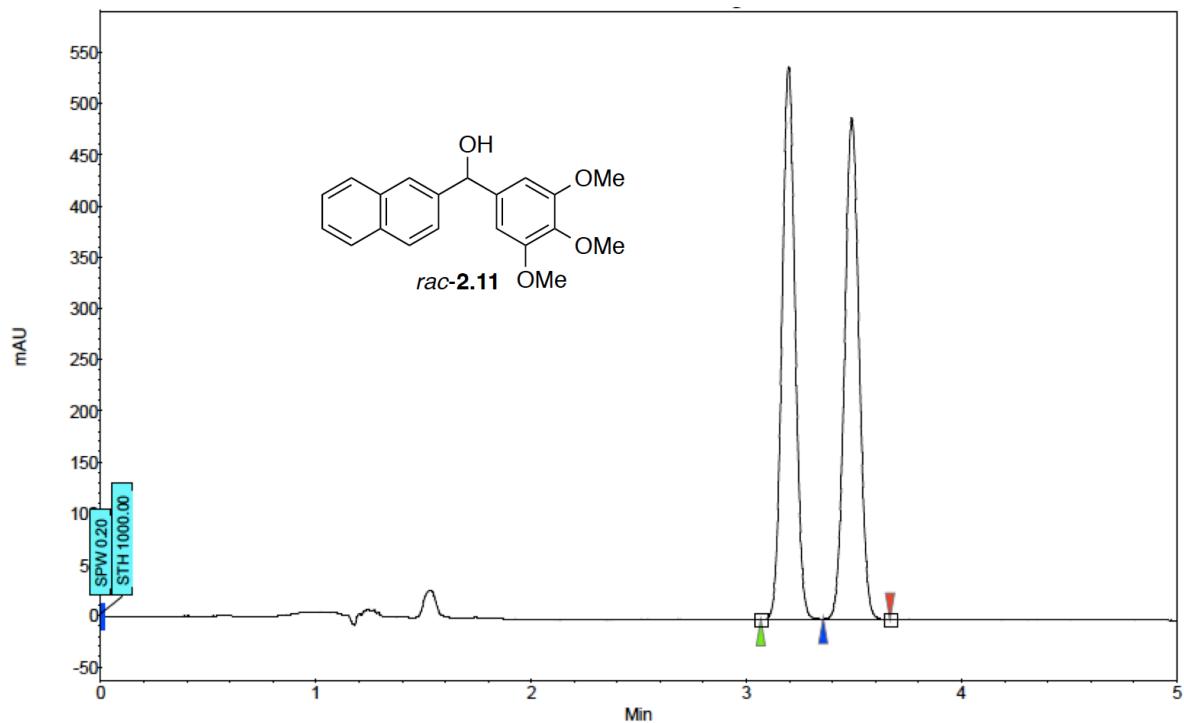
Index	Name	Start	Time	End	RT Offset	Quantity	Height	Area	Area
		[Min]	[Min]	[Min]	[Min]	[% Area]	[ $\mu$ V]	[ $\mu$ V.Min]	[%]
1	UNKNOWN	3.78	3.86	3.98	0.00	2.14	31.8	2.5	2.144
2	UNKNOWN	4.07	4.20	4.40	0.00	97.86	1288.9	115.4	97.856
Total						100.00	1320.7	118.0	100.000



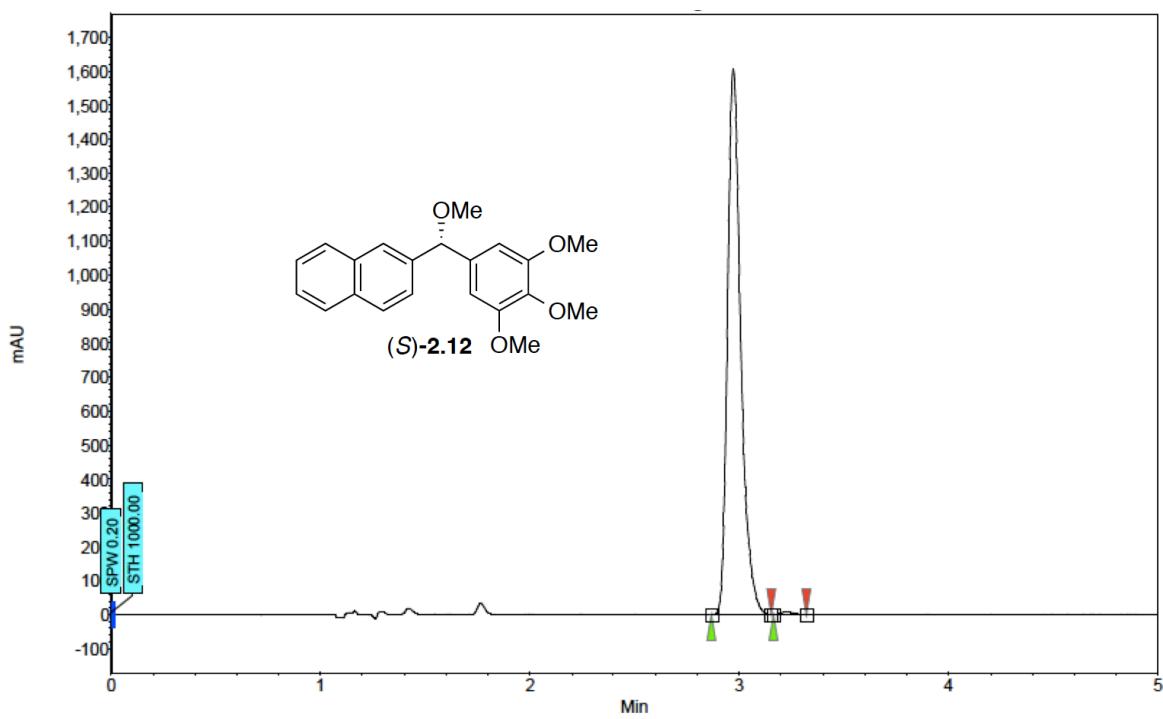
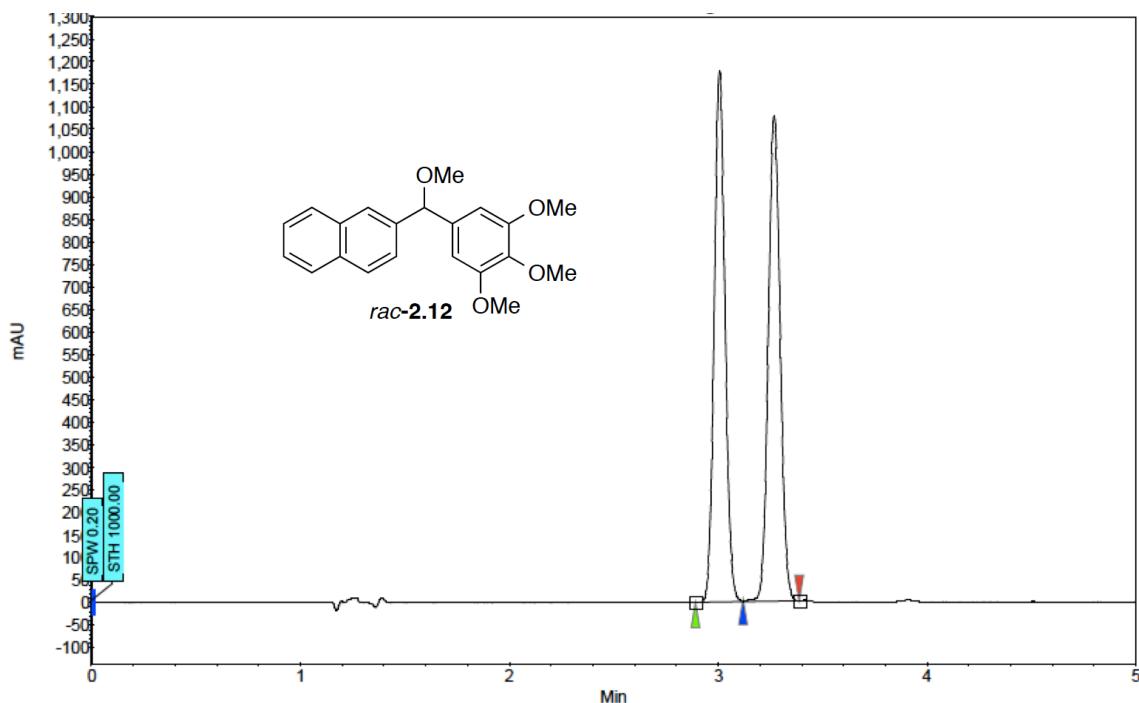
Index	Name	Start Time	Time	End	RT Offset	Quantity	Height	Area	Area
		[Min]	[Min]	[Min]	[Min]	[% Area]	[ $\mu$ V]	[ $\mu$ V.Min]	[%]
1	UNKNOWN	3.71	3.83	4.04	0.00	96.31	1344.1	108.6	96.310
2	UNKNOWN	4.15	4.25	4.37	0.00	3.69	49.8	4.2	3.690
Total						100.00	1393.9	112.8	100.000



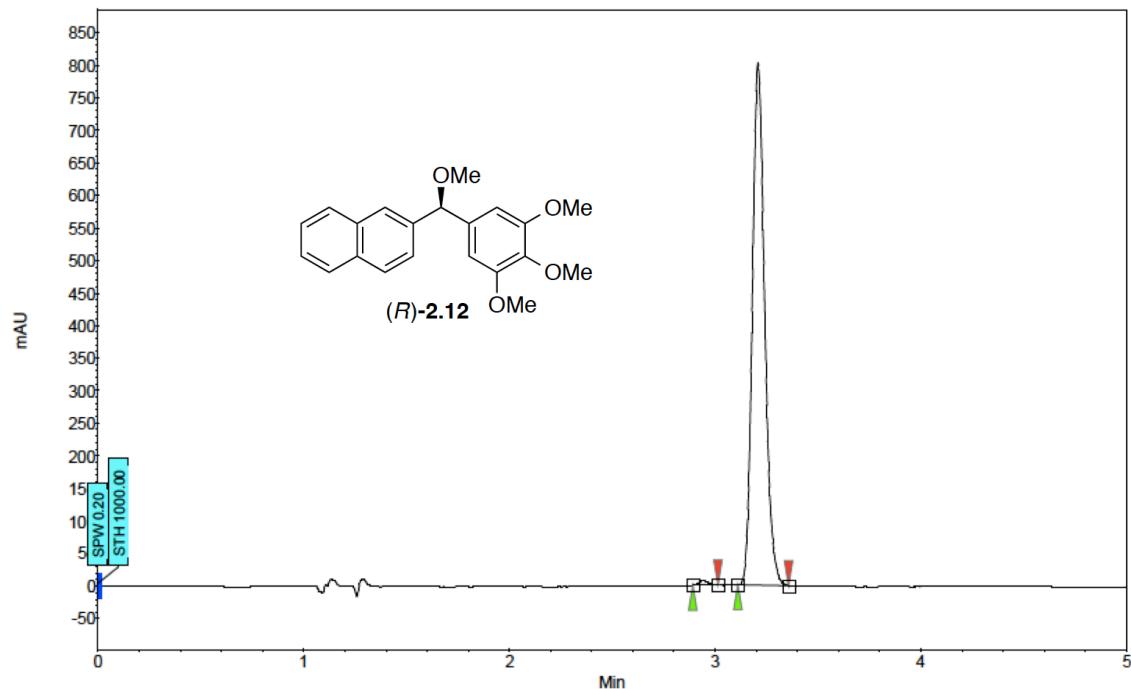
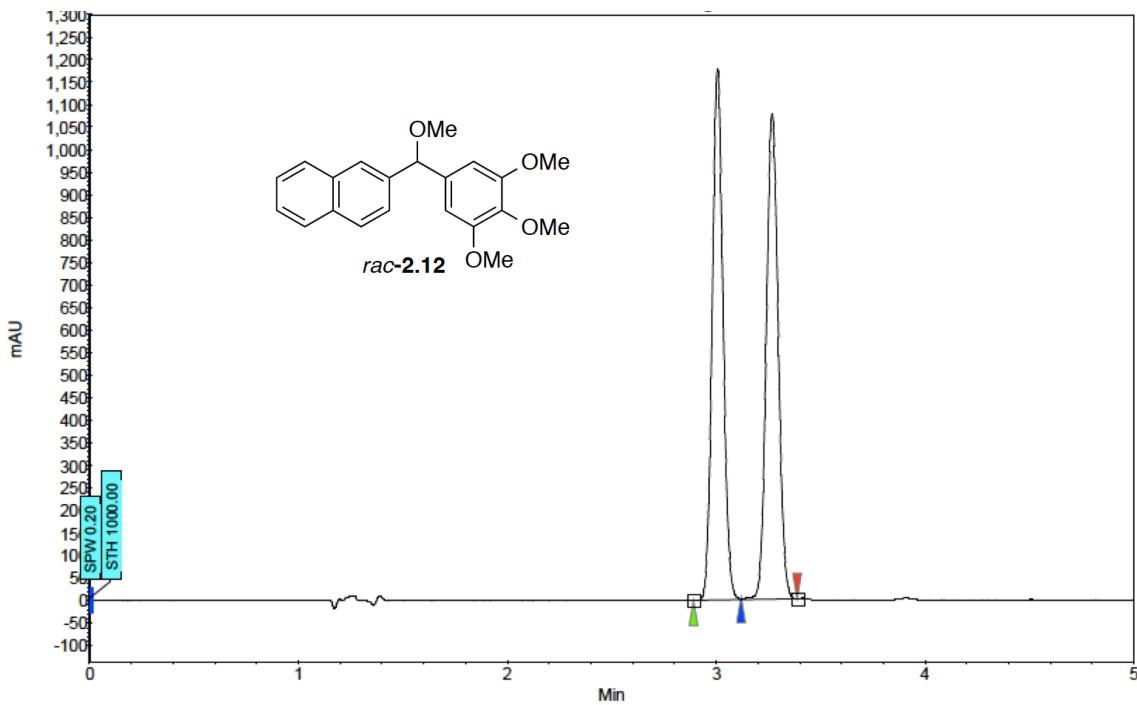
Index	Name	Start Time	Time	End	RT Offset	Quantity	Height	Area	Area
		[Min]	[Min]	[Min]	[Min]	[% Area]	[ $\mu$ V]	[ $\mu$ V.Min]	[%]
1	UNKNOWN	3.16	3.25	3.38	0.00	0.35	9.8	0.7	0.346
2	UNKNOWN	3.42	3.55	3.82	0.00	99.65	2248.5	212.7	99.654
Total						100.00	2258.3	213.4	100.000



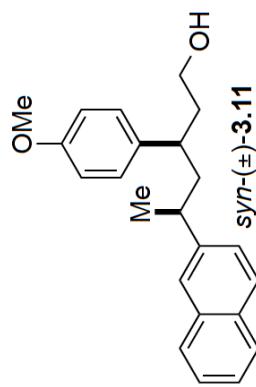
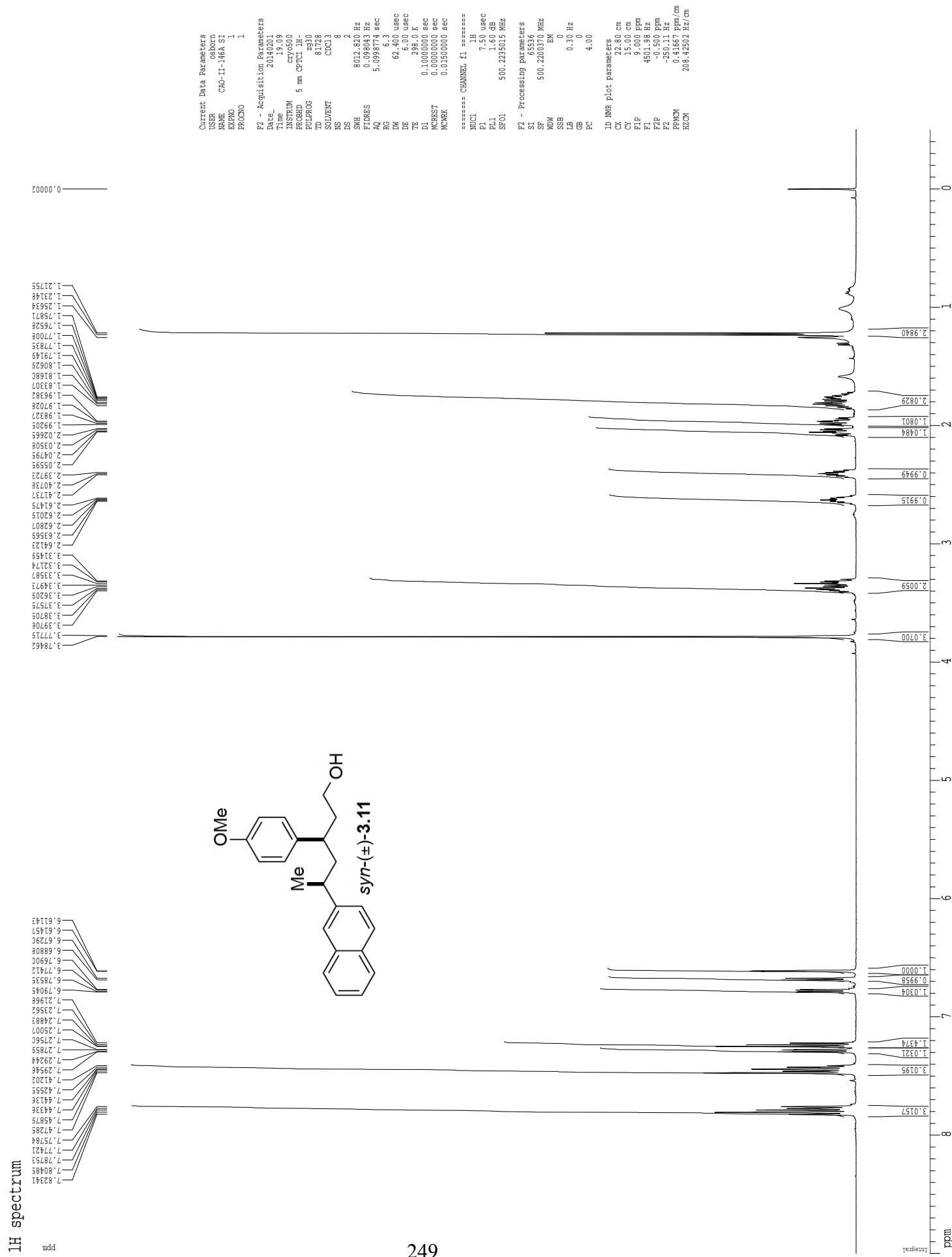
Index	Name	Start [Min]	Time [Min]	End [Min]	RT Offset [Min]	Quantity [% Area]	Height [ $\mu$ V]	Area [ $\mu$ V.Min]	Area [%]
1	UNKNOWN	3.23	3.34	3.53	0.00	99.74	2246.9	190.7	99.738
2	UNKNOWN	3.61	3.67	3.75	0.00	0.26	7.1	0.5	0.262
Total						100.00	2254.1	191.2	100.000



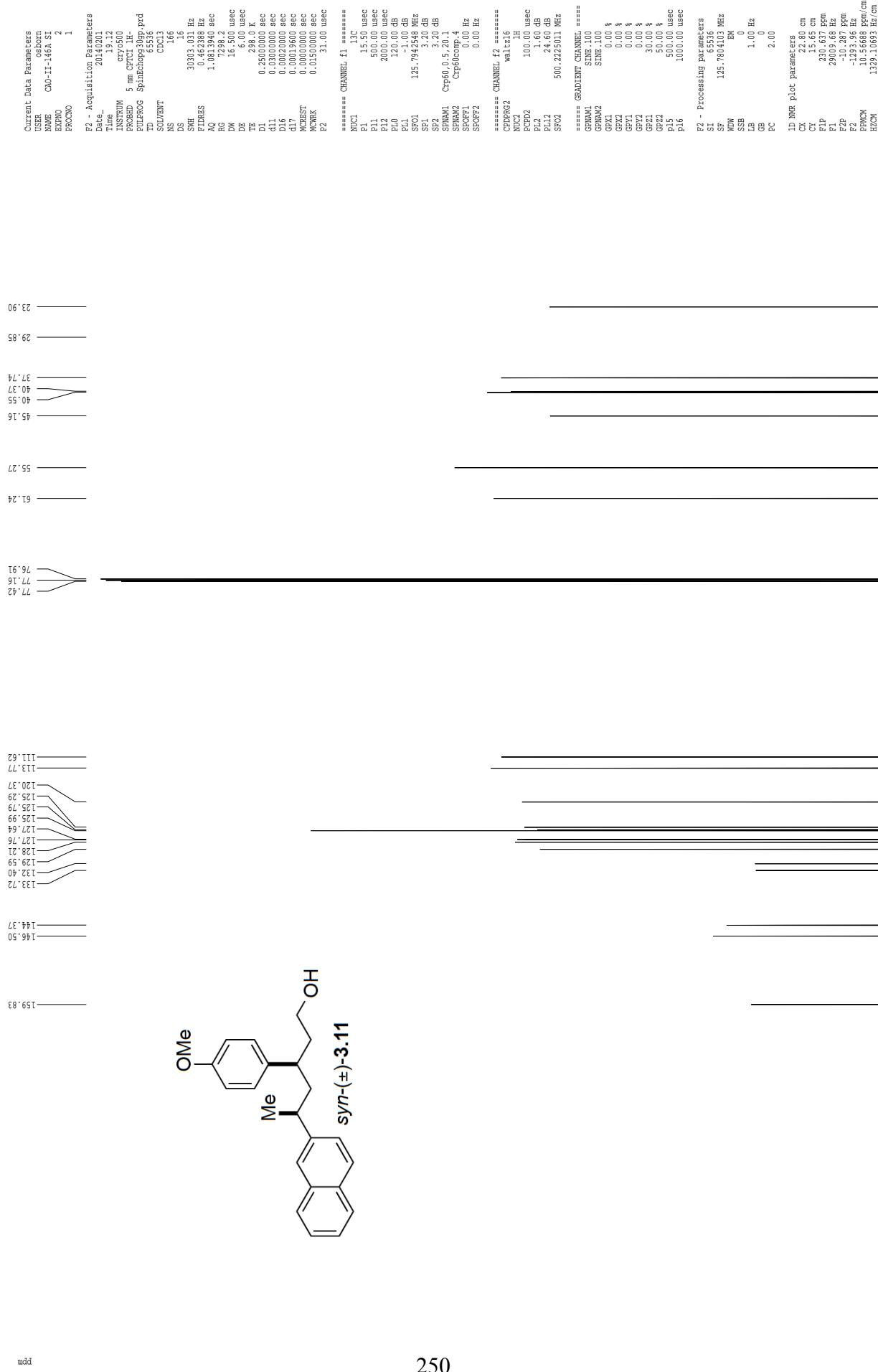
Index	Name	Start [Min]	Time [Min]	End [Min]	RT Offset [Min]	Quantity [% Area]	Height [ $\mu$ V]	Area [ $\mu$ V.Min]	Area [%]
1	UNKNOWN	2.87	2.97	3.15	0.00	99.52	1606.5	117.9	99.518
2	UNKNOWN	3.16	3.23	3.32	0.00	0.48	8.6	0.6	0.482
Total						100.00	1615.1	118.5	100.000



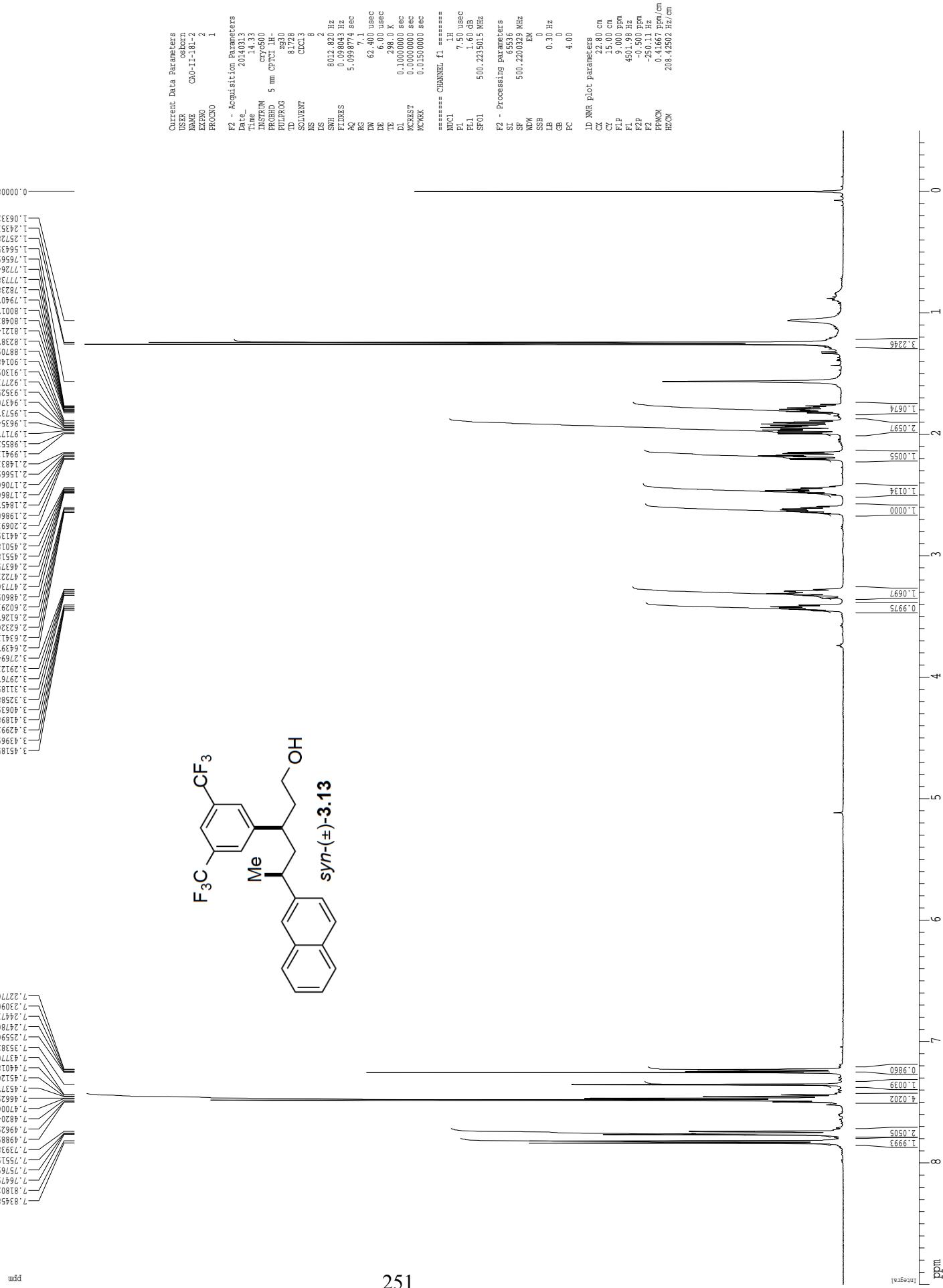
Index	Name	Start [Min]	Time [Min]	End [Min]	RT Offset [Min]	Quantity [% Area]	Height [ $\mu$ V]	Area [ $\mu$ V.Min]	Area [%]
1	UNKNOWN	2.89	2.94	3.01	0.00	0.69	6.7	0.4	0.691
2	UNKNOWN	3.11	3.21	3.36	0.00	99.31	802.1	55.6	99.309
Total						100.00	808.8	56.0	100.000



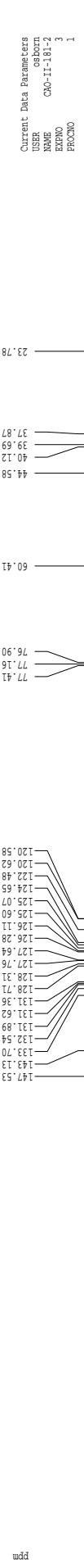
Z-restored spin-echo  $^{13}\text{C}$  spectrum with 1H decoupling



1H spectrum



Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling



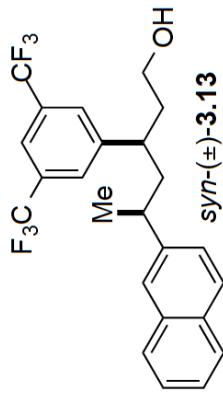
123.78

37.87  
39.69  
40.12  
44.58

60.41

76.90  
77.41

120.26  
122.48  
124.65  
125.07  
125.60  
126.11  
126.28  
127.64  
127.76  
128.31  
128.33  
129.17  
129.55  
131.62  
131.68  
131.70  
132.53  
133.17  
134.13  
134.53



```

Current Data Parameters
USER   :   gsborn
NAME   :   CHO-II-181-2
EXENO  :   3
PRCNO  :   1

F2 - Acquisition Parameters
Date   :   2014/3/3
Time   :   14:37
INSTRUM:   cry500
PROBHD :   5 mm PCD1 1H-
PULPROG:   SpinEditCh30DP.prd
TD     :   65536
SOLVENT:   CHCl3
CR1    :   2048
CR2    :   2048
DS     :   16
SF     :   3003.021 Hz
ETDE   :   0.467338 Hz
TDRES  :   1.003940 sec
AQ     :   11211.3
RG     :   16.500 usec
TE     :   296.1 K
D1     :   0.260000 sec
D11    :   0.000000 sec
D16    :   0.002000 sec
D17    :   0.001960 sec
MC1   :   0.000000 sec
MC2T  :   0.000000 sec
MCNT  :   0.050000 sec
F2R1   :   31.00 usec
F2R2   :   0.00 usec

===== CHANNEL F1 =====
NUCL1 :   ^13C
P1     :   15.30 usec
P11    :   50.00 usec
P12    :   200.00 usec
PL0    :   -120.00 dB
PL1    :   125.794258 MHz
SP01   :   3.20 dB
SP1    :   3.20 dB
SP0M1  :   Crp60/0.5,20.1
SPRM02 :   Crp60/0.5,20.1
CPD1   :   Cpd1cscope,4
SP0FF1:   0.00 Hz
SP0FF2:   0.00 Hz

===== CHANNEL F2 =====
NUCL2 :   ^1H
CPDPG2:   Walz16
P2P1   :   10.00 usec
P2L2   :   24.60 dB
SF02   :   500.2225011 MHz

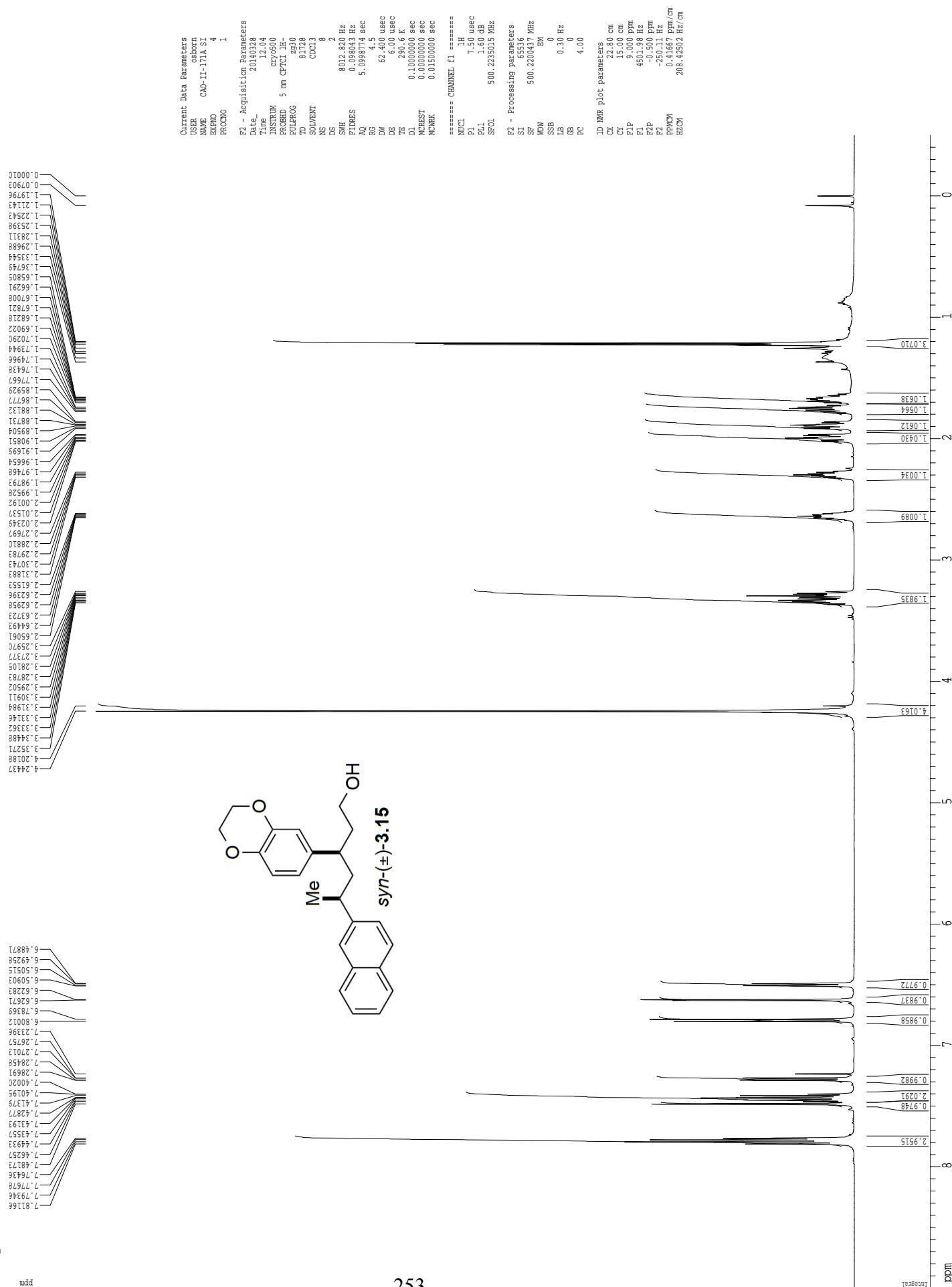
===== GRADIENT CHANNEL =====
GP0M1  :   SINE,100
GP0M2  :   SINE,100
GPX1   :   0.00 %
GPX2   :   0.00 %
GPY1   :   0.00 %
GPY2   :   0.00 %
GPZ1   :   30.00 %
GPZ2   :   50.00 %
GP22   :   50.00 %
P15    :   500.00 usec
P16    :   1000.00 usec

F2 - Processing Parameters
SI     :   65536
SF     :   125.7804080 MHz
WDW   :   EM
SSB   :   0
LB    :   1.00 Hz
GB    :   0
PC    :   2.00

1D NMR Plot Parameters
CX    :   22.80 cm
CY    :   15.65 cm
F1P   :   230.677 ppm
F1    :   230.958 Hz
F2P   :   -10.287 ppm
F2    :   -10.5636 Hz
PPCM  :   10.5668 ppm/cm
HZCM :   1323.10693 Hz/cm

```

CA0-II-171A SI  
1H spectrum



*Syn*-( $\pm$ )-3.15

CAO-II-171A  
Z-restored spin-echo 13C spectrum with 1H decoupling

ppm

23.82  
29.82  
37.59  
40.41  
45.35  
46.49  
64.33  
65.16  
76.90  
77.41  
116.06  
117.10  
118.16  
121.16  
125.24  
125.88  
125.93  
127.69  
128.14  
129.30  
133.69  
138.01  
141.99  
143.57  
144.33

```
Current Data Parameters
USER   : esborn
NAME   : CAO-II-171A.S1
EXNO   : 5
PROCNO : 1

F2 - Acquisition Parameters
Date   : 2014/3/8
Time   : 12:07
INSTRUM: INSPIN
PROBHD : 5 mm PCD1 1H
PULPROG: SpinEcho3D90DP.prd
TD    : 65536
SOLVENT: C6C13
NS    : 12
DS    : 16
SW1   : 2003.021 Hz
SF1   : 0.46338 Hz
ET1   : 0.001390 sec
AQ1   : 16.500 usec
TE    : 29.7 K
D1    : 0.260000 sec
d11   : 0.000000 sec
D16   : 0.002000 sec
t11   : 0.001960 sec
MC1   : 0.000000 sec
MC2T  : 0.000000 sec
MCNT  : 1
DW1   : 0.050000 sec
F2PR1 : 31.00 usec
F2PR2 : 0.00 Hz

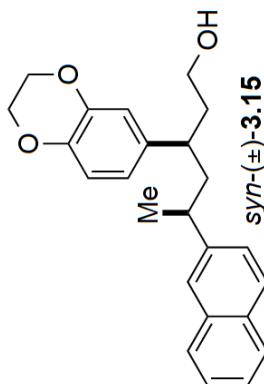
===== CHANNEL F1 =====
NUCL1 : 13C
P1    : 15.50 usec
P11   : 50.00 usec
P12   : 200.00 usec
PL0   : -120.00 dB
PL1   : 125.794258 MHz
SP1   : 3.20 dB
SP2   : 3.20 dB
SPW01 : Crp60.0,5,20.1
SPW02 : Crp60.0,5,20.1
CPD1  : 4
CPD1F1 : 0.00 Hz
CPD1F2 : 0.00 Hz

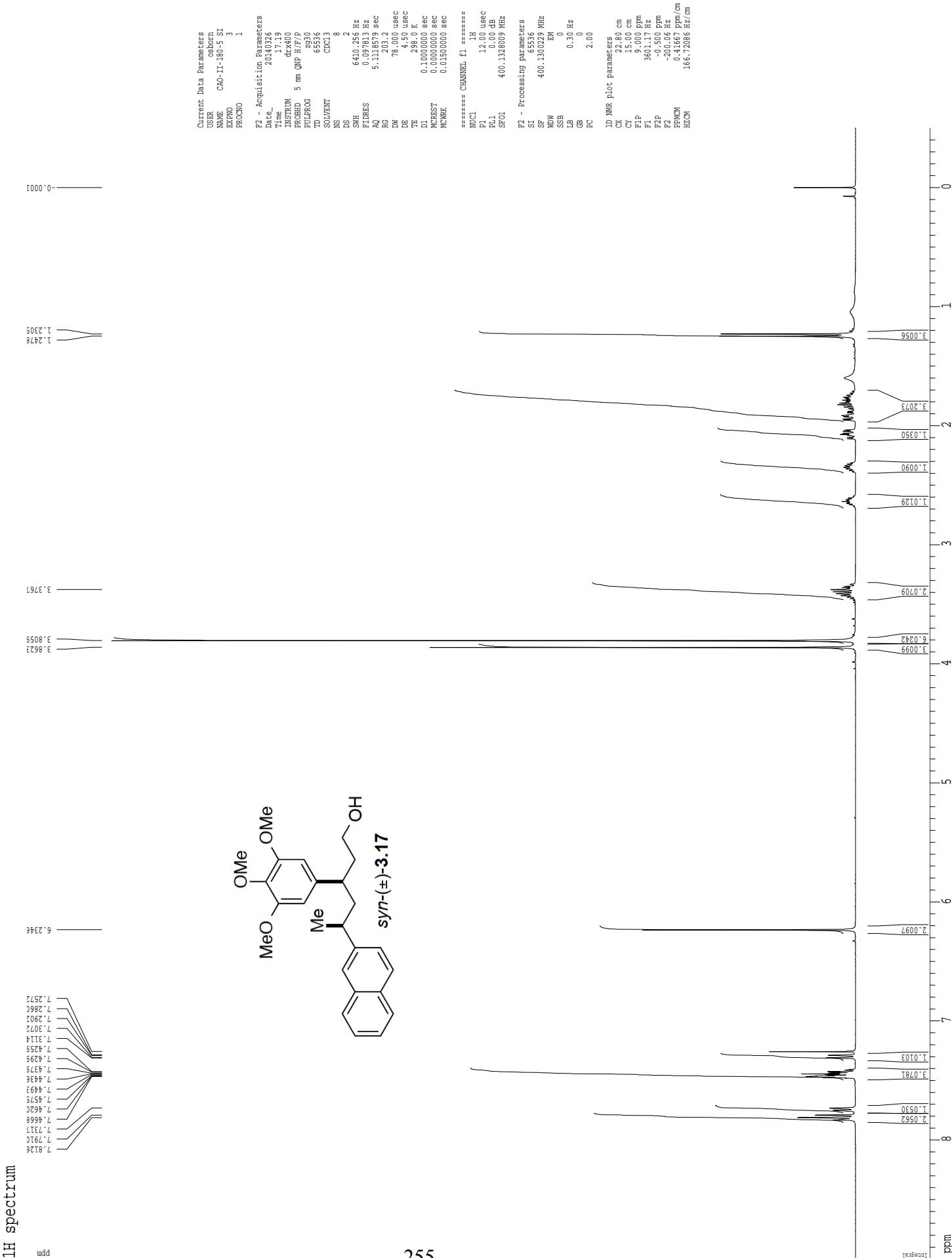
===== CHANNEL F2 =====
NUCL2 : 1H
CPDPG2 : 65536
P2P1  : 10.00 usec
P2L2  : 1.60 dB
PL12  : 24.60 dB
SF02  : 500.2225011 MHz

===== GRADIENT CHANNEL =====
GP0M1 : SINE,100
GP0M2 : SINE,100
GPX1  : 0.00 %
GPX2  : 0.00 %
GPY1  : 0.00 %
GPY2  : 0.00 %
GPZ1  : 30.00 %
GPZ2  : 50.00 %
GP15  : 500.00 usec
GP16  : 1000.00 usec

F2 - Processing Parameters
SI    : 65536
SF    : 125.7804132 MHz
WDW  : EM
SSB  : 0
LB   : 1.00 Hz
GB   : 0
PC   : 2.00

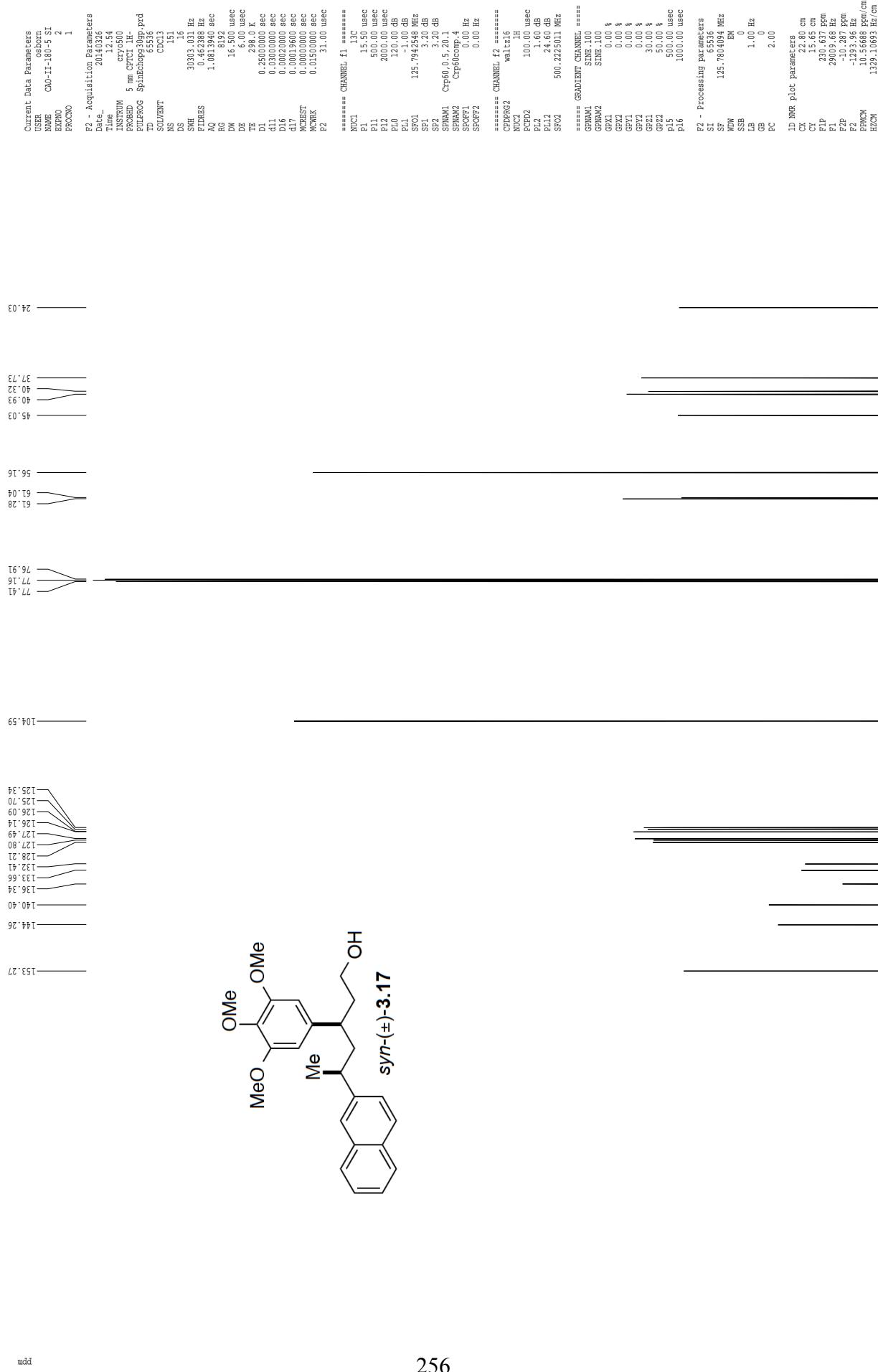
1D NMR Plot Parameters
CX   : 22.80 cm
CY   : 15.65 cm
F1P  : 230.67 ppm
F1I  : 280.9.68 Hz
F2P  : -10.287 ppm
F2I  : -129.3.6 Hz
PPCM : 10.5668 ppm/cm
HZCM: 1325.10706 Hz/cm
```

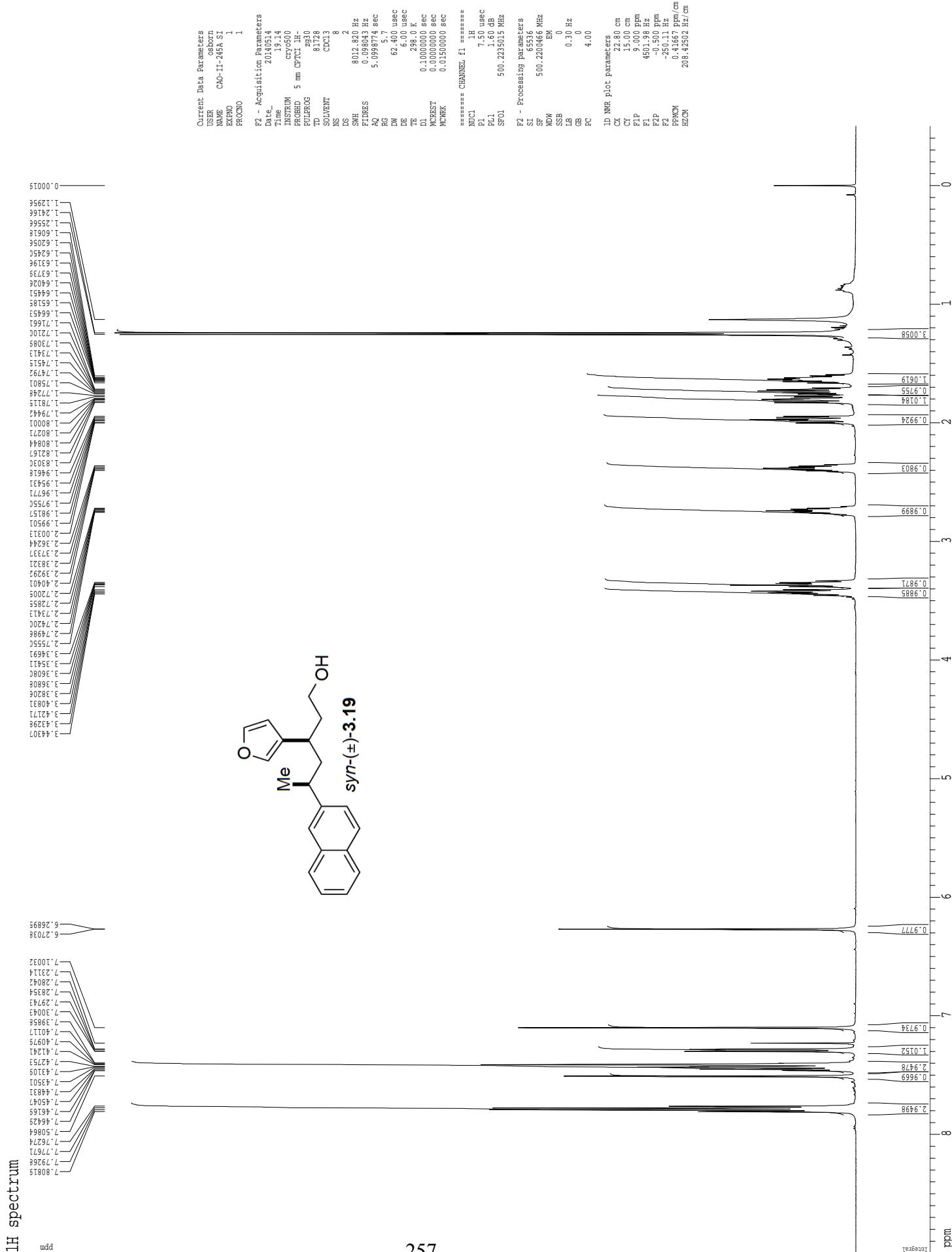




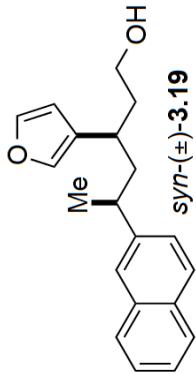
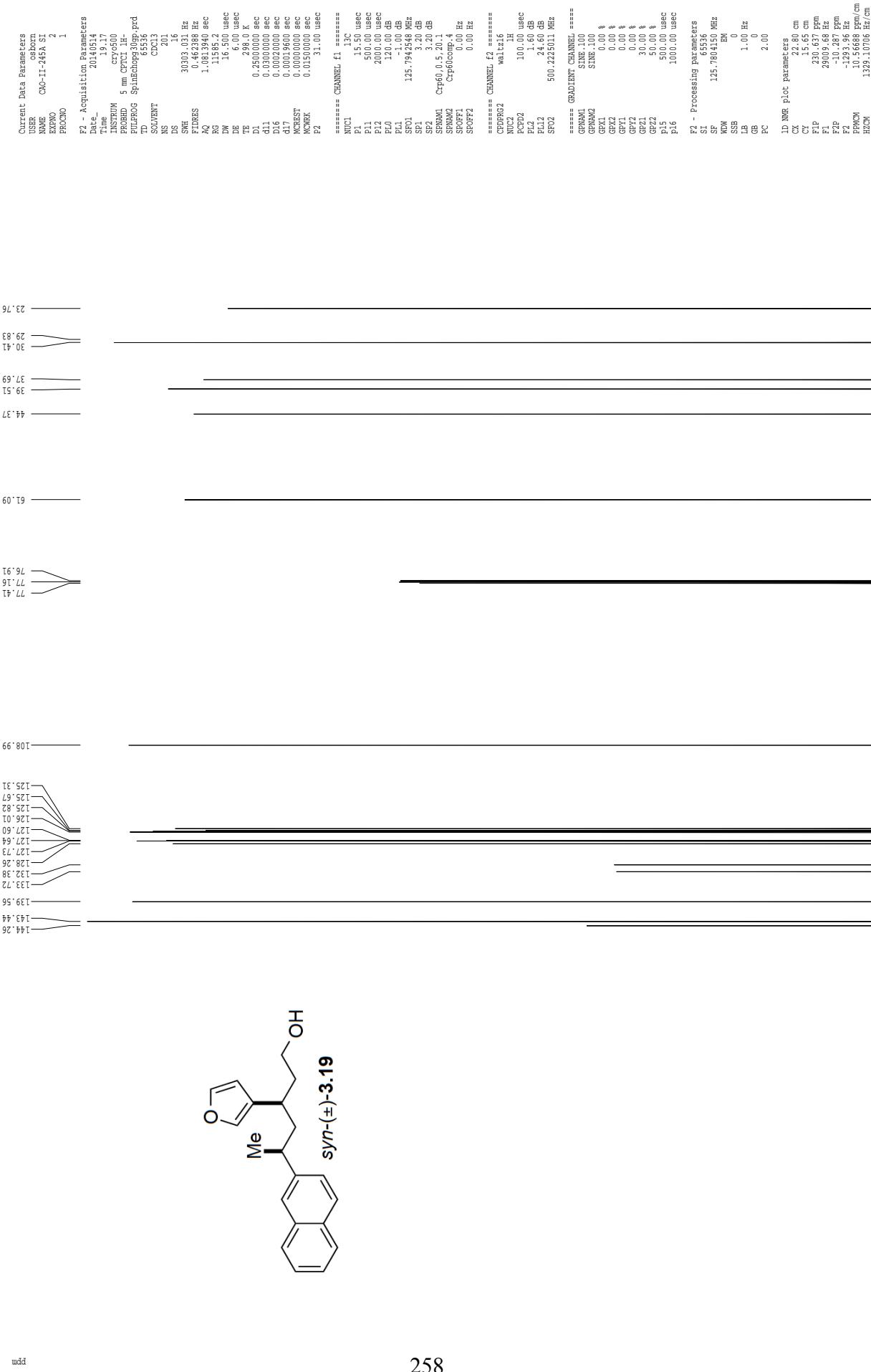
1H spectrum

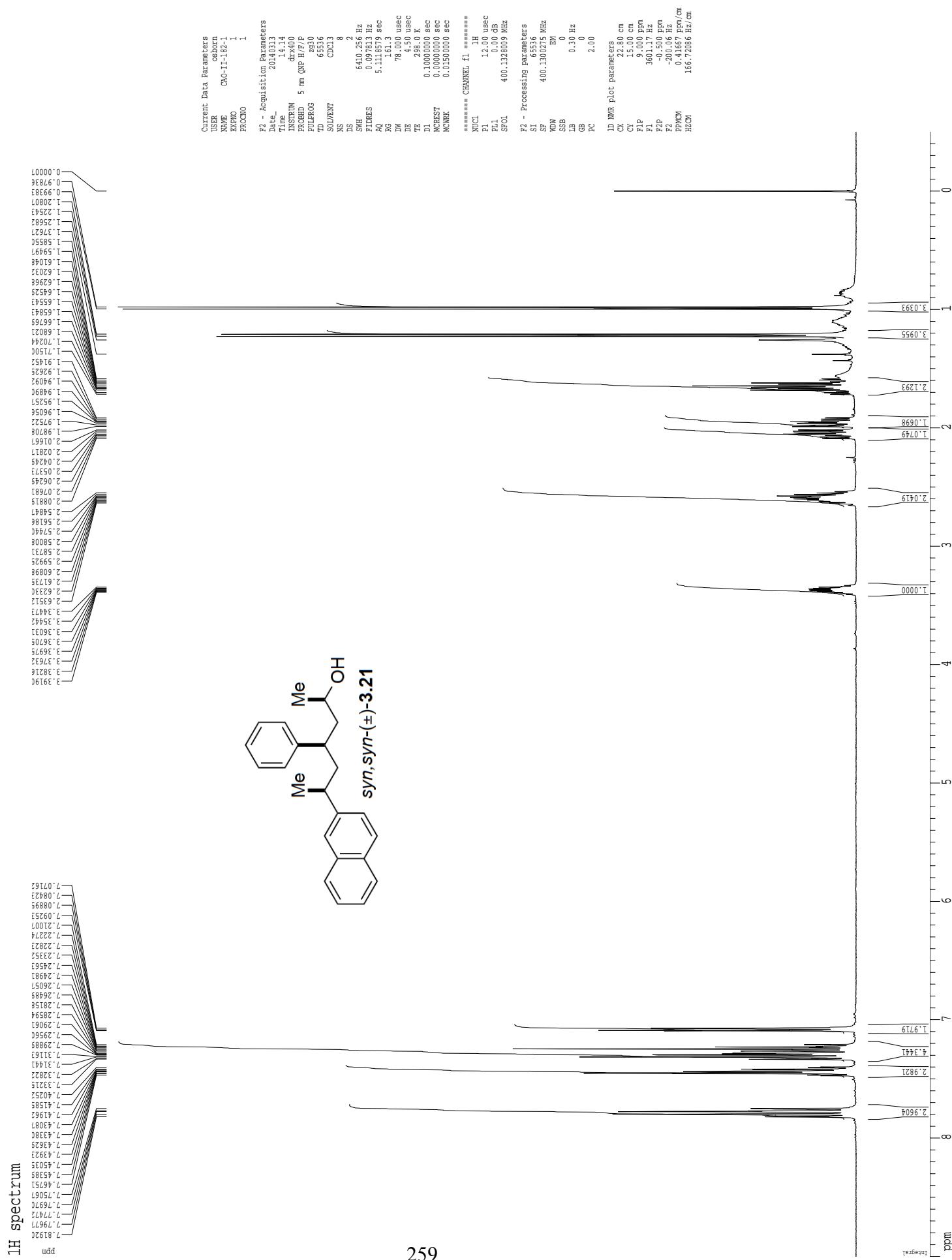
Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling



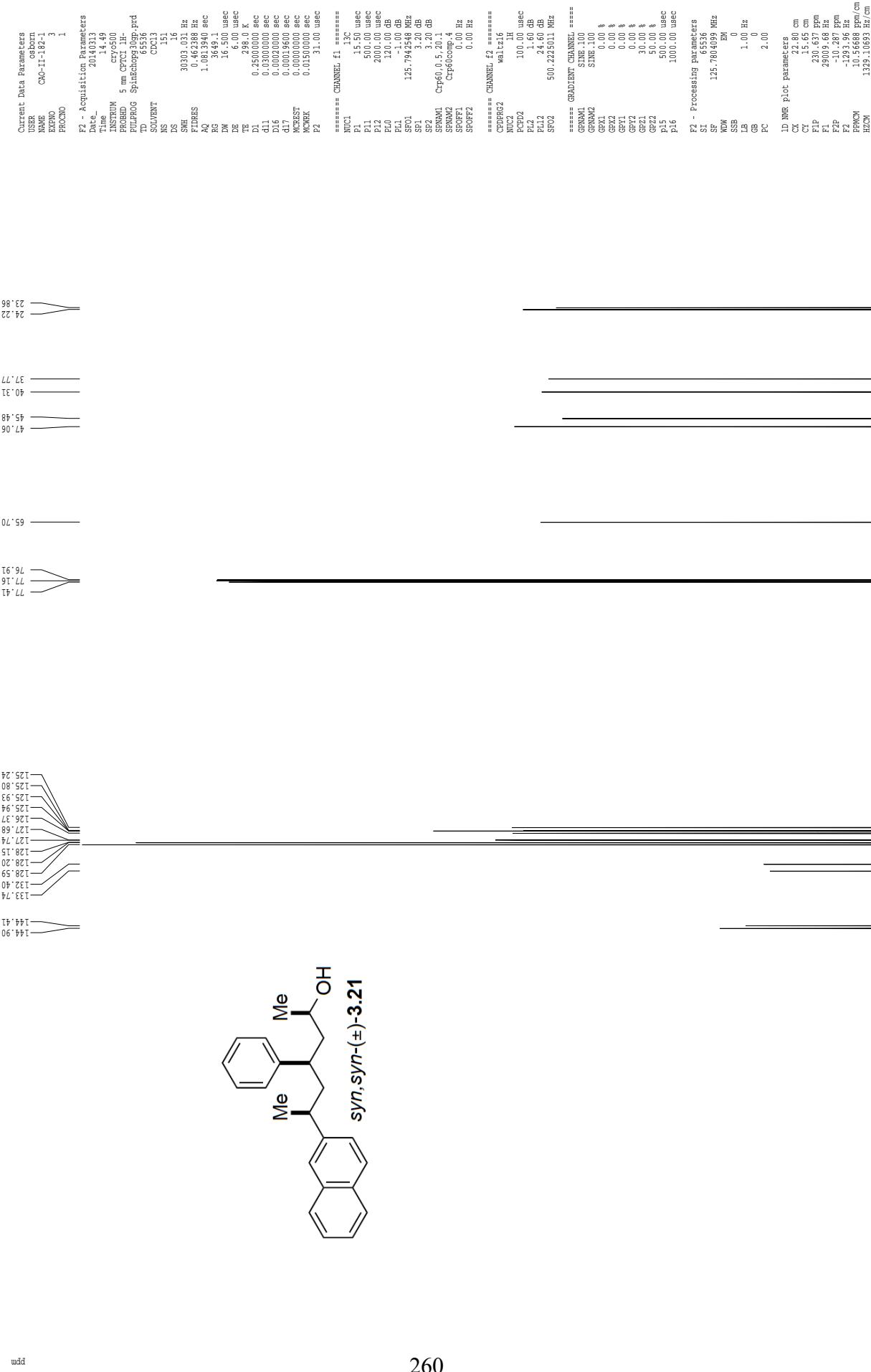


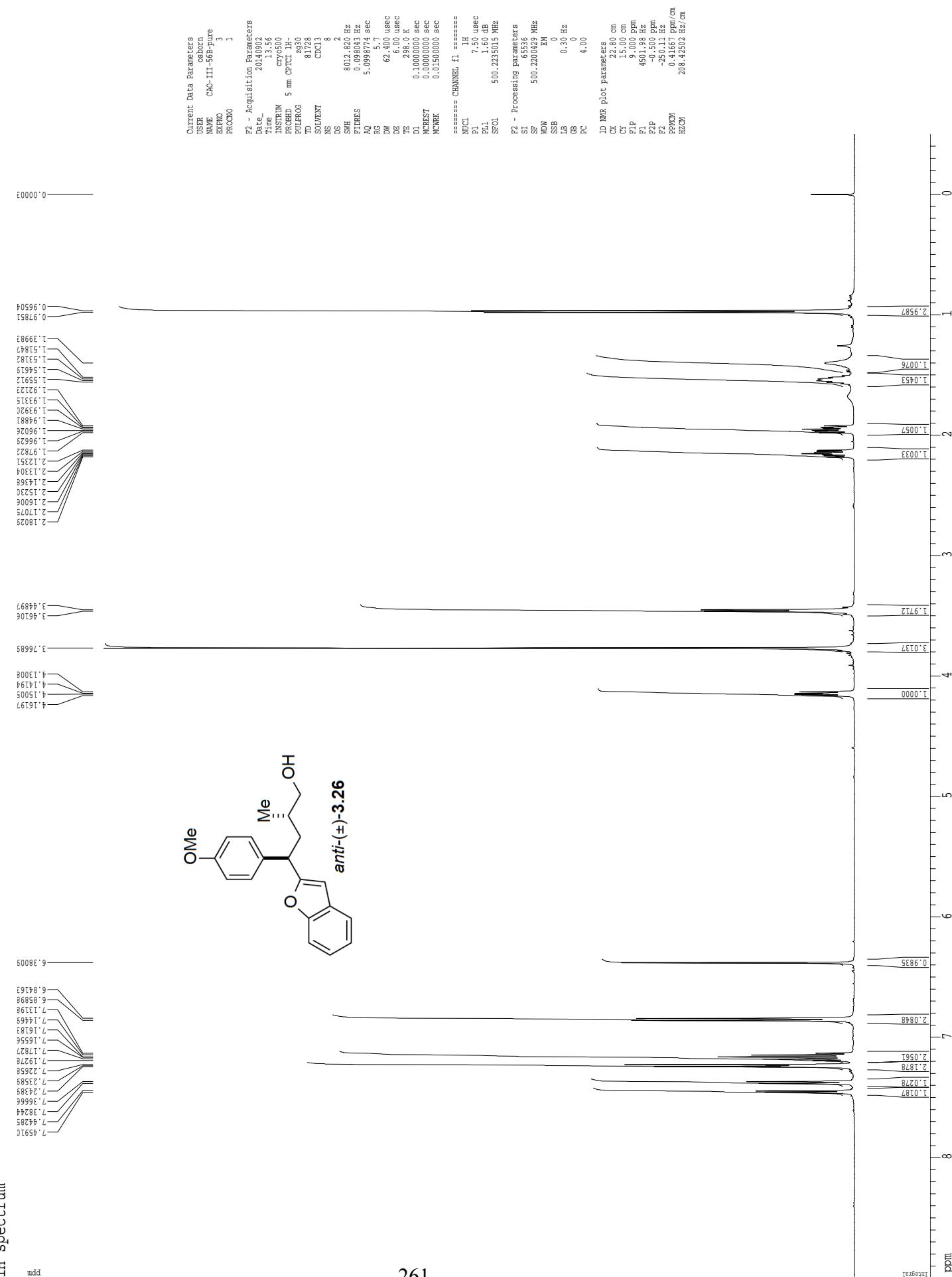
Z-restored spin-echo  $^{13}\text{C}$  spectrum with 1H decoupling





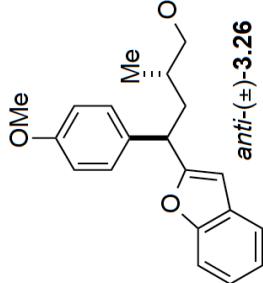
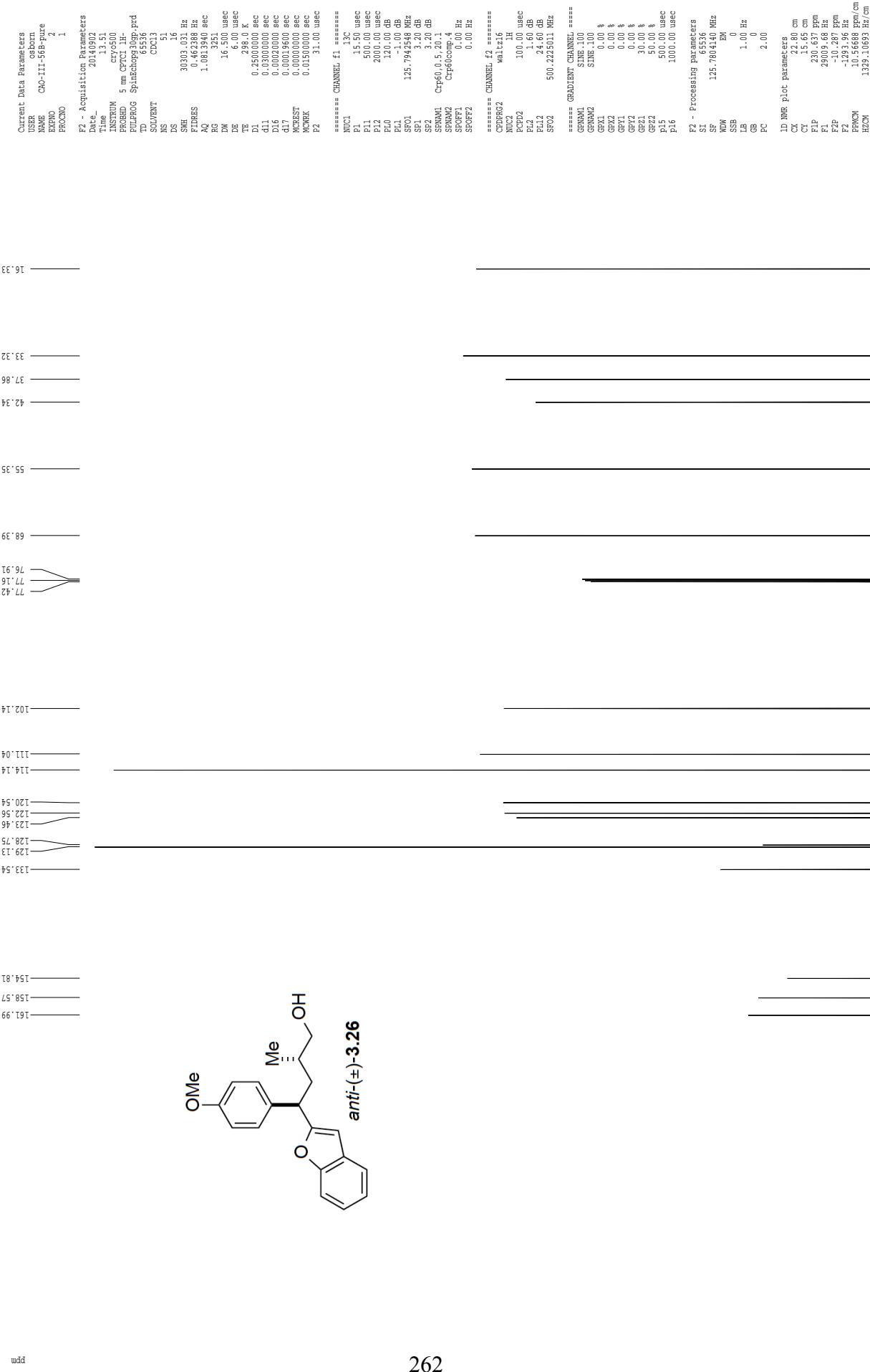
Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling



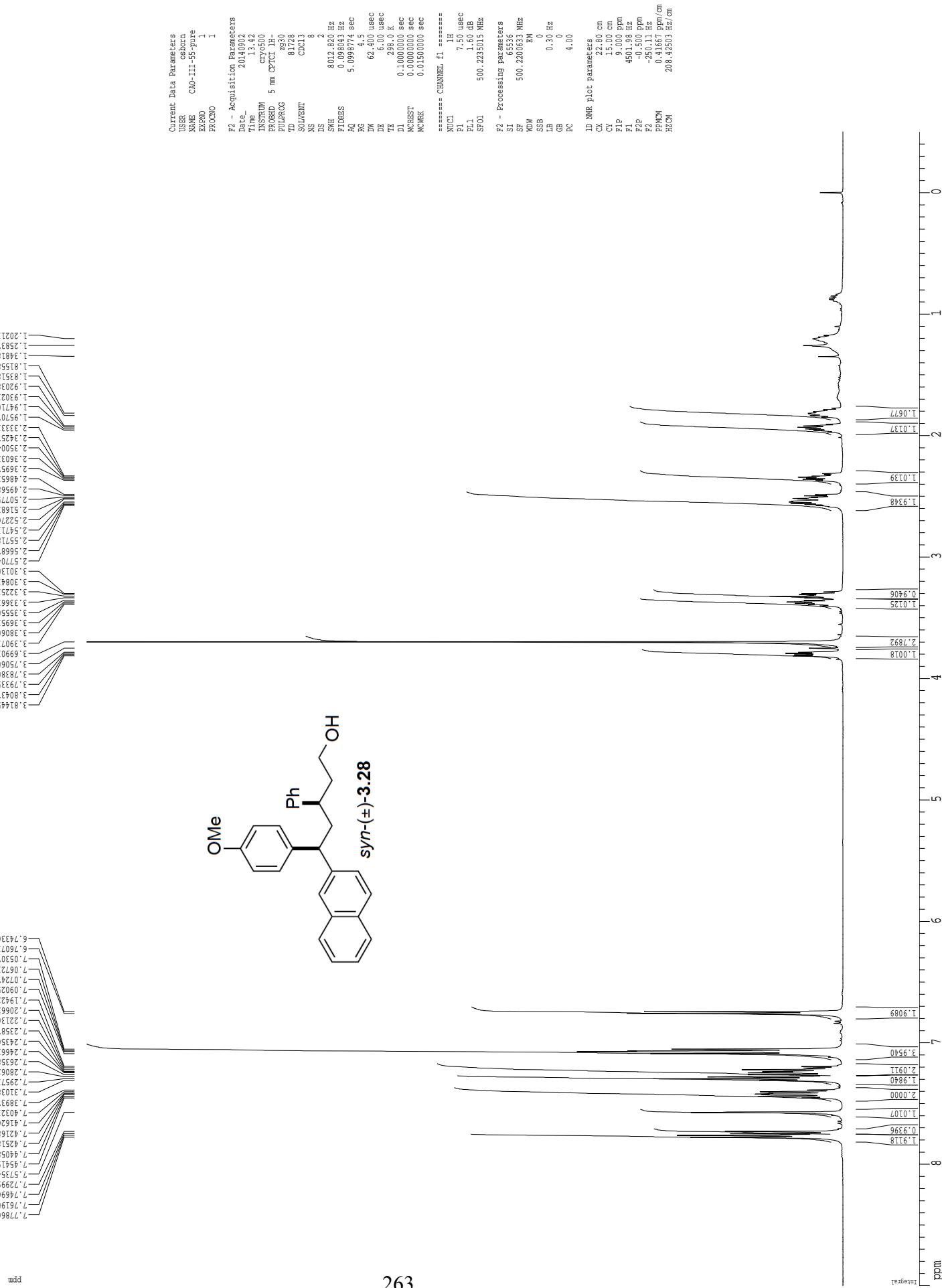


1H spectrum

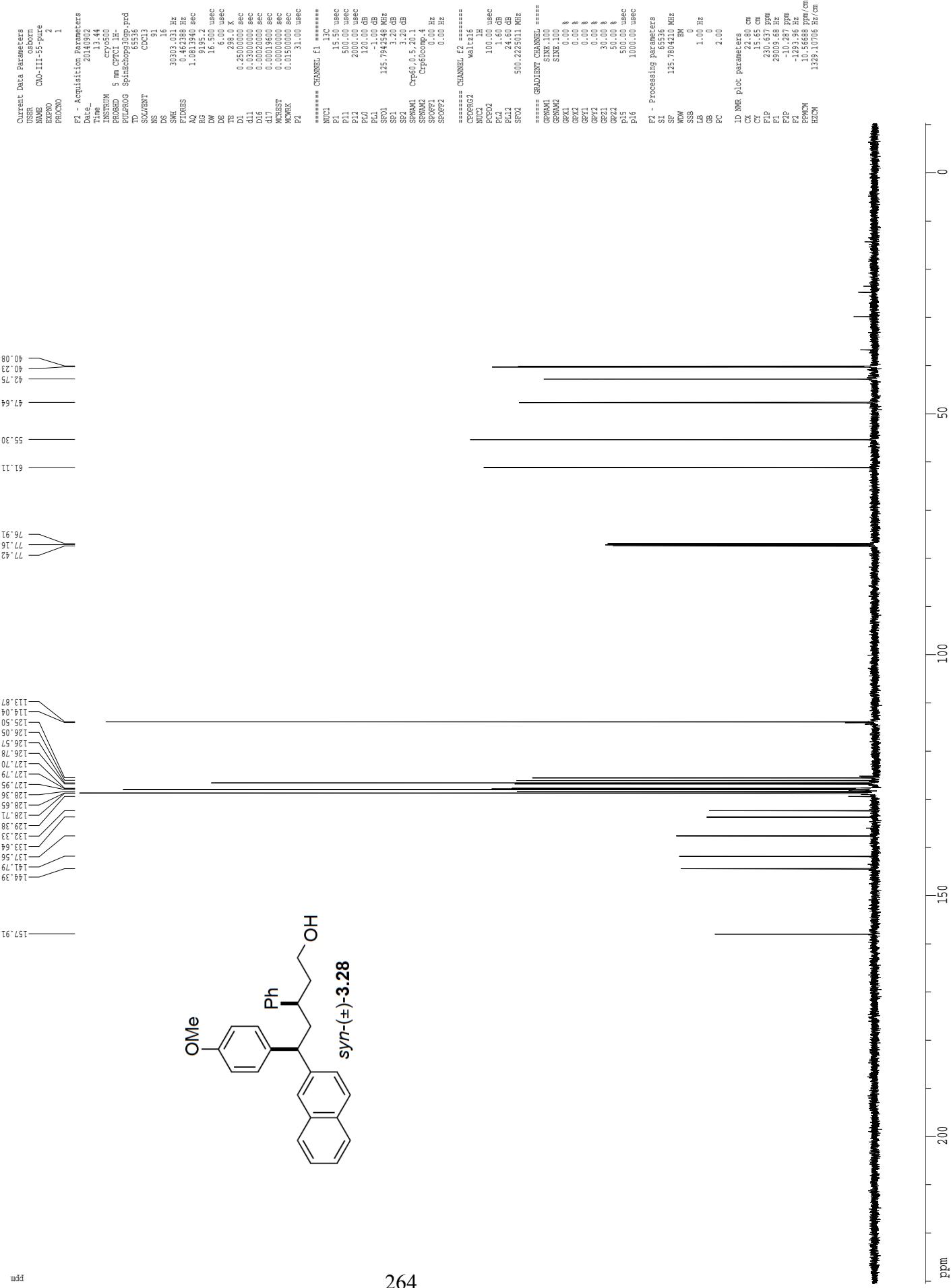
Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

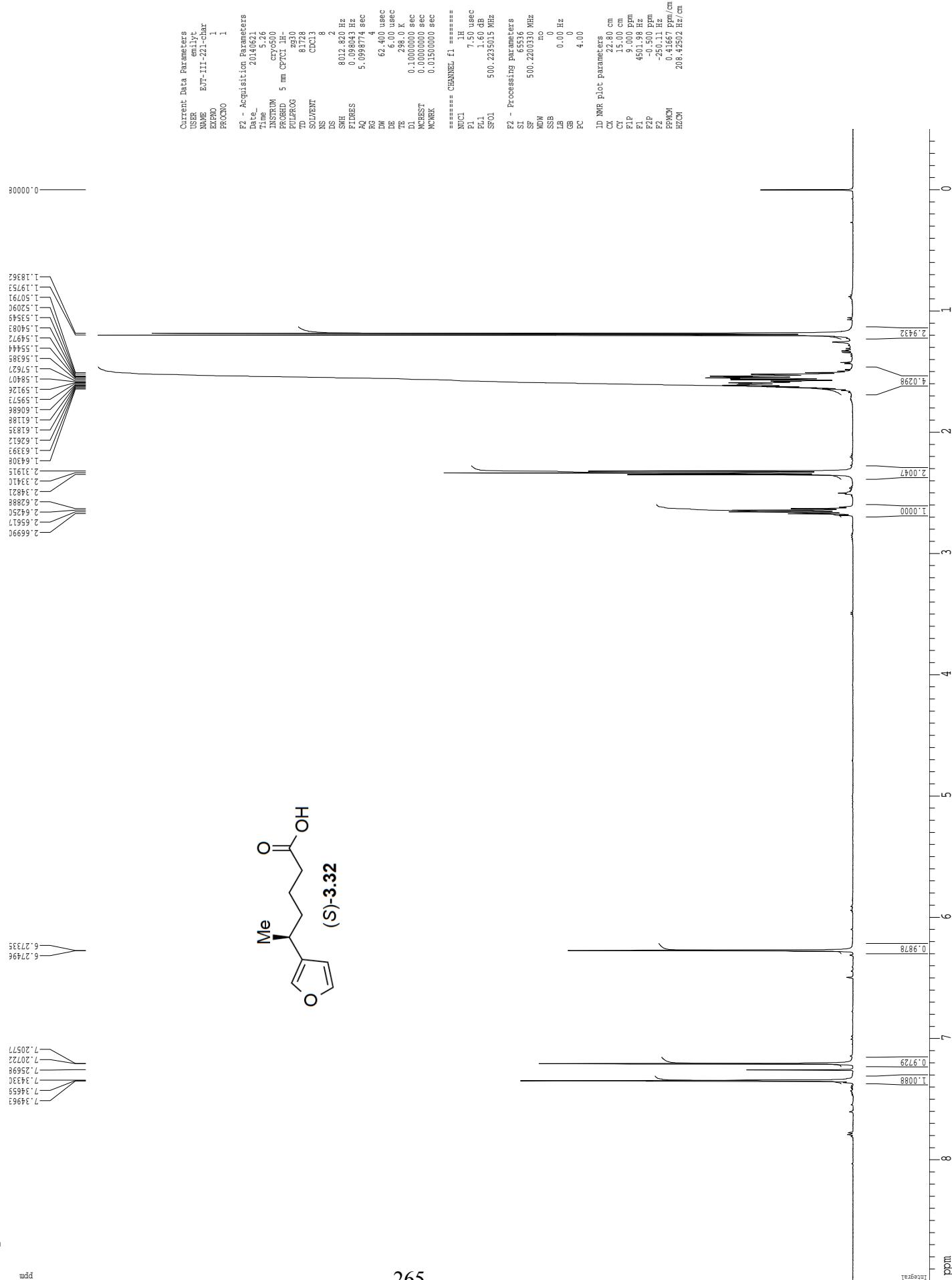


1H spectrum



Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling





1H spectrum

Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

ppm

180.29  
142.95  
138.21  
130.45  
109.47

77.16  
77.41  
76.91

36.91  
34.18  
30.04  
22.42  
21.16

```

Current Data Parameters
  USER          emlit
  NAME          EIT11-221-1har
  EXNO          2
  PRGNO         1

F2 - Acquisition Parameters
  Date        20140621
  Time        5.99
  INSTRUM     cry500
  PROBHD     5 mm CCP1 1H-
  PULPROG    SpinEditCh30P.prd
  TD          65536
  SOLVENT    CFC13
  NS          128
  D1          1.00 usec
  T1           296.0 K
  SWH        0.260000 sec
  FIDRES    0.015390 Hz
 AQ          1.001390 sec
  R1          2945.7
  R2          16.500 usec
  R3          4.00 usec
  TE           296.0 K
  D11         0.000000 sec
  D12         0.000000 sec
  D13         0.000000 sec
  D14         0.0001960 sec
  MC          1024
  N1          0.000000 sec
  NC          0.015000 sec
  PC          0.000000 sec
  P2          31.00 usec

=====
CHANNEL F1 =====
NUCL1      13C
P1          15.30 usec
P11         50.00 usec
P12         200.00 usec
PL0          120.00 dB
PL1          -1.00 dB
SP01        125.794258 MHz
SP1          3.20 dB
SP2          3.20 dB
SPR001      Crp60.0,5,20.1
SPR002      Crp60.0,5,20.1
CPDPR004   Crp60.0,5,20.1
SPCPFF1    0.00 Hz
SPCPFF2    0.00 Hz

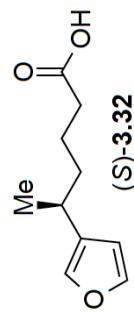
=====
CHANNEL F2 =====
NUCL2      1H
WALTZ16
PCPD2      10.00 usec
PL2          1.60 dB
PL12        24.60 dB
SF02        500.2225011 MHz

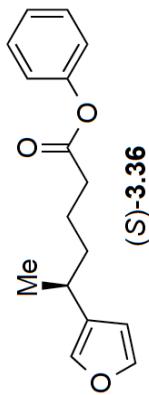
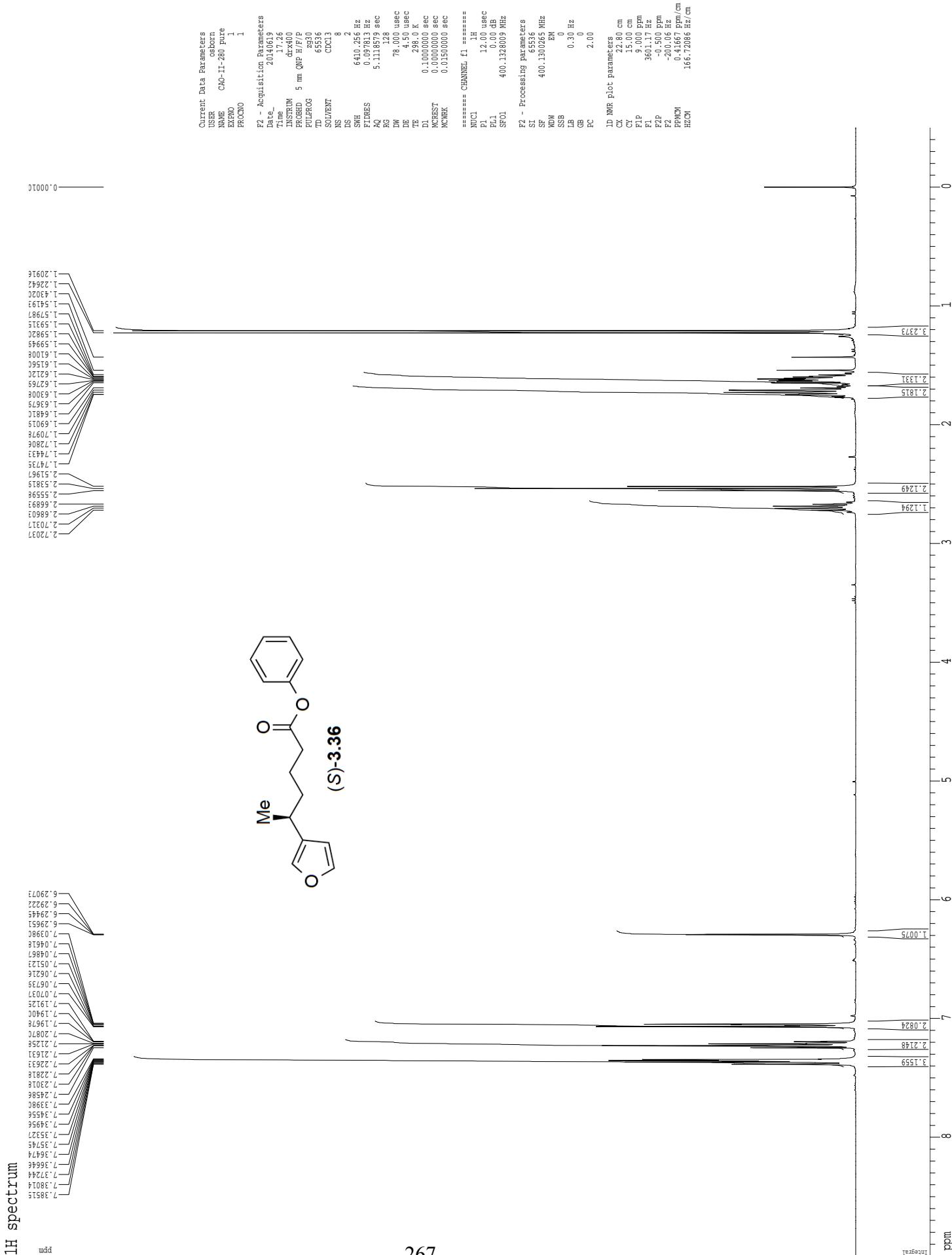
=====
GRADIENT CHANNEL =====
CPDPRG2
GPR001      SINE,100
GPR002      SINE,100
GPX1        0.00 %
GPX2        0.00 %
GPY1        0.00 %
GPY2        0.00 %
GPZ1        30.00 %
GPZ2        50.00 %
GP5         50.00 usec
P15        500.00 usec
P16        1000.00 usec

=====
F2 - Processing Parameters
SI          65536
SF          125.7804099 MHz
WDW        no
SSB        0
LB          0.00 Hz
GB          0
PC          2.00

1D NMR Plot Parameters
CX          22.80 cm
CY          15.65 cm
F1P        200.000 ppm
F1         2515.608 Hz
F2P        0.000 ppm
F2         0.000 Hz
PPCM      8.77133 ppm/cm
HZDW     1103.33691 Hz/cm

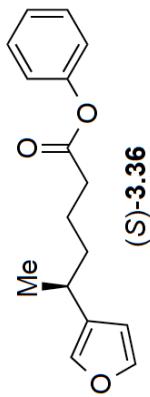
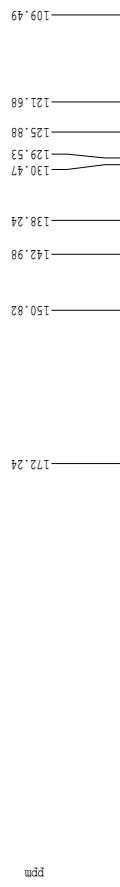
```





1H spectrum

Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling



```

Current Data Parameters
USER      : esborn
NAME      : E77-11-29
EXNO     : 3
PRCENO   : 1

F2 - Acquisition Parameters
Date       : 20140615
Time       : 15:18
INSTRUM  : INSPINW
PROBHD  : 5 mm PCD1 1H
PULPROG : SpinEditCh30DP.prd
TD        : 65536
SOLVENT  : C6C13
NS        : 141
SWH      : 3003.021 Hz
ETRIM    : 0.463388 Hz
AQ        : 1.001390 sec
RG        : 4096
TE        : 16.500 usec
D1        : 300.00 usec
T1        : 296.0 K
D2        : 0.260000 sec
D11       : 0.000000 sec
D16       : 0.002000 sec
D17       : 0.001960 sec
MC1      : 0.000000 sec
MC2K    : 0.0150000 sec
F2R1     : 31.00 usec
F2R2     : 0.00 usec

===== CHANNEL F1 =====
NUC1    : 13C
P1      : 15.30 usec
P11     : 50.00 usec
P12     : 200.00 usec
PL0     : -12.00 dB
PL1     : 125.794258 MHz
SP01    : SP1
SP1     : 3.20 dB
SP001   : Crp60.0,5,20.1
SP002   : Crp60.0,5,20.1
CPD1    : Cpd1cscope,4
SP0FF1  : 0.00 Hz
SP0FF2  : 0.00 Hz

===== CHANNEL F2 =====
NUC2    : 1H
P1      : 1.50 usec
P11     : 50.00 usec
P12     : 200.00 usec
PL1     : -1.00 dB
SP02    : SF
SP2     : 3.20 dB
GP001   : Gp001
GP002   : Gp001
SF02    : 500.2225011 MHz

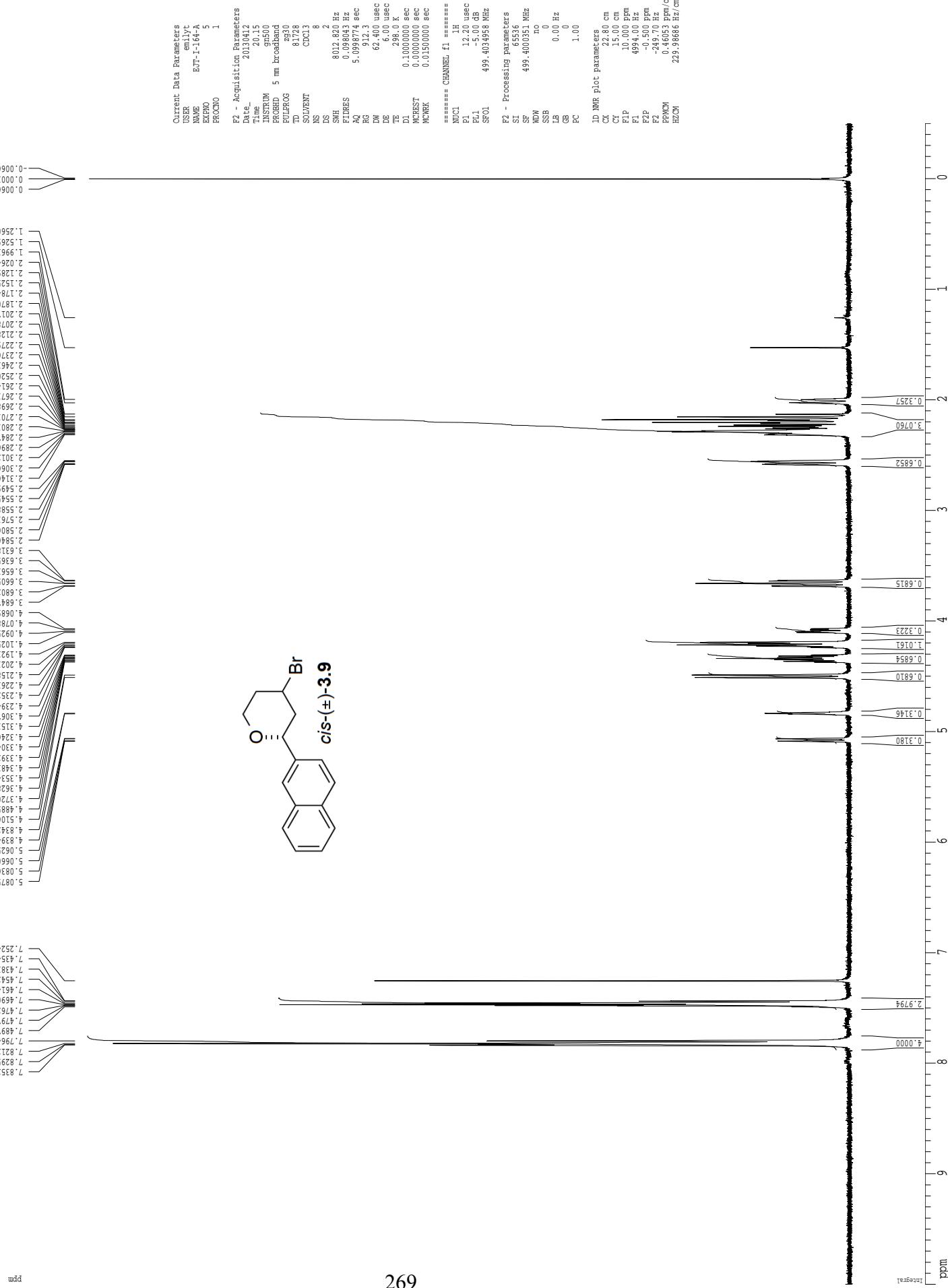
===== GRADIENT CHANNEL =====
CPDPRG2 : SINE,100
GP001   : SINE,100
GP002   : SINE,100
GP1     : 0.00 %
GP2     : 0.00 %
GPV1    : 0.00 %
GPV2    : 0.00 %
GPZ1    : 30.00 %
GPZ2    : 50.00 %
GP22    : 50.00 %
P15     : 500.00 usec
P16     : 1000.00 usec

F2 - Processing Parameters
SI        : 65536
SF        : 125.7804117 MHz
WDW     : EM
SSB      : 0
LB        : 1.00 Hz
GB        : 0
PC        : 2.00

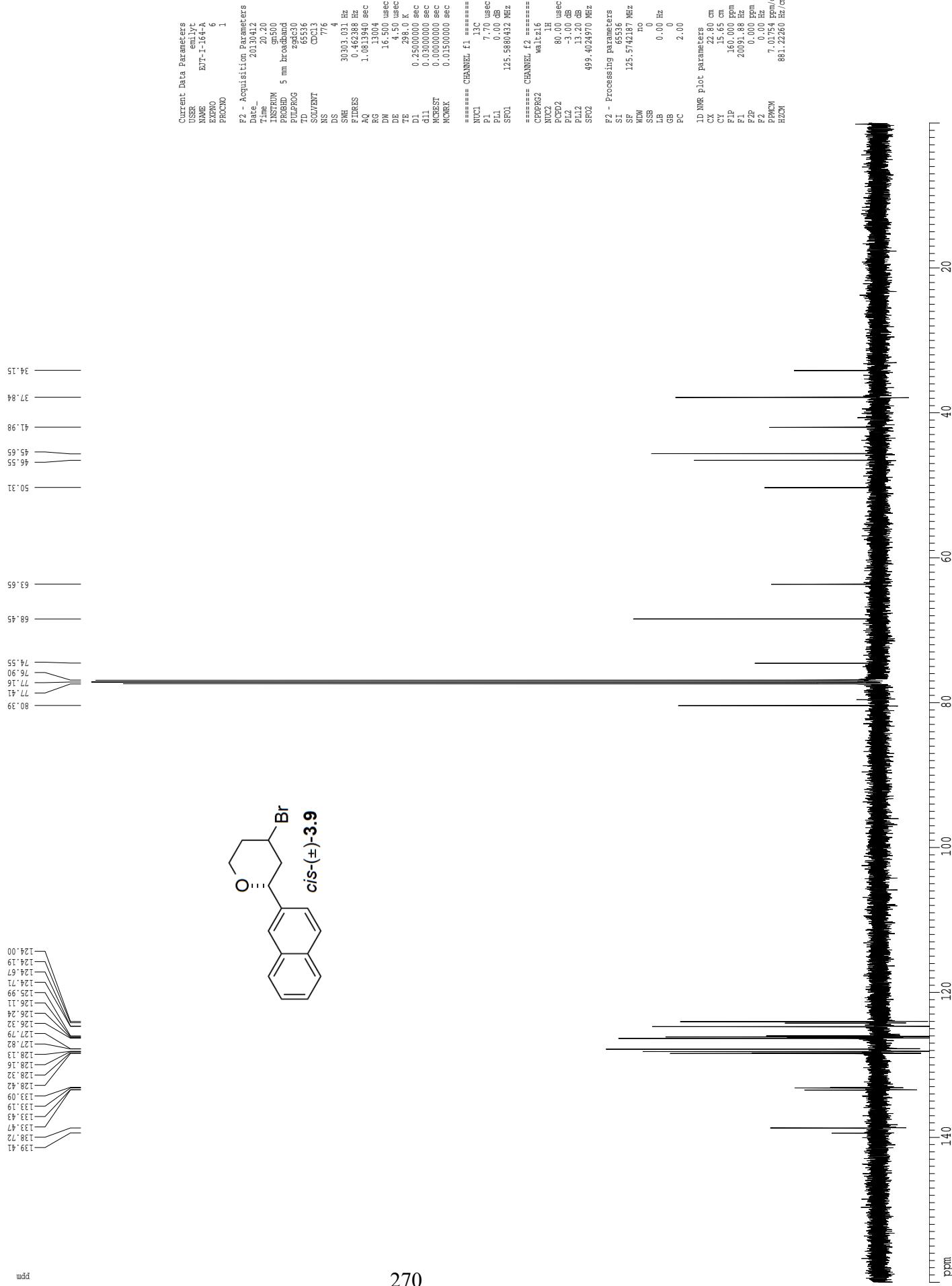
1D NMR Plot Parameters
CX        : 22.80 cm
CY        : 15.65 cm
F1P      : 230.677 ppm
F1       : 230.9.68 Hz
F2P      : -10.287 ppm
F2       : -10.287 Hz
PPCM    : 10.5668 ppm/cm
HZCM   : 1323.1063 Hz/cm

```

1H spectrum

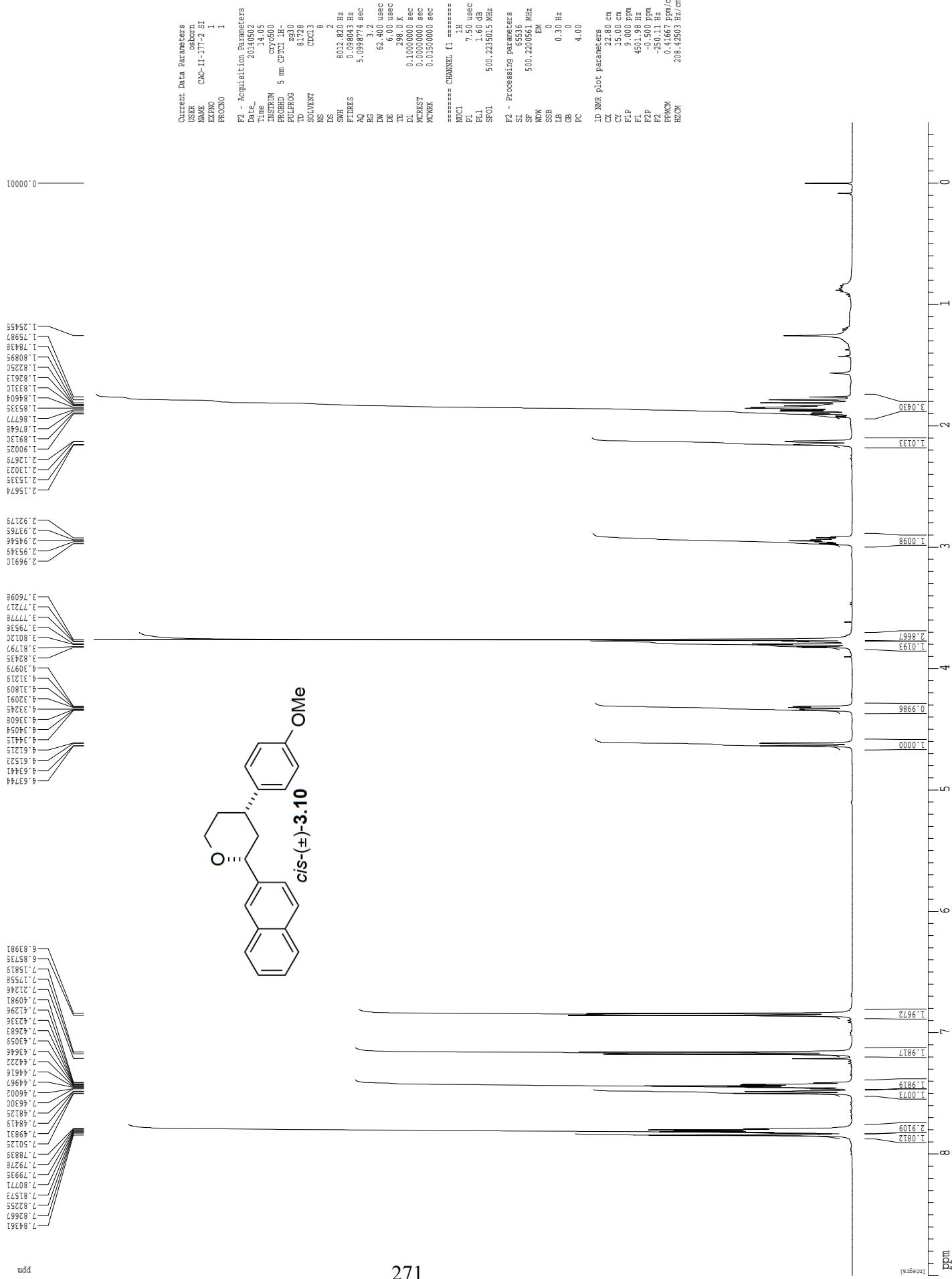


<sup>13</sup>C spectrum with 1H decoupling



<sup>1</sup>H spectrum

0.00001



Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

ppm

158.21  
140.38  
137.75  
133.48  
130.38  
80.09  
77.14  
77.16  
76.91  
68.90  
55.36  
41.83  
41.41  
33.79

Current Data Parameters  
 USER      usborn  
 NAME      Cb-11-177-2 SI  
 EXNO      1  
 PRGNO      1  
 F2 - Acquisition Parameters  
 Date      20140512  
 Time      14:08  
 INSTRUM      cry500  
 PROBHD      5 mm CCP1 1H-  
 PULPROG      SpinEchoes3Dpp.prd  
 TD      65536  
 SOLVENT      CHCl3  
 T1      101  
 NS      16  
 DS      16  
 SWH      3033.021 Hz  
 ETDRS      0.063388 Hz  
 AQ      1.00390 sec  
 TS      128.2  
 DR      16 50.0 usec  
 DE      3.00 1usec  
 TE      296.0 K  
 0.260000 sec  
 D11      0.0000000 sec  
 D16      0.0020000 sec  
 D17      0.0019600 sec  
 NCEST      0.0000000 sec  
 NCNRK      0.0150000 sec  
 F2      31.00 usec

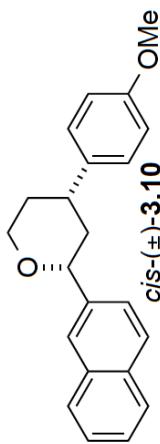
===== CHANNEL F1 =====  
 NUCL       $^{13}\text{C}$   
 P1      15.30 usec  
 P11      50.00 usec  
 P12      200.00 usec  
 PL0      -120.00 dB  
 PLL      125.794258 MHz  
 SP01      SP1 3.20 dB  
 SP1      3.20 dB  
 SPW01      Crp60.0,5,20.1  
 SPW02      Crp60.0,5,20.1  
 SPFF1      0.00 Hz  
 SPFF2      0.00 Hz

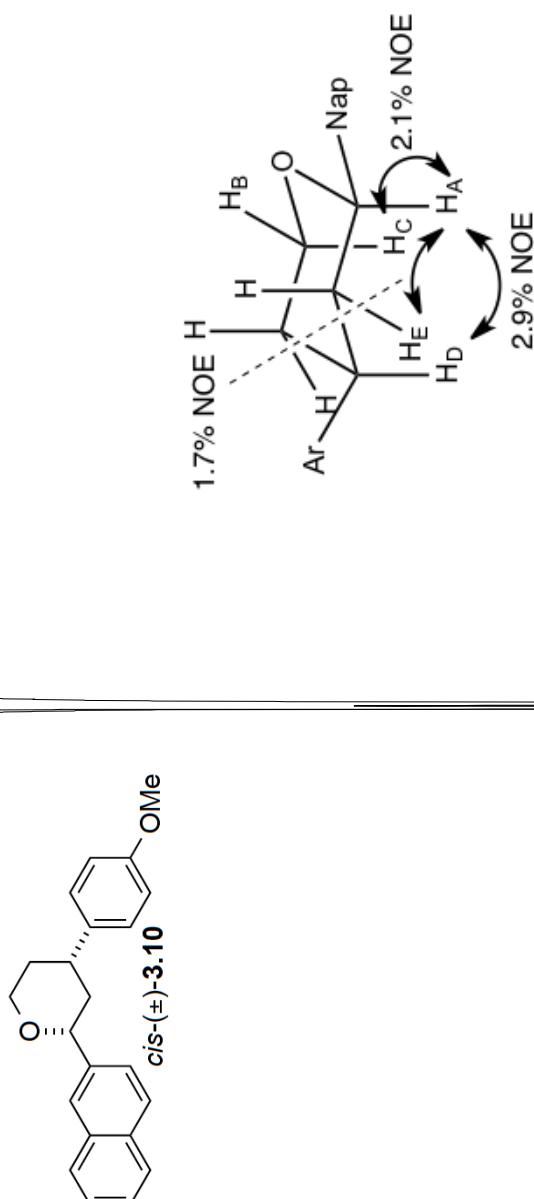
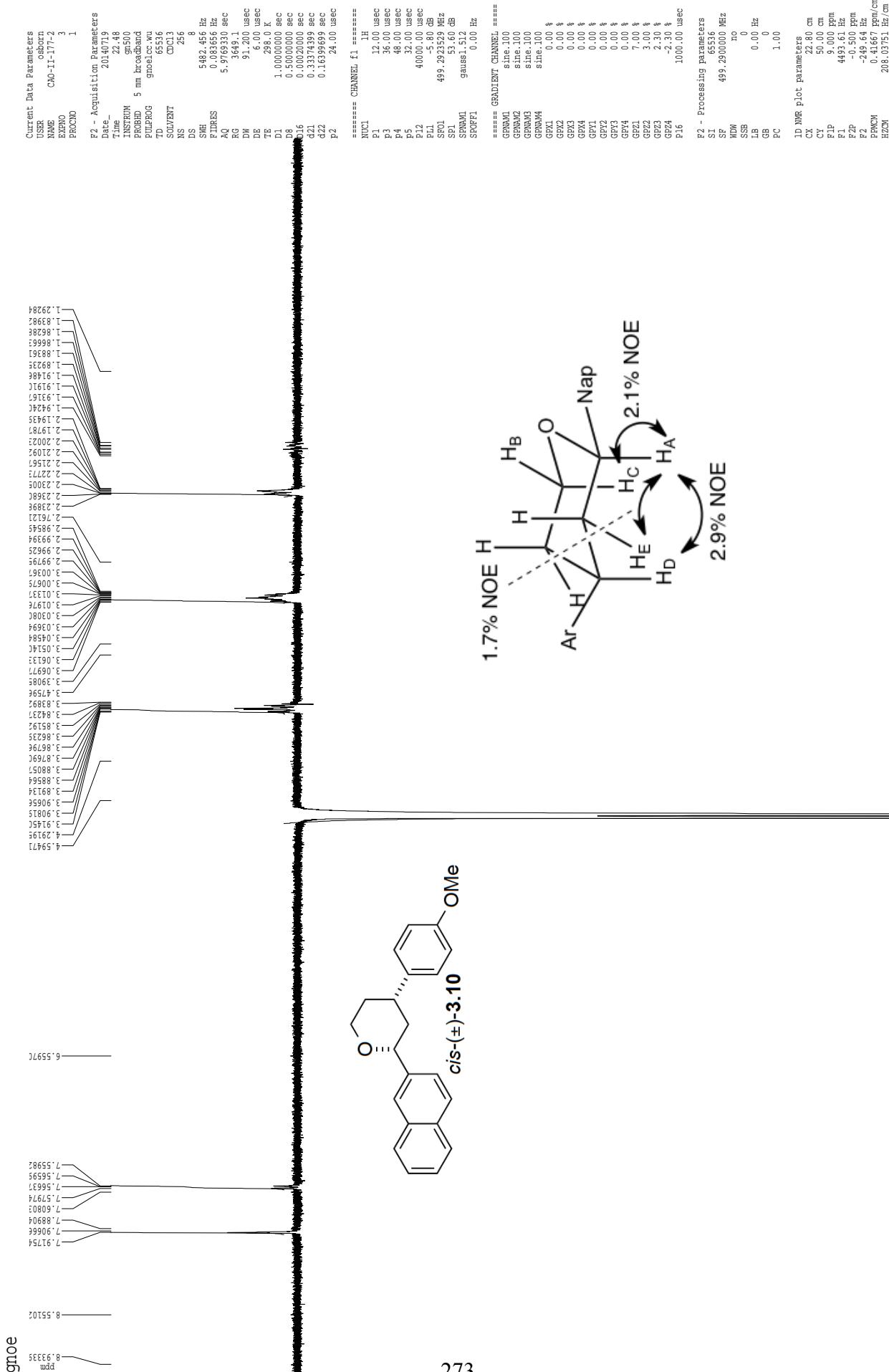
===== CHANNEL F2 =====  
 CPDPRG2  
 NUC2       $^1\text{H}$   
 PCD2      100.00 usec  
 PL2      1.60 dB  
 PLL2      500.2225011 MHz

===== GRADIENT CHANNEL =====  
 GPRM01      SINE,100  
 GPRM02      SINE,100  
 GPX1      0.00 %  
 GPX2      0.00 %  
 GPY1      0.00 %  
 GPY2      0.00 %  
 GPZ1      30.00 %  
 GPZ2      50.00 %  
 GP15      500.00 usec  
 GP16      1000.00 usec

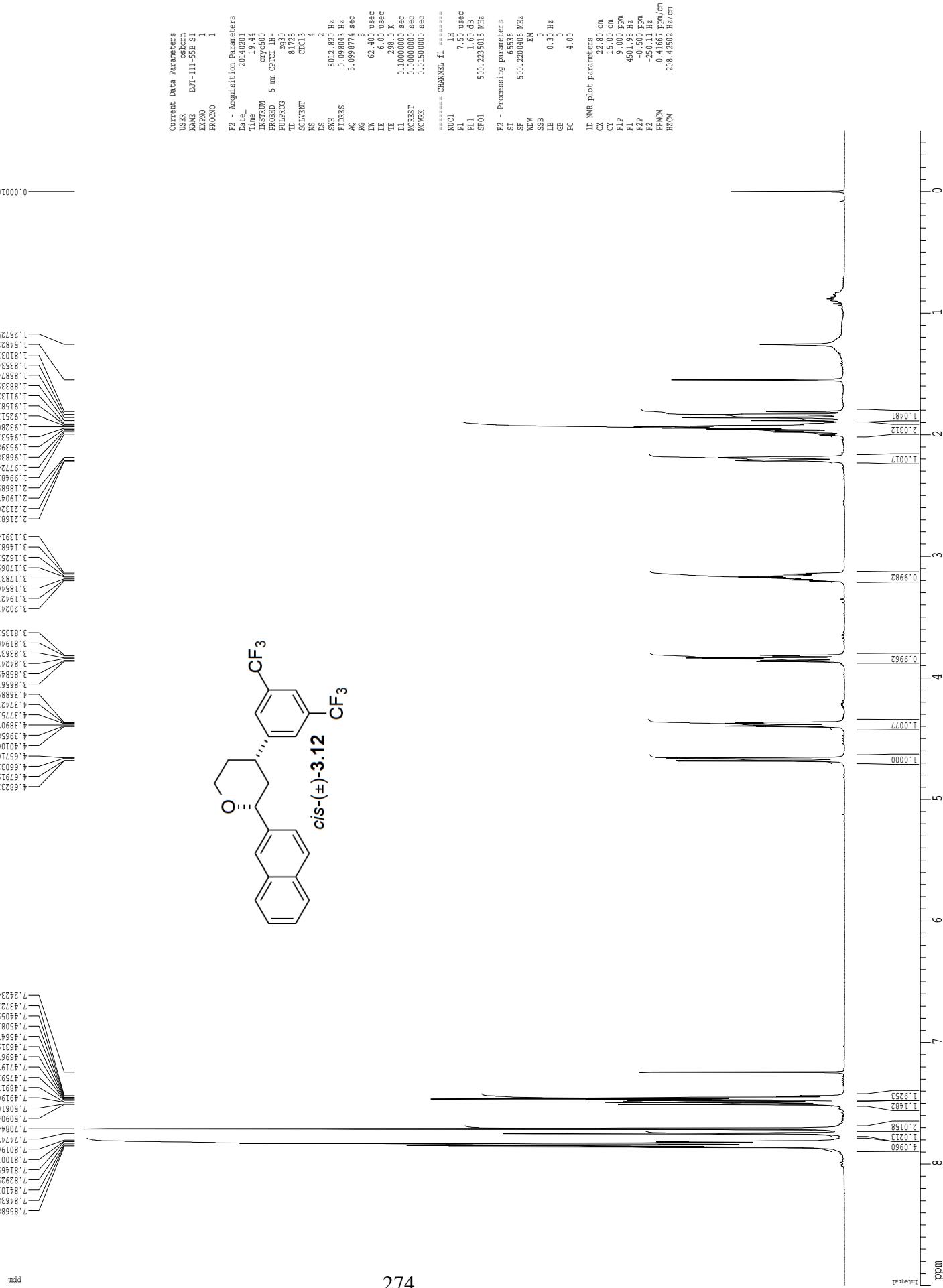
F2 - Processing Parameters  
 SI      65536  
 SF      125.7804132 MHz  
 SW      EM  
 SSB      0  
 LB      1.00 Hz  
 GB      0  
 PC      2.00

1D NMR Plot Parameters  
 CX      22.80 cm  
 CY      15.65 cm  
 F1P      230.67 ppm  
 F1      230.968 Hz  
 F2P      -10.287 ppm  
 F2      -129.346 Hz  
 PPMW      10.5668 ppm/cm  
 HZWM      1323.10706 Hz/cm





1H spectrum



Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

ppm

68.44  
76.90  
77.16  
77.14  
79.82  
120.67  
120.70  
120.73  
122.44  
124.08  
124.59  
126.01  
126.22  
126.76  
127.01  
127.08  
128.11  
128.35  
131.65  
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132.33  
133.12  
133.47  
137.84  
41.24  
42.08  
33.11  
29.85

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Current Data Parameters
USER   : esborn
NAME   : EIT11-55B.S1
PROCNO: 1

F2 - Acquisition Parameters
Date   : 2014/02/01
Time   : 19.46
INSTRUM: INSPINW
PROBHD: 5 mm PCD1 1H-
PULPROG: SpinEchoes30DP.prd
TD    : 65536
SOLVENT: C6C13
NS    : 91
DS    : 16
SWH   : 3003.021 Hz
ETRIM  : 0.063388 Hz
TE    : 0.01390 sec
AQ    : 1.00000 sec
RG    : 128.2
TM    : 16.500 usec
D1    : 0.260000 sec
T1    : 0.000000 sec
D11   : 0.000000 sec
D16   : 0.002000 sec
D17   : 0.001960 sec
MC    : 1
N1    : 0.000000 sec
MCNSK: 0.015000 sec
P2    : 31.00 usec

=====
CHANNEL F1 =====
NUCL1 : 13C
P1    : 15.50 usec
P11   : 50.00 usec
P12   : 200.00 usec
PL0   : -12.00 dB
PL1   : 125.794258 MHz
SP01  : SP1
SP1   : 3.20 dB
SP001 : Crp60.0,5,20.1
SP002 : Crp60.0,5,20.1
SP0FF1: SpcFF2
SPCFF2: 0.00 Hz

=====
CHANNEL F2 =====
NUCL2 : 1H
CPDPG2: SINE16
P2P1  : 10.00 usec
P2L2  : 1.60 dB
PL12  : 500.2225011 MHz
SF02  : 500.2225011 MHz

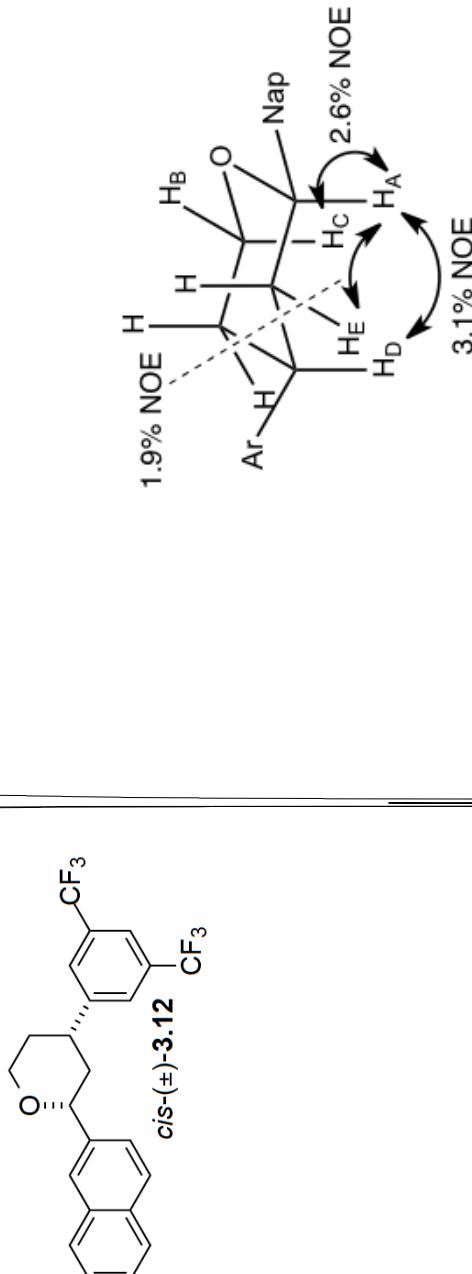
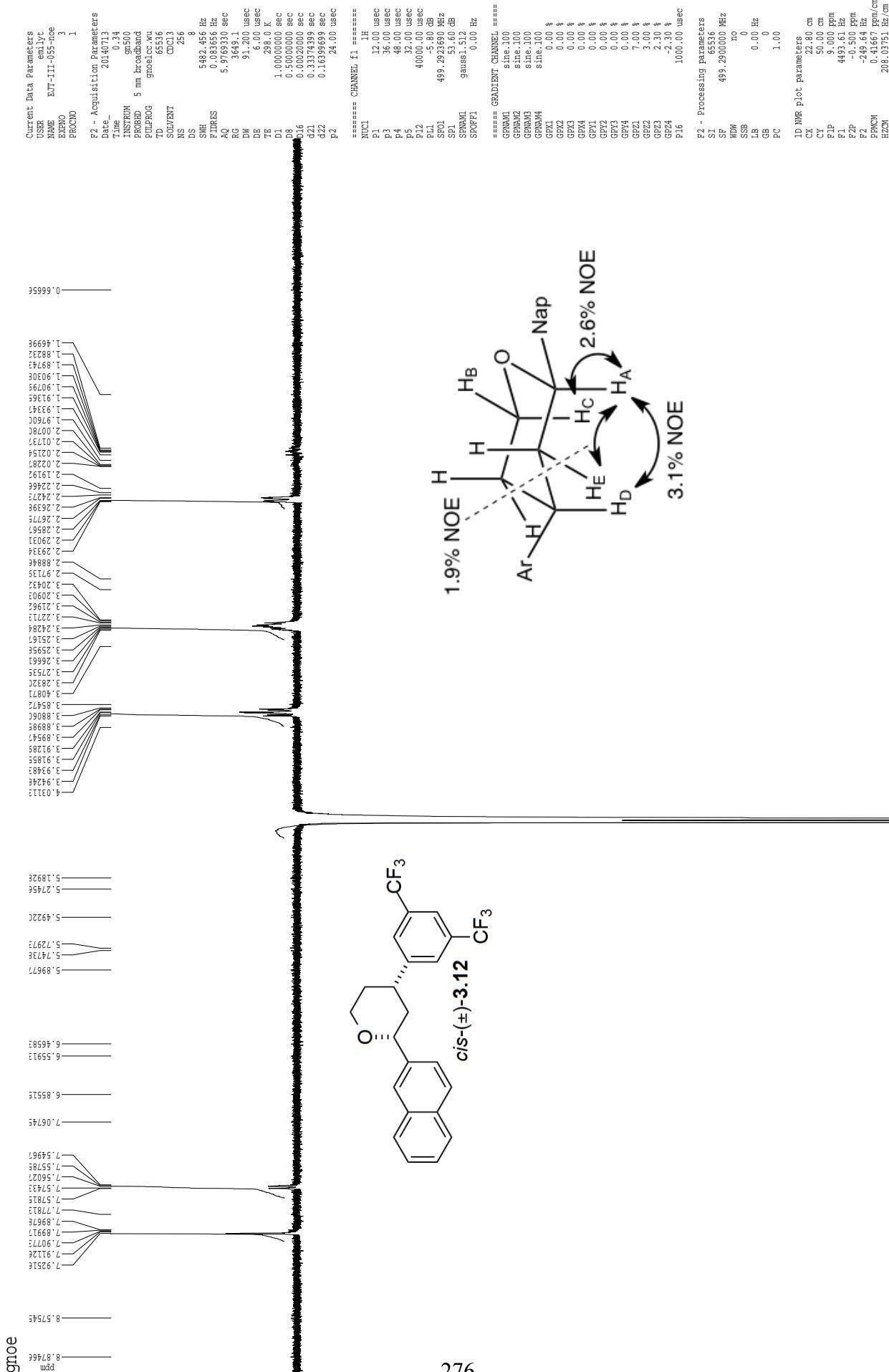
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GRADIENT CHANNEL =====
GP0M1 : SINE100
GP0M2 : SINE100
GPX1  : 0.00 %
GPX2  : 0.00 %
GPY1  : 0.00 %
GPY2  : 0.00 %
GPZ1  : 30.00 %
GPZ2  : 50.00 %
P15   : 50.00 usec
P16   : 1000.00 usec

F2 - Processing Parameters
SI    : 65536
SF    : 125.780413 MHz
WDW   : EM
SSB   : 0
LB    : 1.00 Hz
GB    : 0
PC    : 2.00

1D NMR Plot Parameters
CX    : 22.80 cm
CY    : 15.65 cm
F1P   : 230.67 ppm
F1    : 230.968 Hz
F2P   : -10.287 ppm
F2    : -129.316 Hz
PPCM  : 10.5668 ppm/cm
HZDW : 1323.1063 Hz/cm

```

0  
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100  
150  
200  
250  
ppm



1H spectrum

ppm

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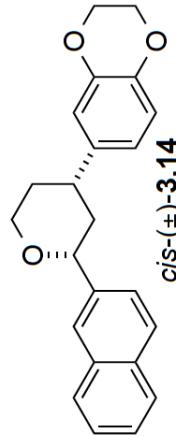
Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

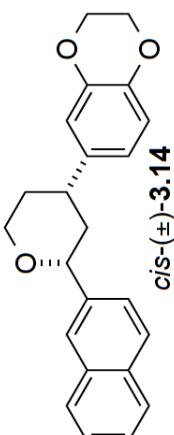
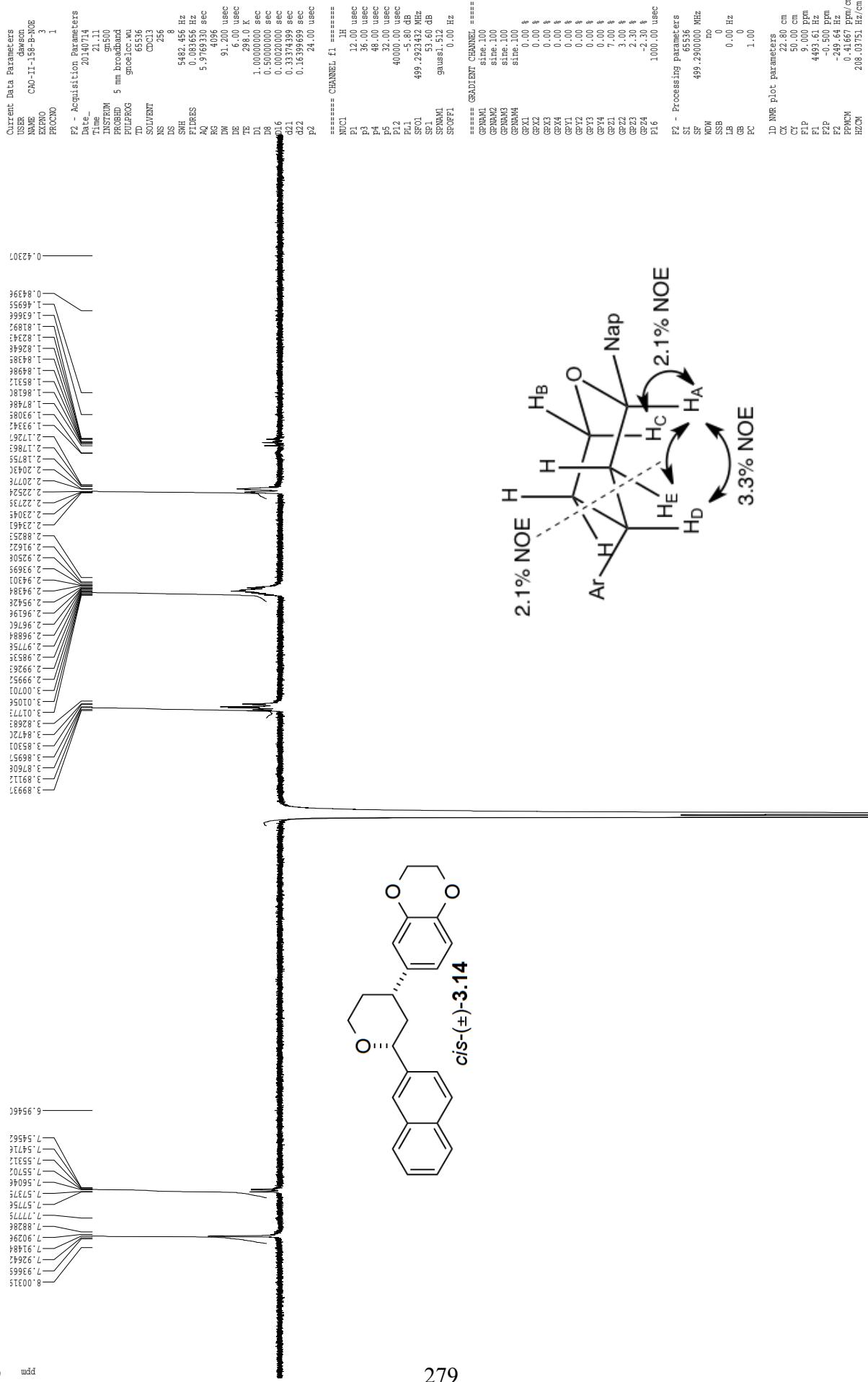
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 Current Data Parameters  
 USER      usborn  
 NAME     Cb-11-158A.S1  
 EXNO      1  
 PRGNO    1  
 F2 - Acquisition Parameters  
 Date    2014/02/05  
 Time    18:16  
 INSTRUM  cry500  
 PROBHD  5 mm CCP1 1H-  
 PULPROG SpinEchoes30P-prd  
 TD       65536  
 SOLVENT   NS  
 PCP1    128  
 T2      10.00 usec  
 TE      296.0 K  
 SRF    30303.021 Hz  
 FIDRES 0.061338 Hz  
 AQ      1.00 sec  
 TS      128.2  
 DR      16.50 usec  
 DE      3.00 usec  
 T90    0.260000 sec  
 D11    0.000000 sec  
 D16    0.002000 sec  
 D17    0.001960 sec  
 NCEST 0.000000 sec  
 MCNSK 0.015000 sec  
 F2      31.00 usec  
 SPCFF2 0.00 Hz  
 ======  
 ===== CHANNEL F1 ======  
 NUC1      13C  
 P1       15.30 usec  
 P11      50.00 usec  
 P12      200.00 usec  
 PL0      -12.00 dB  
 PLL      125.794258 MHz  
 SP01     3.20 dB  
 SP1      3.20 dB  
 SPW01 Crp60/0.5,20.1  
 SPW02 Crp60/0.5,20.1  
 Crp60/0.5,20.1  
 SPCFF1 0.00 Hz  
 SPCFF2 0.00 Hz  
 ===== CHANNEL F2 ======  
 CPDPRG2 1H  
 NUC2      13C  
 PCD2 100.00 usec  
 PL2      1.60 dB  
 PLL2 500.2225011 MHz  
 SF02  
 ===== GRADIENT CHANNEL ======  
 GPRM01 SINE,100  
 GPRM02 SINE,100  
 GPX1 0.00 %  
 GPX2 0.00 %  
 GPY1 0.00 %  
 GPY2 0.00 %  
 GPZ1 30.00 %  
 GPZ2 50.00 %  
 P15 500.00 usec  
 P16 1000.00 usec  
 ===== Processing Parameters ======  
 SI       65536  
 SF      125.780412 MHz  
 WDW      EM  
 SSB      0  
 LB      1.00 Hz  
 GB      0  
 PC      2.00  
 1D NMR Plot parameters  
 CX      22.80 cm  
 CY      15.65 cm  
 F1P    230.67 ppm  
 F1    230.9.68 Hz  
 F2P   -10.287 ppm  
 F2   -129.3.16 Hz  
 PPMW 10.5668 ppm/cm  
 HZWM 1323.1063 Hz/cm  
 =====

33.66  
 41.54  
 41.69  
 41.74

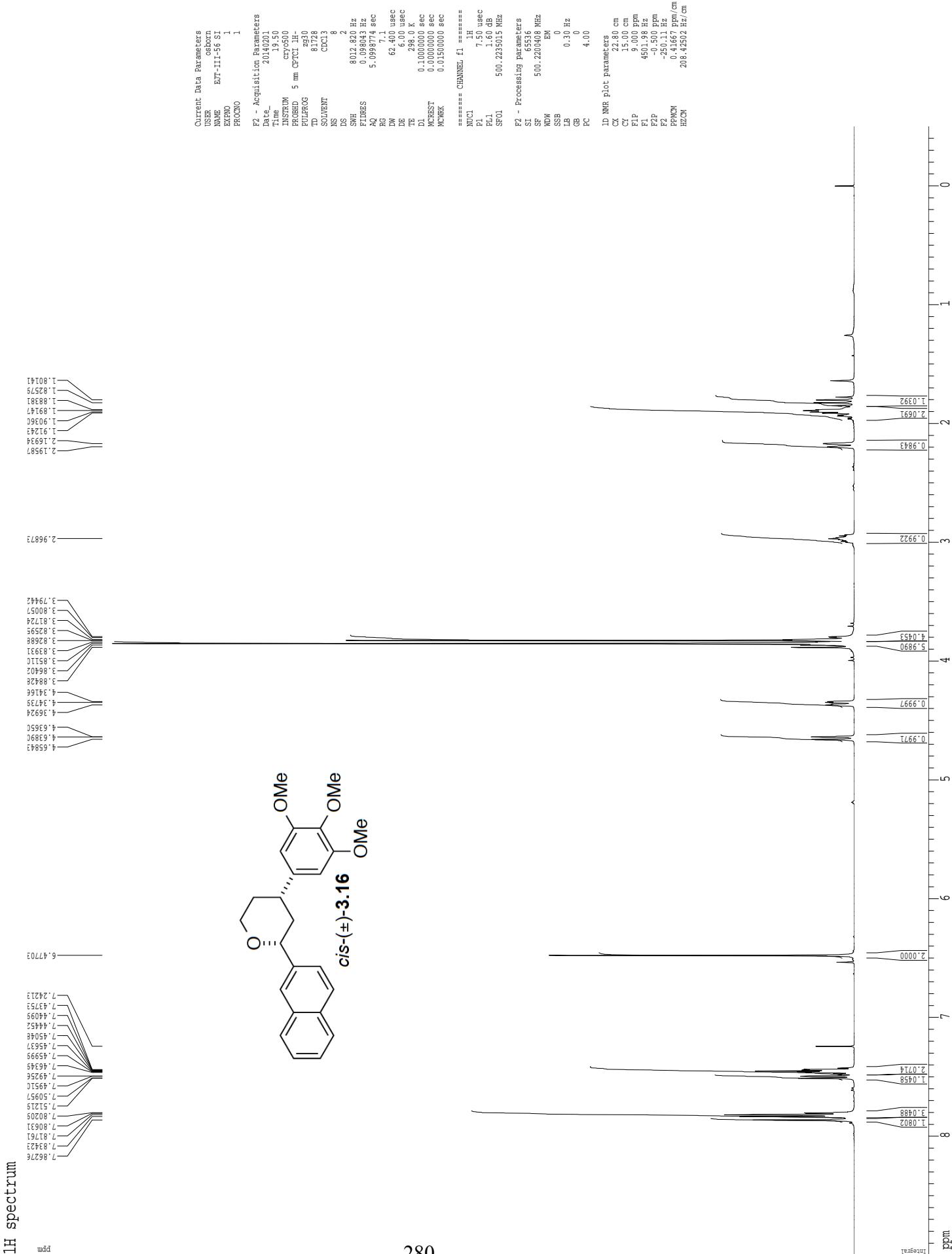
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 76.91  
 77.16  
 77.44  
 80.08

115.55  
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 124.36  
 124.45  
 125.88  
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 142.10  
 143.55





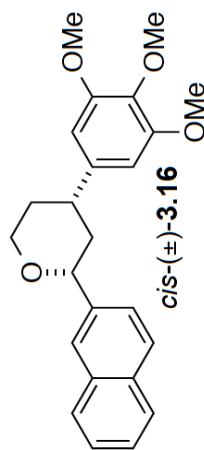
279



Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

ppm

153.35  
141.43  
140.23  
136.53  
133.44  
132.96  
132.15  
128.07  
127.14  
125.83  
124.29  
124.20  
103.73  
104.21  
68.79  
60.94  
56.26  
56.18  
41.77  
41.76  
33.71



```

Current Data Parameters
USER          : esborn
NAME         : E7-111-56 SI
EXNO        : 1
PRCENO      : 1
F2CNO       : 1

F2 - Acquisition Parameters
Date        : 2014/02/01
Time        : 19:52
INSTRUM    : cryo500
PROBHD   : 5 mm CCP1 1H-
PULPROG  : SpinEchoes30GP.prd
TD        : 65536
SOLVENT   : CCl4
NUC1      : 13C
DS        : 16
SWH      : 3003.021 Hz
ETRIM     : 0.46338 Hz
TE        : 1.00 sec
AQ        : 2100.6
RG        : 16.500 usec
DW        : 4.00 usec
DSSW     : 296.0 K
D1        : 0.260000 sec
D11       : 0.000000 sec
D16       : 0.002000 sec
D17       : 0.001960 sec
MC1      : 0.000000 sec
MC2      : 0.000000 sec
MC3      : 0.000000 sec
P2K1     : 31.00 usec
P2K2     : 31.00 usec

===== CHANNEL F1 =====
NUC1      : 13C
P1        : 15.50 usec
P11       : 50.00 usec
P12       : 200.00 usec
PL0       : 120.00 dB
PL1       : -1.00 dB
SP01     : 125.794258 MHz
SP1       : 3.20 dB
SP2       : 3.20 dB
SPRWA1  : Crp60.0,5,20.1
SPRWA2  : Crp60.0,5,20.1
CPDPRC4 : Cpd60.0,5,20.1
SPCFF1  : 0.00 Hz
SPCFF2  : 0.00 Hz

===== CHANNEL F2 =====
NUC2      : 1H
CPDPG2  : SINE,1.00
PCPD2   : 10.00 usec
PL2       : 1.60 dB
PL12      : 24.60 dB
SF02     : 500.2225011 MHz

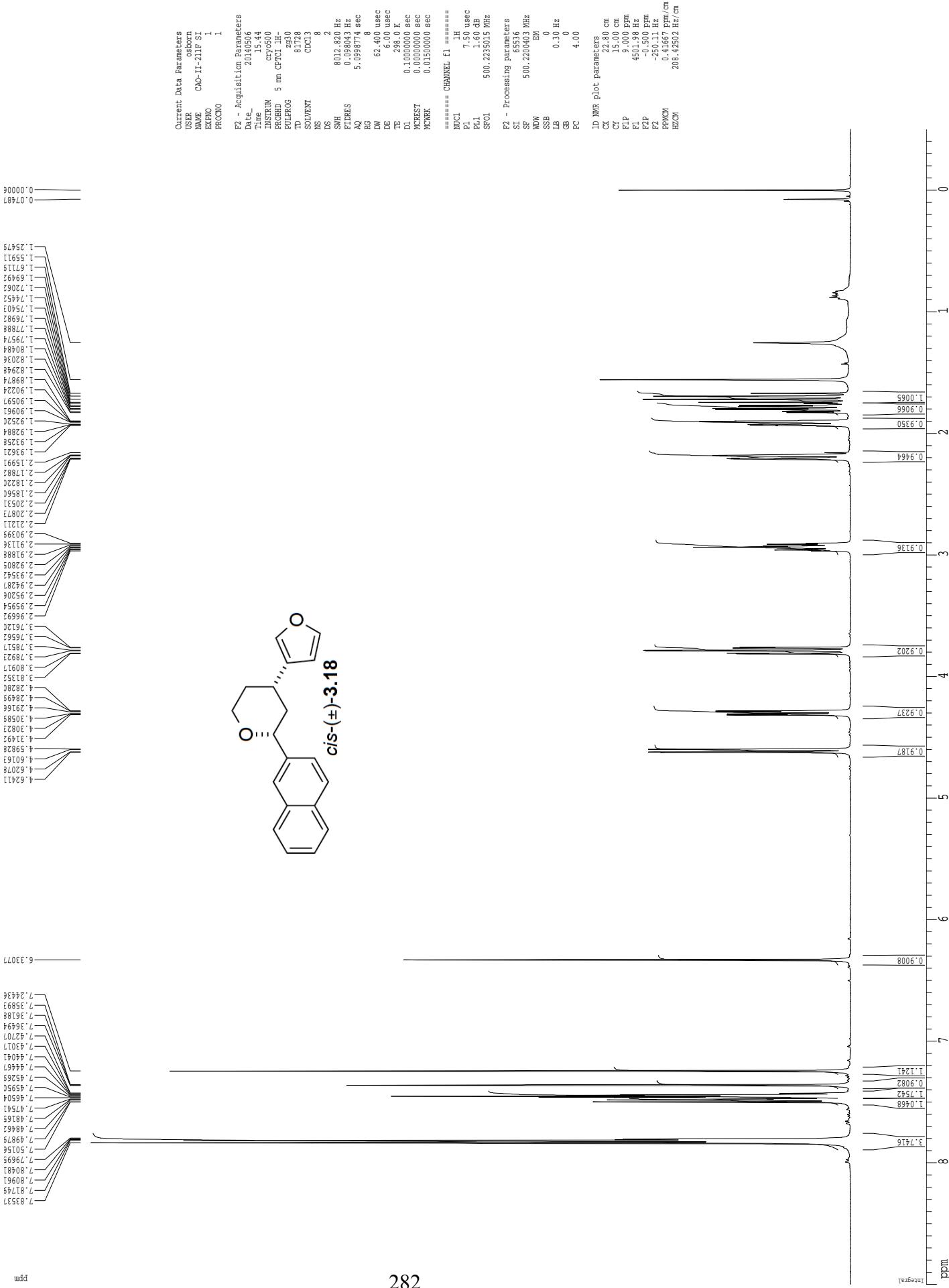
===== GRADIENT CHANNEL =====
CPDPG1  : SINE,1.00
GPRAW1  : 65536
GPRAW2  : 65536
GPX1    : 0.00 %
GPX2    : 0.00 %
GPY1    : 0.00 %
GPY2    : 0.00 %
GPZ1    : 30.00 %
GPZ2    : 50.00 %
GP22    : 50.00 %
P15     : 500.00 usec
P16     : 1000.00 usec

F2 - Processing Parameters
SI        : 65536
SF        : 125.7804173 MHz
WDW      : EM
SSB      : 0
LB        : 1.00 Hz
GB        : 0
PC        : 2.00

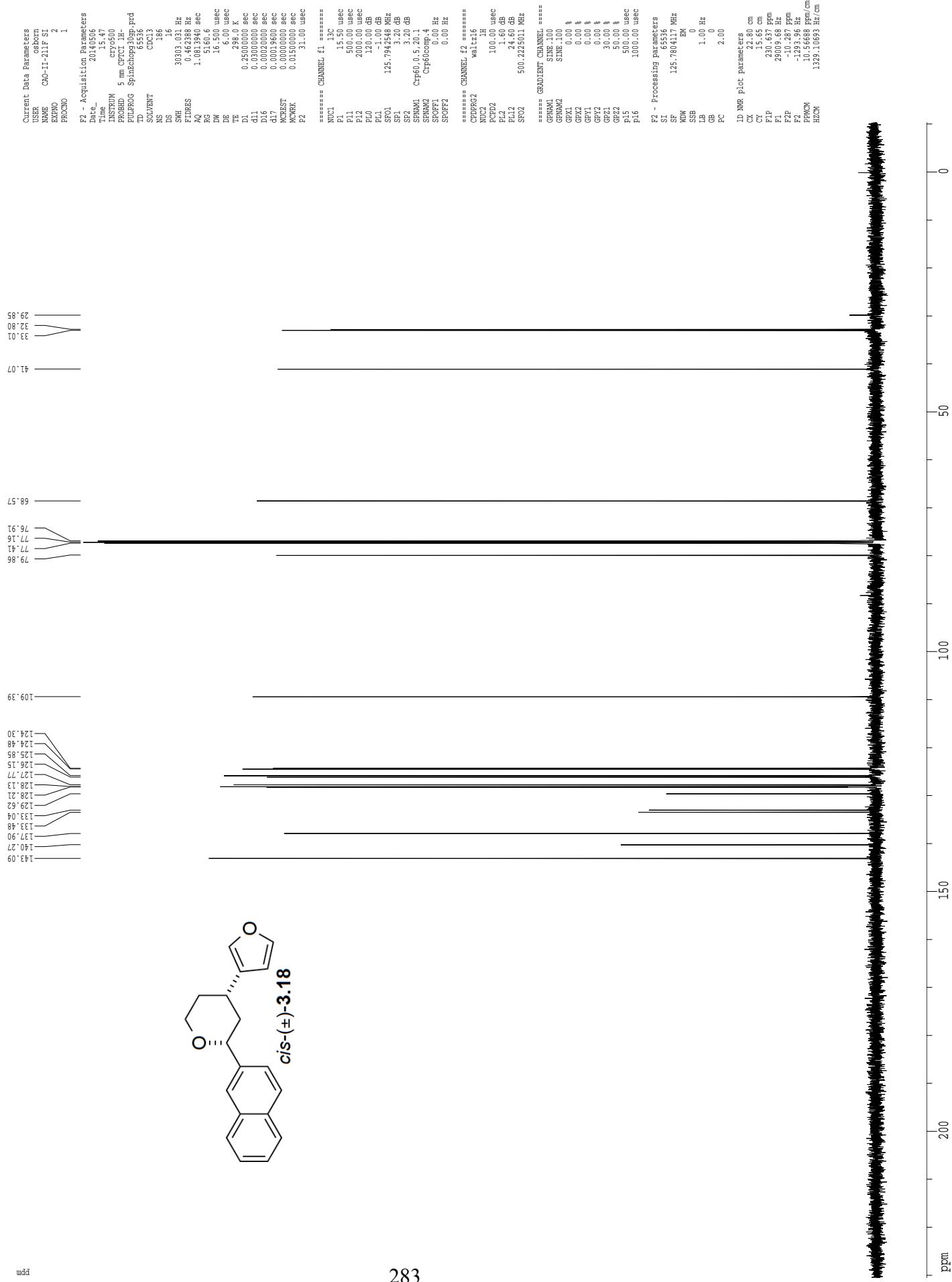
1D NMR Plot parameters
CX        : 22.80 cm
CY        : 15.65 cm
F1P      : 230.67 ppm
F1       : 230.9.68 Hz
F2P      : -10.287 ppm
F2       : -129.3.16 Hz
PPCM    : 10.5668 ppm/cm
HZCM   : 1325.10706 Hz/cm

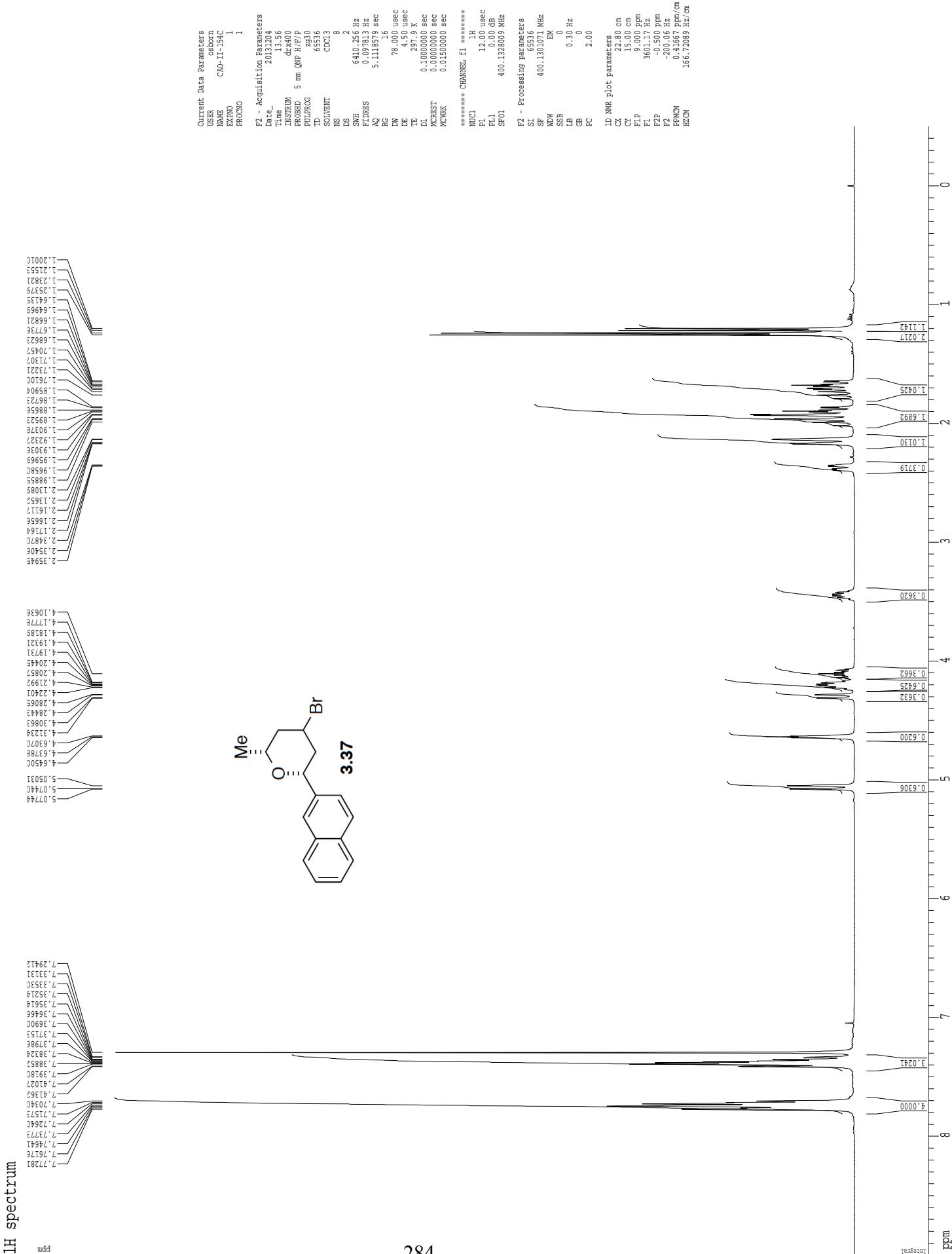
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1H spectrum



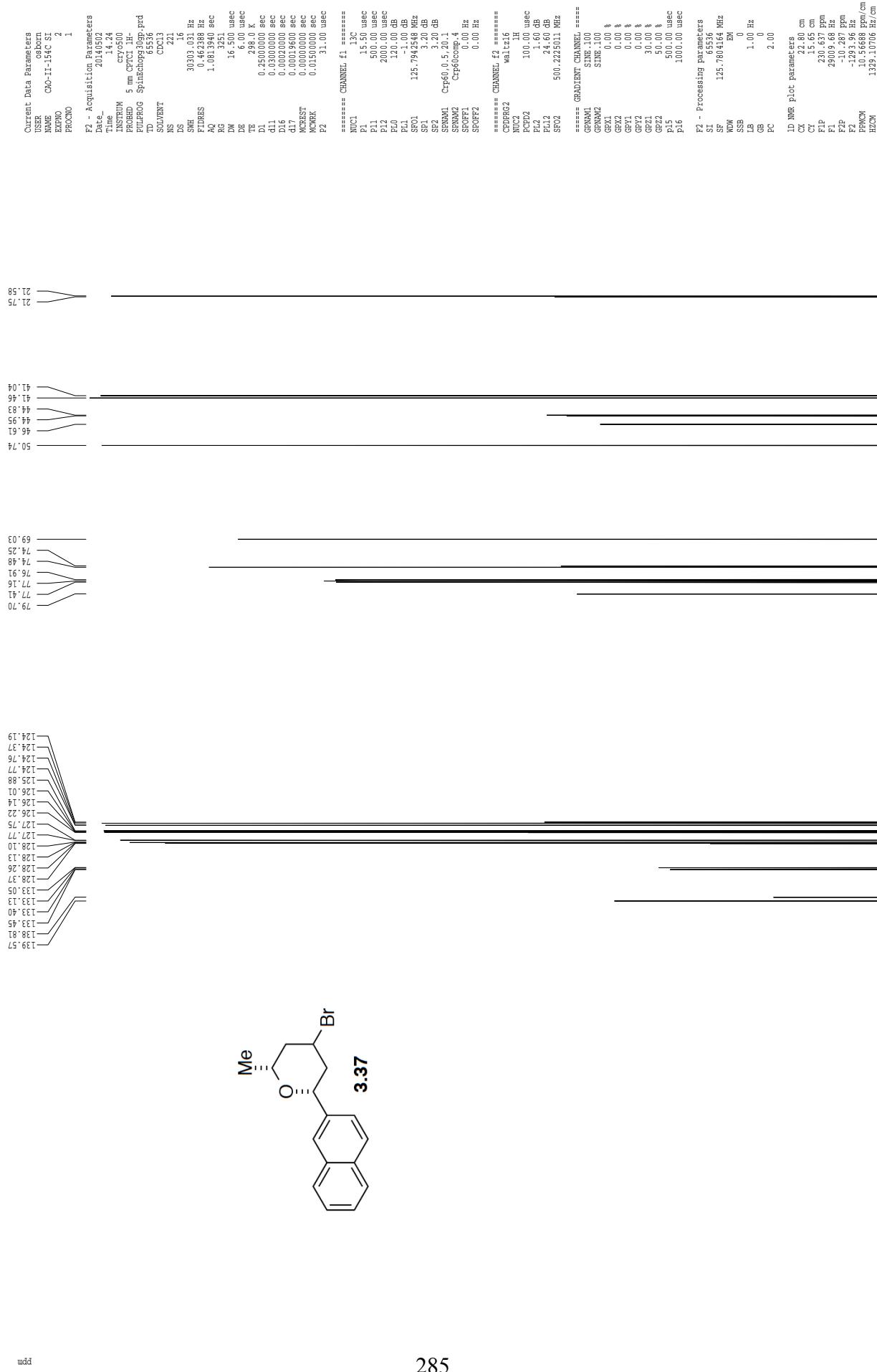
Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling



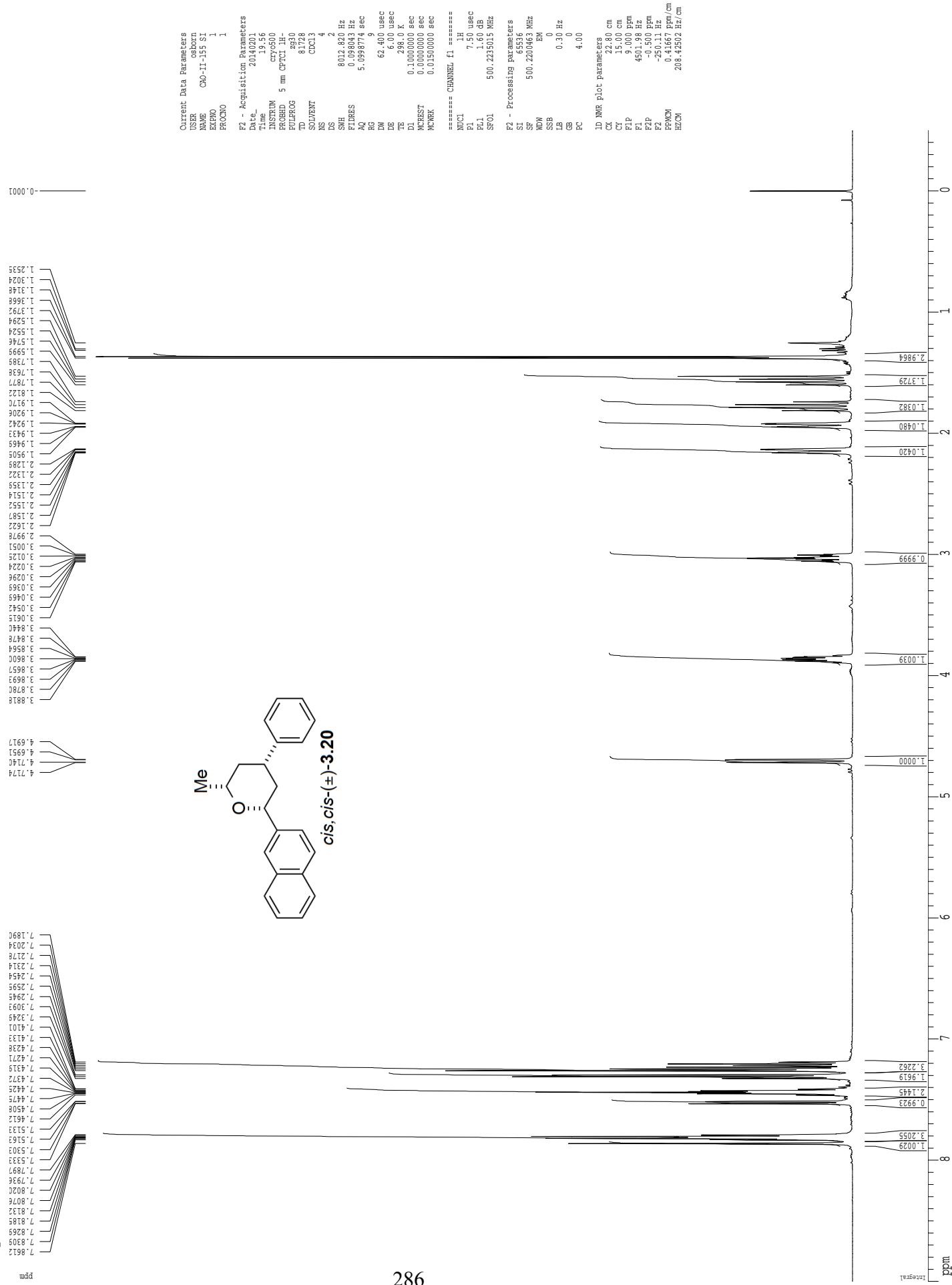


1H spectrum

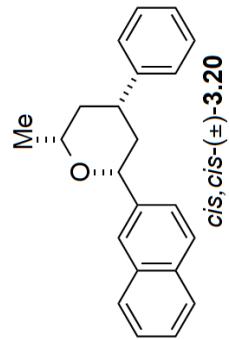
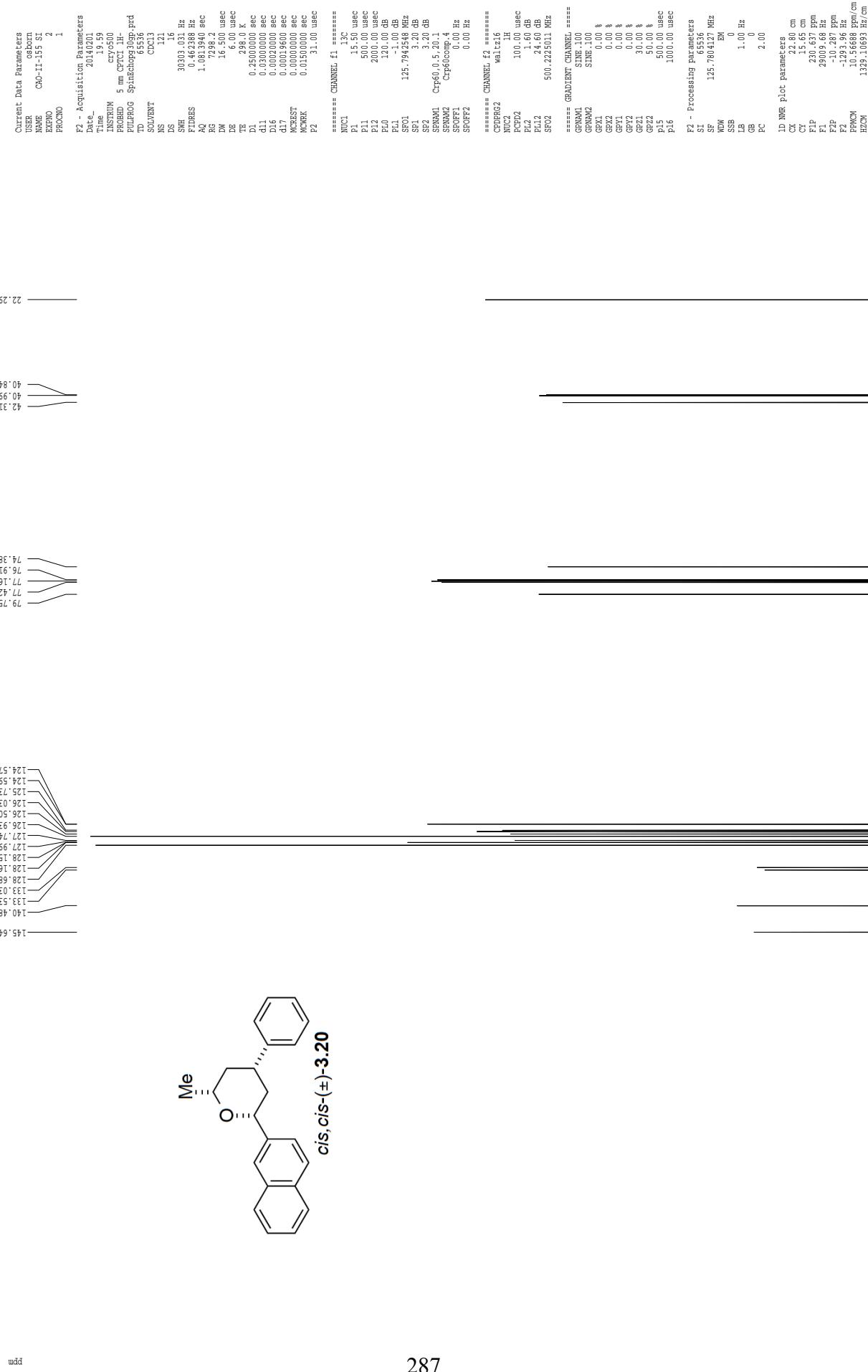
Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

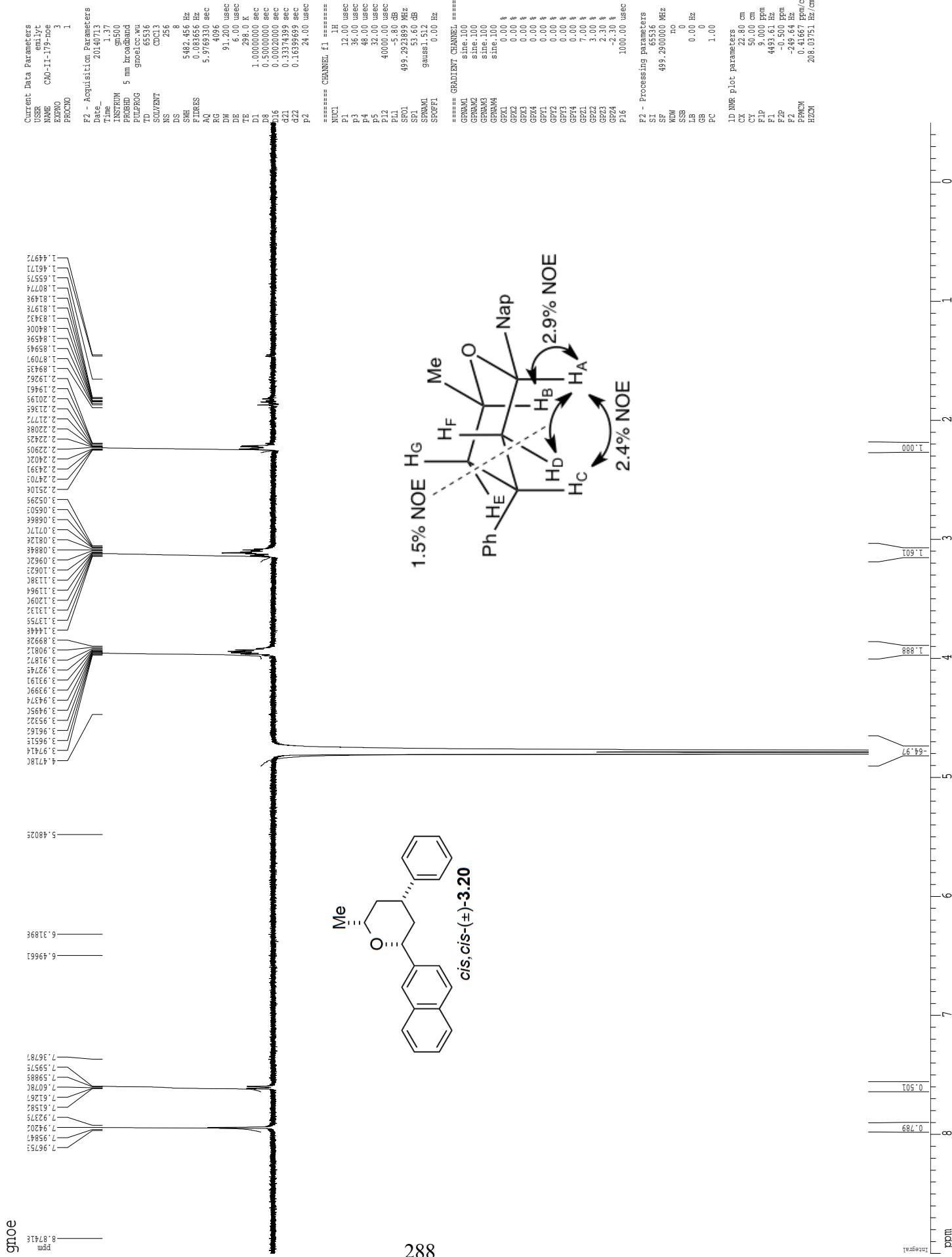


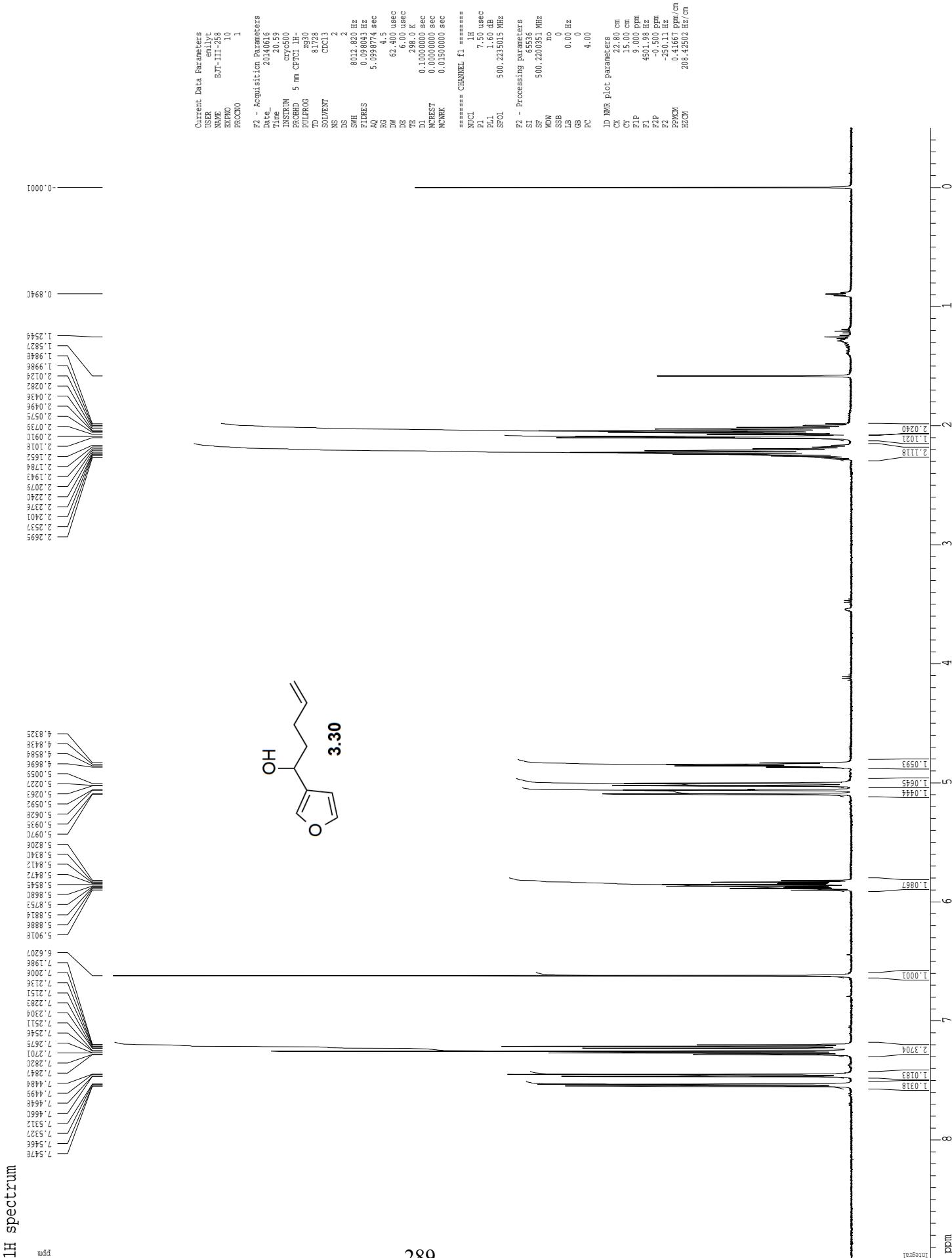
<sup>1</sup>H spectrum



Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

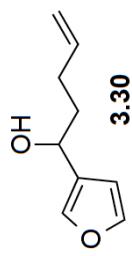






1H spectrum

Z-restored spin-echo  $^{13}\text{C}$  spectrum with LH decoupling



Current Data Parameters  
 TISER en1yt  
 NAME EIT-III-29  
 EXPNO 6  
 PROCN 1

F2 - Acquisition Parameters  
 Date\_ 2014/6/16  
 Time\_ 20:27  
 INSTRUM cryo500  
 PROBID 5 mm CPTCI 1H-  
 PULPROG Spinchages3Dpp3rd  
 TD 5536  
 SOLVENT CDCl3  
 NS 148  
 TS 16  
 SWH 3033.131 Hz  
 FIDRES 0.442388 Hz  
 AQ 1.083340 sec  
 RS 2551  
 DR 16.500 usec  
 D1 6.00 usec  
 T2 298.0 usec  
 TE 0.2500000 sec  
 D1 1.50 usec  
 P1 50.00 usec  
 P2 200.00 usec  
 PL0 120.00 dB  
 PL1 125.00 dB  
 SR01 125.7942548 MHz  
 SP1 3.20 dB  
 SP2 Cr60,1.5,20.1  
 SPRA01  
 SPRA02  
 SPCOMP4  
 SPOFF1 0.00 Hz  
 SPOFF2 0.00 Hz

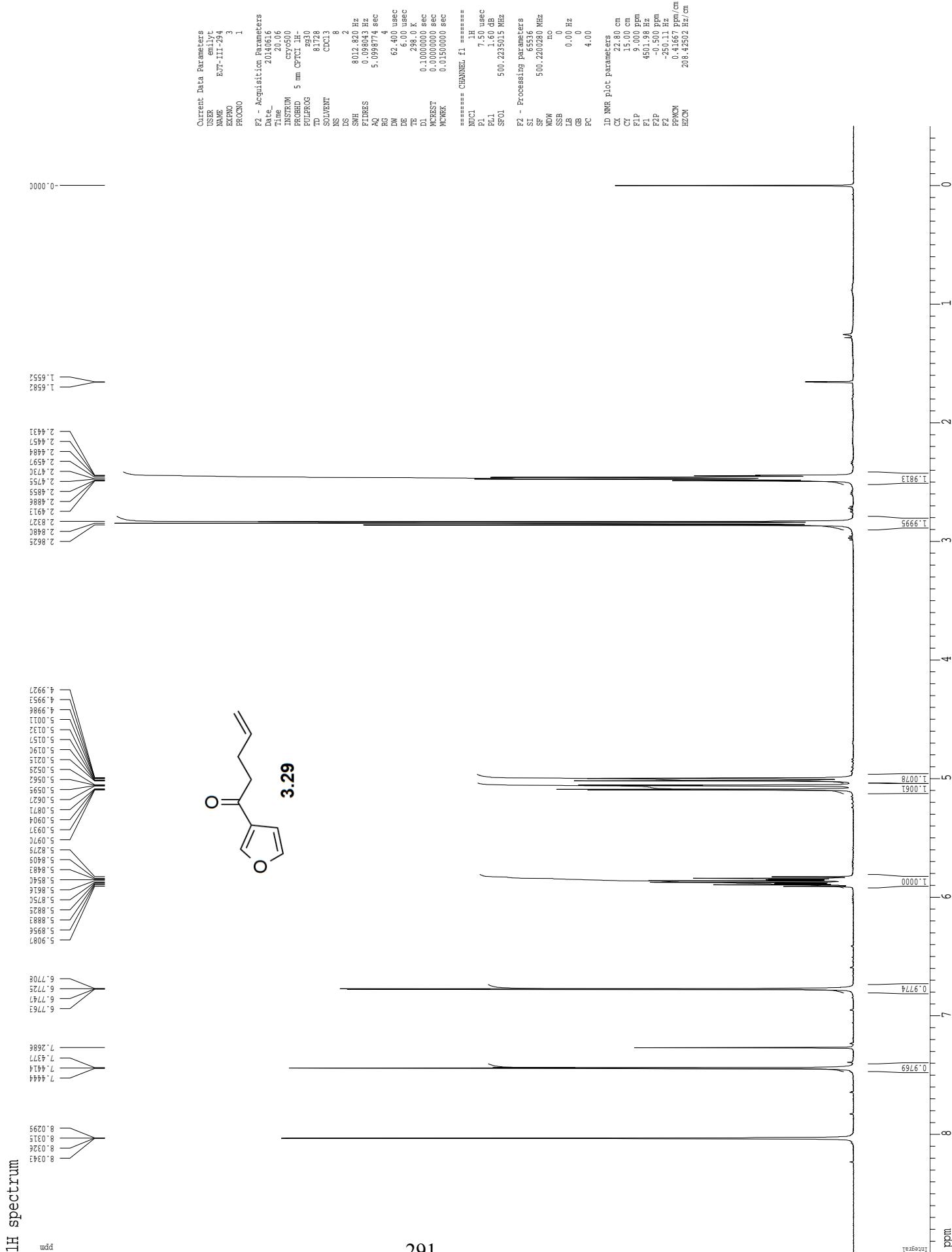
===== CHANNEL f1 ======  
 NUC1 13C  
 P1 1.50 usec  
 P2 50.00 usec  
 PL0 200.00 usec  
 PL1 120.00 dB  
 SR01 125.7942548 MHz  
 SP1 3.20 dB  
 SP2 Cr60,1.5,20.1  
 SPRA01  
 SPRA02  
 SPCOMP4  
 SPOFF1 0.00 Hz  
 SPOFF2 0.00 Hz

===== CHANNEL f2 ======  
 CDPGR2 SINE,100  
 NUC2 1H  
 GEX1 0.00 %  
 GEP2 100.00 usec  
 PL2 1.60 dB  
 PL12 24.60 dB  
 SP02 500.2232011 MHz

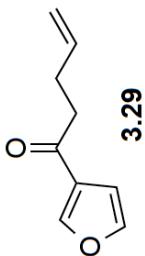
===== GRADIENT CHANNEL =====  
 GERM1 SINE,100  
 GERM2 SINE,100  
 SF 0.00 %  
 GEY1 0.00 %  
 GEY2 0.00 %  
 GEZ1 30.00 %  
 GEZ2 50.00 %  
 P15 500.00 usec  
 P16 1000.00 usec

F2 - Processing parameters  
 SI 65536  
 MDW no  
 SSB 0  
 LB 0.00 Hz  
 GB 0  
 PC 2.00

1D NMR plot parameters  
 CX 22.80 cm  
 CY 15.65 cm  
 F1P 251.6.08 Hz  
 F2P 0.000 Hz  
 F2P 8.77193 ppm/cm  
 PPMW 1103.33691 Hz/cm



Z-restored spin-echo  $^{13}\text{C}$  spectrum with LH decoupling



47.21

108.74

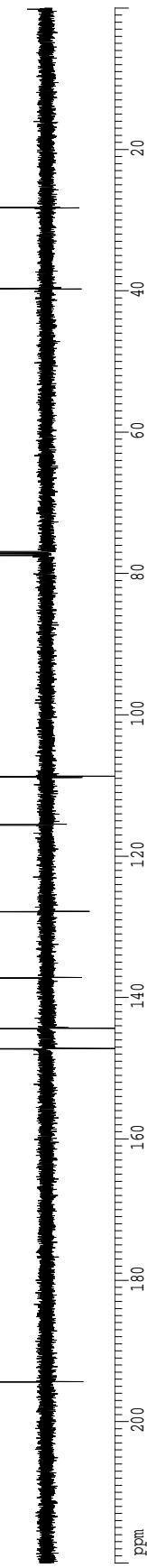
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77.16  
77.42

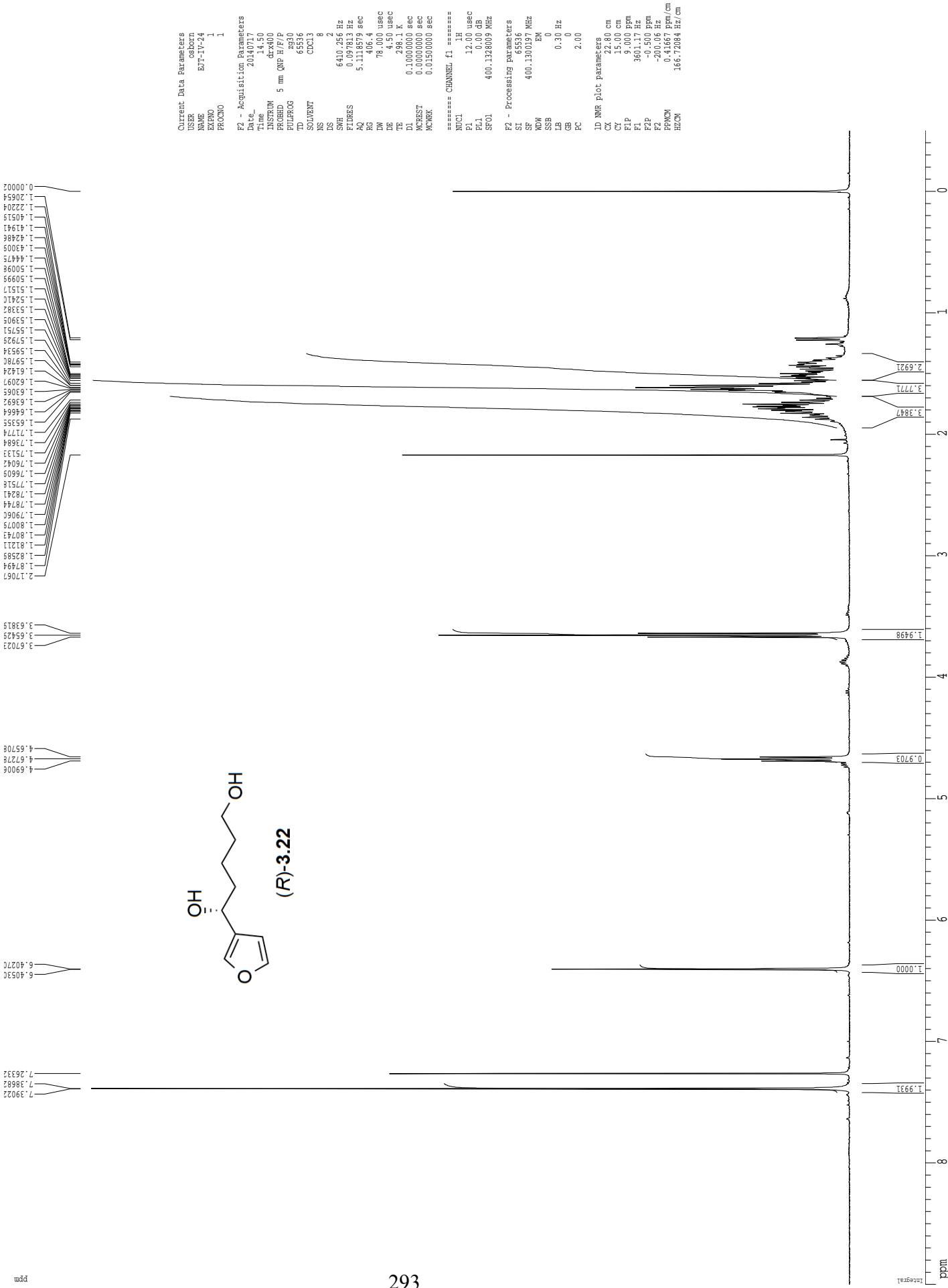
59.65 —

28.21

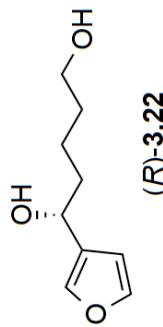
Current Data Parameters	
USER	EST-111-234
EXPERIMENT	4
FACRON	1
F2 - Acquisition Parameters	
DATE	2014/06/20
TIME	20:10
INSTRUM	5 mm CPMG T1H- T2H-FID
FIDRES	0.03946 sec
RG	516.0
TW	16.500 sec
SOLVENT	CDCl3
NS	140
DW	16.00 usec
DE	2.00 usec
TW	2.980 sec
TD	25000000 sec
TE	0.4300000 sec
TM	0.0002000 sec
TR	0.0001960 sec
TCR	0.0000000 sec
W1	0.0150000 sec
W2	31.00 usec
CHANNEL F1 =====	
NUC1	13C
P1	15.50 usec
P1	500.00 usec
P1	200.00 usec
P1	120.00 usec
P1	100.00 usec
P1	75.454 sec
SP1	3.20 dB
SP1	2.00 dB
SP1	1.00 dB
SP1	0.00 dB
CPDPR2	Walsh16
SP2	0.00 dB
SP2	20.00 sec
SP2	10.00 sec
SP2	5.00 sec
SP2	2.50 sec
SP2	1.00 sec
SP2	0.50 sec
SP2	0.25 sec
SP2	0.125 sec
CHANNEL F2 =====	
GRAD1	GRADIENT CHANNEL
GRAD2	GRADIENT CHANNEL
GRAD1	SINE,100
GRAD2	SINE,100
GP1	0.00 sec
GP2	0.00 sec
GP3	0.00 sec
GP4	0.00 sec
GP5	0.00 sec
GP6	0.00 sec
GP7	0.00 sec
GP8	0.00 sec
GP9	0.00 sec
GP10	0.00 sec
GP11	0.00 sec
GP12	0.00 sec
GP13	0.00 sec
GP14	0.00 sec
GP15	0.00 sec
GP16	0.00 sec
F2 - Processing parameters	
S1	65536
SF	125.76499 MHz
NDW	no
NSB	0
LB	0.00 Hz
GB	0
PC	2.00
1D NMR pilot parameters	
CX	22.80 cm
CY	15.65 cm
CP	220.00 Hz
F1	271.65 Hz
F2	0.0000 Hz
F2	0.0000 Hz
PPM0	9.4491 ppm
HOM	123.4567 ppm



1H spectrum



Z-restored spin-echo  $^{13}\text{C}$  spectrum with LH decoupling



wdc

Current Data Parameters

USER	eborn
NAME	ETr-24
EXPO	4
PROCNO	1

F2 - Acquisition Parameters

Date_	20140719
Time_	21:54
INSTRUM	5 mm CP/CPS 1H
PROBID	8G
FIDPRE	8094.2038F.psd
ID	65356
SOLVENT	CDC13
NS	426
DS	16
SWH	30301.51 Hz
TDRES	0.46288 Hz
AQ	1.063340 sec
RG	296.3
DW	16.300 usec
DE	6.00 usec
TE	296.0 K
D1	0.250000 sec
D11	0.030000 sec
D16	0.002000 sec
d17	0.00019600 sec
MCBEST	0.000000 sec
NCURV	0.0150000 sec
P2	31.00 usec

===== CHANNEL f1 =====

NUC1	13C
P1	15.50 usec
P11	500.00 usec
P12	2000.00 usec
PL0	120.00 dB
PL1	-1.00 dB
SP01	125.794248 MHz
SP1	3.20 dB
SPNAME1	Crf60.5.201.1
SPNAME2	Crf61cmcp.4
SPRF1	0.00 Hz
SPRF2	0.00 Hz

===== CHANNEL f2 =====

CPDPG2	1H
NPC2	1H
FCPD2	100.00 usec
PL2	1.60 dB
PL12	24.60 dB
SP2	500.2235011 MHz

===== GRADIENT CHANNEL =====

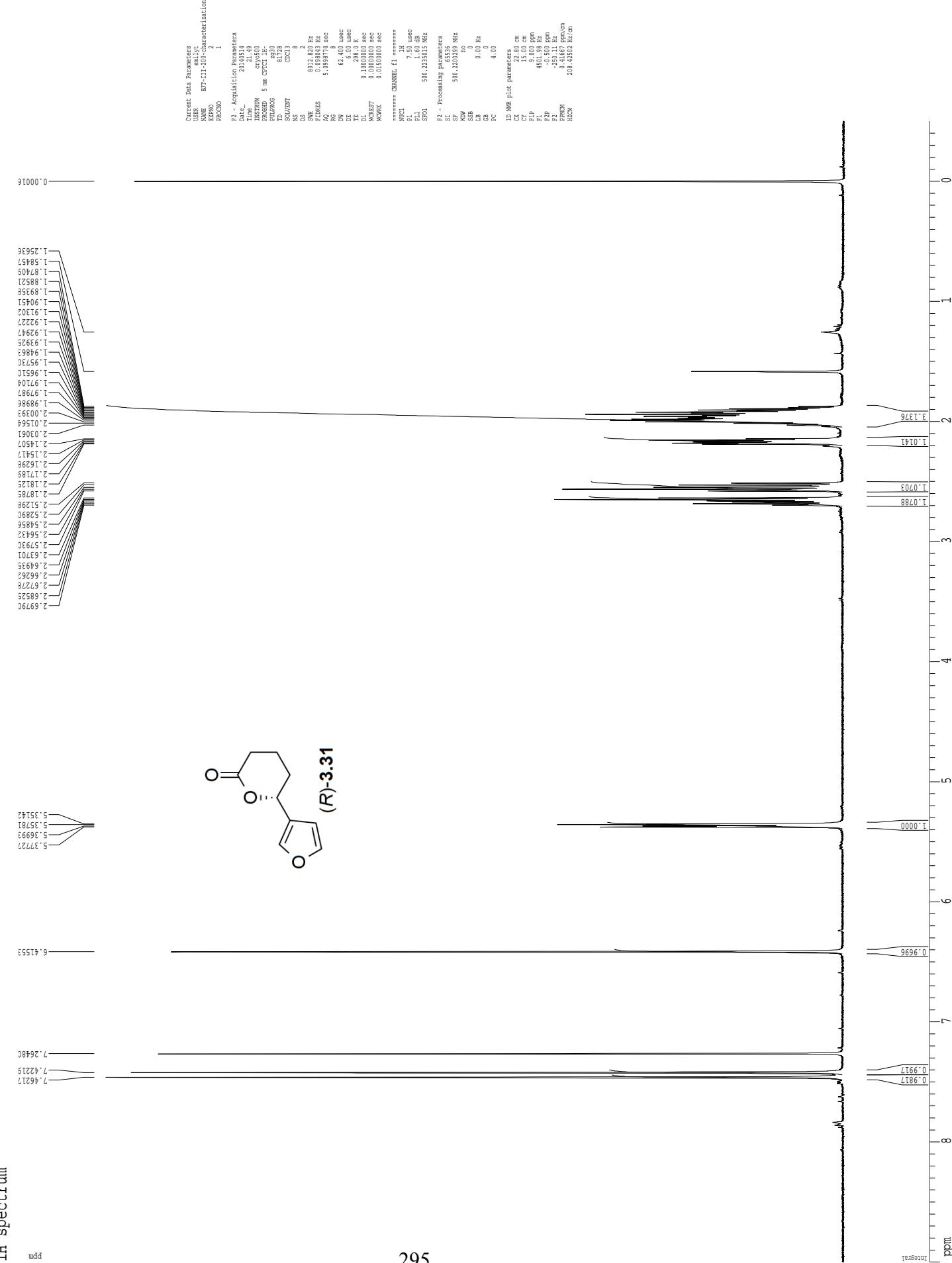
GRDM1	SINE 100
GRDM2	SINE 100
GP11	0.00 %
GP12	0.00 %
GP21	0.00 %
GP22	30.00 %
GP15	50.00 %
P16	1000.00 usec

F2 - Processing parameters

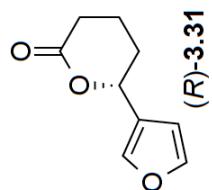
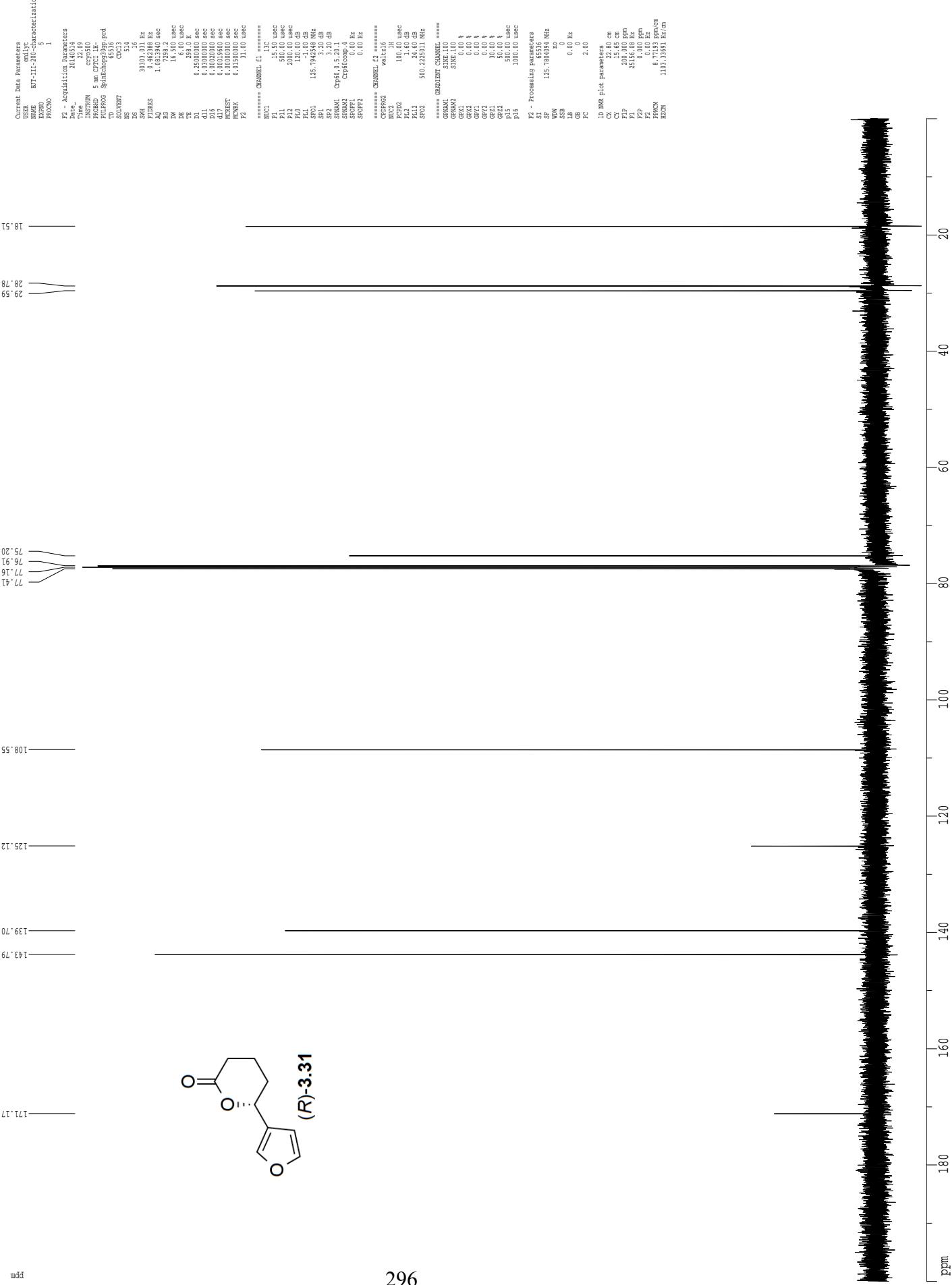
S1	65356
SF	125.7804065 MHz
NDW	1
SSB	0
LB	1.00 Hz
GB	0
FC	2.00

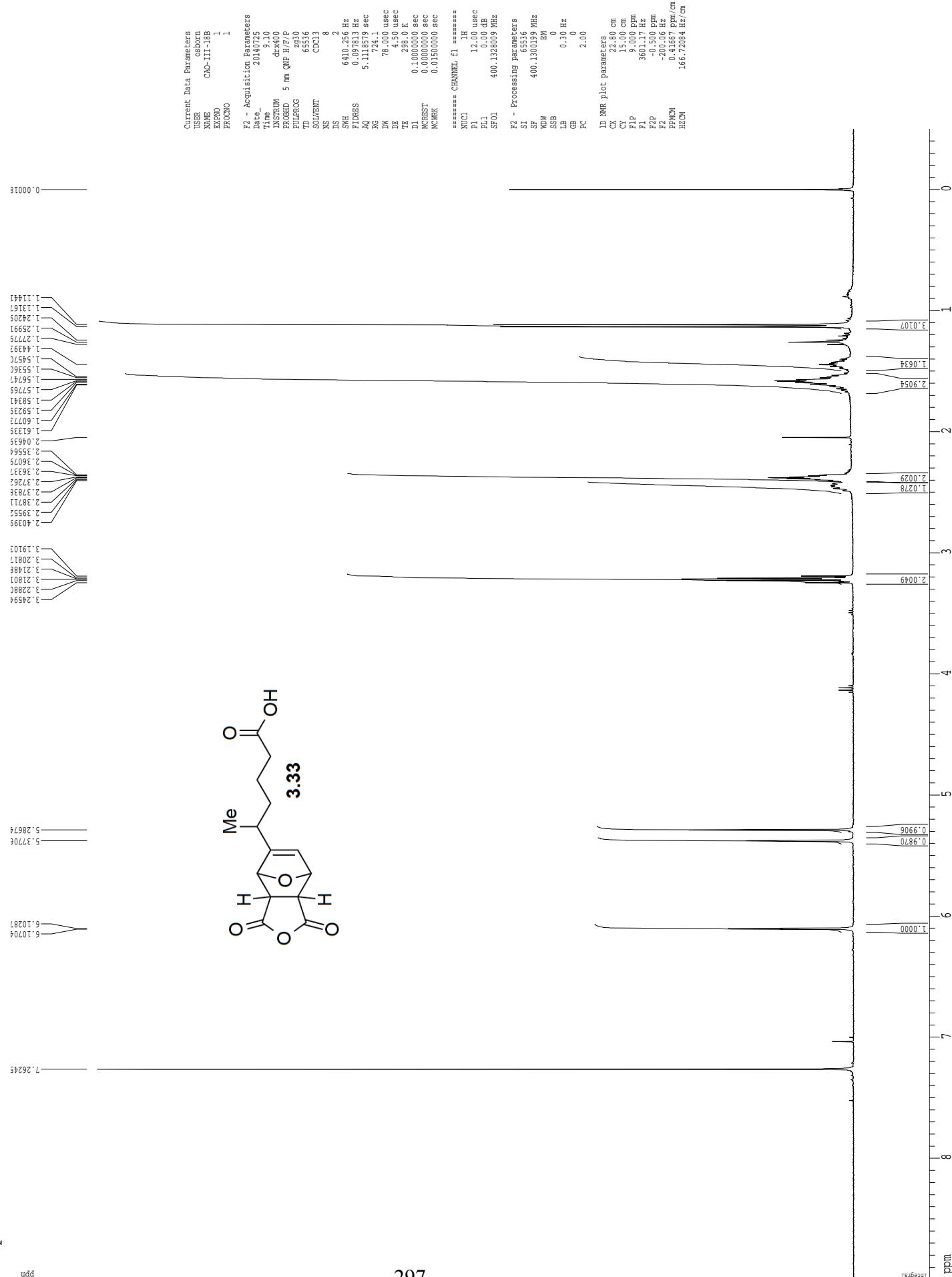
1D NMR plot parameters

CX	15.80 cm
CY	15.65 cm
F1P	230.637 ppm
F1	2809.68 Hz
F2P	-10.287 ppm
F2	-129.396 Hz
PPCM	10.16688 ppm/cm
HZCM	12.10693 Hz/cm

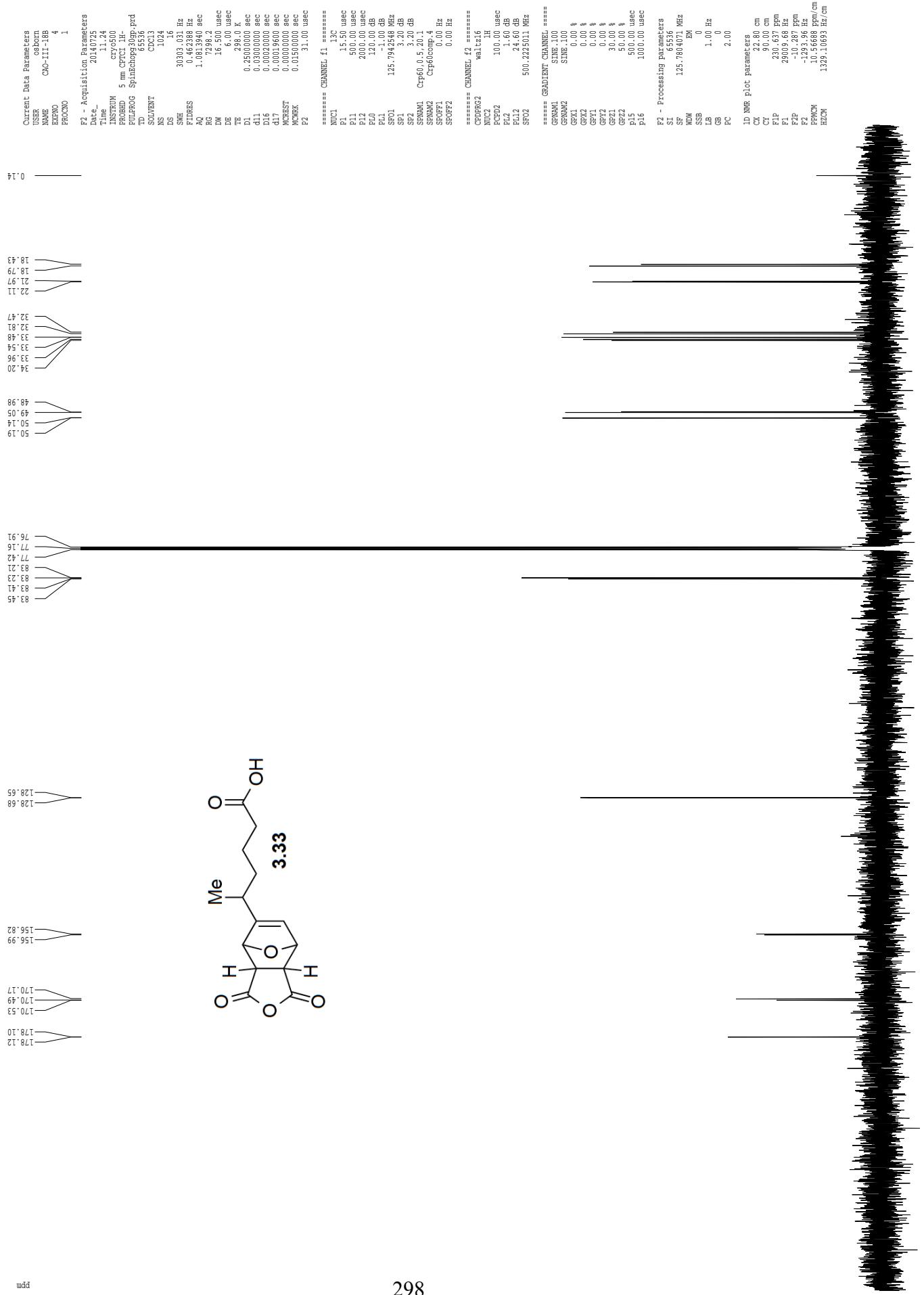


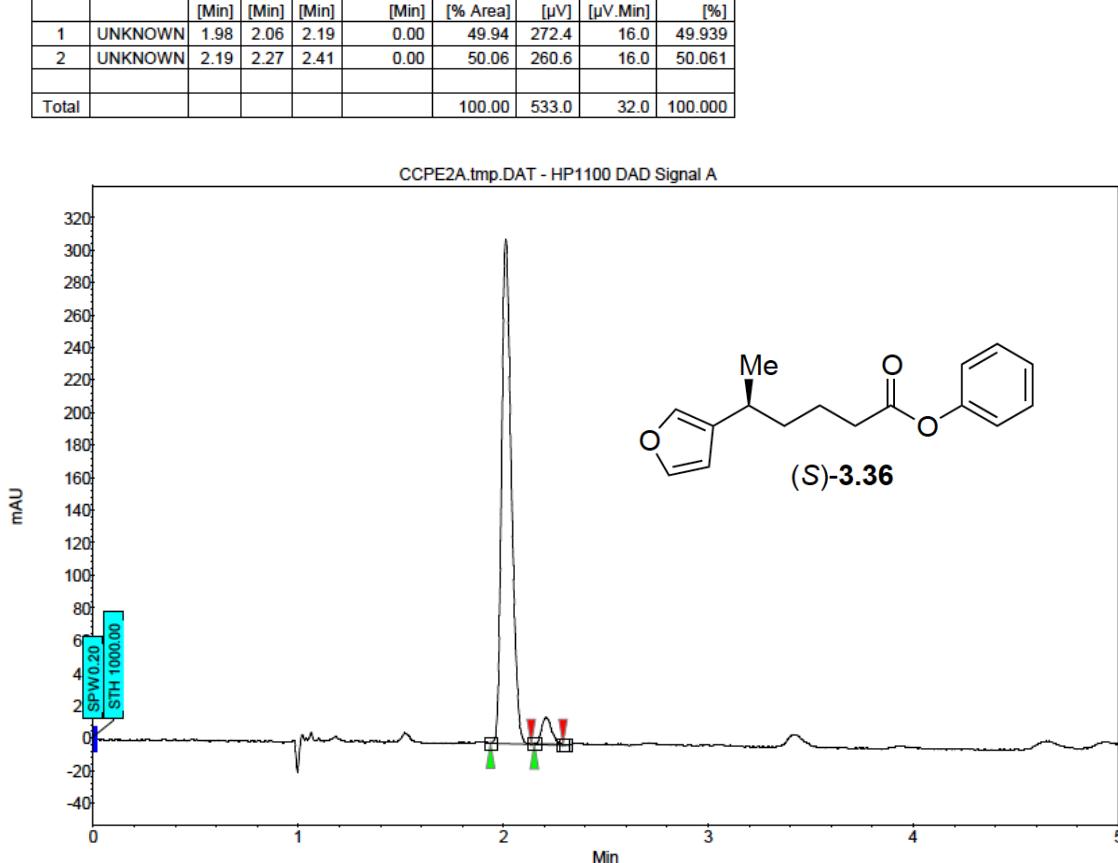
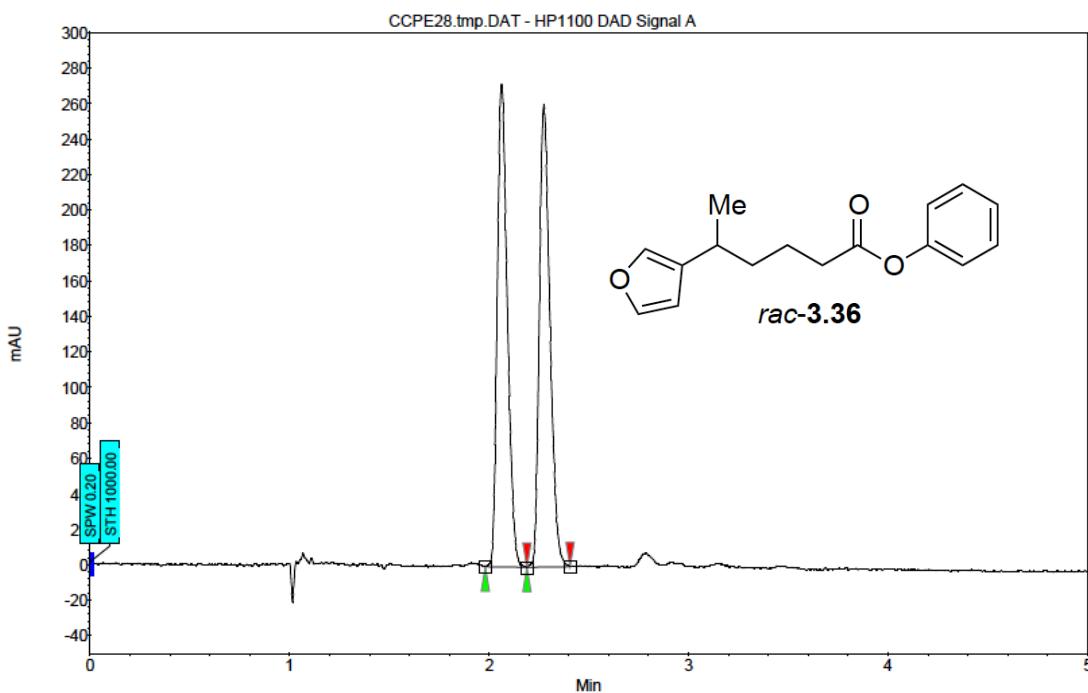
Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

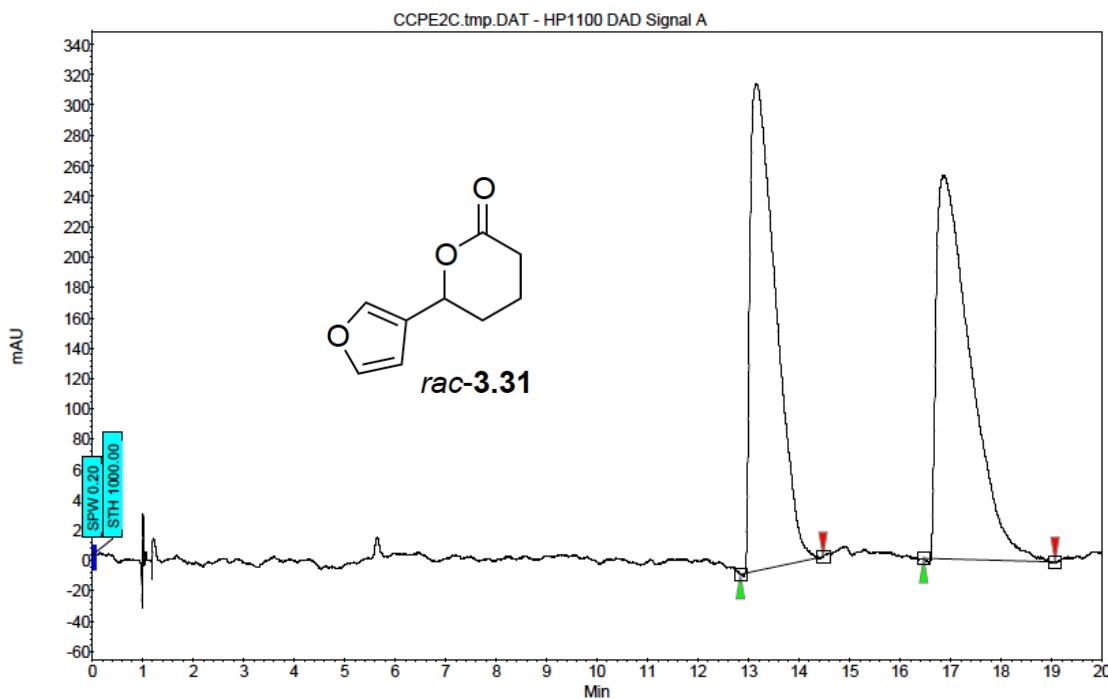




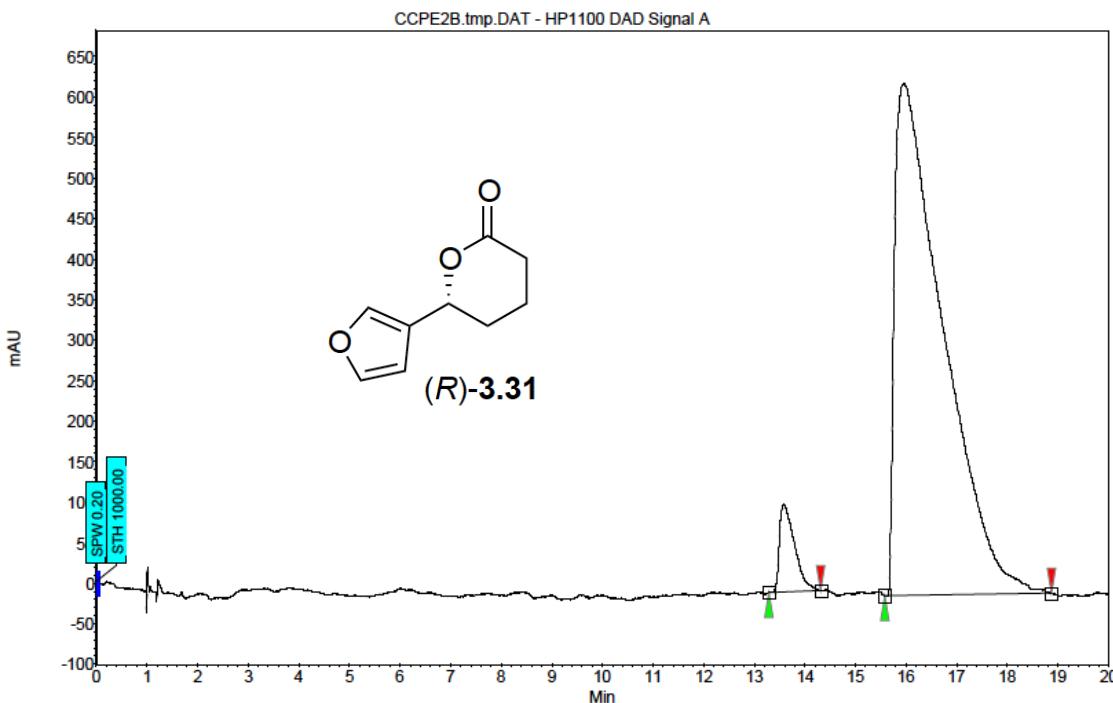
Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $1\text{H}$  decoupling

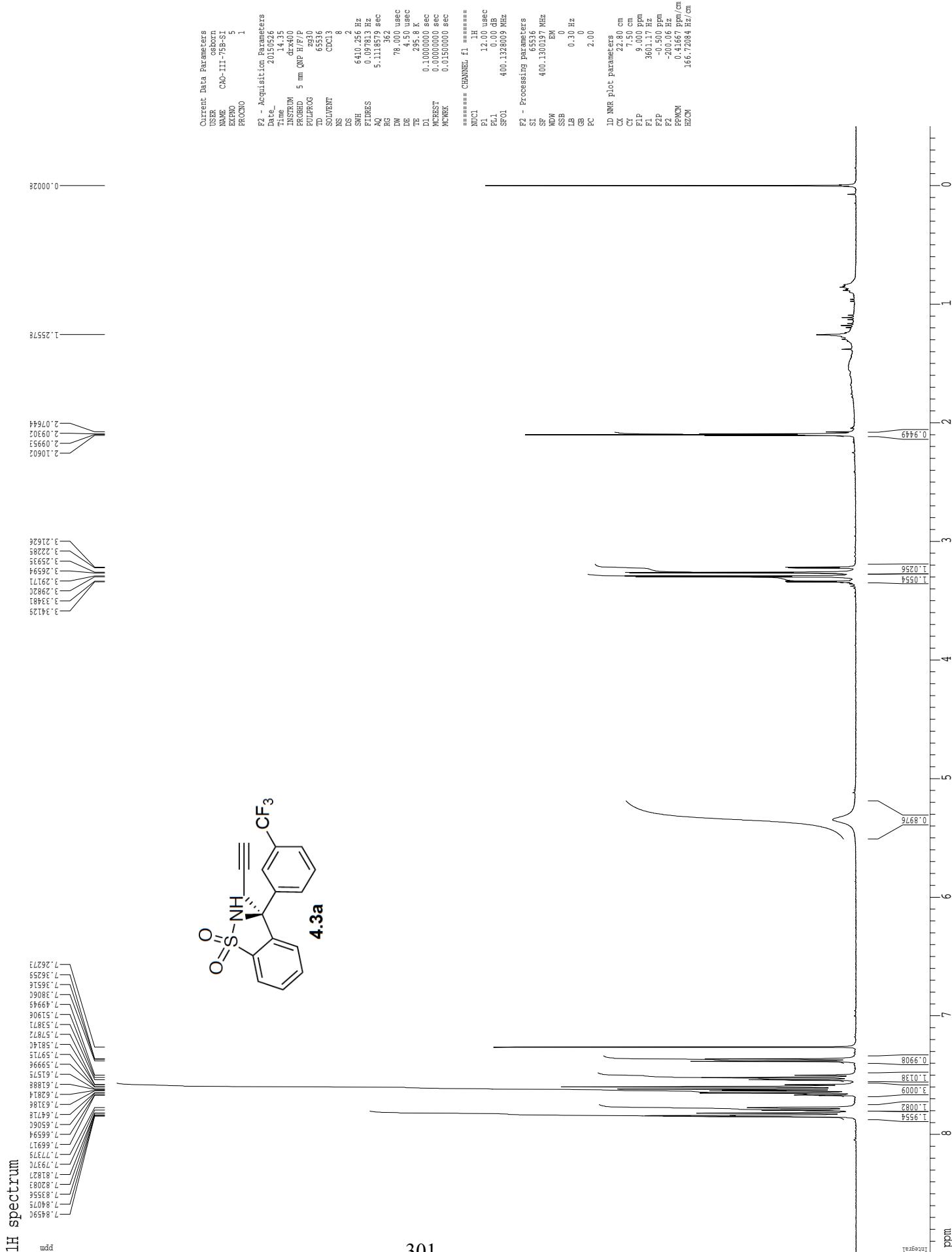






Index	Name	Start [Min]	Time [Min]	End [Min]	RT Offset [Min]	Quantity [% Area]	Height [µV]	Area [µV.Min]	Area [%]
1	UNKNOWN	12.83	13.15	14.48	0.00	48.70	321.1	187.4	48.704
2	UNKNOWN	16.46	16.88	19.07	0.00	51.30	252.7	197.4	51.296
Total					100.00	573.8	384.8	100.000	





1H spectrum

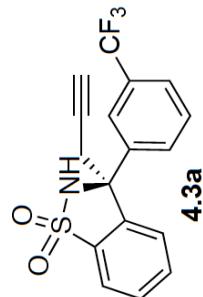
Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

=====  
 Current Data Parameters  
 USER      usborn  
 NAME      Cb-111-75-B-SI  
 EXPNO     4  
 PRGNO    1  
 F2 - Acquisition Parameters  
 Date      20150525  
 Time      19.13  
 INSTRUM  INSPINW  
 PROBHD  5 mm PCD1 1H-  
 PULPROG SpinEchoes30DP.prd  
 TD        65536  
 SOLVENT   NS  
 PCP1     291  
 J1        16  
 DS        0  
 SWH      3033.031 Hz  
 FIDRES  0.0462388 Hz  
 AQ        1.01390 sec  
 R261     16.500 usec  
 DR        16.500 usec  
 DE        10.00 usec  
 TE        296.1 K  
 D1        0.260000 sec  
 d11      0.000000 sec  
 D16      0.002000 sec  
 d17      0.001960 sec  
 NCEST   0.000000 sec  
 MCNRRK 0.0150000 sec  
 F2R      33.10 usec  
 SPCFF2 0.00 Hz  
 ======  
 ===== CHANNEL F1 ======  
 NUC1      13C  
 P1        16.55 usec  
 P12      50.00 usec  
 P120     200.00 usec  
 PLL      120.00 dB  
 SP01     -1.00 dB  
 SP1      125.794258 MHz  
 SP2      2.70 dB  
 SPW01   Crp60.0,5,20.1  
 SPW02   Crp60.0,5,20.1  
 Crp60.0,5,20.1  
 SPCFF1 0.00 Hz  
 SPCFF2 0.00 Hz  
 ===== CHANNEL F2 ======  
 CPDPRG2  Walz16  
 NUC2      1H  
 GPPM1     SINE,100  
 GPPM2     SINE,100  
 GPPX1    0.00 %  
 GPPX2    0.00 %  
 GPY1     0.00 %  
 GPY2     0.00 %  
 GPZ1     30.00 %  
 GPZ2     50.00 %  
 GP22     50.00 %  
 P15      50.00 usec  
 P16      1000.00 usec  
 ===== GRADIENT CHANNEL ======  
 GPPM1     SINE,100  
 GPPM2     SINE,100  
 GPPX1    0.00 %  
 GPPX2    0.00 %  
 GPY1     0.00 %  
 GPY2     0.00 %  
 GPZ1     30.00 %  
 GPZ2     50.00 %  
 ===== PROCESSING PARAMETERS ======  
 CX        22.80 cm  
 CY        35.00 cm  
 F1P      230.637 ppm  
 F1        230.938 Hz  
 LB        1.00 Hz  
 GB        0  
 PC        2.00  
 1D NMR Plot Parameters  
 CX        22.80 cm  
 CY        35.00 cm  
 F1P      230.637 ppm  
 F1        230.938 Hz  
 LB        1.00 Hz  
 F2P      -10.287 ppm  
 F2        -10.287 ppm  
 FPPCM    10.5668 ppm/cm  
 HZCM    1323.1063 Hz/cm

31.52

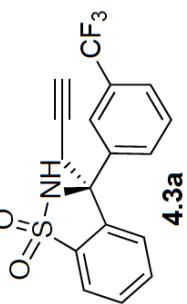
66.67  
 73.93  
 76.93  
 77.14  
 77.34  
 77.73

121.79  
 122.28  
 123.22  
 123.28  
 123.31  
 124.05  
 124.69  
 125.66  
 129.75  
 130.34  
 130.38  
 131.55  
 133.16  
 135.09  
 141.61  
 144.61



19F spectrum with 1H decoupling

ddd



```

Current Data Parameters
USER          osborn
NAME         C10-III-75B-SI
EXNO          7
PRCNO         1

P2 - Acquisition Parameters
Date        20150526
Time        20.53
INSTRUM   qcpmg
PROBODIM  5 mm QNP H/F/P
PULPROG  zg3ghe130
TD        65536
SOLVENT    CDCl3
PCP1       31
DS           4
SWH       75187.969 Hz
ETW       1.147277 Hz
TDRES     0.435864 sec
AQ        3649.1
RG        6.650 usec
DW        9.46 usec
DE        298.0 K
TE        2.000000 sec
D1        0.030000 sec
Q1        0.0002000 sec
Q2        0.0002000 sec

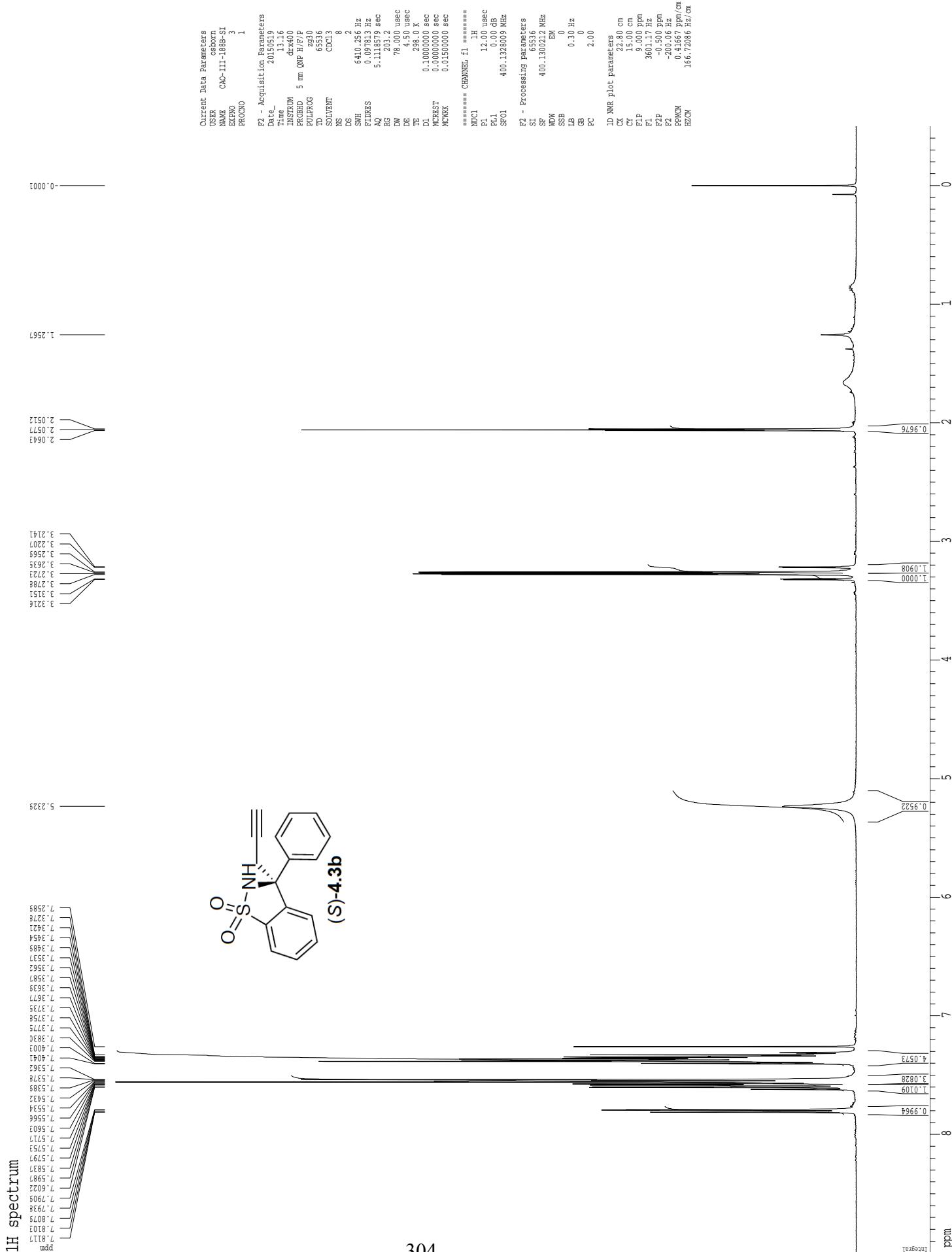
===== CHANNEL f1 =====
NUCL1      19F
P1        22.50 usec
PL1      376.4646491 MHz
SF01

===== CHANNEL f2 =====
CPDPRG2
NUC2      1H
P2CD2    90.00 usec
PL2      120.00 dB
PL12     17.70 dB
SF02    400.1320007 MHz

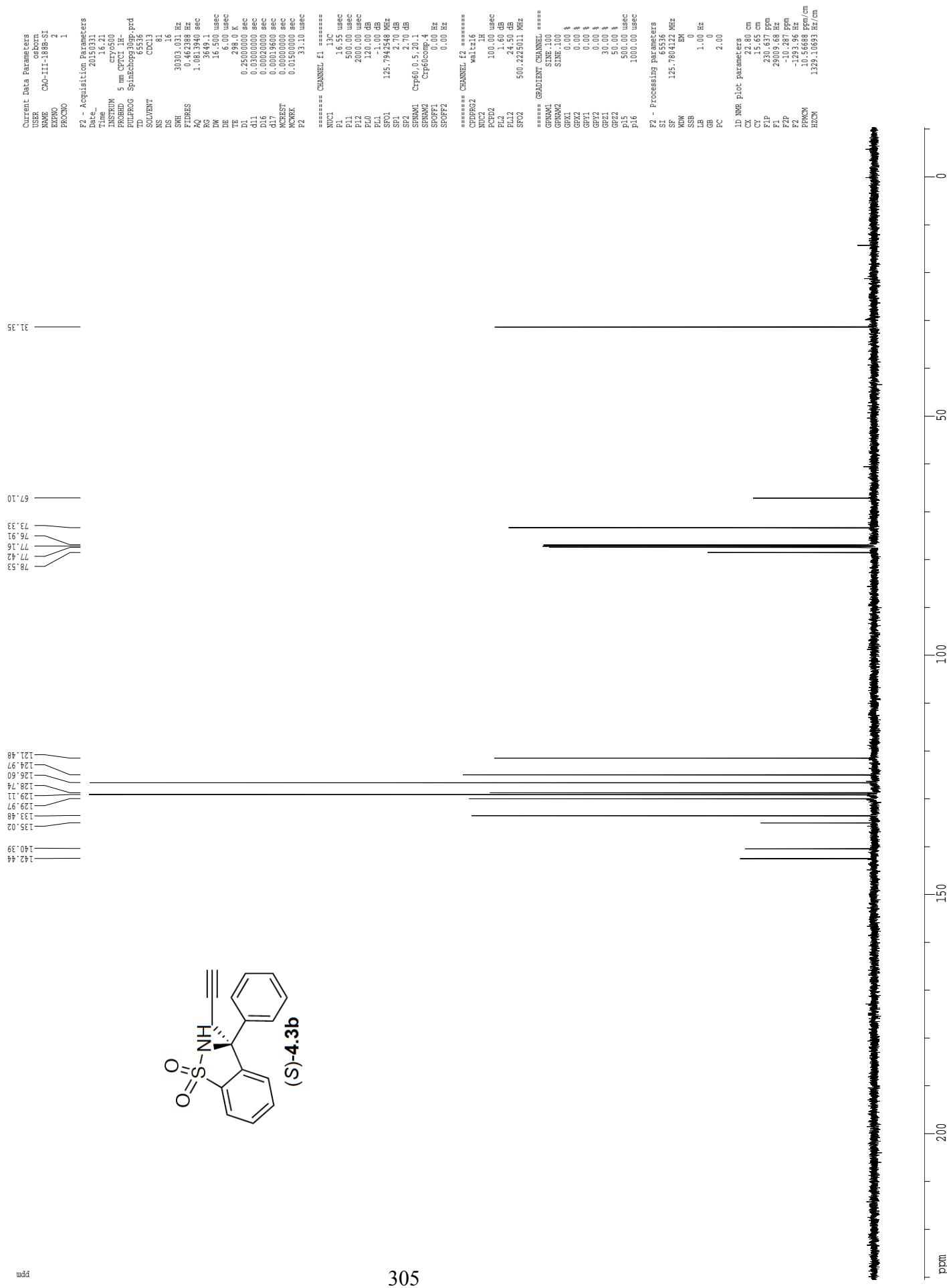
P2 - Processing parameters
SI        65536
SF      376.4933852 MHz
WDW
SSB      0
LB      0.30 Hz
GB      0
PC      1.00

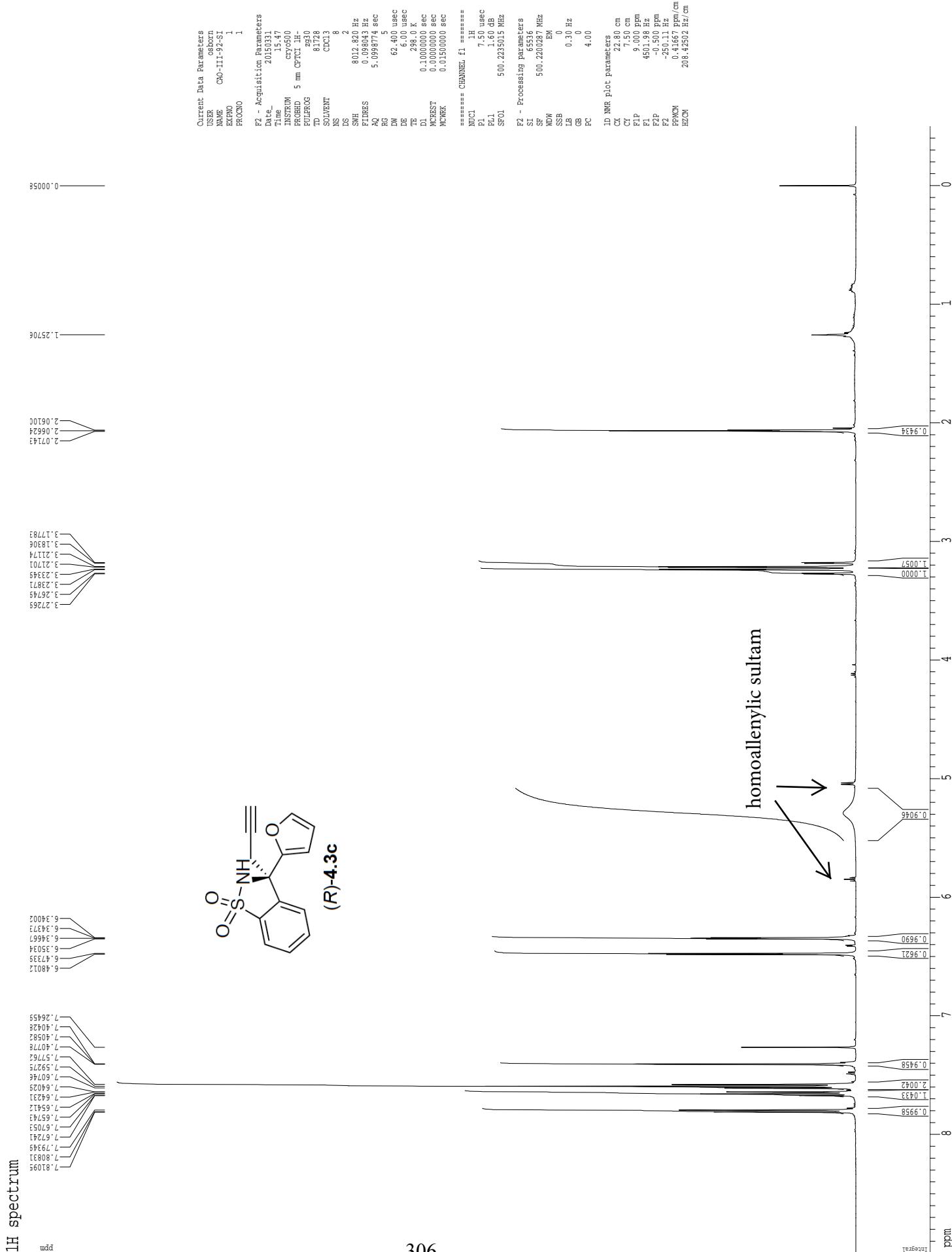
```

HZCM 3153.99376 Hz/cm

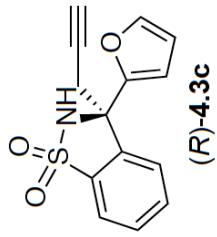
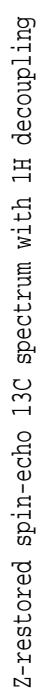


Z-restored spin-echo  $^{13}\text{C}$  spectrum with LH decoupling

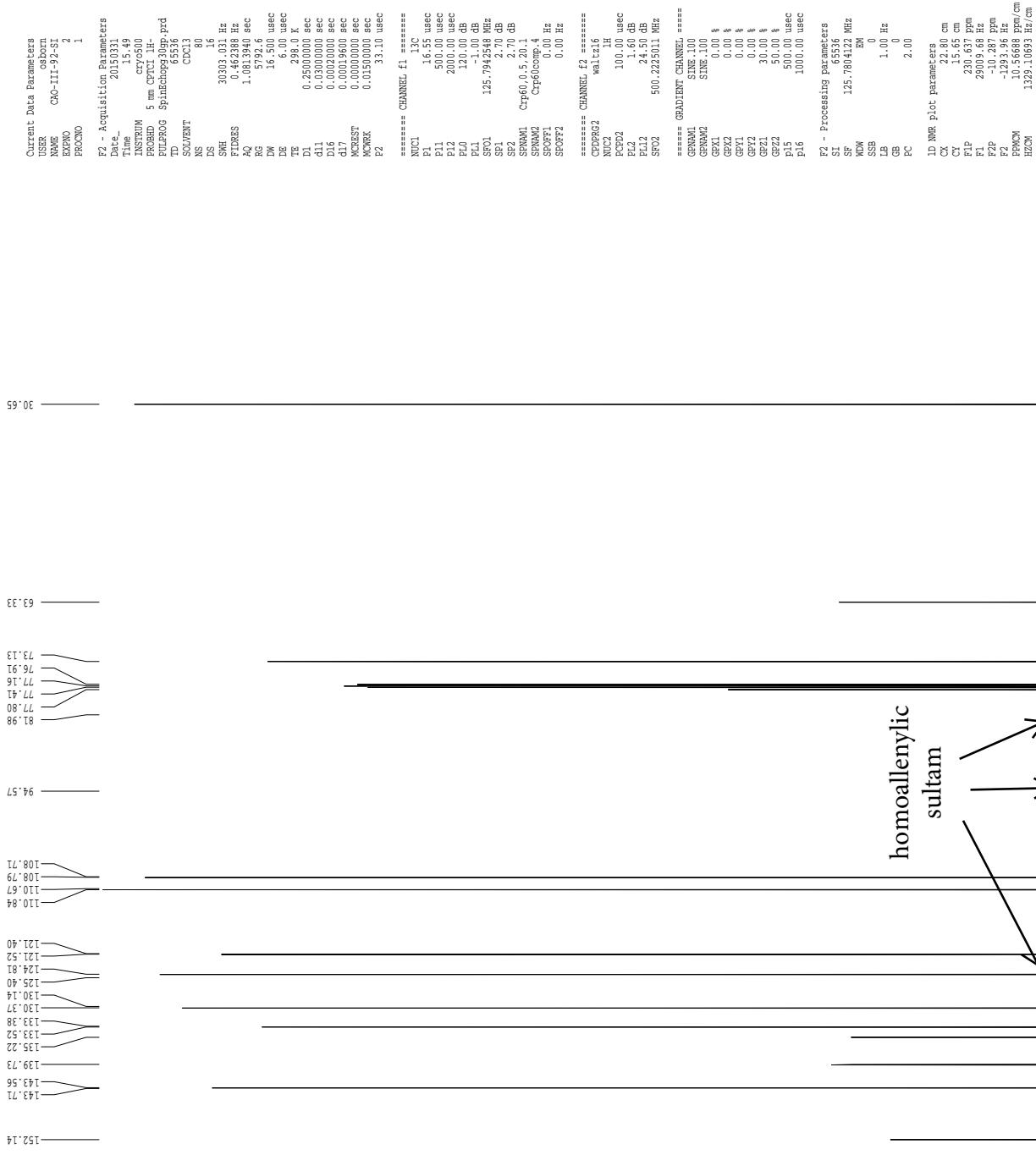




1H spectrum

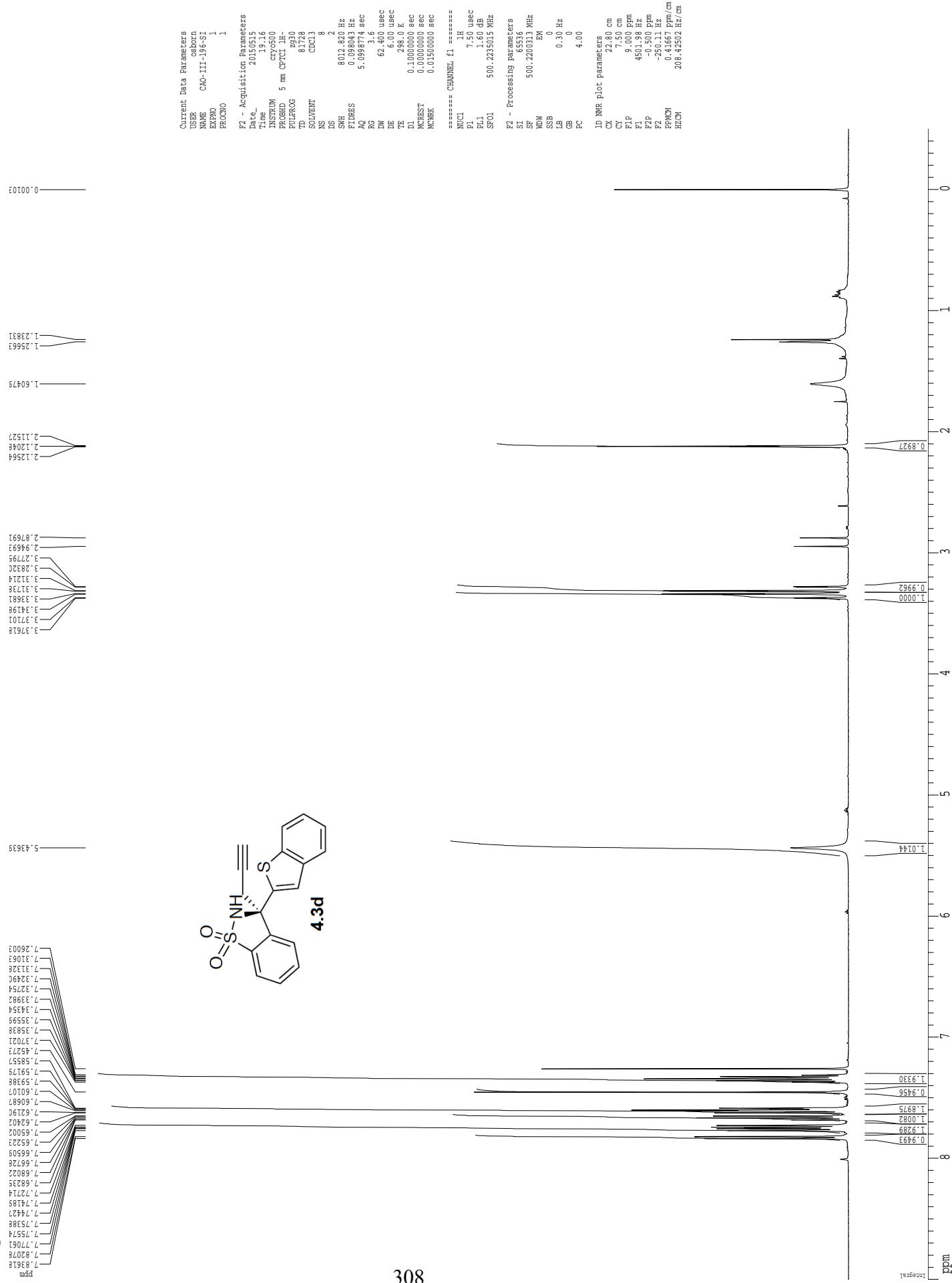


(R)-4.3c



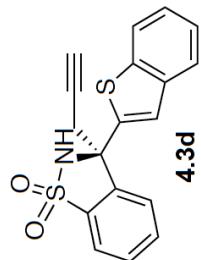
307

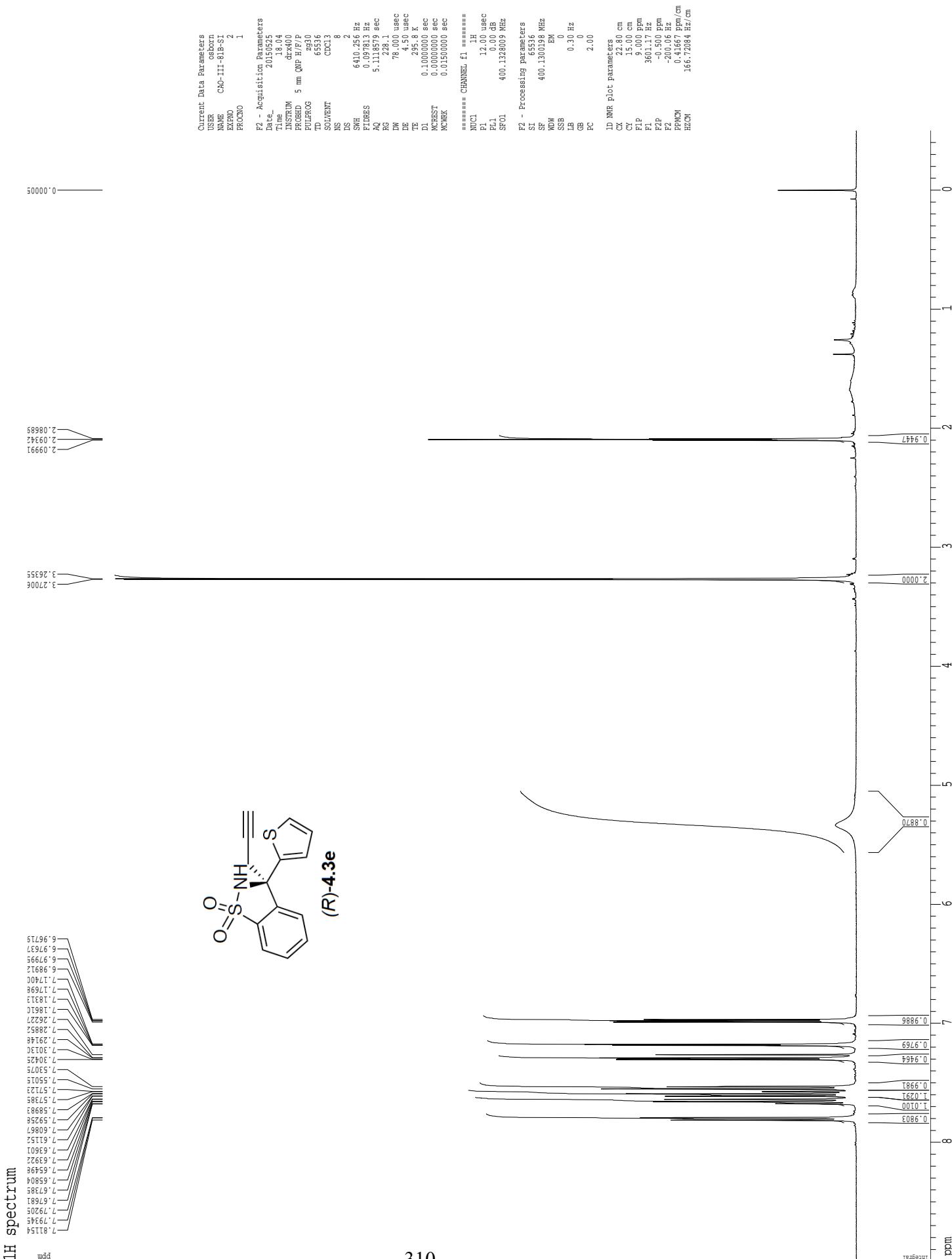
<sup>1</sup>H spectrum



Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

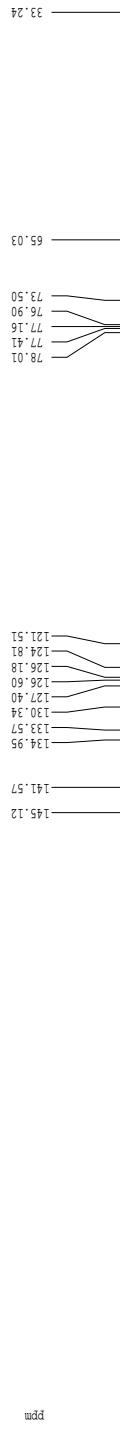
=====  
 Current Data Parameters  
 USER      usborn  
 NAME     Cb-111-196-SI  
 EXNO      1  
 PRCNO  
 F2CNO  
 F2 - Acquisition Parameters  
 Date    20150515  
 Time    19.19  
 INSTRUM  cry500  
 PROBHD  5 mm CCP1 1H-  
 PULPROG SpinEchoes30P.prd  
 TD       65536  
 SOLVENT  
 CPMG1    200  
 NS       2048  
 DS       16  
 SWH    3003.021 Hz  
 FIDRES 0.062398 Hz  
 AQ      1.00390 sec  
 TS       128.2  
 DR       16.50 usec  
 DE       3.00 usec  
 TE       296.0 K  
 D1       0.260000 sec  
 d11      0.000000 sec  
 D16      0.002000 sec  
 t11      0.001960 sec  
 NCEST  
 MCNSTK  
 PCPRK    0.050000 sec  
 F2R  
 F2       33.10 usec  
 ======  
 ====== CHANNEL F1 ======  
 NUC1      13C  
 P1       1.65 usec  
 P11      50.00 usec  
 P12      200.00 usec  
 PLO      120.00 dB  
 PLL      -1.00 dB  
 SP01     128.794258 MHz  
 SP1      2.70 dB  
 SP2      2.70 dB  
 SPW01    Crp60.0,5,20.1  
 SPW02    Crp60.0,5,20.1  
 SPFF1    0.00 Hz  
 SPFF2    0.00 Hz  
 ====== CHANNEL F2 ======  
 CPDPRG2  
 NUC2      1H  
 PCPD2    10.00 usec  
 PL2      1.60 dB  
 PLL2    24.50 dB  
 SF02    500.2225011 MHz  
 ====== GRADIENT CHANNEL ======  
 GPRM01    SINE.100  
 GPRM02    SINE.100  
 GPX1      0.00 %  
 GPX2      0.00 %  
 GPY1      0.00 %  
 GPY2      0.00 %  
 GPZ1      30.00 %  
 GPZ2      50.00 %  
 GP5      50.00 usec  
 P15      100.00 usec  
 P16      1000.00 usec  
 ====== Processing Parameters ======  
 SI       65536  
 SF       125.7804094 MHz  
 WDW      EM  
 SSB      0  
 LB       1.00 Hz  
 GB       0  
 PC       2.00  
 1D NMR Plot Parameters  
 CX       22.80 cm  
 CY       15.65 cm  
 F1P      230.67 ppm  
 F1       2300.968 Hz  
 F2P      -10.287 ppm  
 F2       -129.346 Hz  
 PPCM    10.5668 ppm/cm  
 HZWM    1323.10693 Hz/cm  
 ======  
 32.85  
 65.37  
 73.77  
 76.93  
 77.16  
 77.44  
 77.77  
 121.68  
 122.49  
 122.99  
 124.26  
 124.77  
 125.35  
 130.53  
 133.66  
 134.49  
 135.29  
 139.39  
 140.59  
 145.50  
 ppm





1H spectrum

Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling



```

Current Data Parameters
USER   usborn
NAME   C6-111-81-B-SI
EXNO   5
PROCNO 1
F2 - Acquisition Parameters
Date   20150525
Time   18:52
INSTRUM  INSPINW
PROBHD  5 mm CCP1 1H
PULPROG SpinEchoes3DPP.prd
TD    65536
SOLVENT
NUC1  C13
PCP1  251
DS    16
SWH   3033.021 Hz
ETRIM
PTIMES
AQ    0.01390 sec
RG    128.2
TE    16.50 usec
T1    296.0 K
TDZ   0.26000 sec
D11
D16
D17
MCNEST
MCNRR
P2    33.10 usec

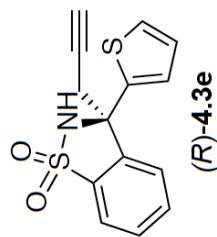
=====
CHANNEL F1 =====
NUC1  13C
P1    16.55 usec
P12   50.00 usec
P10   200.00 usec
PL0
PL1   -1.00 dB
SP01  125.794258 MHz
SP1
SP2
SPR001 Crp60.0,5,20.1
SPR002 Crp60.0,5,20.1
SP0FF1 0.00 Hz
SP0FF2 0.00 Hz

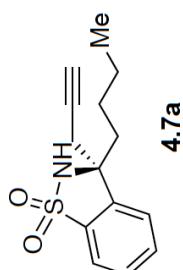
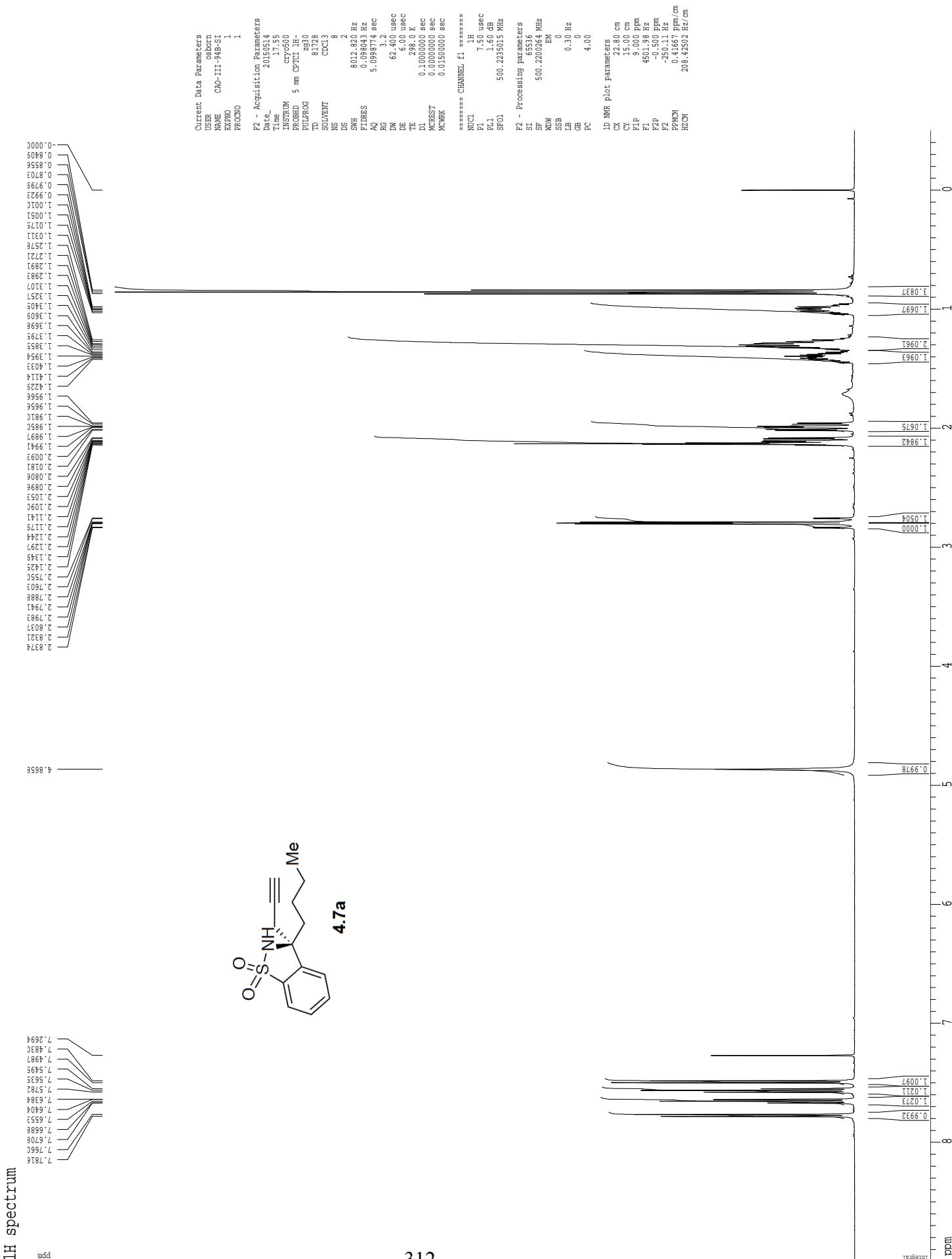
=====
CHANNEL F2 =====
NUC1  13C
P1    16.55 usec
P12   50.00 usec
PL0
PL1   -1.00 dB
SP01  125.794258 MHz
SP1
SP2
SPR001 Crp60.0,5,20.1
SPR002 Crp60.0,5,20.1
SP0FF1 0.00 Hz
SP0FF2 0.00 Hz

=====
GRADIENT CHANNEL =====
CPDPRG2
NUC2
GP0M1
GP0M2
SME.100
SF    65536
WDW
SSB
LB    1.00 Hz
GB
PC    2.00

1D NMR Plot Parameters
CX    22.80 cm
CY    15.65 cm
F1P   230.637 ppm
F1    230.968 Hz
F2P   -10.287 ppm
F2    -129.346 Hz
PPCM  10.5668 ppm/cm
HZDW  1323.10693 Hz/cm

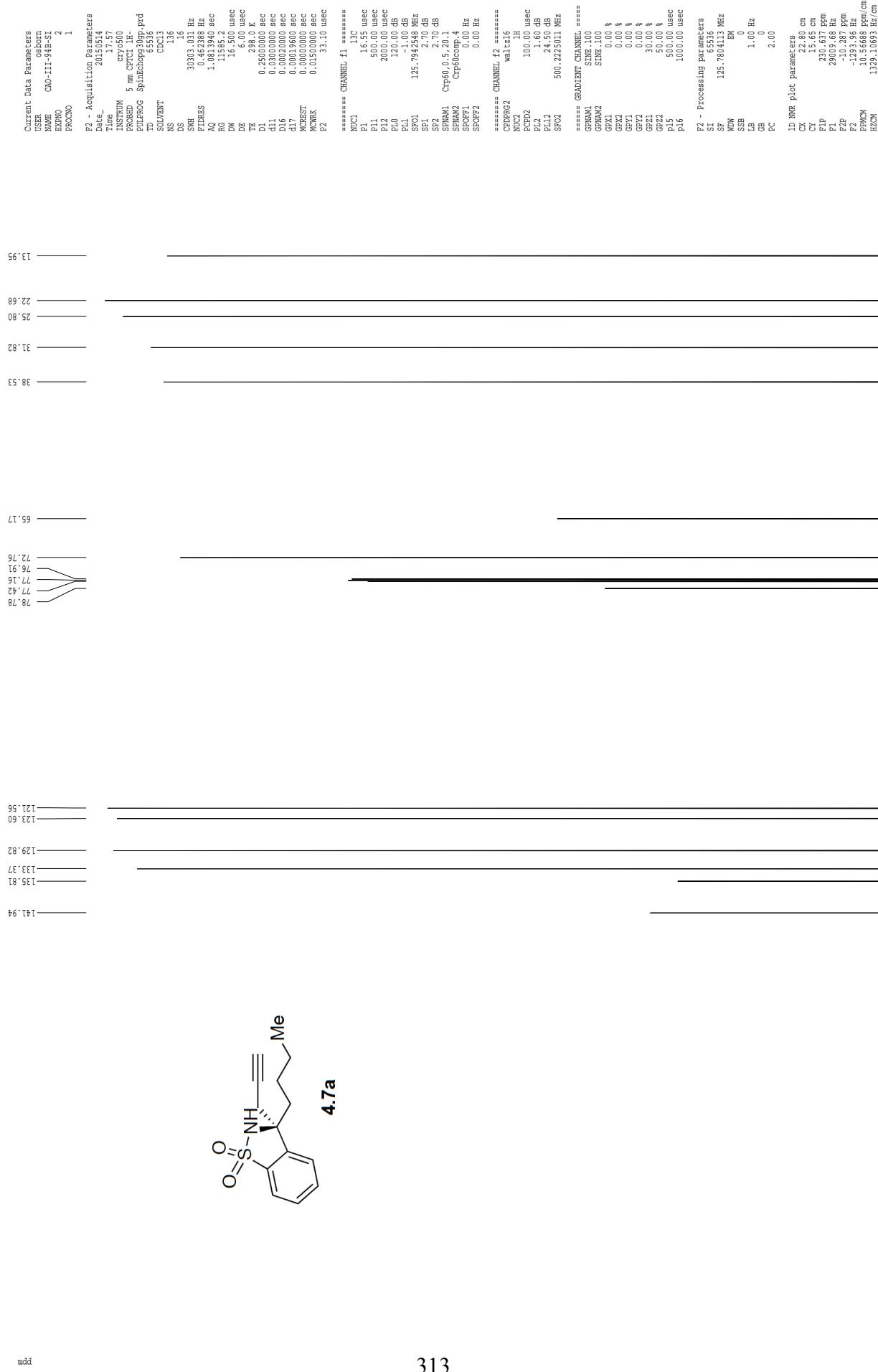
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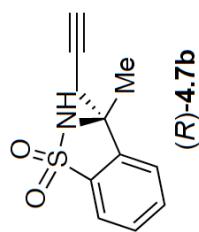
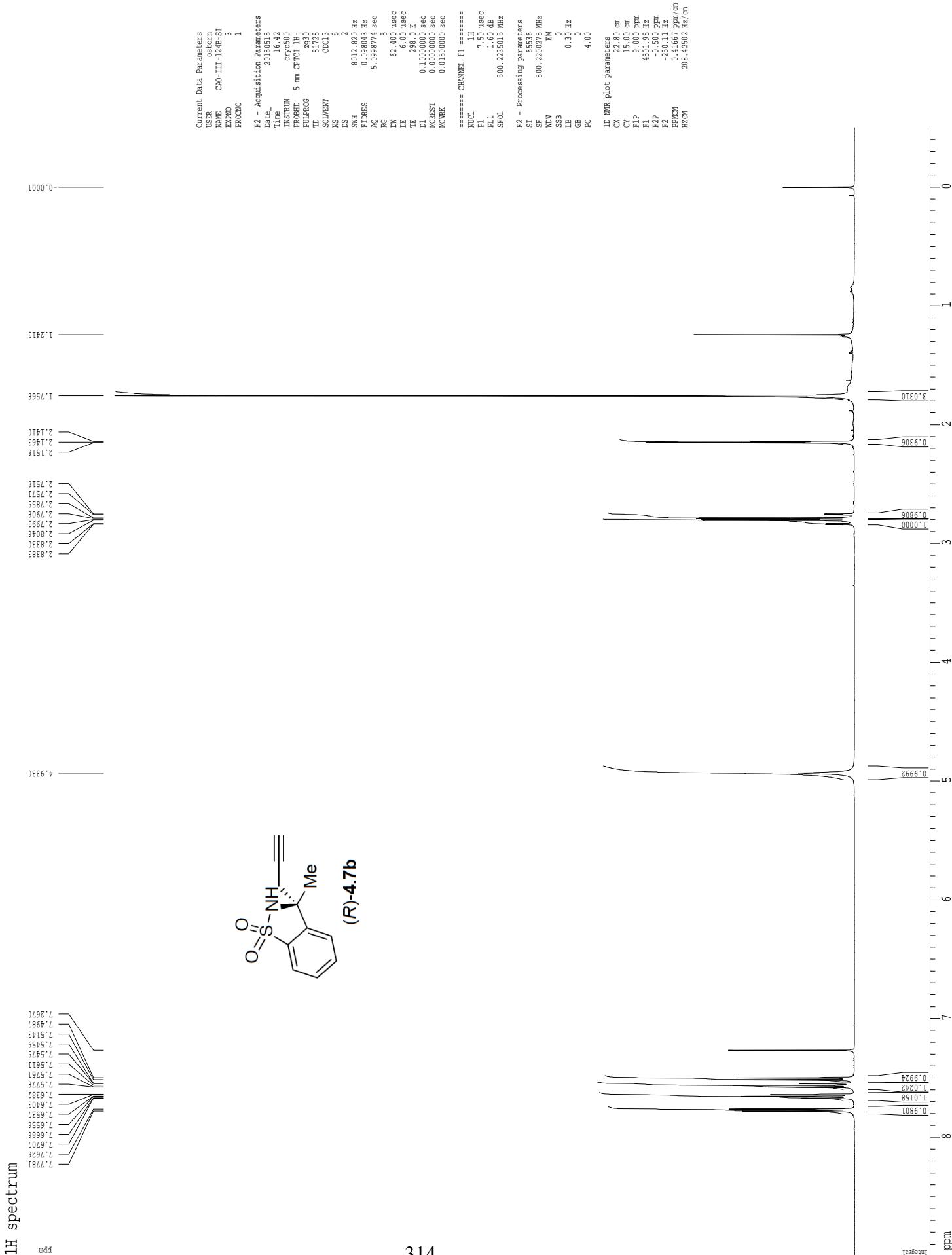




1H spectrum

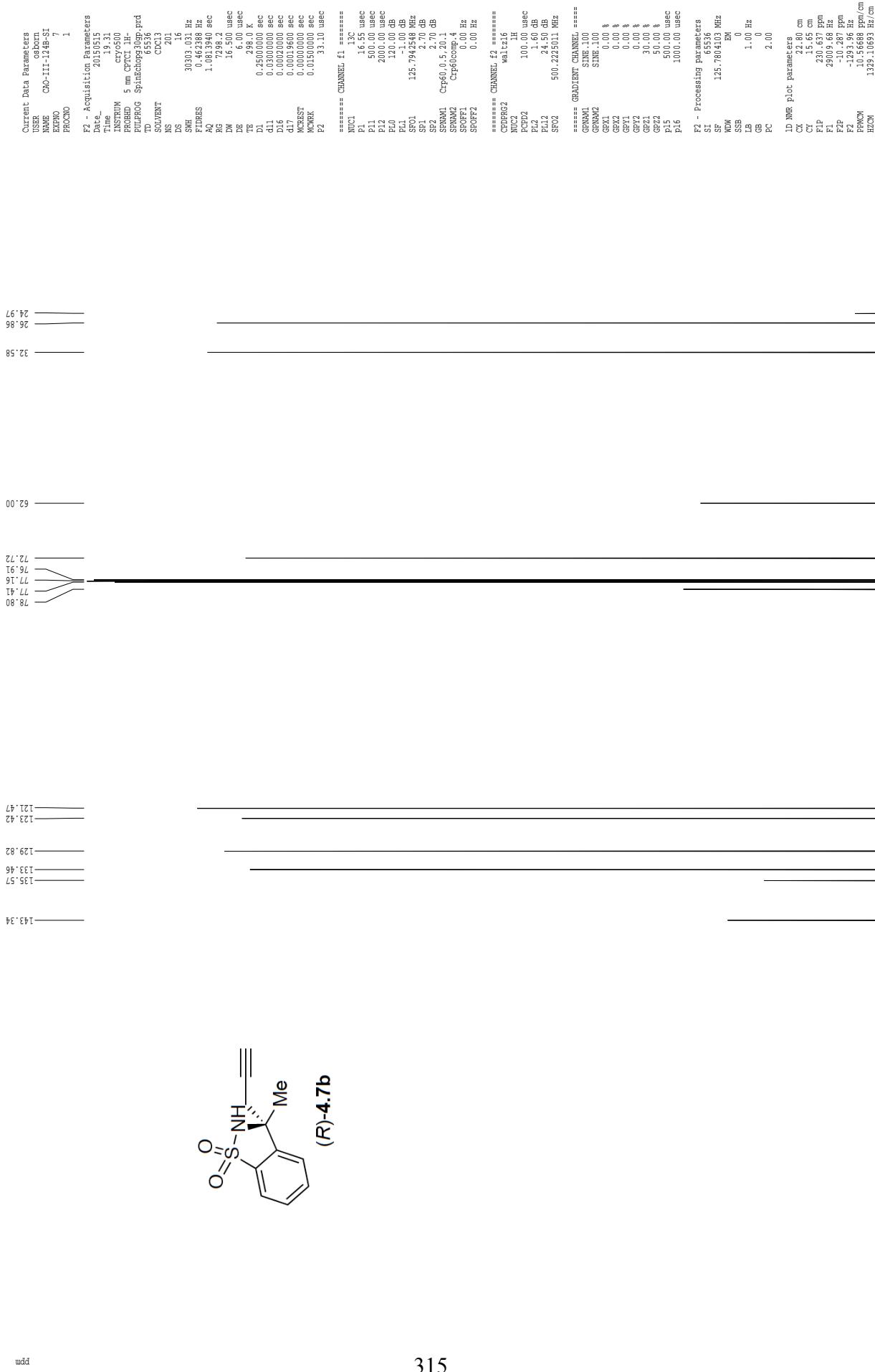
Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

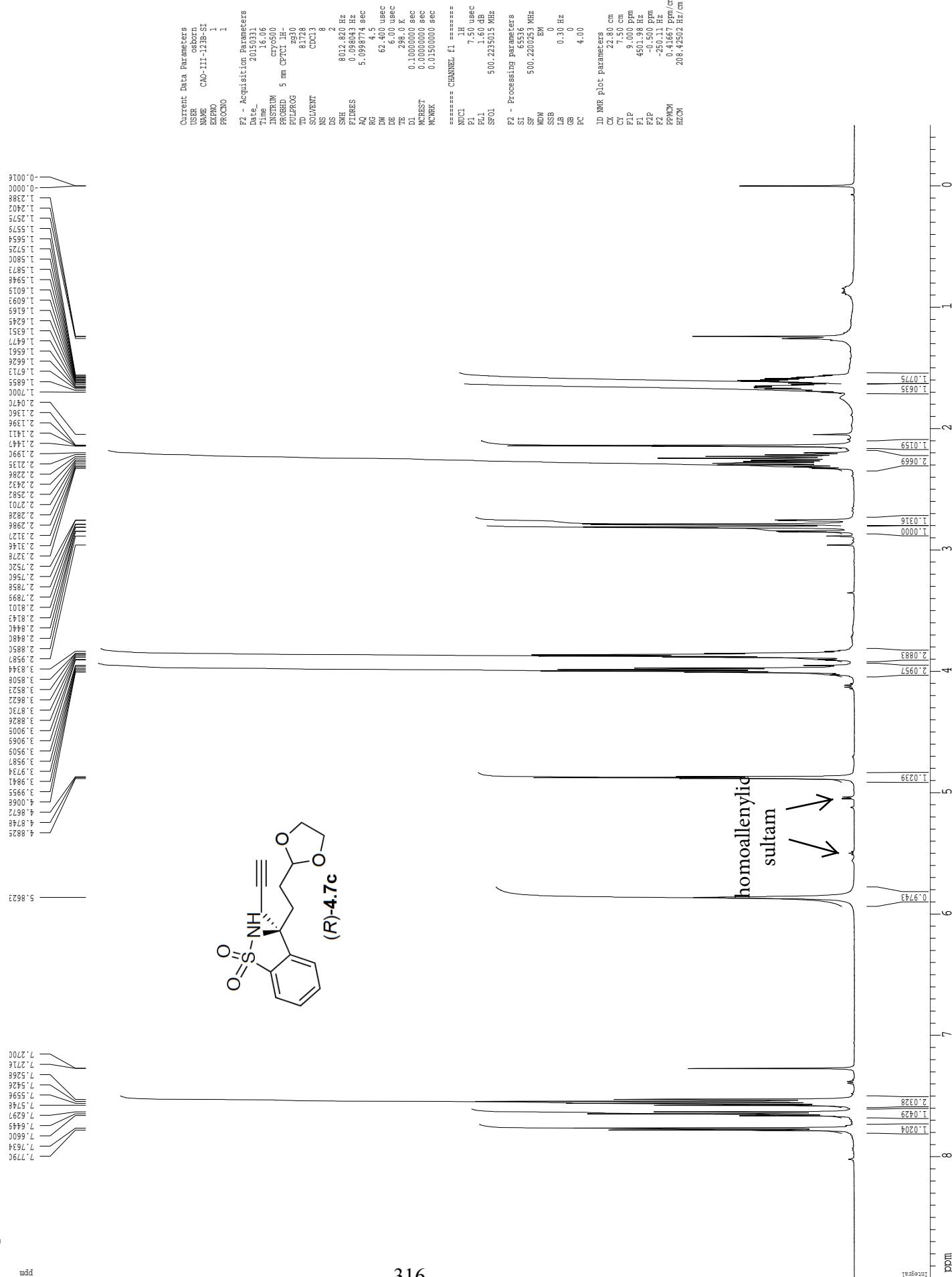




1H spectrum

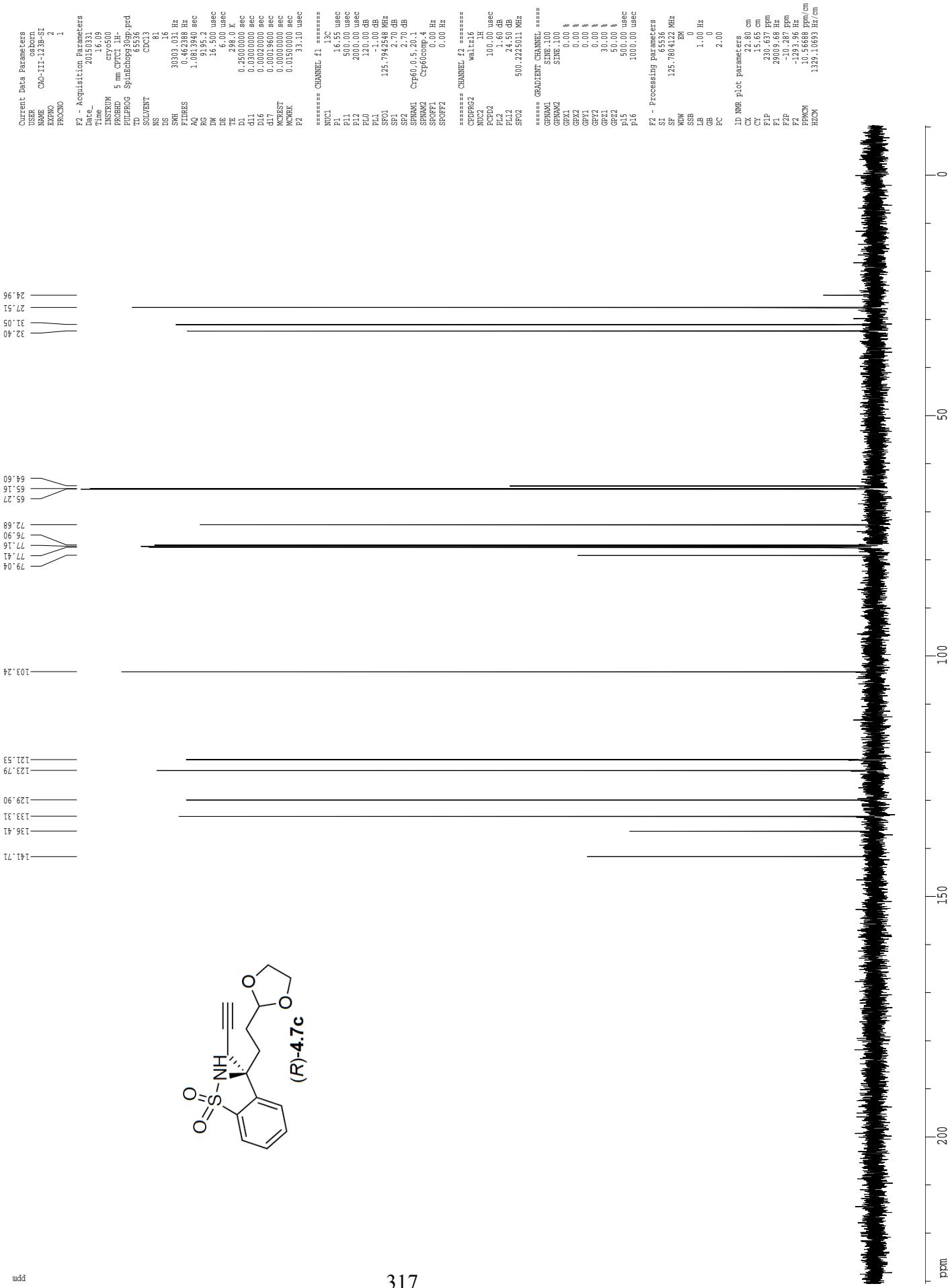
Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling



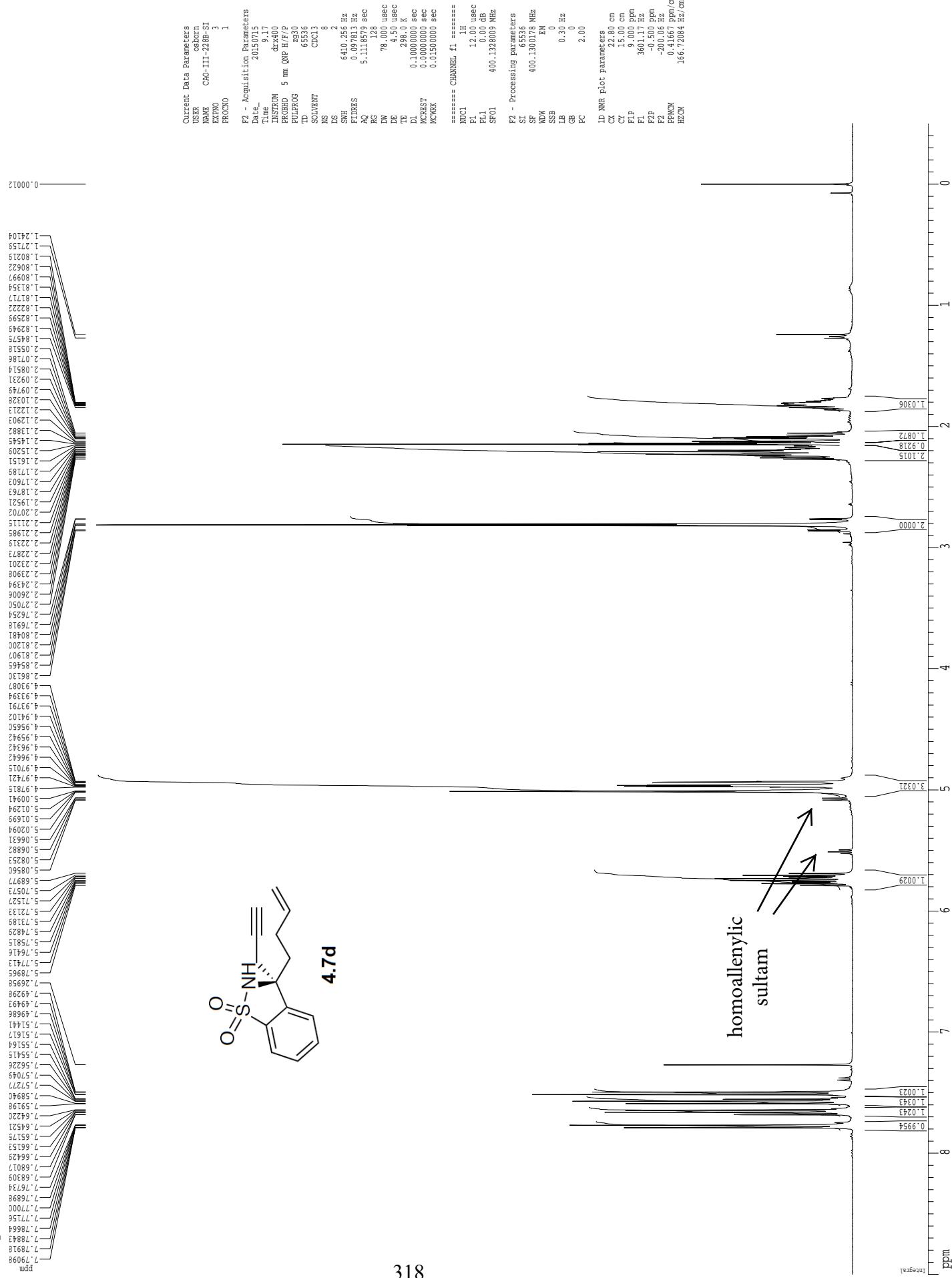


1H spectrum

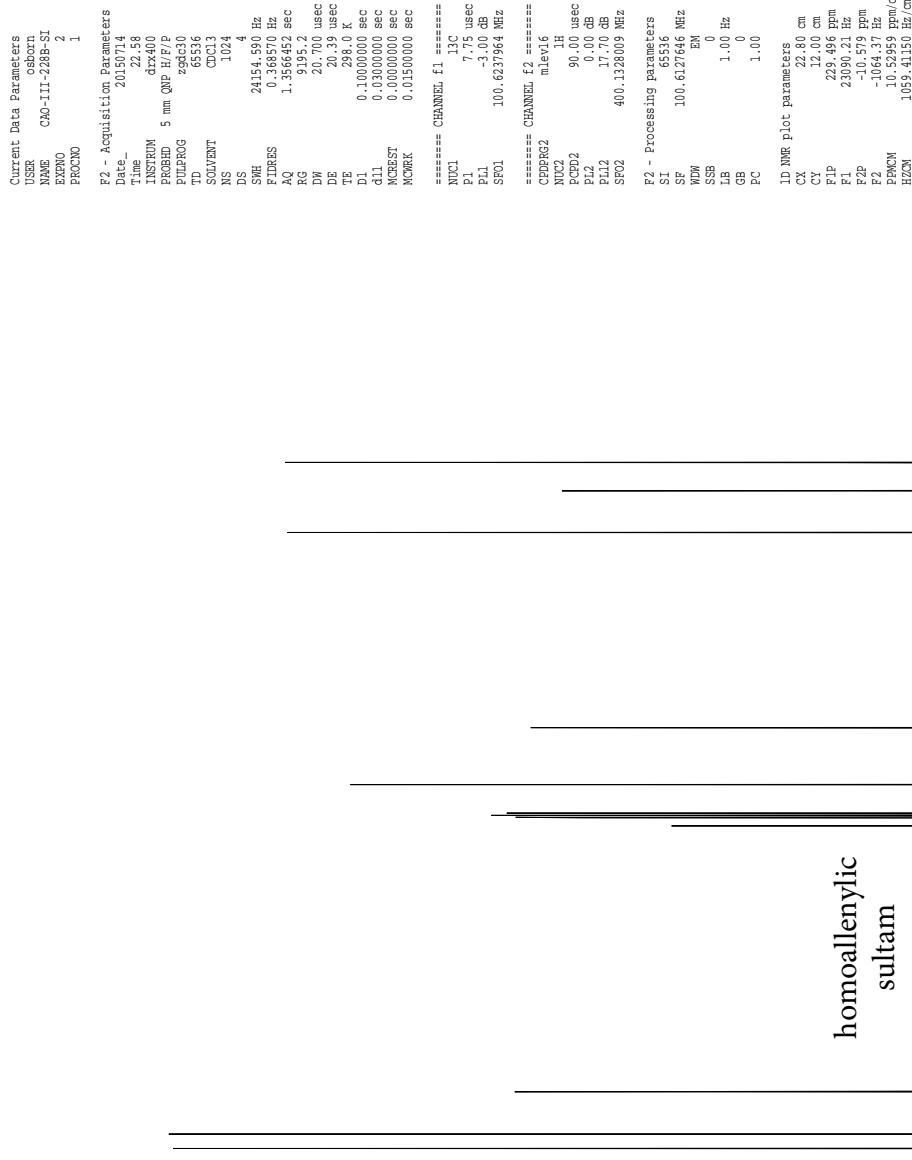
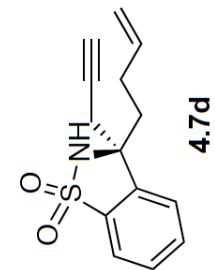
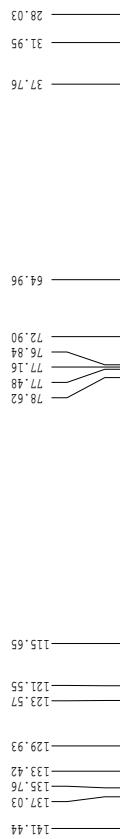
Z-restored spin-echo  $^{13}\text{C}$  spectrum with 1H decoupling

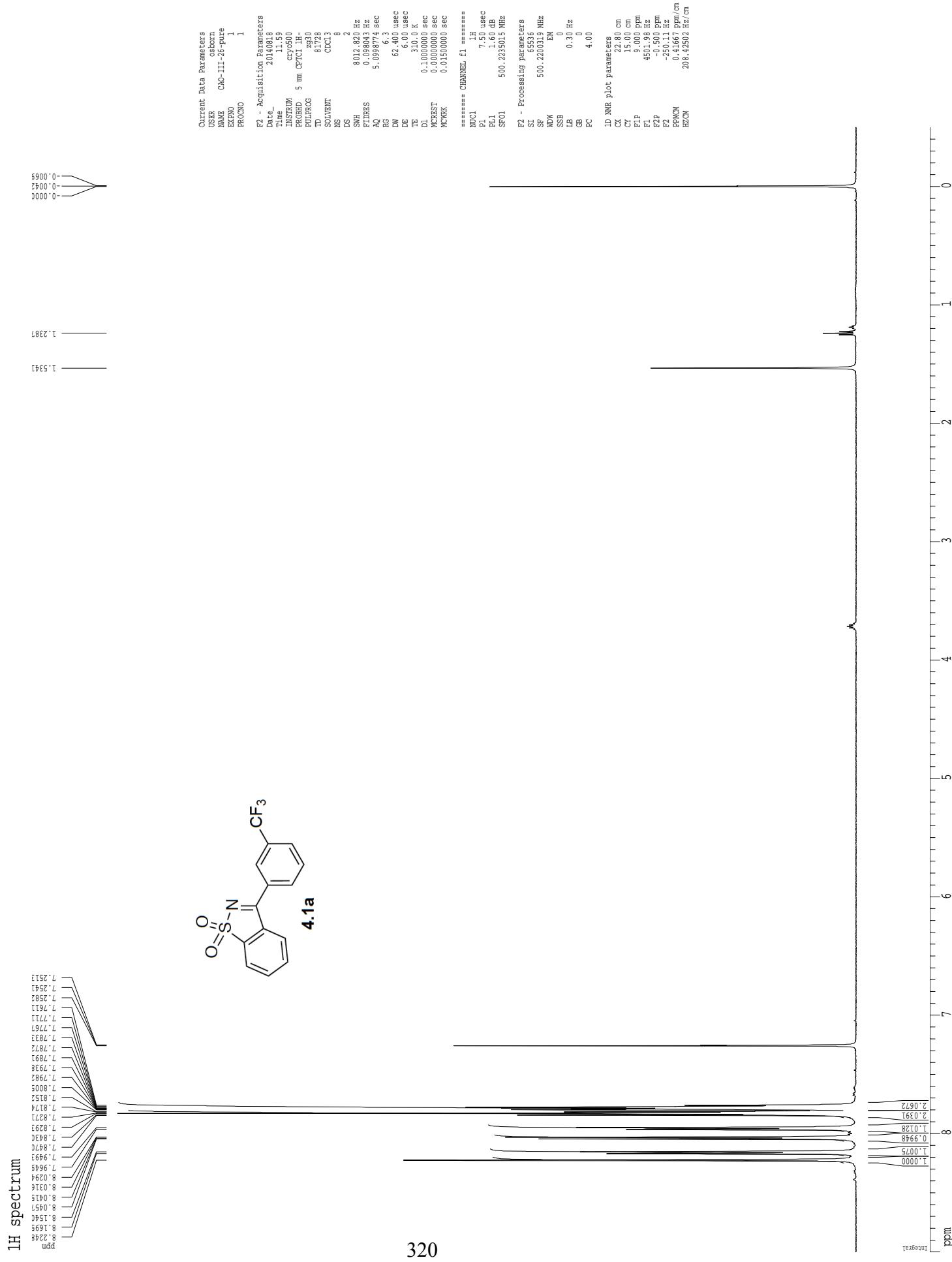


<sup>1</sup>H spectrum



z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

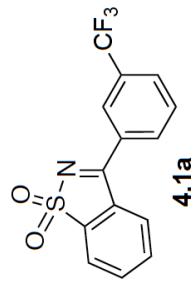




1H spectrum

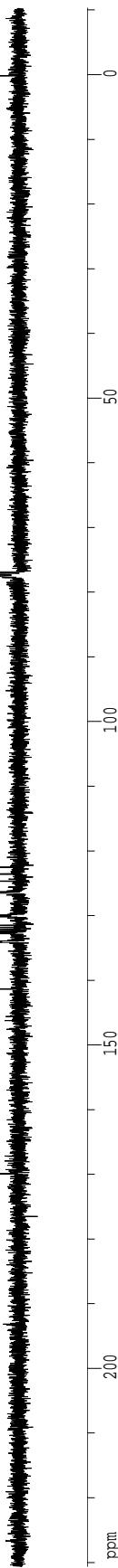


—169.88



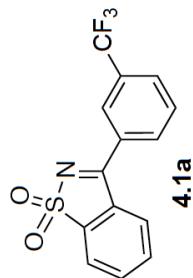
76.91  
77.16  
77.41

Current Data Parameters									
User	Job	Date	Time	Instrument	Prob	SpinBeads	3D�	PCD	PPC
GEO-III-26-june	2	2010-06-26	10:00:00	5 mm cryo300	5 mm	1H	3D�	300	300
EPICRO	1			PROB					
				SPINPDRG					
				TD					
				SOLVENT					
NS	331			DCD13					
DS	16								
SRH	30303.031	HZ							
FFIDRES	0.4428180	sec							
RG	1.0813940	sec							
DW	16.500	used							
DE	6.00	used							
TE	2500000.000	sec							
D1	0.0300000	sec							
D11	0.0300000	sec							
D16	0.0003000	sec							
D17	0.0003000	sec							
MOREST	0.0000000	sec							
WORK	0.0150000	sec							
P2	0.0000000	sec							
	31.00	used							
<b>==== CHANNEL f1 =====</b>									
NUC1	13C								
SP1	15.50	used							
SP2	500.00	used							
SP12	100.00	used							
SP10	20.00	used							
SP11	10.00	used							
SP13	3.00	used							
SP14	2.00	used							
SP15	1.00	used							
SP16	0.50	used							
SP17	0.30	used							
SP18	0.20	used							
SP19	0.10	used							
SP20	0.05	used							
<b>==== CHANNEL f2 =====</b>									
NUC2	1H								
CPDPFG2	0.5	used							
CPDPFG2	0.5	used							
FCP2	10.00	used							
FCP22	1.00	used							
FE1	1.68	48							
FE2	24.68	48							
FE11	20.00	used							
FE22	50.00	used							
PF1	50.00	used							
PF2	100.00	used							
<b>==== GRADIENT SINE =====</b>									
GRAD1	1.00	used							
GRAD2	1.00	used							
GR1	0.00	0							
GR2	0.00	0							
GR3	0.00	0							
GR4	0.00	0							
GR5	0.00	0							
GR6	0.00	0							
GR7	0.00	0							
GR8	0.00	0							
GR9	0.00	0							
GR10	0.00	0							
<b>==== GRADIENT SINE-PPC =====</b>									
PPC1	1.00	used							
PPC2	1.00	used							
PPC3	1.00	used							
PPC4	1.00	used							
PPC5	1.00	used							
PPC6	1.00	used							
PPC7	1.00	used							
PPC8	1.00	used							
PPC9	1.00	used							
PPC10	1.00	used							
<b>==== 1D NMR plot parameters =====</b>									
CX	22.80	cm							
CX	15.65	cm							
SD	231.637	ppm							
ND	289.068	ppm							
SSB	0	ppm							
WDW	-10.387	ppm							
GB	-12.387	ppm							
PC	0.00	ppm							
<b>==== 2D NMR processing parameters =====</b>									
EM	125.7804393	MHz							
EN	129.09935	MHz							
PPM	12.00	ppm							
HEOM	12.00	ppm							



19F spectrum with 1H decoupling

ddd



```

Current Data Parameters
USER          osborn
NAME         CHO-III-26-SI
EXPNO        2
PRCNO        1

P2 - Acquisition Parameters
Date_        20150530
Time_        19:49
INSTRUM     qxr400
PROBODIM   5 mm QNP H/F/P
PULPROG    zg3f90g130
TD          65536
SOLVENT      CDCl3
NS           31
DS            4
SWH         75187.969 Hz
ETW        1.147277 Hz
TDRES       0.435864 sec
AQ          3649.1
RG          6.650 usec
DW          9.46 usec
DE          298.0 K
TE          2.000000 sec
D1          0.030000 sec
D11         0.0000200 sec
D12         0.0000200 sec

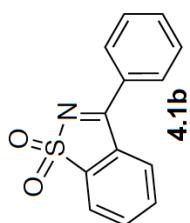
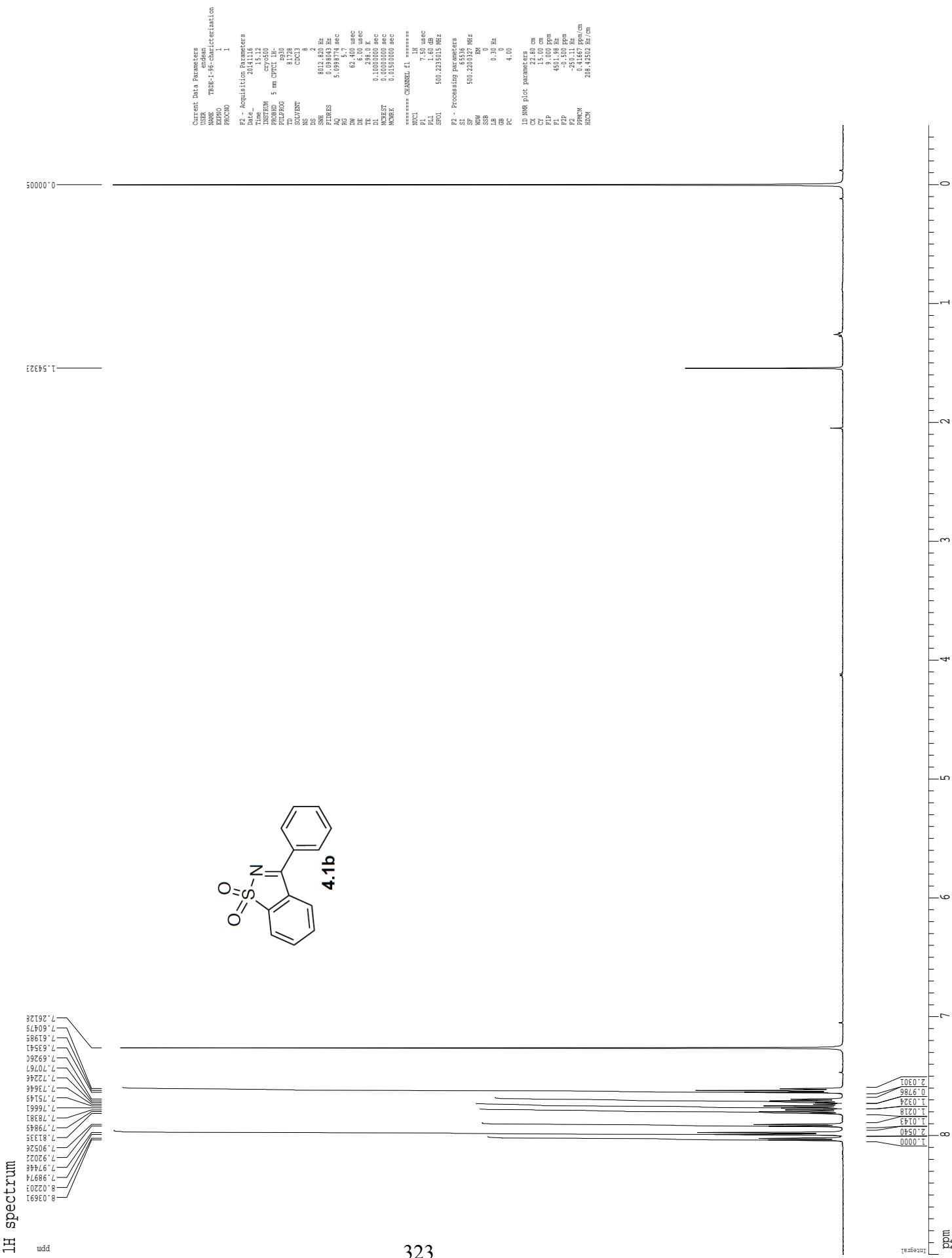
===== CHANNEL f1 =====
NUCL1       19F
P1          22.50 usec
PL1         -6.00 dB
SF1        376.4646491 MHz
SF2        400.1320007 MHz

===== CHANNEL f2 =====
CPDPRG2    waltz16
NUC2        1H
P2CD2      90.00 usec
PL2        120.00 dB
PLL2       17.70 dB
SF2        400.1320007 MHz

P2 - Processing parameters
SI          65536
SF        376.493351 MHz
WDW        EM
SSB          0
LB          0.30 Hz
GB          0
PC          1.00

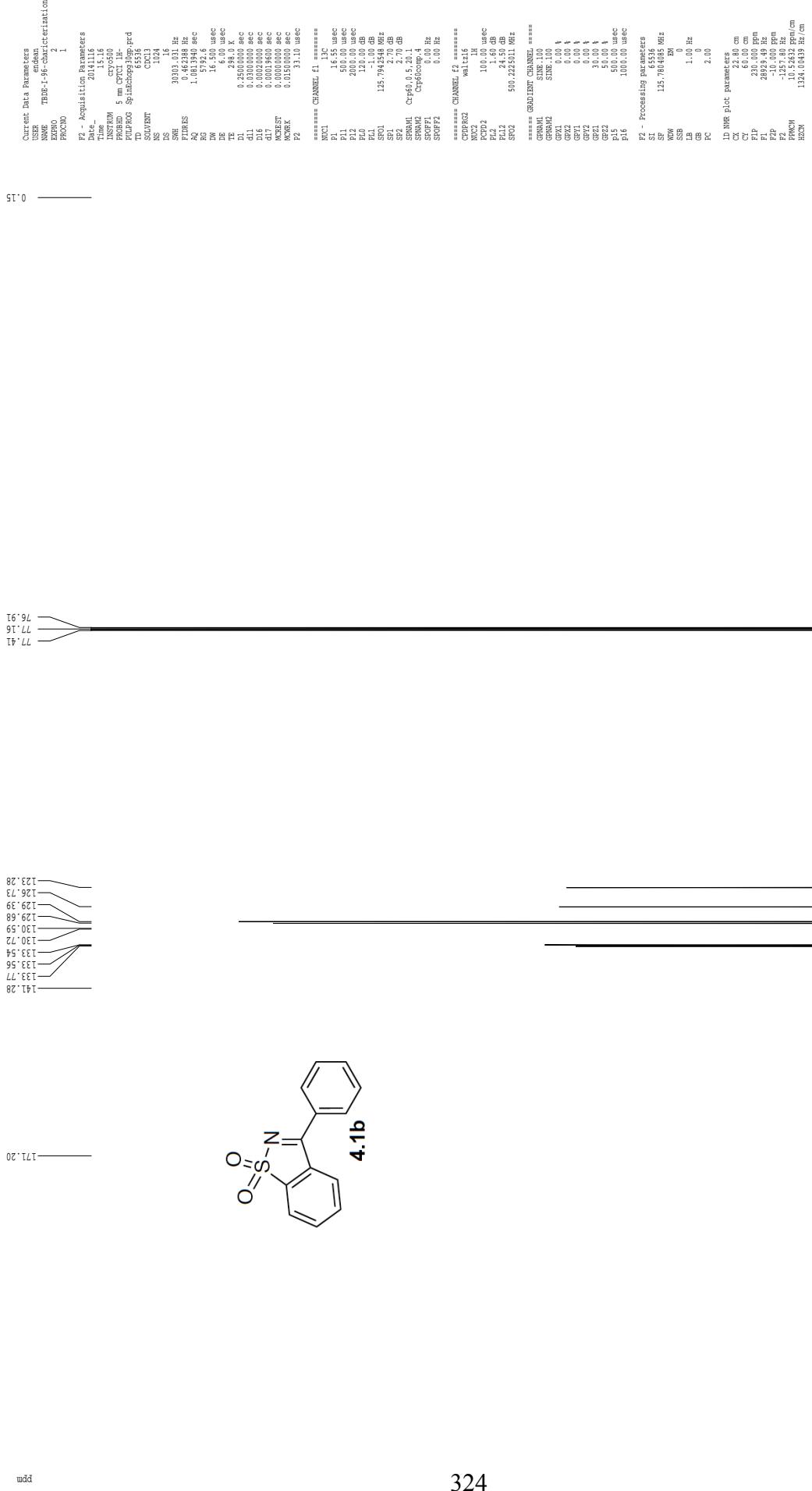
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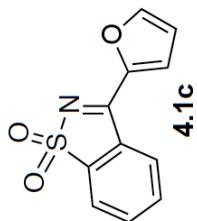
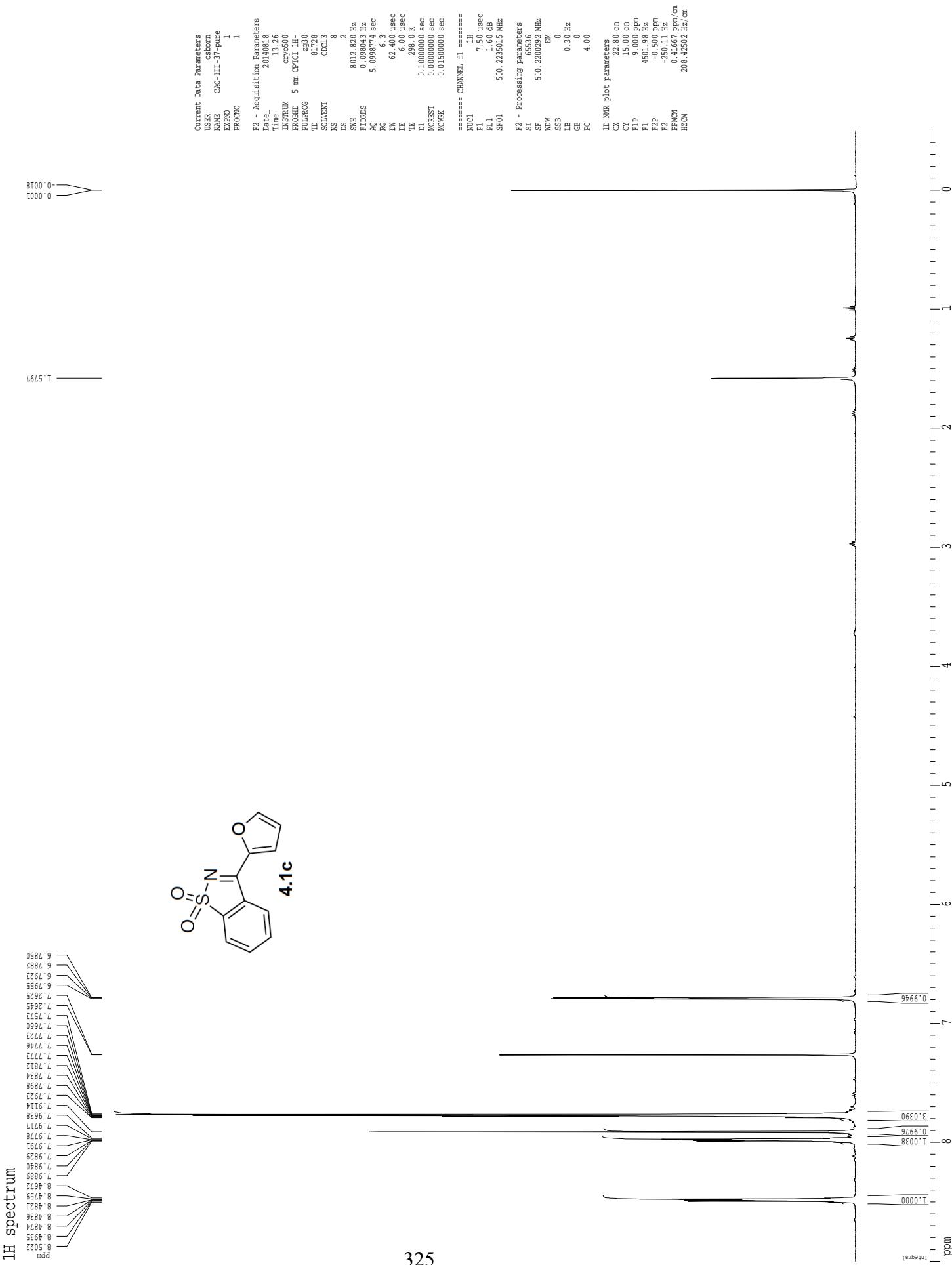




1H spectrum

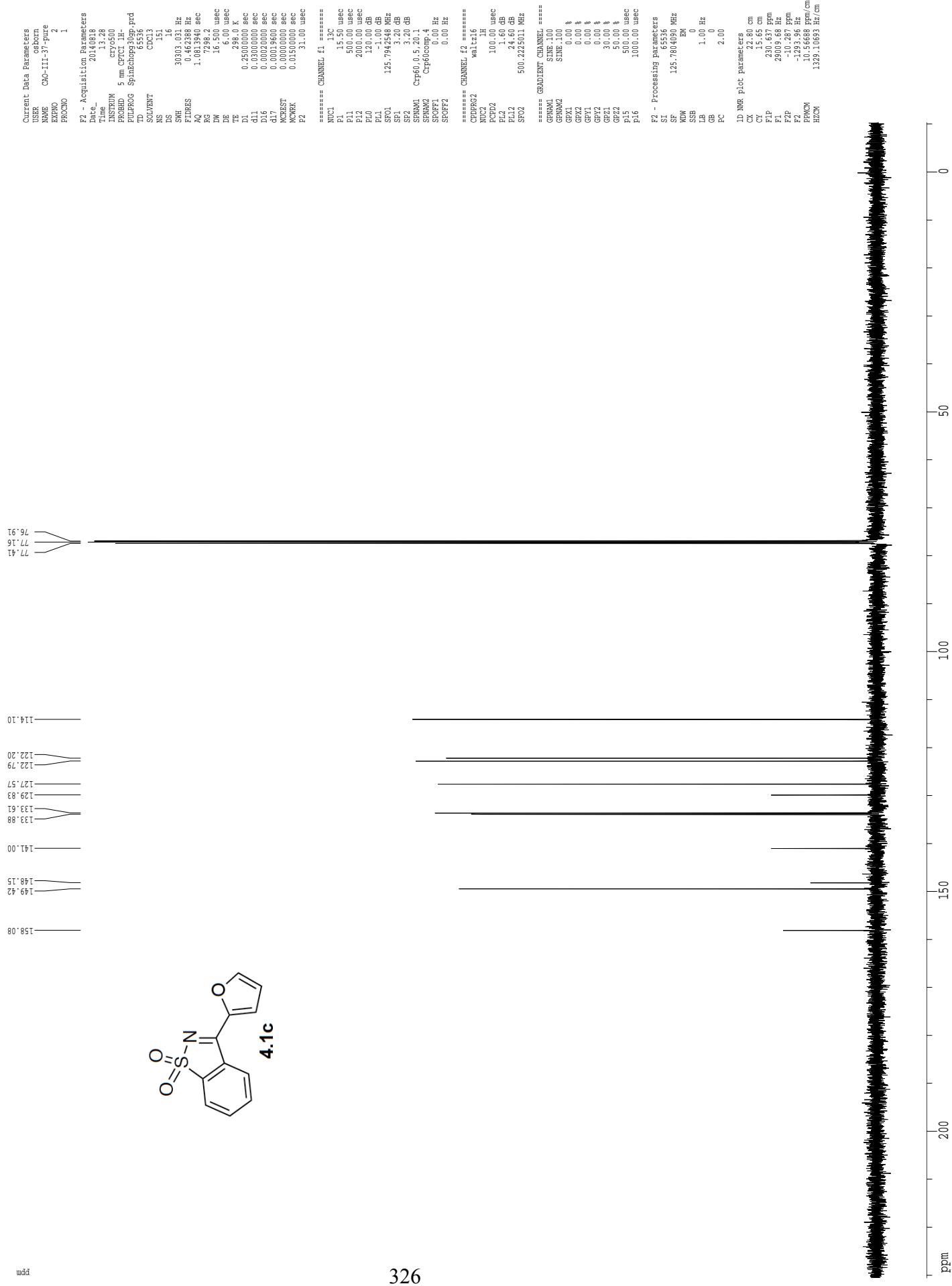
Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling



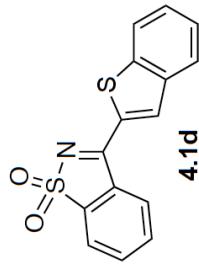
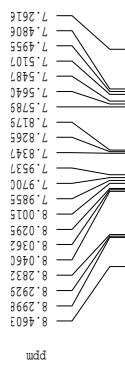


1H spectrum

Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling



1H spectrum



```

Current Data Parameters
USER          obourn
NAME          CMA-III-1b4-S1
EXNO          3
PROCNO        1

P2 - Acquisition Parameters
Date_        20100529
Time         17.41
INSTRUM     cryo500
PROBHD      5 mm CPTCI 1H-
PULPROG    TD
TD          81728
SOLVENT      CDCl3
NS           8
DS           2
SWH         8012.820 Hz
ETRIM       0.09804 Hz
TE          5.099871 sec
RG           6.3
DW           62.400 usec
DE           6.00 usec
TE          298.0 K
D1          0.1000000 sec
MCBEST      0.0000000 sec
MCRK        0.0150000 sec

=====
CHANNEL f1
=====
NUCL      1H
P1          7.50 usec
PL1        1.60 dB
SFOL      500.222501 MHz
SFQ1

F2 - Processing parameters
SI           65536
SF          500.2220009 MHz
WDW        EM
SSB          0
LB          0.30 Hz
GB          0
PC          4.00

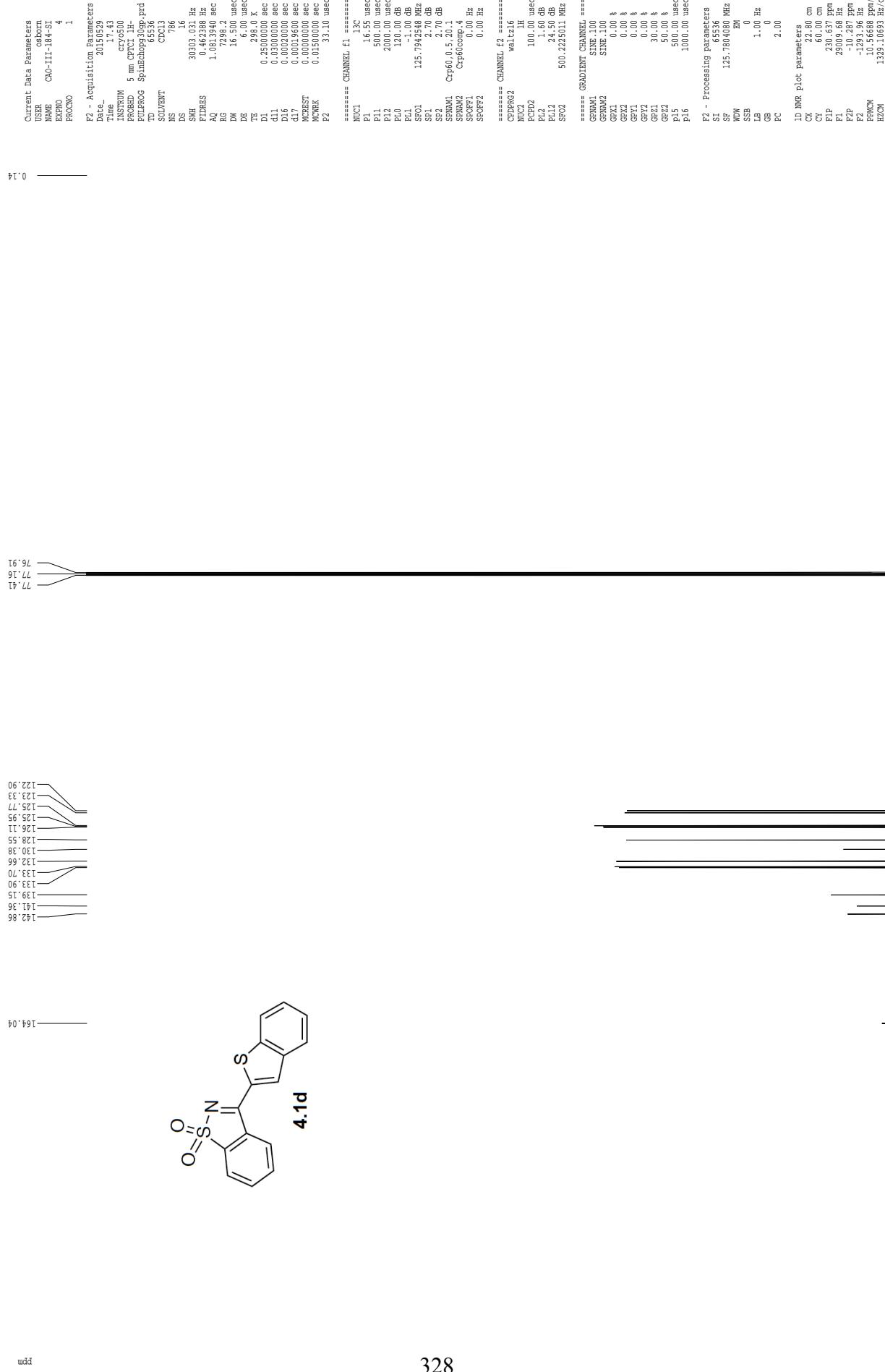
```

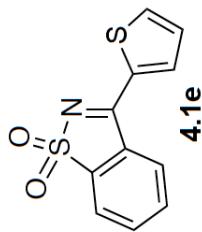
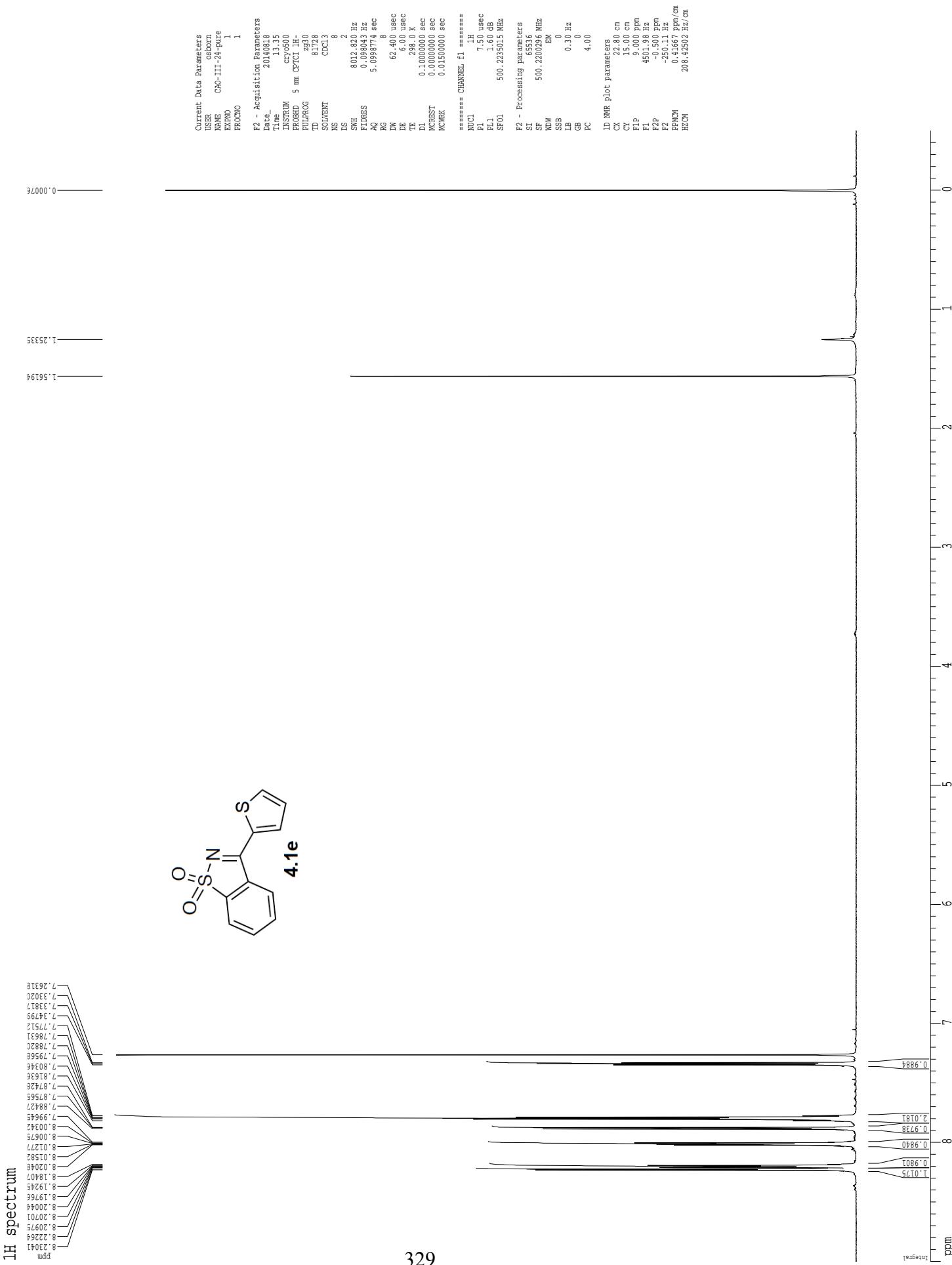
```

1D NMR plot parameters
CX           22.80 cm
CY           15.00 cm
F1P         9.000 Ppm
F1          451.98 Hz
F2P        -0.500 Ppm
F2         -250.11 Hz
PPM        0.11667 Ppm/cm
HZCM      208.412502 Hz/cm

```

Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling





1H spectrum

Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

ppm

0.14

163,12

141,40

135,76

135,27

133,78

133,96

133,59

133,49

132,23

123,16

123,16

125,68

129,16

130,49

133,16

133,16

77,43

77,16

76,91

**4.1e**

```

Current Data Parameters
USER          ssbmr
NAME          CHCl3-24-pure
EXNO          2
PRCNO         1

F2 - Acquisition Parameters
Date        20140818
Time        13:37
INSTRUM    INSPINW
PROBHD   5 mm CCP1 1H-
PULPROG  SpinEchoes3DPB.prd
TD        65536
SOLVENT    CS2
PC1       216
DS        16
TE        296.0 K
SF        3030.021 Hz
ETRIM      0.053238 Hz
AQ        1.00000 sec
RG        16.500 usec
DW        16.500 usec
D1        0.260000 sec
D11       0.000000 sec
D16       0.0002000 sec
D17       0.0001960 sec
MC1EST    0.000000 sec
MCNMRK   0.0150000 sec
P2K2      31.00 usec

=====
CHANNEL F1 =====
NUC1      13C
P1        15.50 usec
P11       50.00 usec
P12       200.00 usec
PL0       120.00 dB
PL1       -1.00 dB
SP01     125.794258 MHz
SP1       3.20 dB
SP2       3.20 dB
SPR001   Crp60,0.5,20.1
SPR002   Crp60,0.5,20.1
CPDPR04  Crp60,0.5,20.1
SP0FF1   0.00 Hz
SP0FF2   0.00 Hz

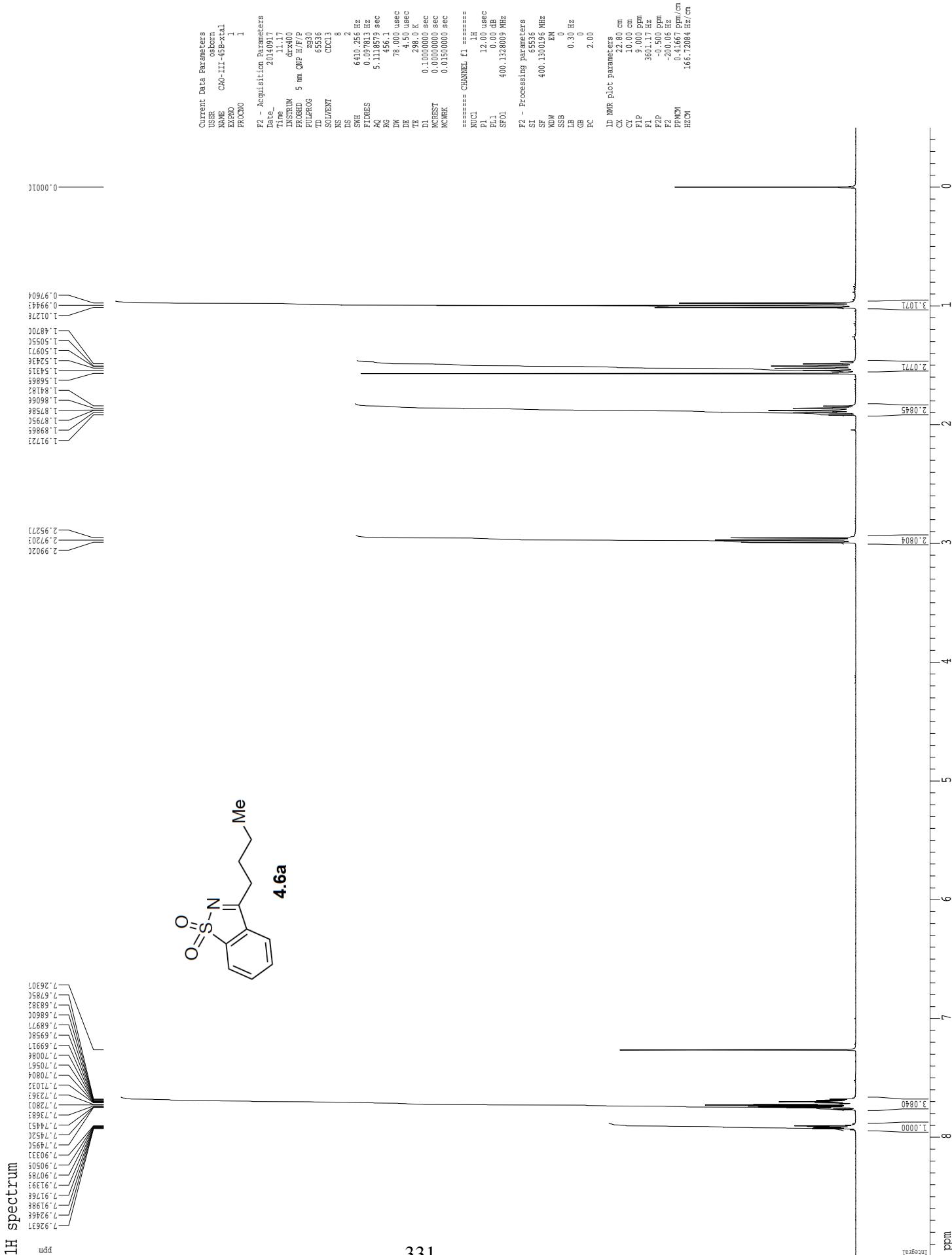
=====
CHANNEL F2 =====
NUC2      1H
CPDPG2   waltz16
P2P2      10.00 usec
P12       1.60 dB
PL12      500.2225011 MHz
SF02      500.2225011 MHz

=====
GRADIENT CHANNEL =====
GP0M1     SINE,100
GP0M2     SINE,100
GPX1      0.00 %
GPX2      0.00 %
GPY1      0.00 %
GPY2      0.00 %
GPZ1      30.00 %
GPZ2      50.00 %
GP15     500.00 usec
GP16     1000.00 usec

F2 - Processing Parameters
SI        65536
SF        125.780405 MHz
WDW      EM
SSB      0
LB        1.00 Hz
GB      0
PC        2.00

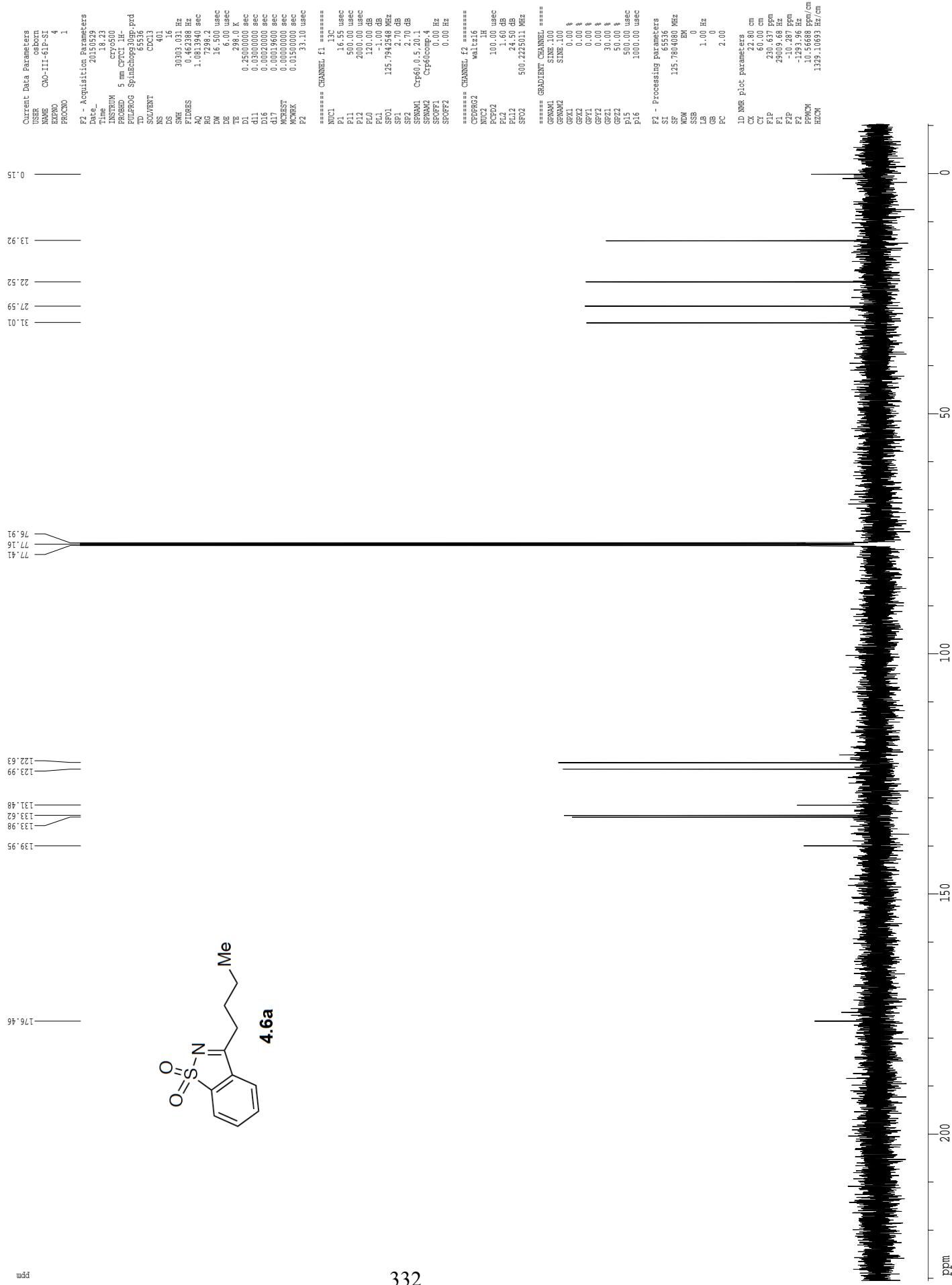
1D NMR Plot parameters
CX        22.80 cm
CY        35.00 cm
F1P      230.637 ppm
F1       2300.968 Hz
F2P      -10.287 ppm
F2       -10.287 Hz
PPCM    10.56688 ppm/cm
HZCM   1323.10693 Hz/cm

```

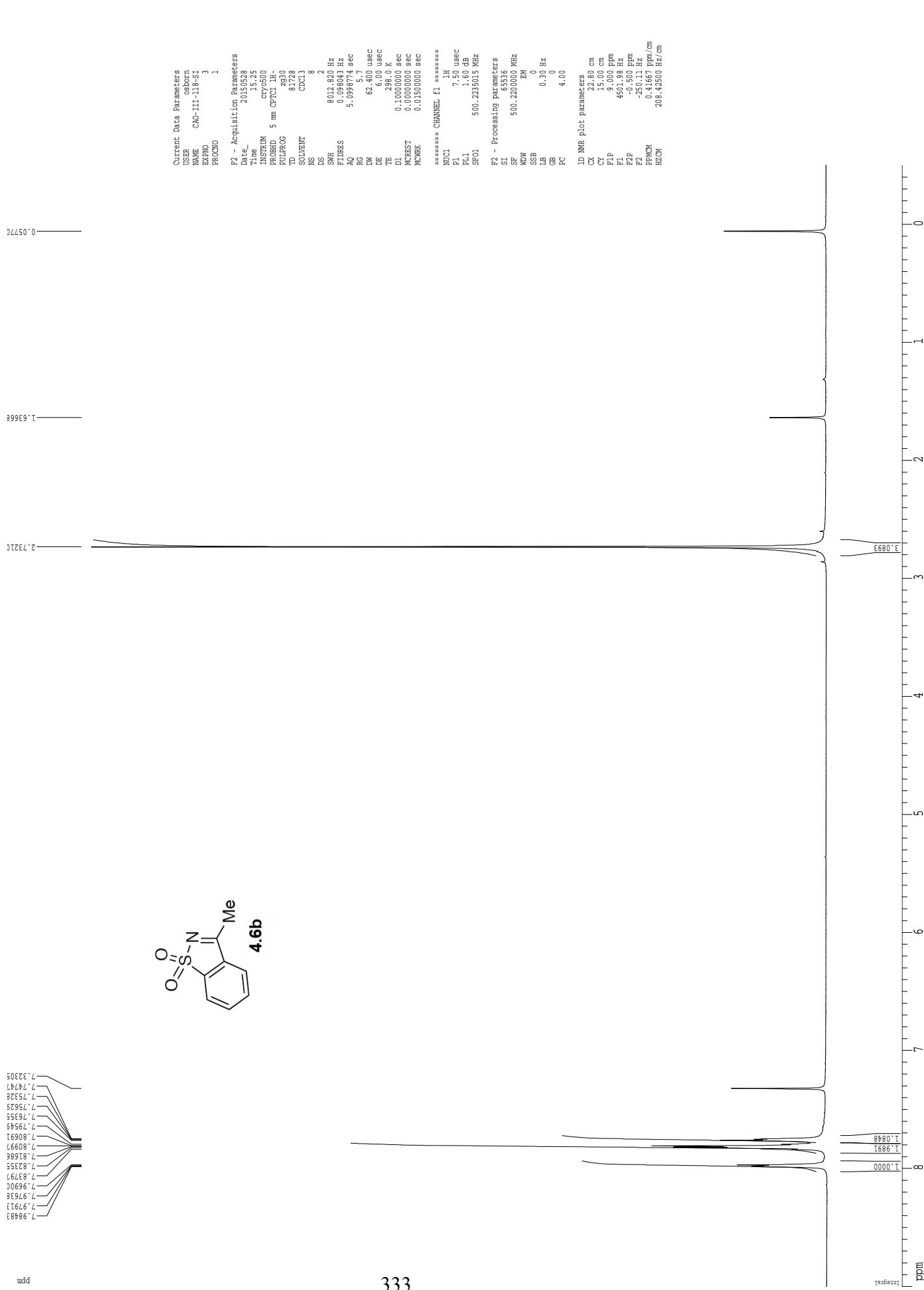
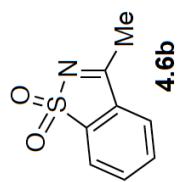


1H spectrum

Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

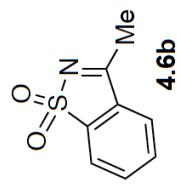
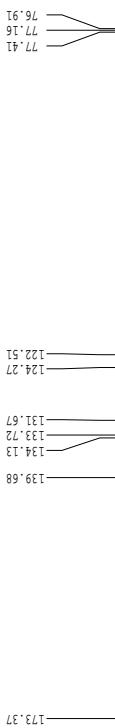


1H spectrum



Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

ddd



```

Current Data Parameters
USER   usborn
NAME  Cb-111-118-51
EXNO   1
PROCNO 1

F2 - Acquisition Parameters
Date_ 20150528
Time_ 15.18
INSTRUM cryo500
PROBHD 5 mm CCP1 1H-
PULPROG SpinEchoes300P.prd
TD    65536
SOLVENT
NUC1 13C
PC1 231
DS    16
SWH 3033.031 Hz
ETRIM 0.46338 Hz
AQ 1.00390 sec
RG 3112
TE   296.0 usec
D1   0.260000 sec
D11  0.000000 sec
D16  0.002000 sec
D17  0.001960 sec
MC1 0.000000 sec
MC2T 0.000000 sec
MCNTK 0.0150000 sec
P2K 33.10 usec

=====
CHANNEL F1 =====
NUC1 13C
P1   16.55 usec
P11  50.00 usec
P12  200.00 usec
PL0   12.00 dB
PL1   -1.00 dB
SP01  125.794258 MHz
SP1   2.70 dB
SP2   2.70 dB
SPW01 Crp60.0,5,20.1
SPW02 Crp60.0,5,20.1
CPD1 0.00 Hz
SPCF1 0.00 Hz
SPCF2 0.00 Hz

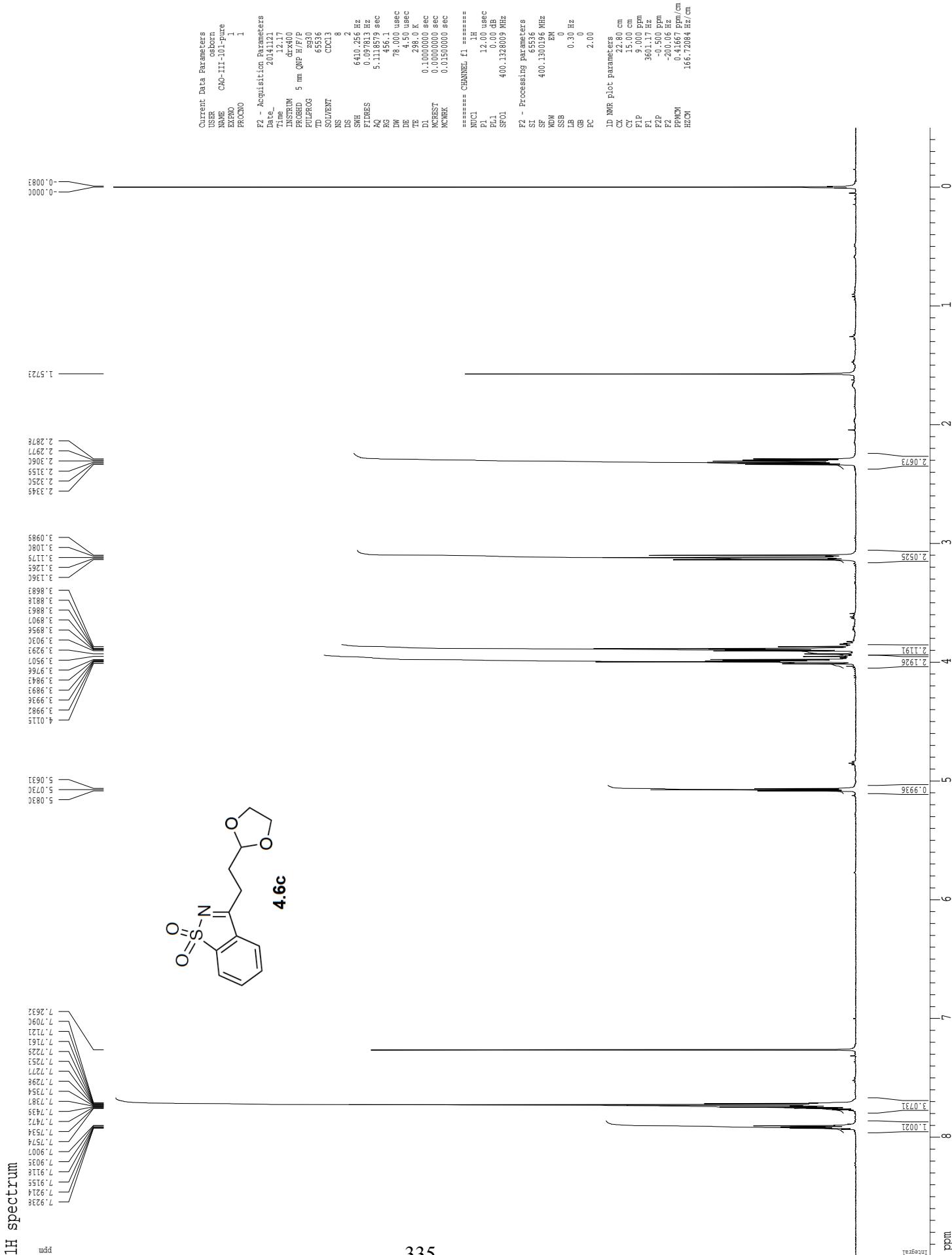
=====
CHANNEL F2 =====
CPDPRG2
NUC2 1H
P2D2 100.00 usec
P2L2 1.60 dB
PL12 24.50 dB
SF02 500.2225011 MHz

=====
GRADIENT CHANNEL =====
GP0M1 SINE.100
GP0M2 SINE.100
GPX1 0.00 %
GPX2 0.00 %
GPY1 0.00 %
GPY2 0.00 %
GPZ1 30.00 %
GPZ2 50.00 %
GP15 500.00 usec
GP16 1000.00 usec

F2 - Processing Parameters
SI    65536
SF   125.780413 MHz
WDW   EM
SSB   0
LB    1.00 Hz
GB    0
PC    2.00

1D NMR Plot parameters
CX    22.80 cm
CY    15.65 cm
F1P   230.67 ppm
F1    230.9.68 Hz
F2P   -10.287 ppm
F2    -129.3.16 Hz
PPCM  10.5668 ppm/cm
HZDW  1323.1063 Hz/cm

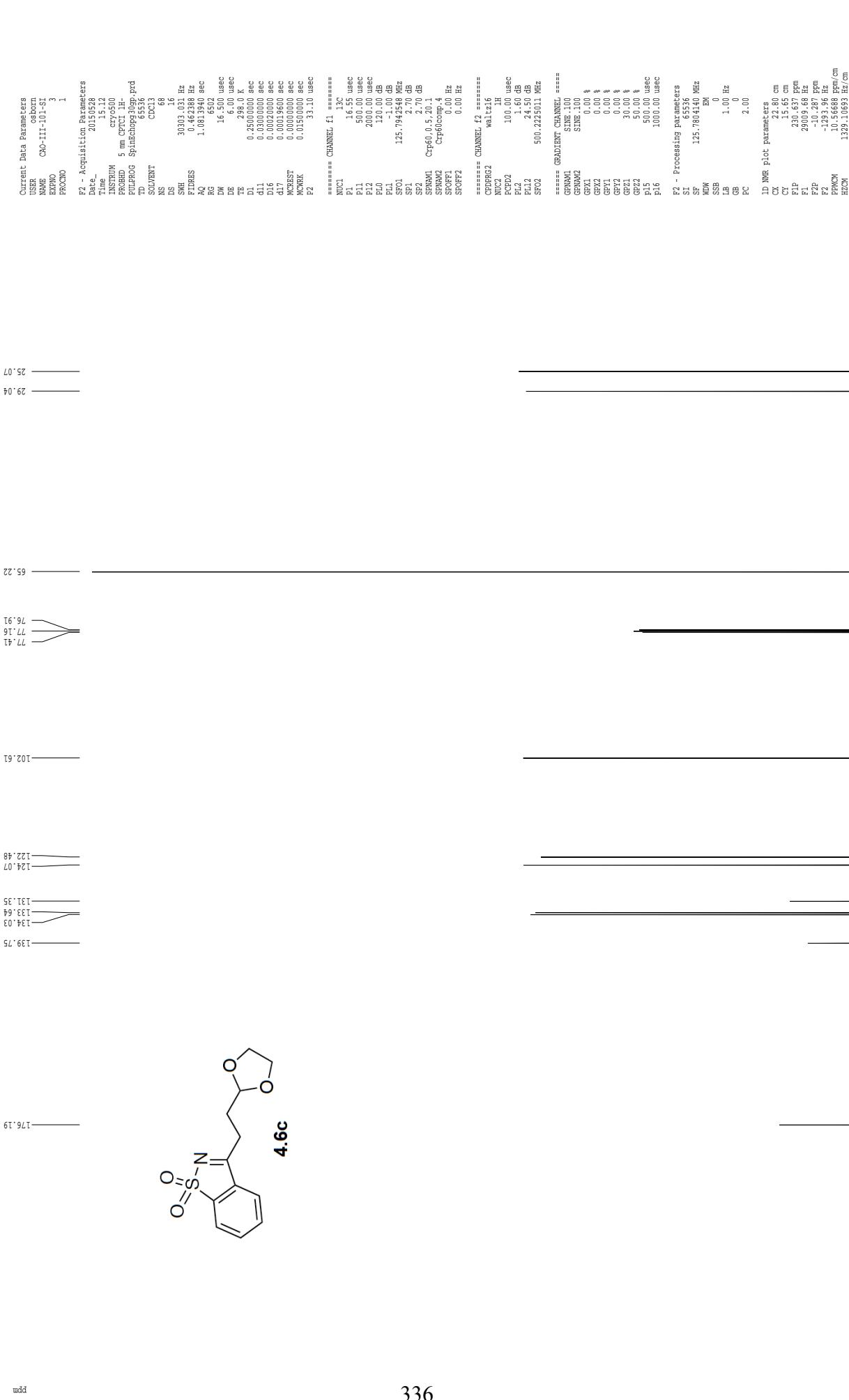
```

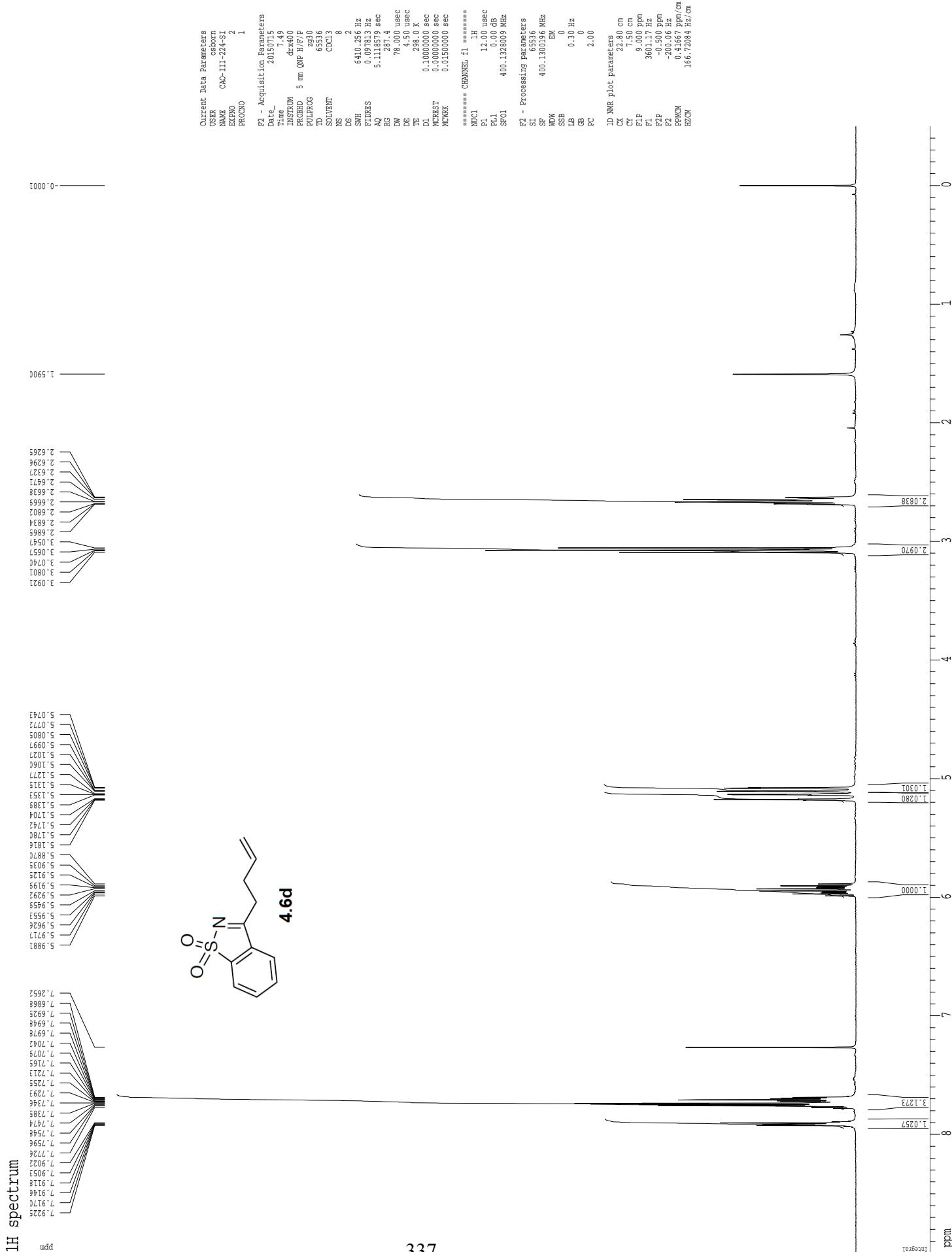


1H spectrum

Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

ppm





<sup>13</sup>C spectrum with 1H decoupling

ppm

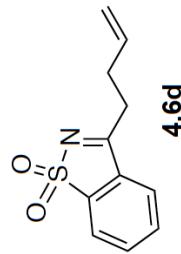
29.20  
30.62

76.84  
77.16  
77.48

116.71

122.61  
123.98

131.33  
133.72  
134.06  
135.91  
139.89



```

Current Data Parameters
USER          osborn
NAME         CNA-III-224-SI
EXNO          3
PRCNO         1

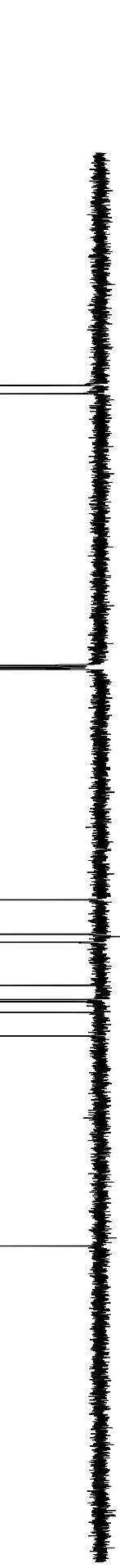
P2 - Acquisition Parameters
Date        20150715
Time        7.57
INSTRUM   QNP H/F/P
PROBHD    5 mm
PULPROG  zg3g30
TD        65536
SOLVENT   CDCl3
NS           1024
DS            4
SWH       24154.590 Hz
ETRATES   0.368570 Hz
AQ        1.3566452 sec
RG        13004
DW        20.700 usec
DE        20.39 usec
TE        298.0 K
D1        0.1000000 sec
d11      0.0300000 sec
MTCRST   0.0000000 sec
MCRK     0.0150000 sec

=====
CHANNEL f1 =====
NUC1        13C
PL1        7.75 usec
PL1        -3.00 dB
SFOL     100.6237964 MHz

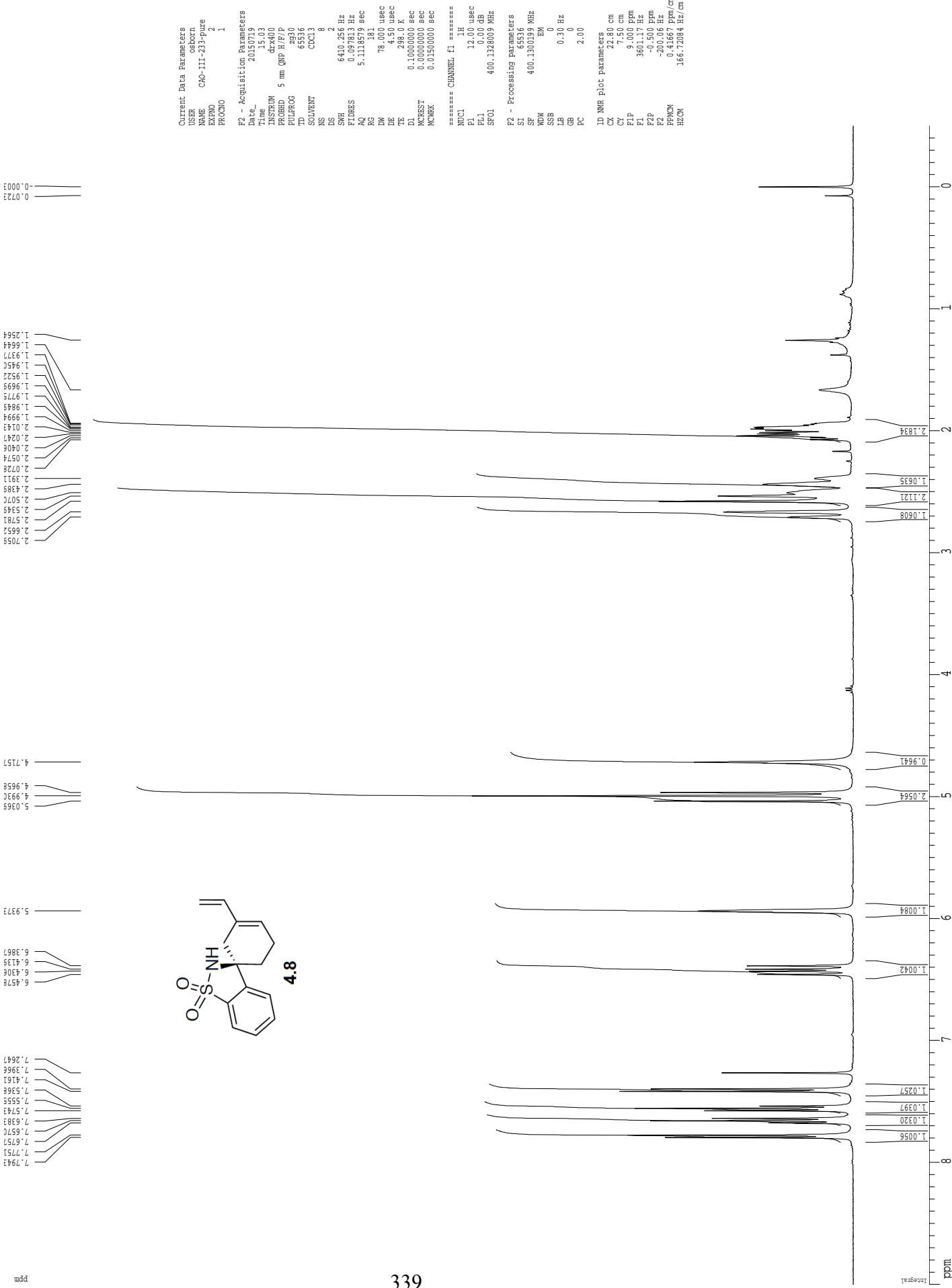
=====
CHANNEL f2 =====
CPDPG2      mlyr16
NUC2        1H
PL2        90.00 usec
PL2        0.00 dB
PL12      17.70 dB
SFQ2     400.1328009 MHz

P2 - Processing parameters
SI        65536
SF      100.6127606 MHz
WDW        EM
SSB          0
LB        1.00 Hz
GB          0
PC        1.00

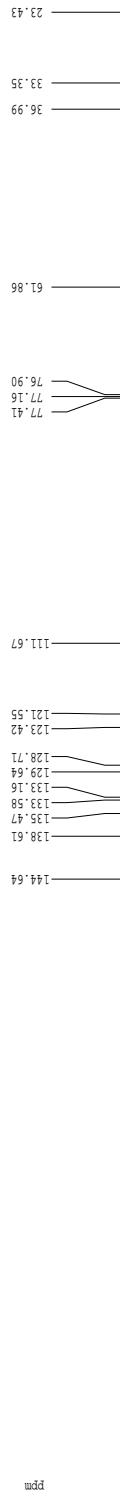
```



1H spectrum



Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling



```

Current Data Parameters
USER          usborn
NAME         Cb-111-23-51
EXNO          1
PROCNO        1
P1CNUO

F2 - Acquisition Parameters
Date        2015/07/9
Time       16.12
INSTRUM   INSPINW
PROBHD   5 mm CCP1 1H-
PULPROG  SpinEchoes3DP.prd
TD        65536
SOLVENT      CS2
PCP1      100
PCP2      100
DW        16.00 usec
TE        296.0 K
D1        0.260000 sec
d11      0.000000 sec
D16      0.002000 sec
d17      0.001500 sec
MC1EST    0.000000 sec
MCNMRK   0.015000 sec
P2K2      33.10 usec

=====
CHANNEL F1 =====
NUCL1     13C
P1        16.55 usec
P12      50.00 usec
P10      200.00 usec
PL0      120.00 dB
PL1      -1.00 dB
SP01     125.794258 MHz
SP1      2.70 dB
SP2      2.70 dB
SPR0M1   Crp60.0,5,20.1
SPR0M2   Crp60.0,5,20.1
CPD1     0.00 Hz
SPCPFF2  0.00 Hz

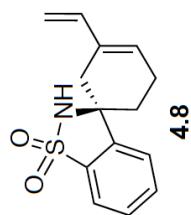
=====
CHANNEL F2 =====
NUCL2     1H
CPDPRG2  waltz16
P2CNUO
P2D2      10.00 usec
P2L2      1.60 dB
PL12      24.50 dB
SF02     500.2225011 MHz

=====
GRADIENT CHANNEL =====
GP0M1     SINE.100
GP0M2     SINE.100
GPX1      0.00 %
GPX2      0.00 %
GPY1      0.00 %
GPY2      0.00 %
GPZ1      30.00 %
GPZ2      50.00 %
P15      50.00 usec
P16      100.00 usec

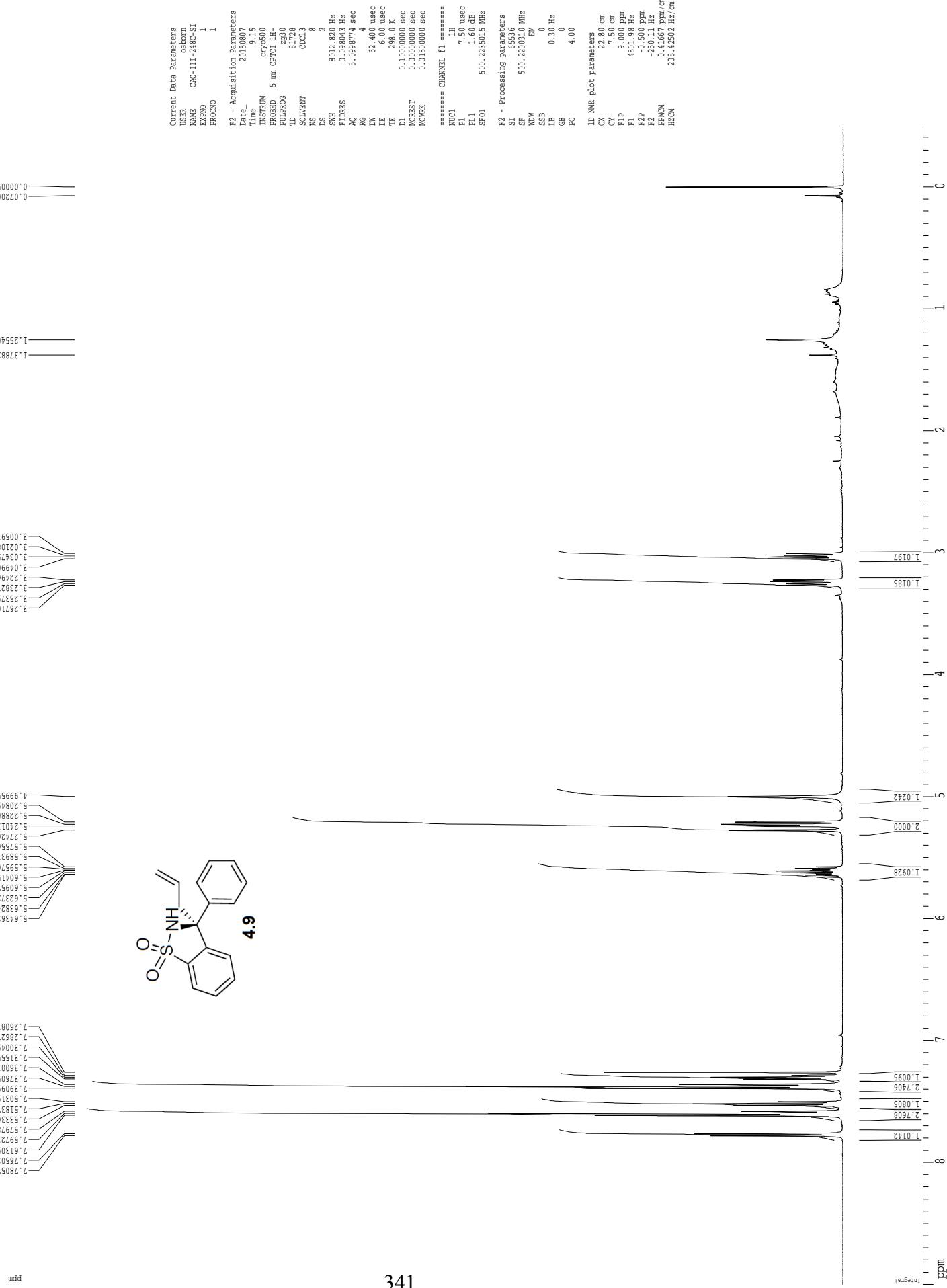
F2 - Processing Parameters
SI        65536
SF        125.7804122 MHz
WDW      EM
SSB      0
LB        1.00 Hz
GB      0
PC        2.00

1D NMR Plot Parameters
CX        22.80 cm
CY        15.65 cm
F1P      230.677 ppm
F1       230.958 Hz
F2P      -10.287 ppm
F2       -10.287 Hz
PPCM    10.56688 ppm/cm
HZCM   1325.10693 Hz/cm

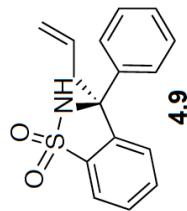
```

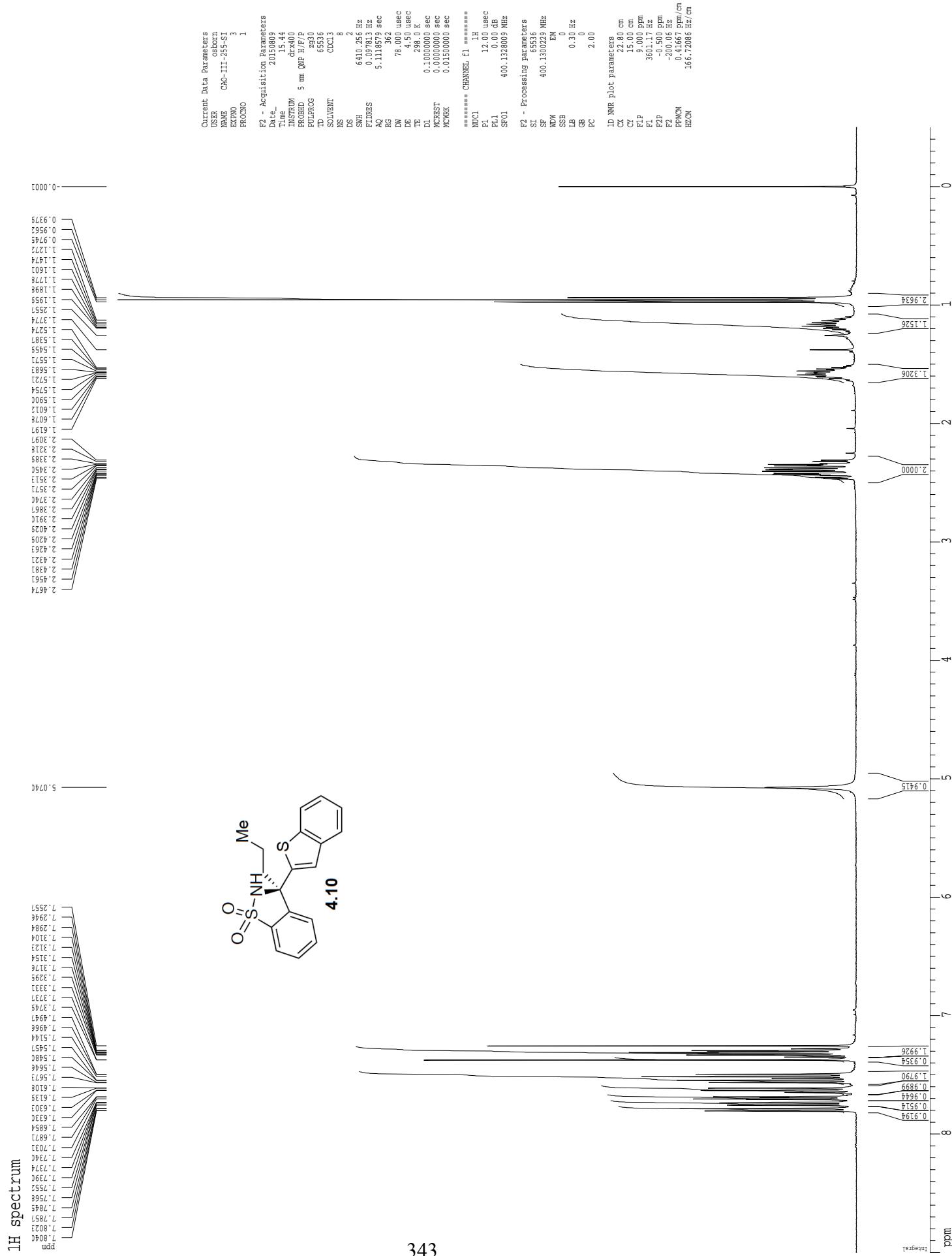


1H spectrum

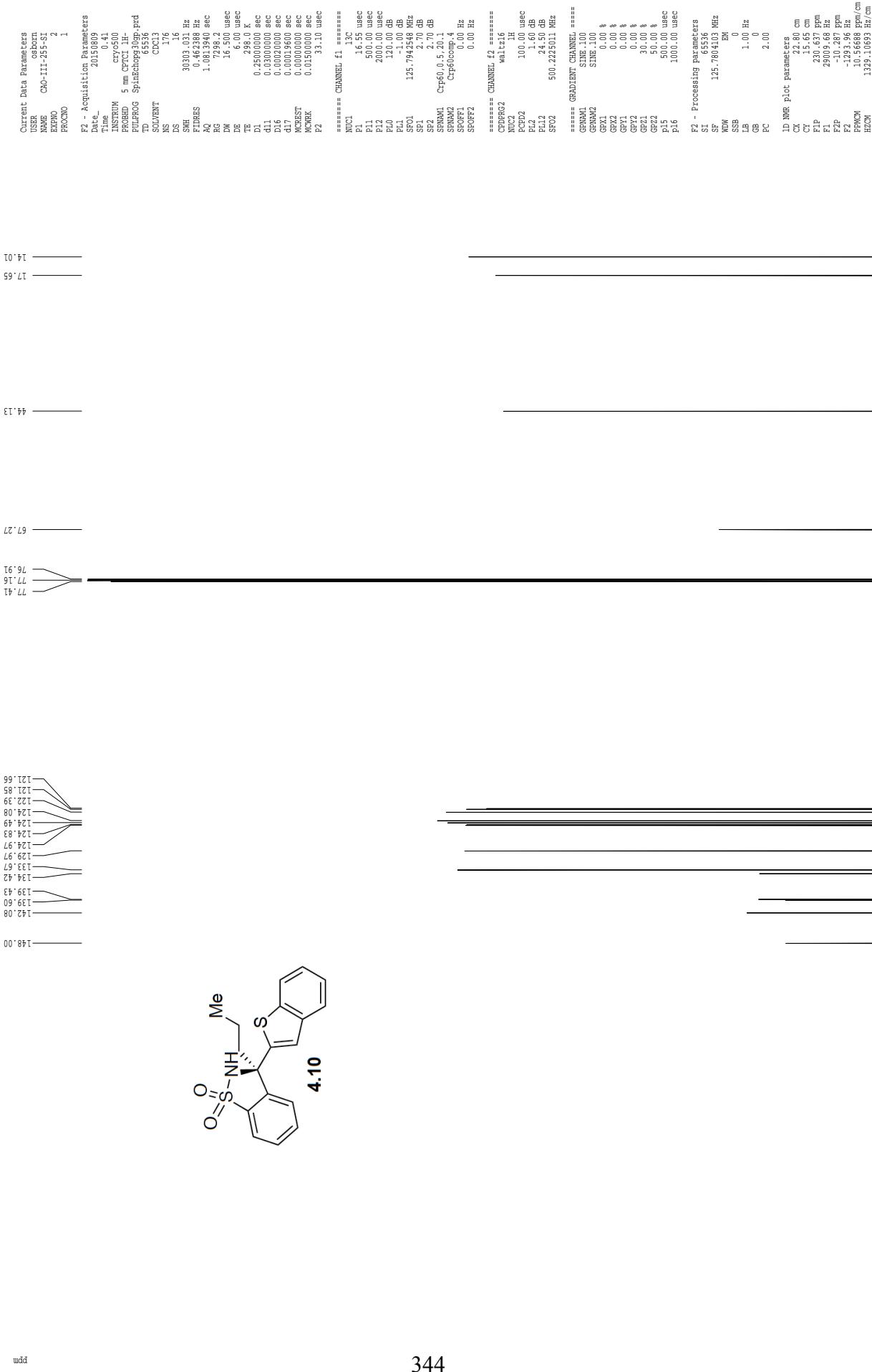


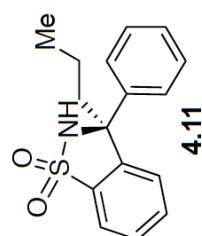
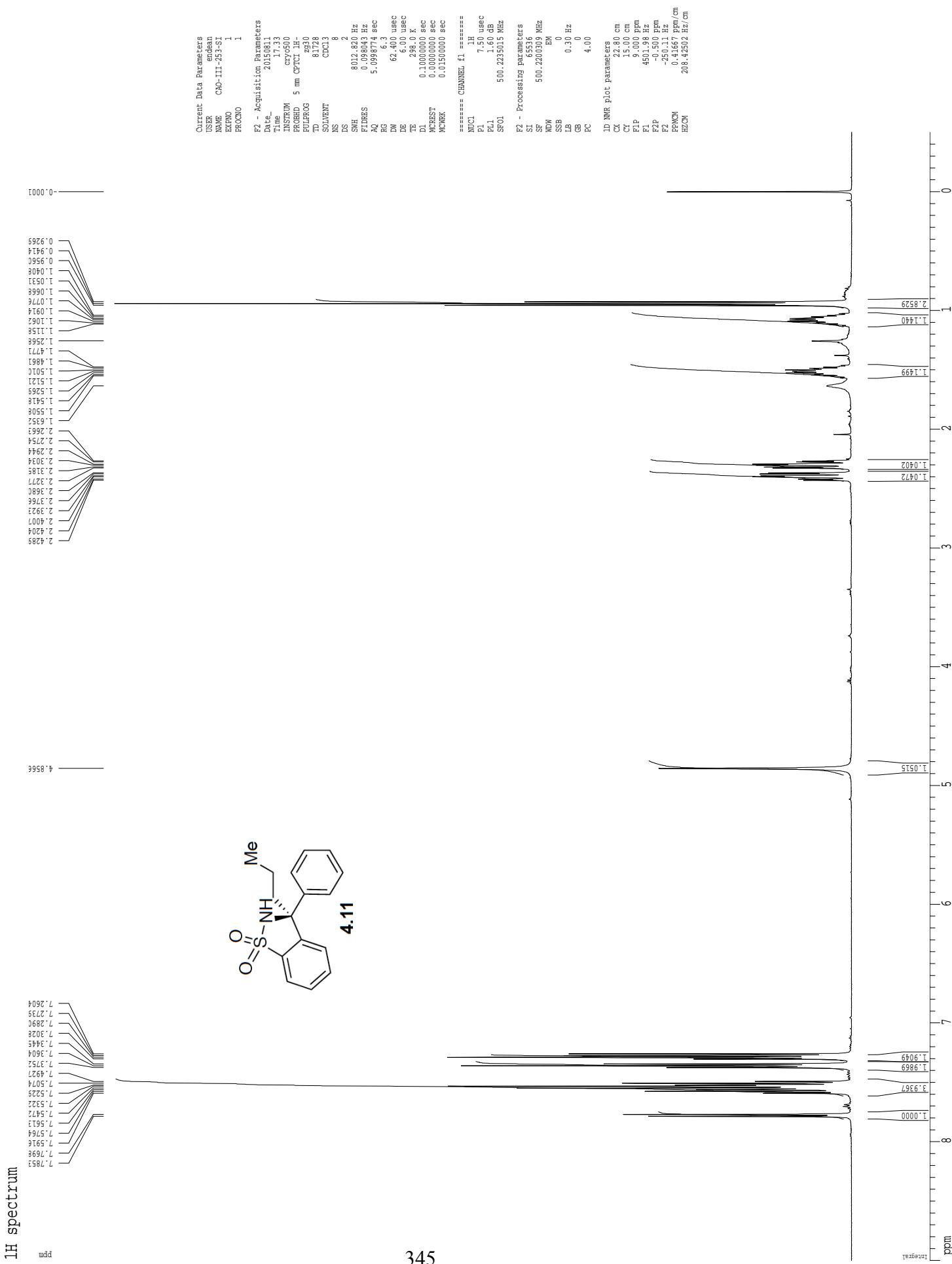
Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $1\text{H}$  decoupling



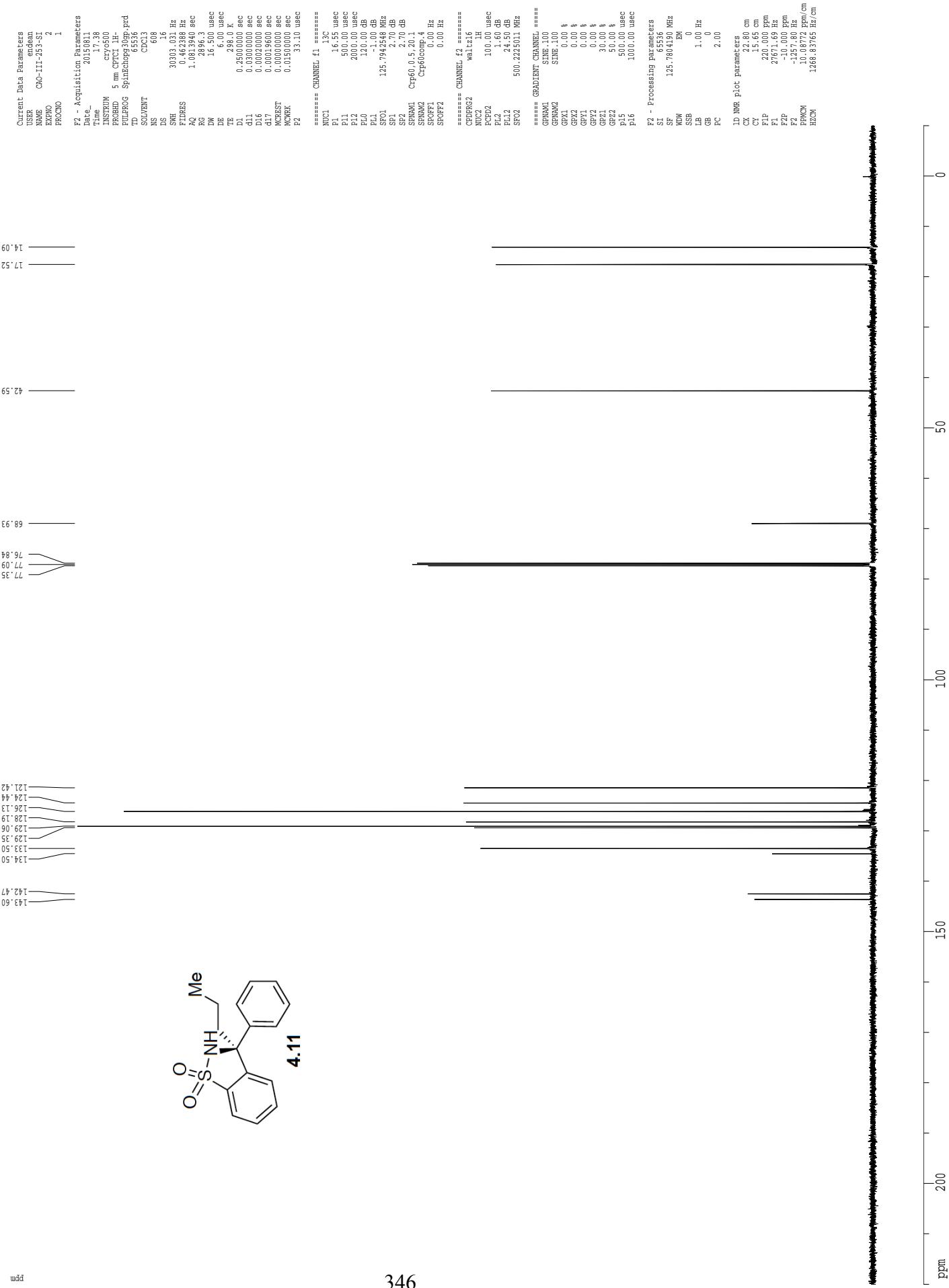


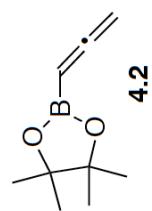
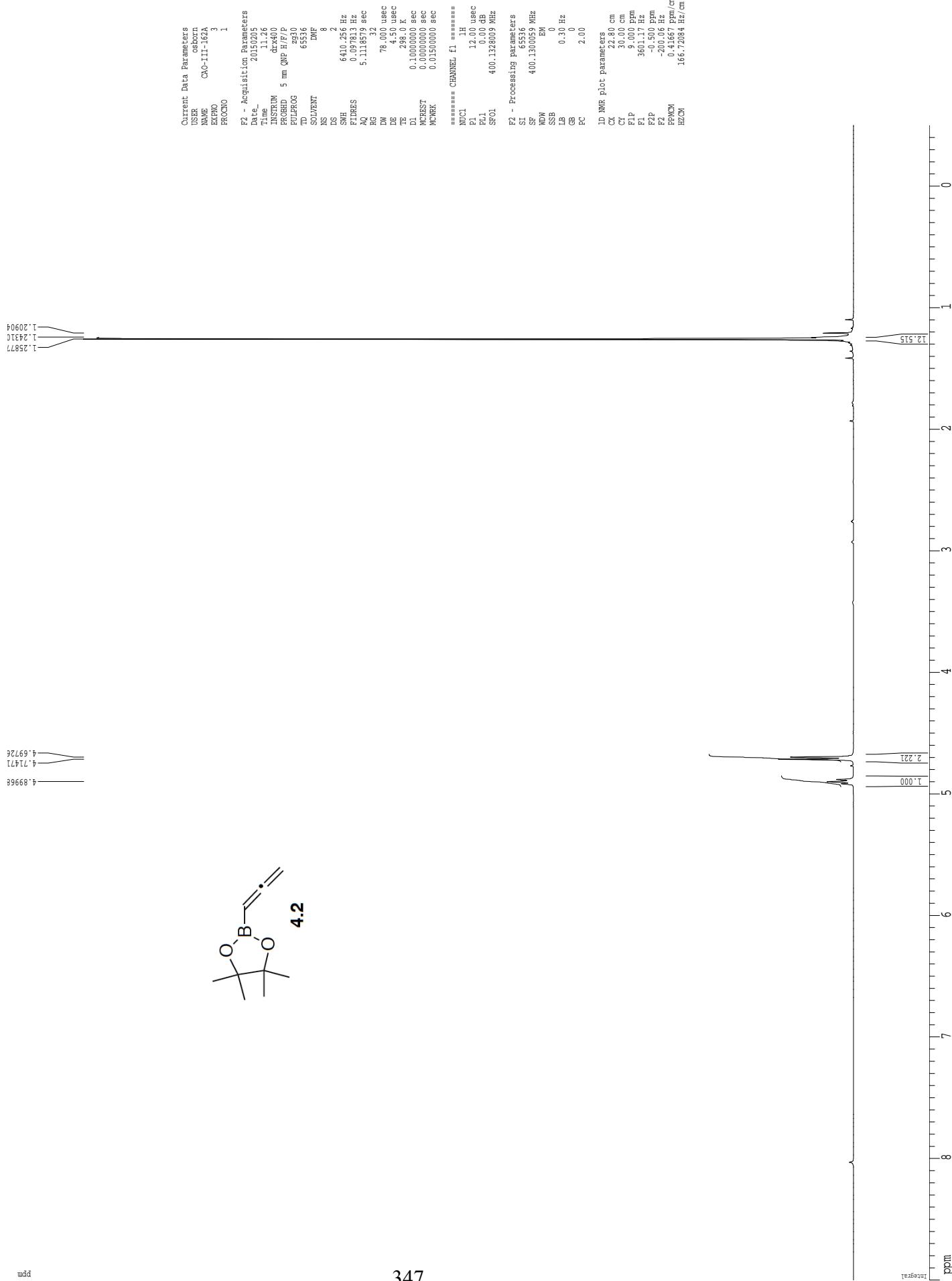
Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling





Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

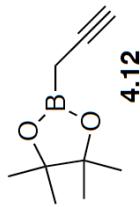
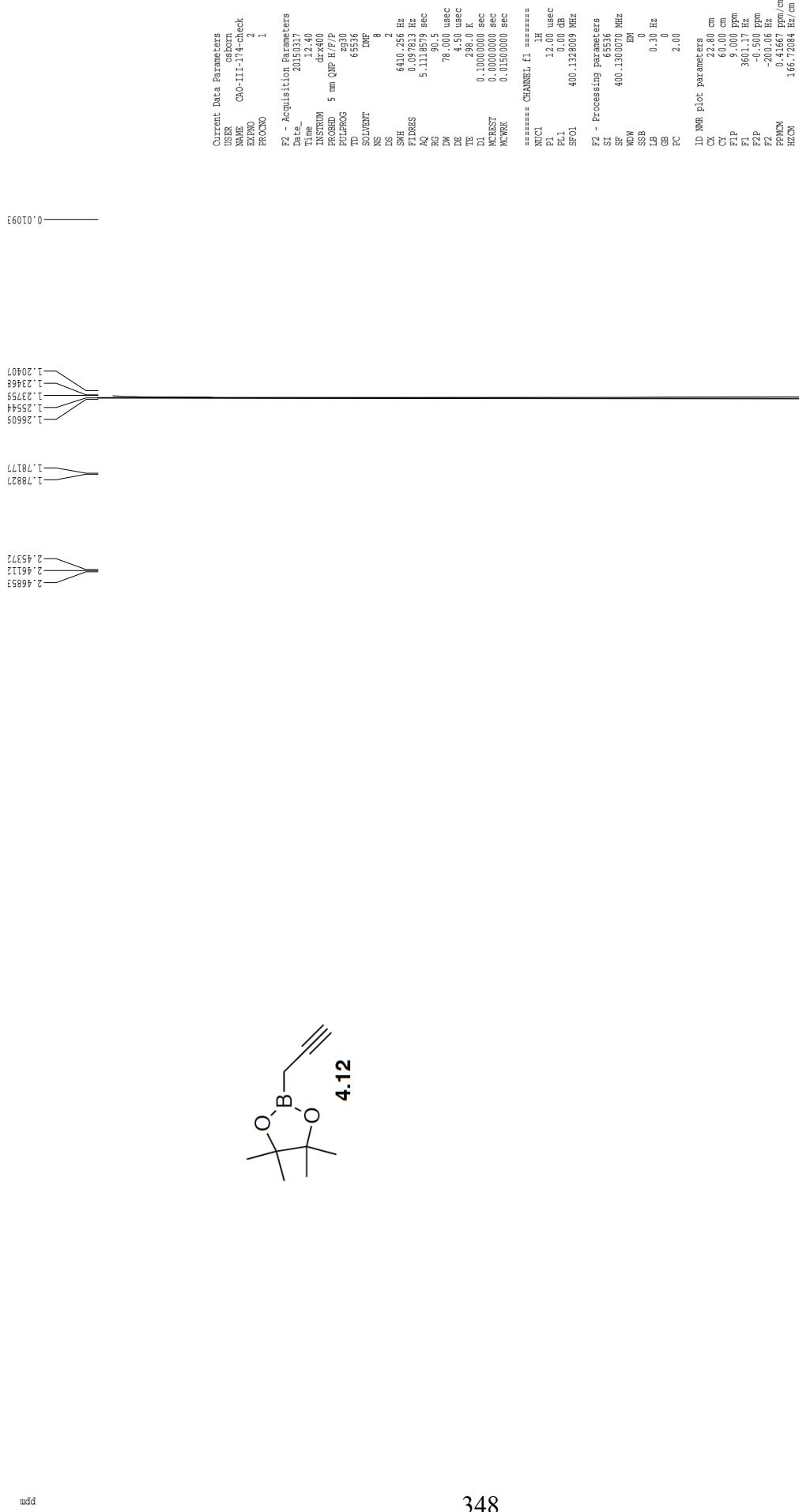




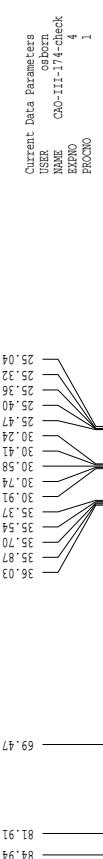
1H spectrum

1H spectrum

0.01093



Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling



```
Current Data Parameters
USER          osborn
NAME          CH-111-174-check
EXNO          4
PRCNO         1

F2 - Acquisition Parameters
Date        2015/3/18
Time       12:57
INSTRUM    cry500
PROBHD   5 mm CCP1 1H-
PULPROG  SpinEchoes300P.prd
TD        65536
SOLVENT    CC13
CR1       100
CR2       100
DS        16
SF        3003.021 Hz
ETRIM      0.05328 Hz
TE        296.0 us
D1        0.260000 sec
d11      0.000000 sec
D16      0.002000 sec
d17      0.001960 sec
MC1      0.000000 sec
MC2T     0.000000 sec
MC3K     0.0150000 sec
P2        33.10 us
P2SCW    296.0 K

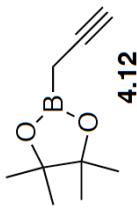
===== CHANNEL F1 =====
NUC1      13C
P1        16.55 us
P11      50.00 us
P12      200.00 us
PL0      -120.00 dB
PL1      125.794258 MHz
SP1      2.70 dB
SP2      2.70 dB
SPW01    Crp60.0,5,20.1
SPW02    Crp60.0,5,20.1
CPDPR1   CpdScope,4
SPCFF1  0.00 Hz
SPCFF2  0.00 Hz

===== CHANNEL F2 =====
NUC2      1H
CPDPG2   waltz16
P2CD2    10.00 us
P12      1.60 dB
PL12     24.50 dB
SF02    500.2225011 MHz

===== GRADIENT CHANNEL =====
GPRA01    SINE,100
GPRA02    SINE,100
GPX1      0.00 %
GPX2      0.00 %
GPY1      0.00 %
GPY2      0.00 %
GPZ1      30.00 %
GPZ2      50.00 %
GP22     500.00 us
P15      1000.00 us
P16      1000.00 us

F2 - Processing Parameters
SI        65536
SF        125.780304 MHz
WDW      EM
SSB      0
LB        1.00 Hz
GB      0
PC        2.00

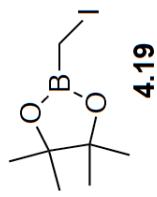
1D NMR Plot Parameters
CX        22.80 cm
CY        35.00 cm
F1P      230.67 ppm
F1       230.945 Hz
F2P      -10.287 ppm
F2      -129.346 Hz
PPCM    10.5668 ppm/cm
HZDW   1323.10596 Hz/cm
```



1H spectrum

ppm

350



```

Current Data Parameters
USER      osborn
NAME      QM-1-123 SI
EXNO      1
PROCNO   1

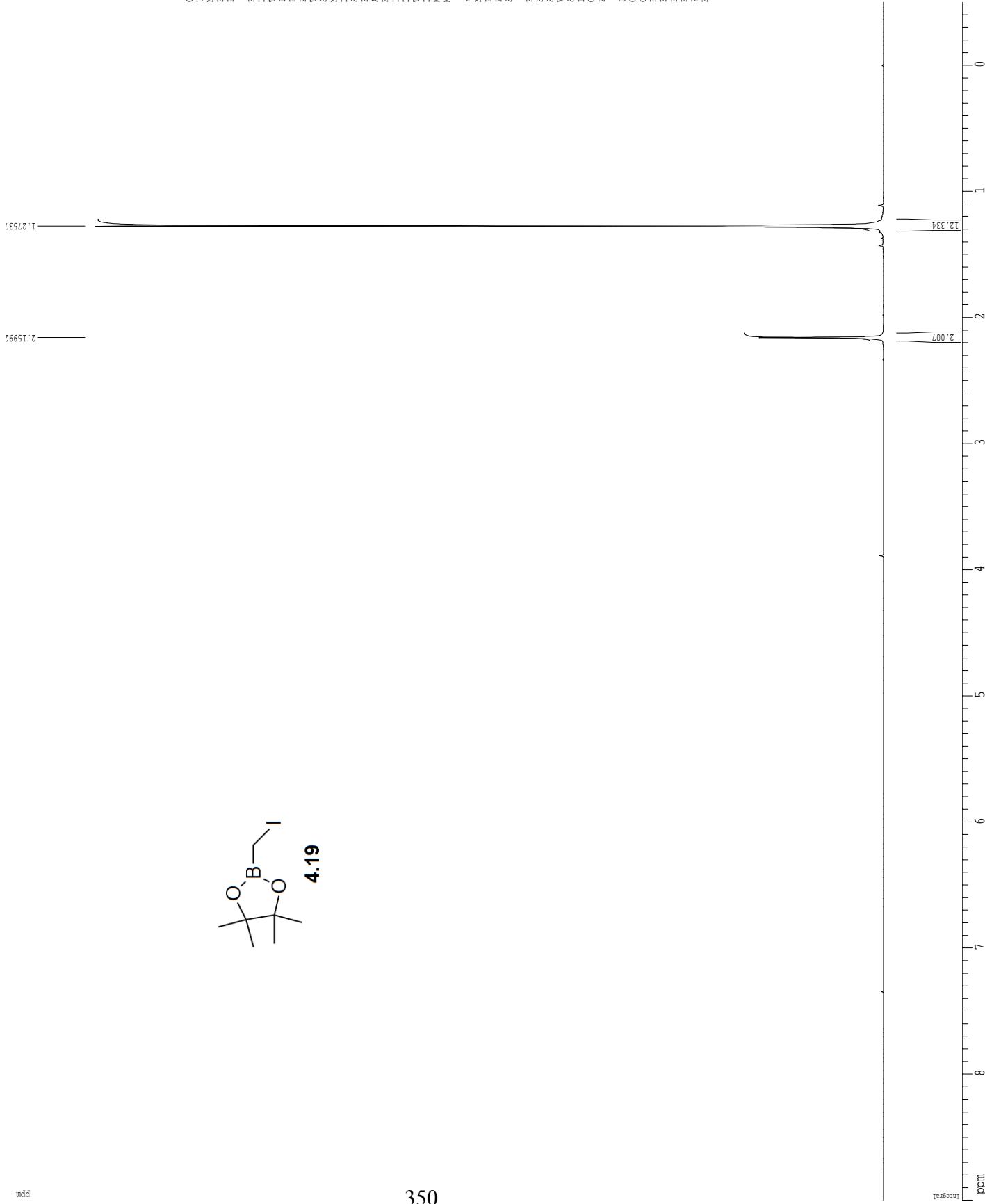
F2 - Acquisition Parameters
Date      20131017
Time      9.45
INSTRUM  drx400
PROBID   5 mm QNP H/F/P
PULPROG  2330
TD       65536
SOLVENT  CDCl3
NS        8
DS        2
SWH     6410.056 Hz
FIDRES  0.07713 Hz
AQ      5.118379 sec
RG      16
DW      78.000 usec
DE      4.50 usec
TE      298.0 K
D1    0.1000000 sec
MCHEST  0.0000000 sec
MCWRK  0.0150000 sec

=====
CHANNEL f1
=====
N1C1      1H
P1      12.00 usec
PL1     -0.60 dB
SF01    400.1380010 MHz

F2 - Processing parameters
SI      65536
SP      400.129970 kHz
WDW      EN
SSB      0
LB      0.30 Hz
GB      0
PC      2.00

1D NMR plot parameters
CX      22.80 cm
CY      15.00 cm
F1P    9.0000 ppm
F1     3601.17 Hz
F2P    -0.500 ppm
F2     -200.06 Hz
PPCM    0.14167 ppm/cm
HCEN    166.12083 Hz/cm

```



Z-restored spin-echo  $^{13}\text{C}$  spectrum with  $^1\text{H}$  decoupling

zdd

24.481

84.131

```

Current Data Parameters
USER   osborn
NAME   CH0-I-123.SI
EXENO   8
PRCNO   1

F2 - Acquisition Parameters
Date 2012.01.07
Time 19:38
INSTRUM cryo500
PROBHD 5 mm CCP1 1H-
PULPROG SpinEditCh30DP.prd
TD    65536
SOLVENT C6C13
NS     16
DS     16
TE    296.0 K
SWH 0.260000 sec
ETRIM 0.065398 Hz
ETRIM2 1.000000 sec
AQ    128.2
RG    16.50 usec
DR    16.50 usec
DE    3.00 usec
TE    296.0 K
D1    0.260000 sec
d11   0.000000 sec
D16   0.002000 sec
d17   0.001960 sec
MC    1
MCINT 0.000000 sec
MCNTRK 0.0150000 sec
F2R1 31.00 usec
SPCFF2 0.00 Hz

===== CHANNEL F1 =====
NUC1 13C
P1   15.50 usec
P12  50.00 usec
P1J 200.00 usec
PL0   12.00 dB
PL1   -1.00 dB
SP01 125.794258 MHz
SP1   3.20 dB
SP2   3.20 dB
SPR0M1 Crp60.0,5,20.1
SPR0M2 Crp60.0,5,20.1
Cp60cscope.4
SPCFF1 0.00 Hz
SPCFF2 0.00 Hz

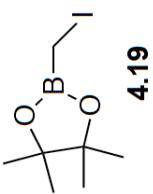
===== CHANNEL F2 =====
CPDPRG2 waltz16
NUC2 1H
PCPD2 100.00 usec
PL2   1.60 dB
PL12  24.60 dB
SF02 500.2225011 MHz

===== GRADIENT CHANNEL =====
GP0M1 SINE.100
GP0M2 SINE.100
GPX1 0.00 %
GPX2 0.00 %
GPY1 0.00 %
GPY2 0.00 %
GPZ1 30.00 %
GPZ2 50.00 %
GP15 500.00 usec
P16 1000.00 usec

F2 - Processing Parameters
SI    65536
SF    125.7804190 MHz
WDW   EM
SSB   0
LB    1.00 Hz
GB    0
PC    2.00

1D NMR Plot Parameters
CX    22.80 cm
CY    15.65 cm
F1P   20.67 ppm
F1    280.968 Hz
F2P   -10.287 ppm
F2    -129.316 Hz
PPCM  10.5668 ppm/cm
HZCM 1325.10706 Hz/cm

```



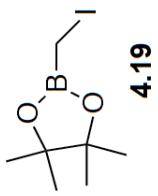
**4.19**



11B spectrum with 1H decoupling with background suppression

31.709

ppm



Current Data Parameters  
USER osborn  
NAME C4O-1-123 SI  
EXPNO 5  
PRCNO 1

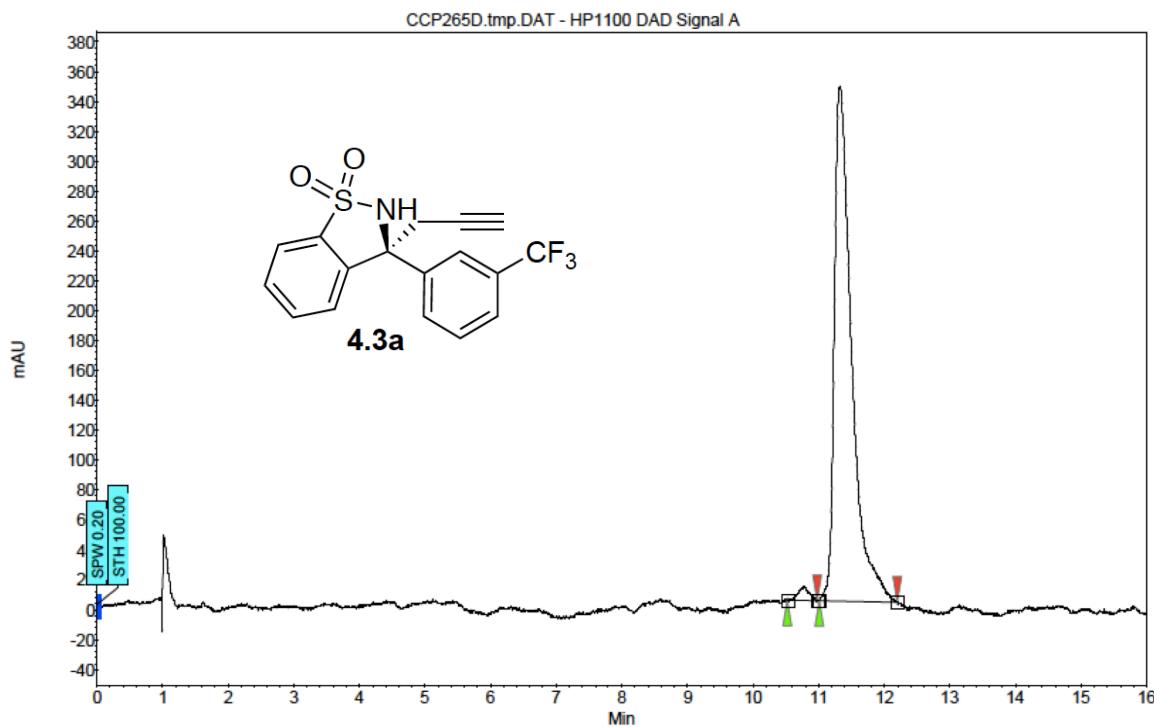
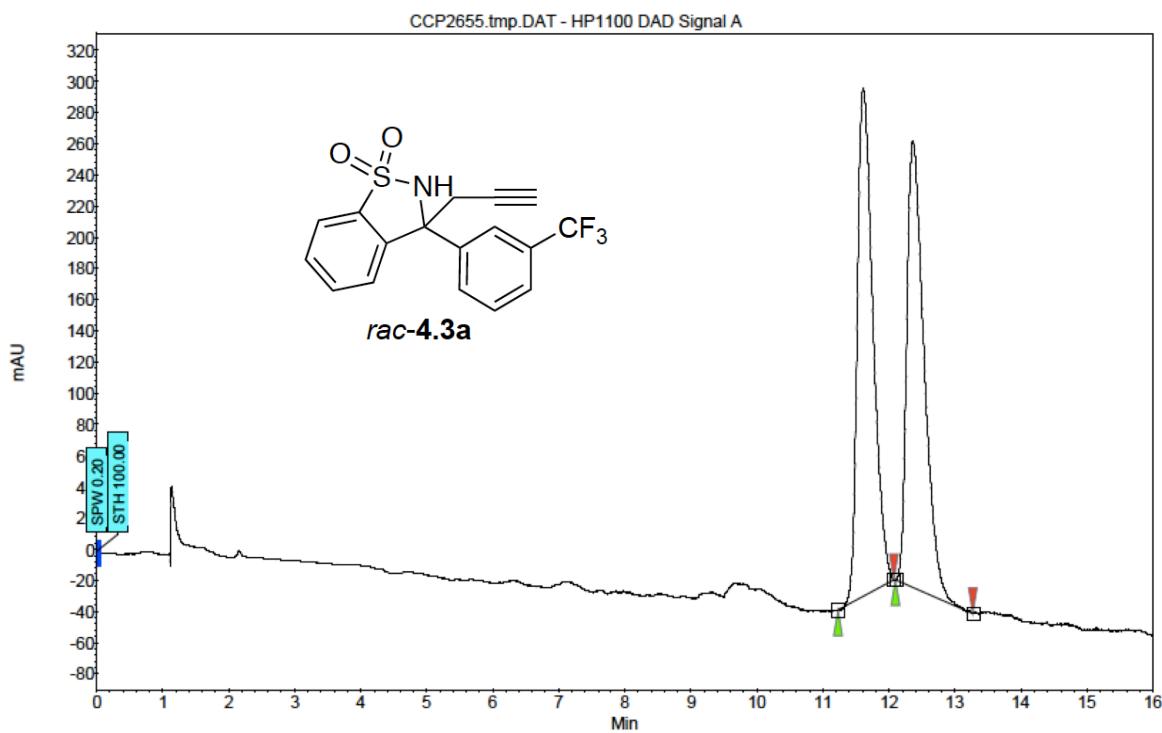
F2 - Acquisition Parameters  
Date 20/12/07  
Time 10:21  
INSTRUM 3G500  
PROBOD 5 mm broadband  
PULPROG 2gps  
TD 65536  
SOLVENT CDCl<sub>3</sub>  
NS 64  
DS 4  
SWH 37037.035 Hz  
FIDRES 0.565140 Hz  
AQ 0.884760 sec  
RG 90.5  
DW 13.500 usec  
DE 6.00 usec  
TE 298.0 K  
D1 1.000000 sec  
MESTET 0.000000 sec  
MCRRK 0.150000 sec

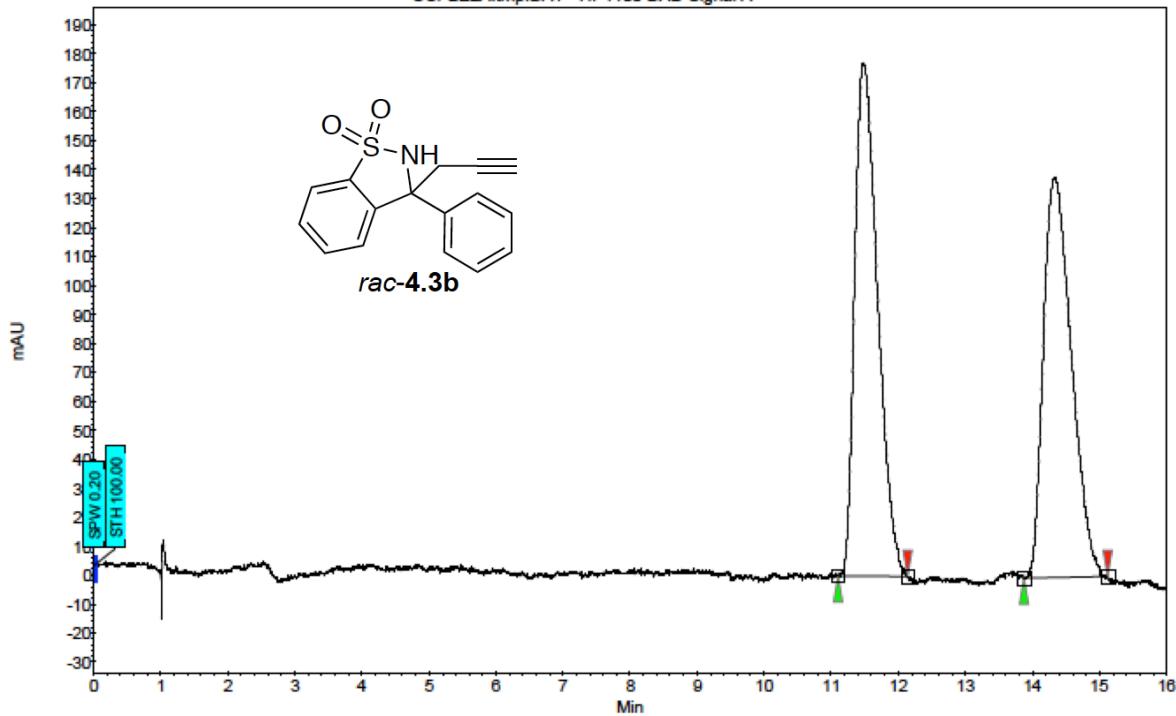
===== CHANNEL f1 =====  
NUCL 11B  
SI 65536  
P1 8.65 usec  
P2 17.30 usec  
PL1 -3.00 dB  
SFOL 160.2273660 MHz

F2 - Processing parameters  
SI 65536  
SF 160.227321 MHz  
WDW EN  
SSB 0  
LB 2.00 Hz  
GB 0  
PC 2.00

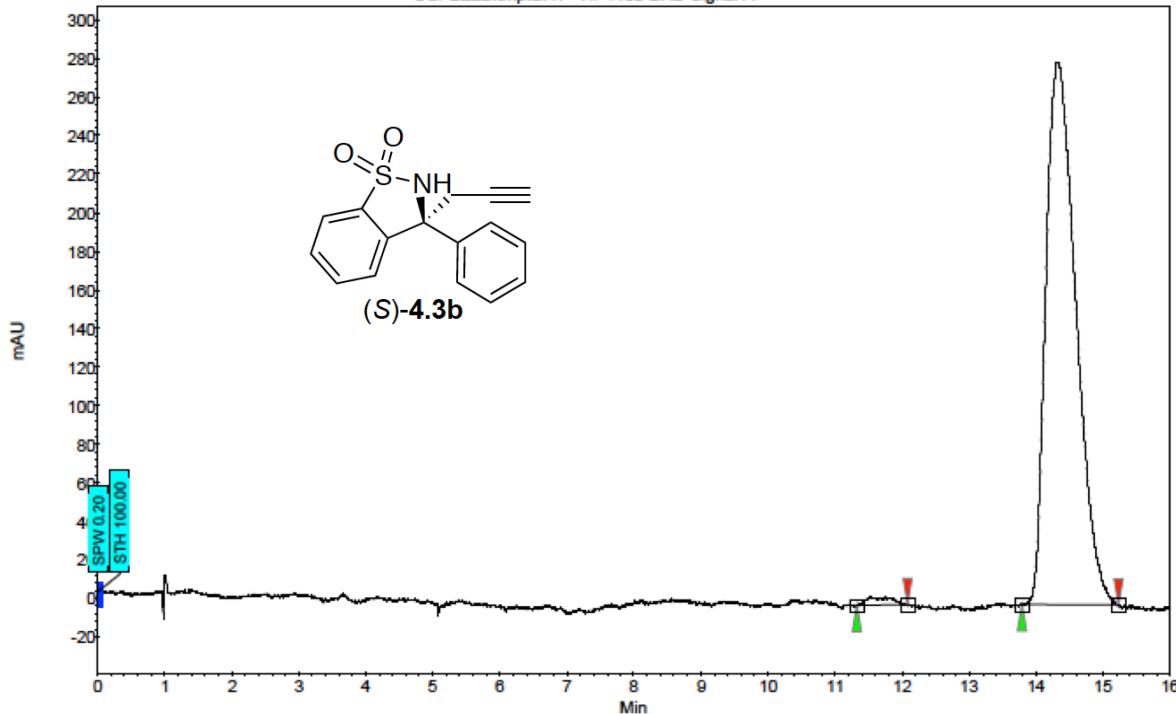
1D NMR plot parameters  
CX 22.80 cm  
CY 15.00 cm  
F1P 115.601 ppm  
F1 16522.36 Hz  
F2P -115.552 ppm  
F2 -195.14.67 Hz  
PFCM 10.13229 Hz/cm  
HZCM 1624.43140 Hz/cm

-100  
-50  
0  
50  
100  
ppm

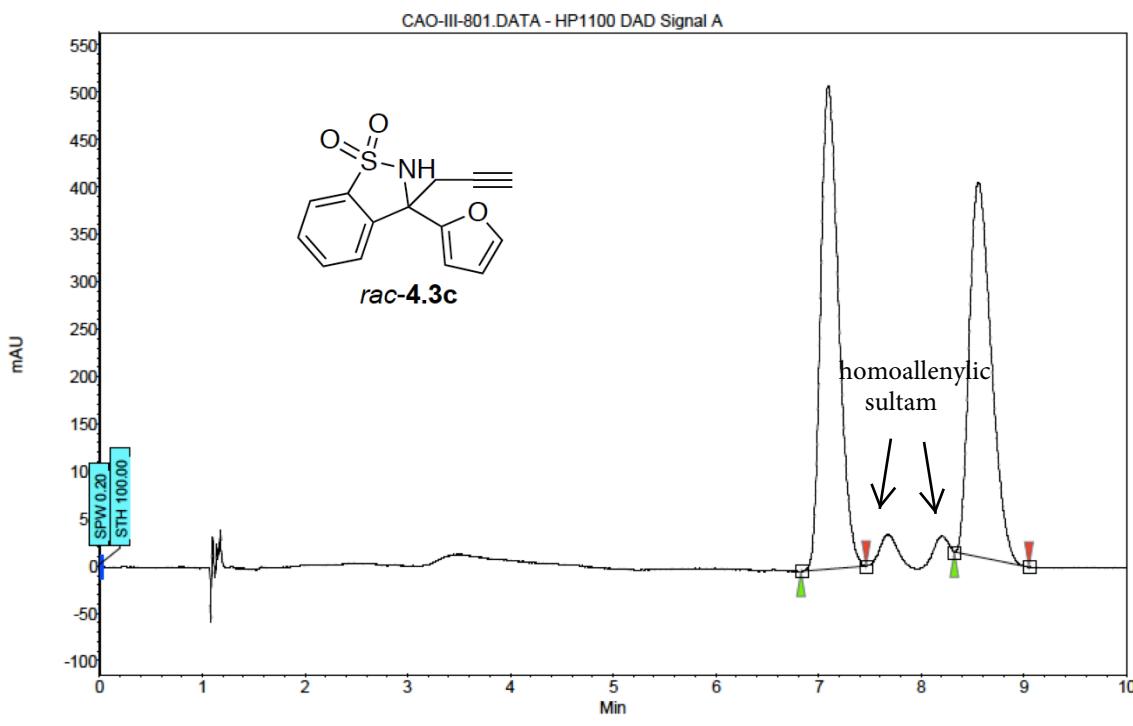




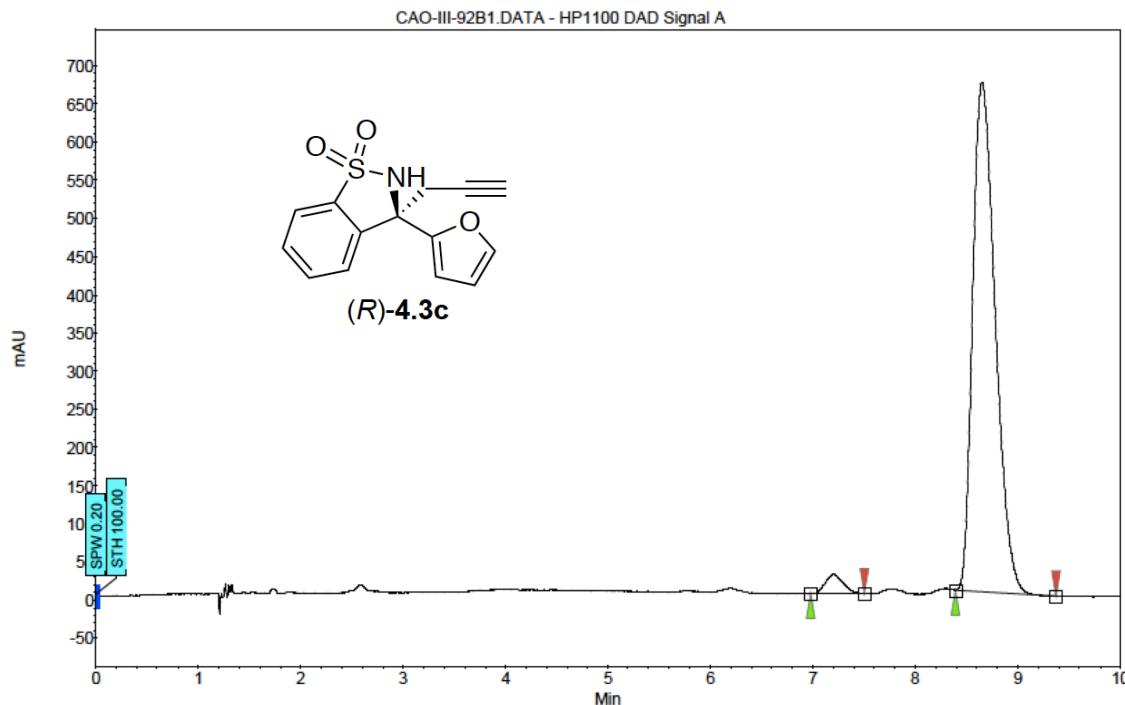
Index	Name	Start	Time	End	RT Offset	Quantity	Height	Area	Area
		[Min]	[Min]	[Min]	[Min]	[% Area]	[ $\mu$ V]	[ $\mu$ V.Min]	[%]
1	UNKNOWN	11.10	11.48	12.14	0.00	51.26	177.2	70.1	51.263
2	UNKNOWN	13.87	14.33	15.12	0.00	48.74	138.2	66.6	48.737



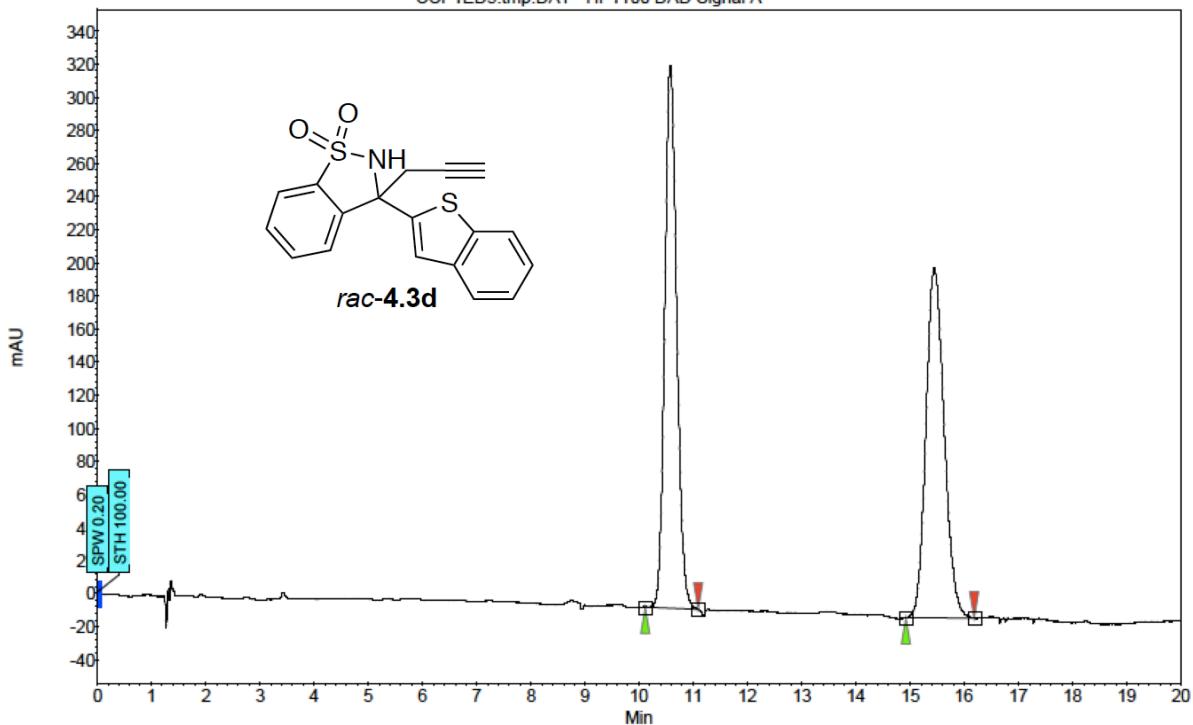
Index	Name	Start	Time	End	RT Offset	Quantity	Height	Area	Area
		[Min]	[Min]	[Min]	[Min]	[% Area]	[ $\mu$ V]	[ $\mu$ V.Min]	[%]
1	UNKNOWN	11.33	11.54	12.08	0.00	1.19	4.6	1.8	1.190
2	UNKNOWN	13.79	14.32	15.23	0.00	98.81	281.6	146.1	98.810
Total						100.00	286.2	147.9	100.000



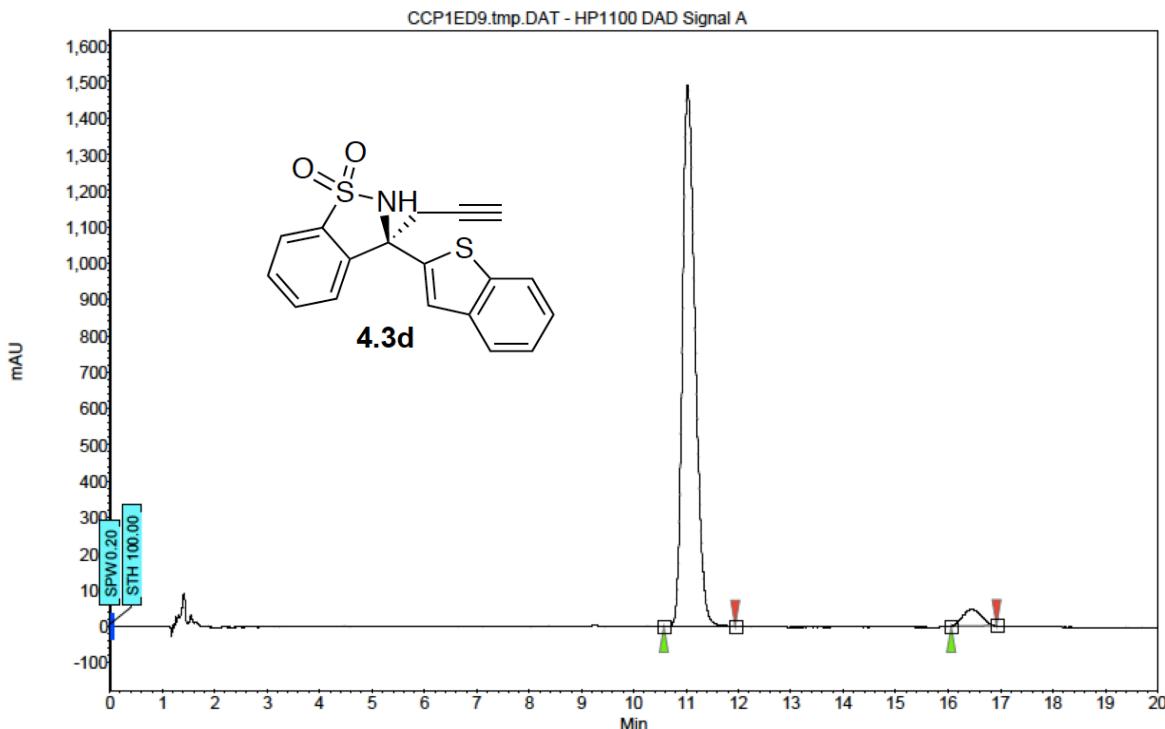
Index	Name	Start [Min]	Time [Min]	End [Min]	RT Offset [Min]	Quantity [% Area]	Height [µV]	Area [µV.Min]	Area [%]
1	UNKNOWN	6.83	7.09	7.46	0.00	51.48	510.0	104.0	51.480
2	UNKNOWN	8.32	8.55	9.05	0.00	48.52	395.3	98.0	48.520
Total						100.00	905.3	202.0	100.000

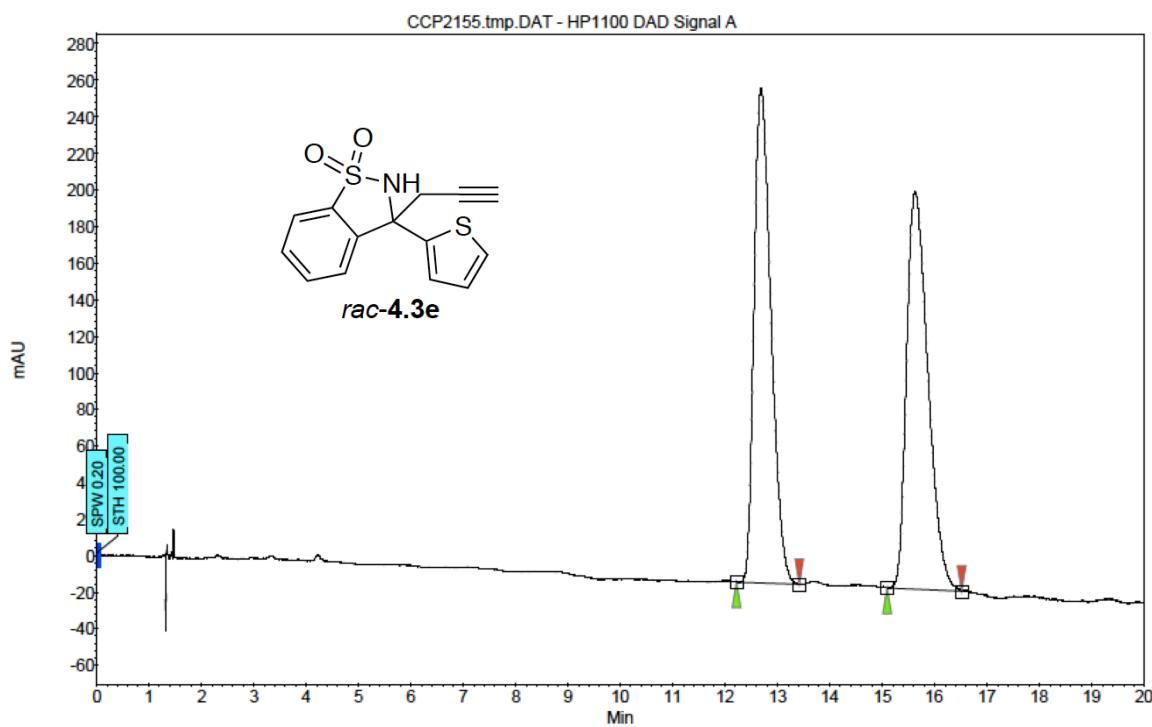


Index	Name	Start [Min]	Time [Min]	End [Min]	RT Offset [Min]	Quantity [% Area]	Height [µV]	Area [µV.Min]	Area [%]
1	UNKNOWN	6.98	7.20	7.50	0.00	2.78	24.7	4.9	2.778
2	UNKNOWN	8.39	8.65	9.37	0.00	97.22	667.3	170.0	97.222
Total						100.00	692.0	174.8	100.000

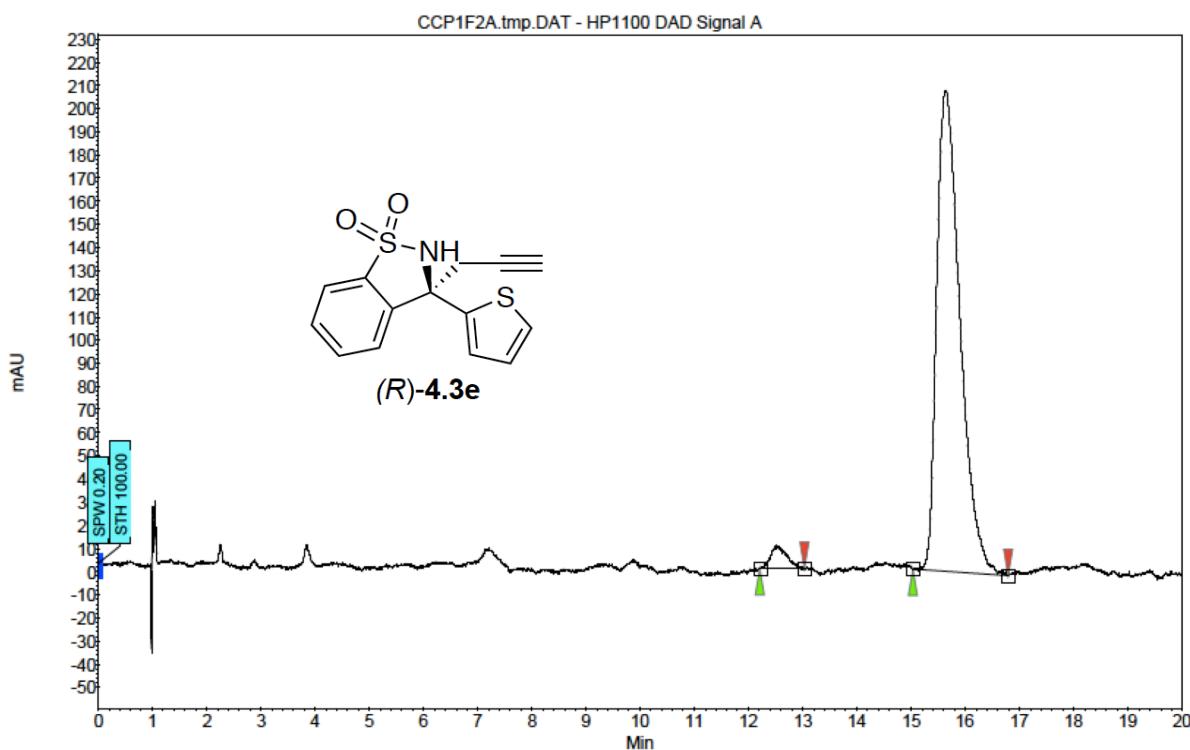


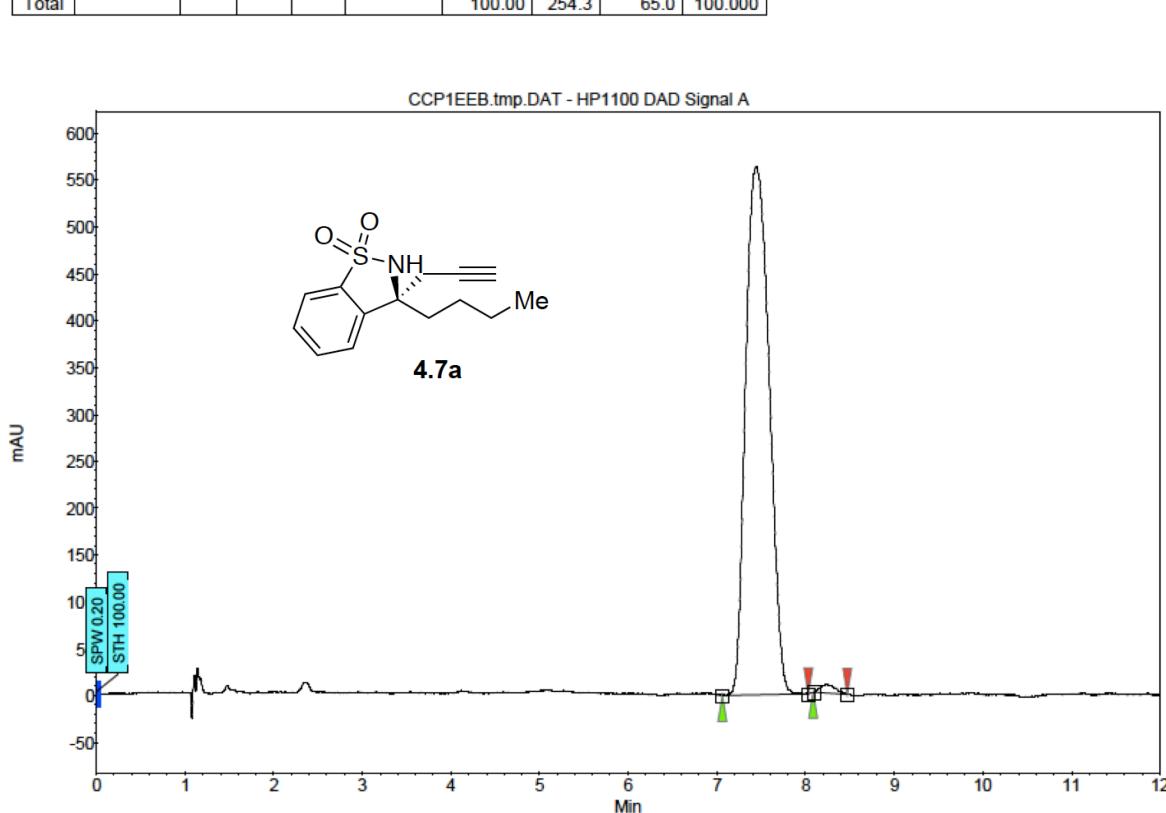
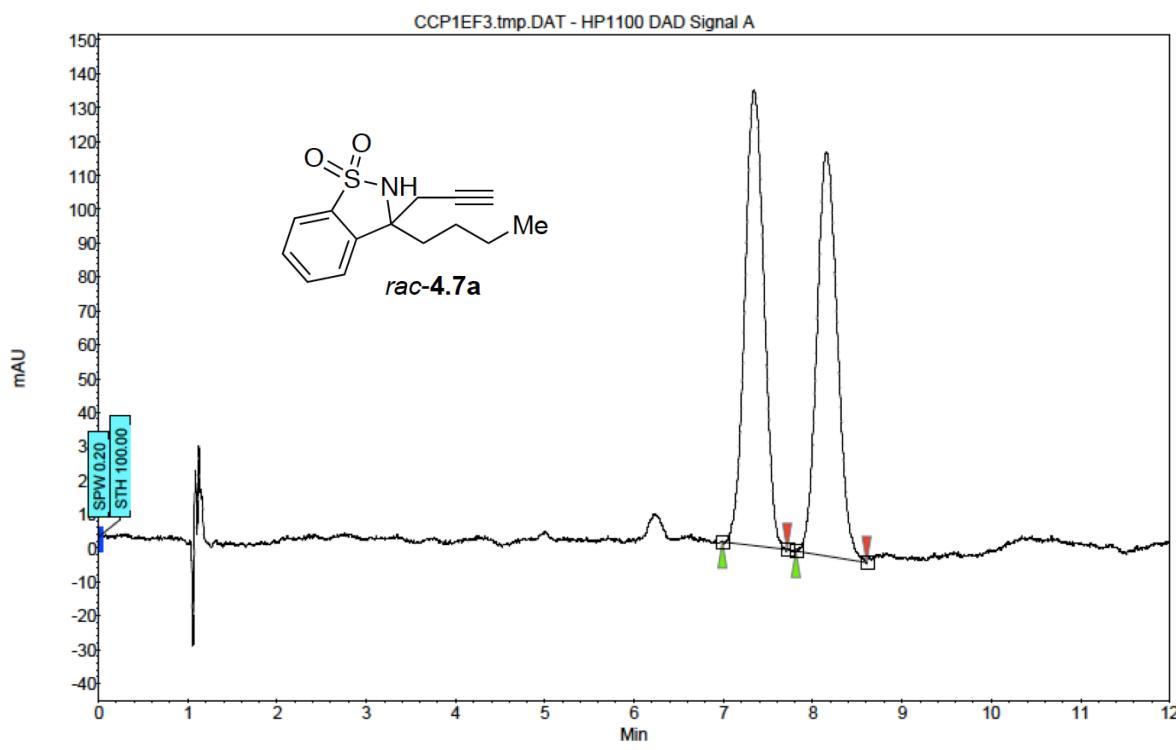
Index	Name	Start [Min]	Time [Min]	End [Min]	RT Offset [Min]	Quantity [% Area]	Height [µV]	Area [µV.Min]	[%]
1	UNKNOWN	10.12	10.58	11.09	0.00	50.44	328.0	83.7	50.437
2	UNKNOWN	14.92	15.45	16.18	0.00	49.56	211.6	82.3	49.563
Total						100.00	539.6	166.0	100.000

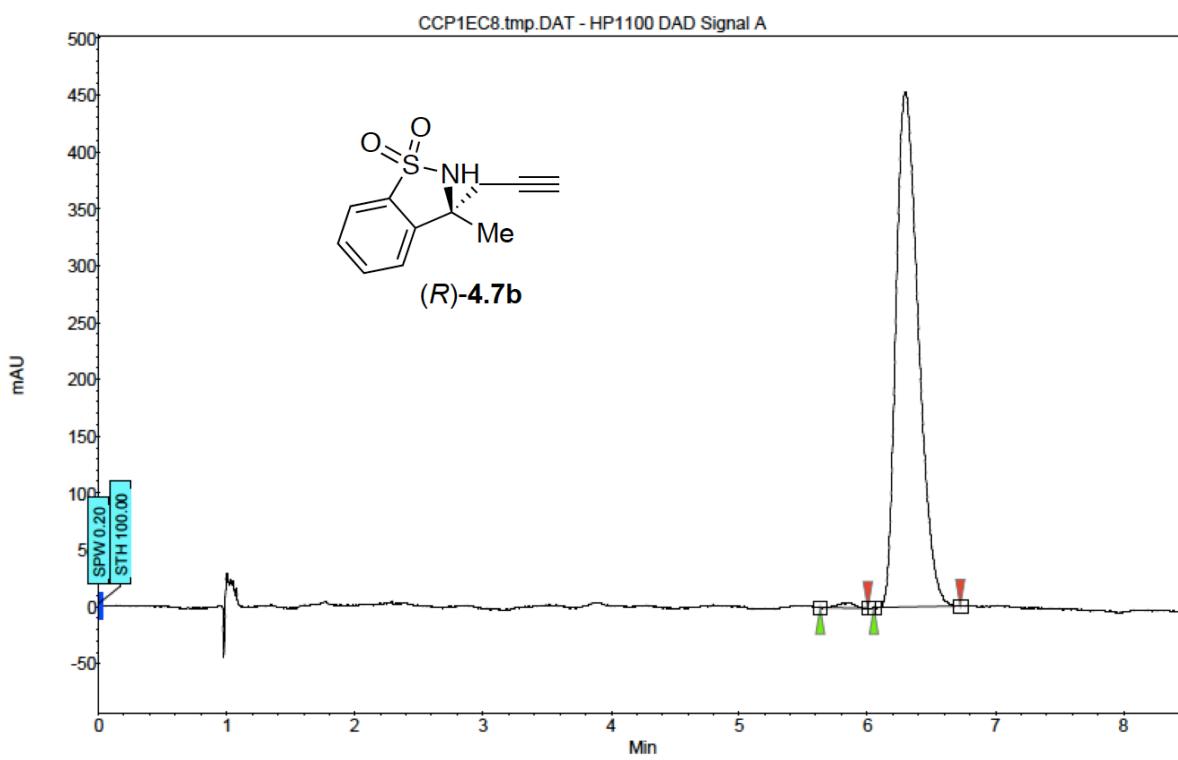
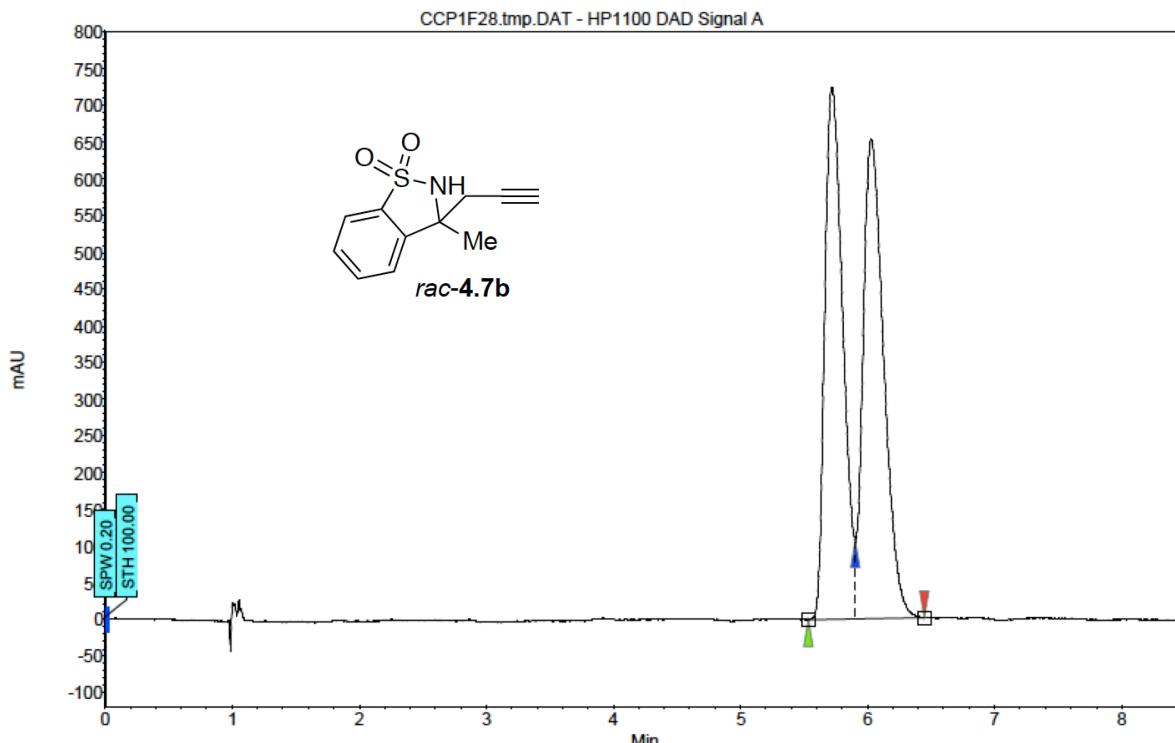


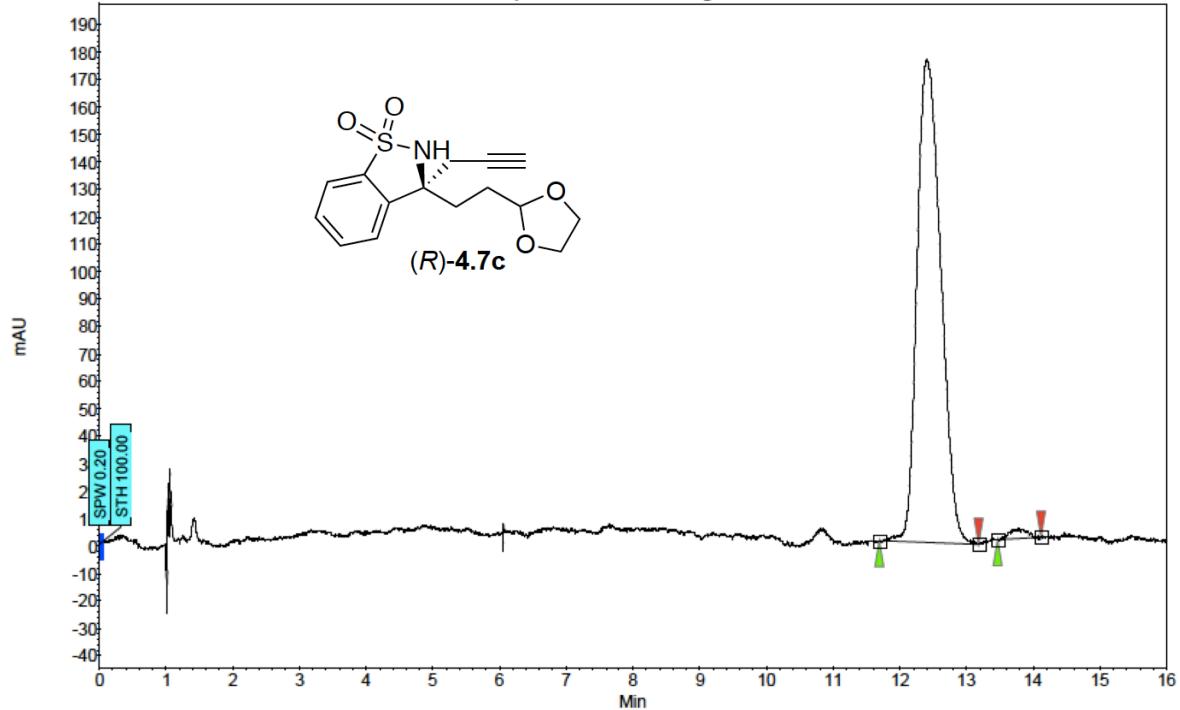
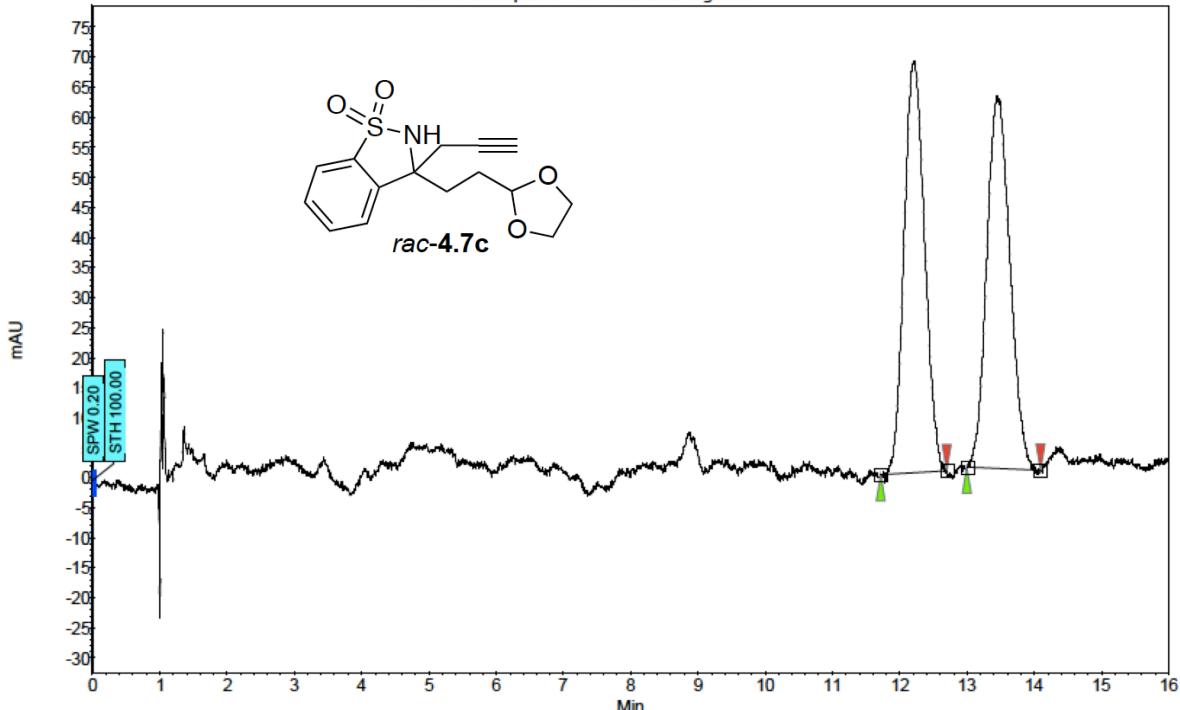


Index	Name	Start [Min]	Time [Min]	End [Min]	RT Offset [Min]	Quantity [% Area]	Height [µV]	Area [µV.Min]	Area [%]
1	UNKNOWN	12.22	12.68	13.42	0.00	49.78	270.8	101.5	49.777
2	UNKNOWN	15.09	15.63	16.51	0.00	50.22	217.6	102.4	50.223
Total						100.00	488.4	203.9	100.000

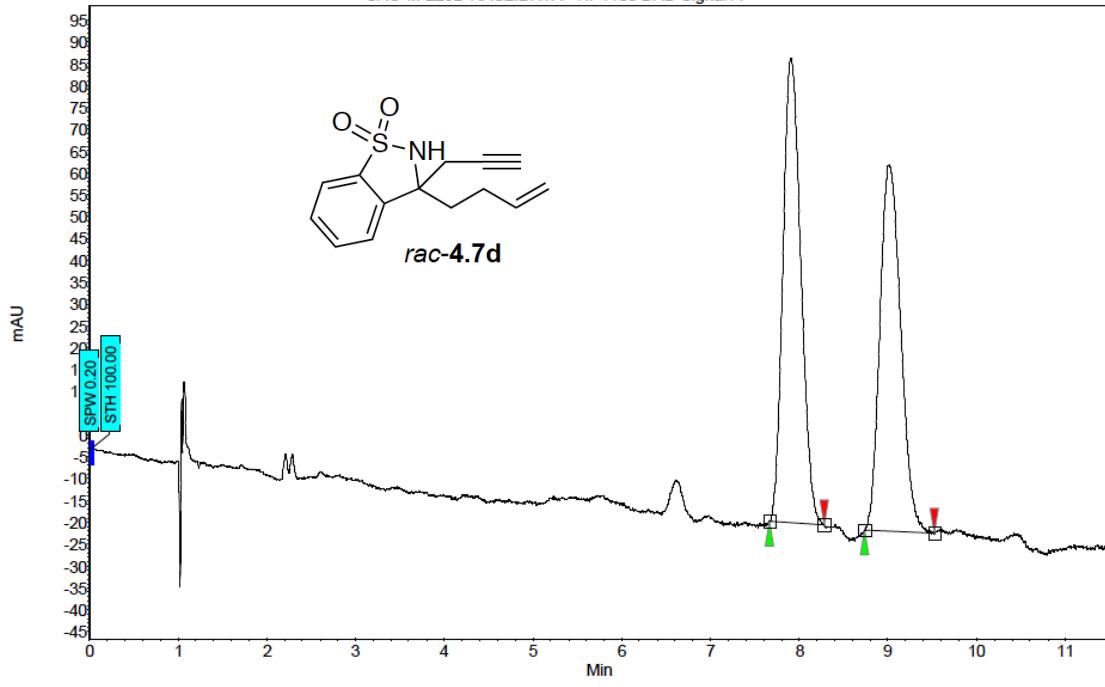




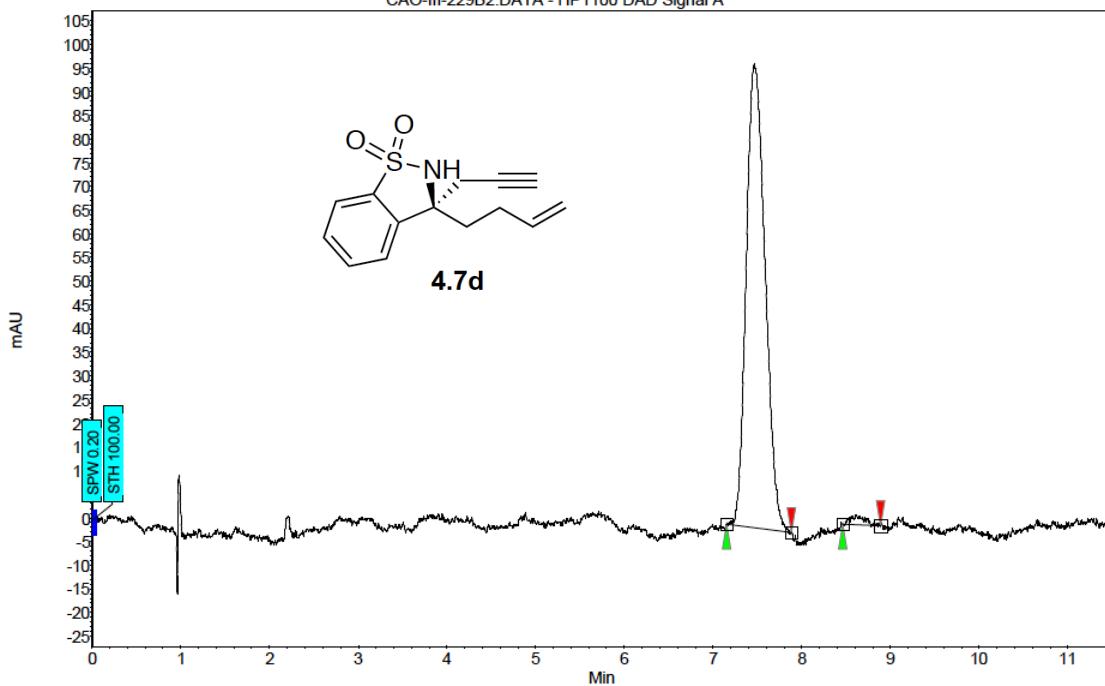




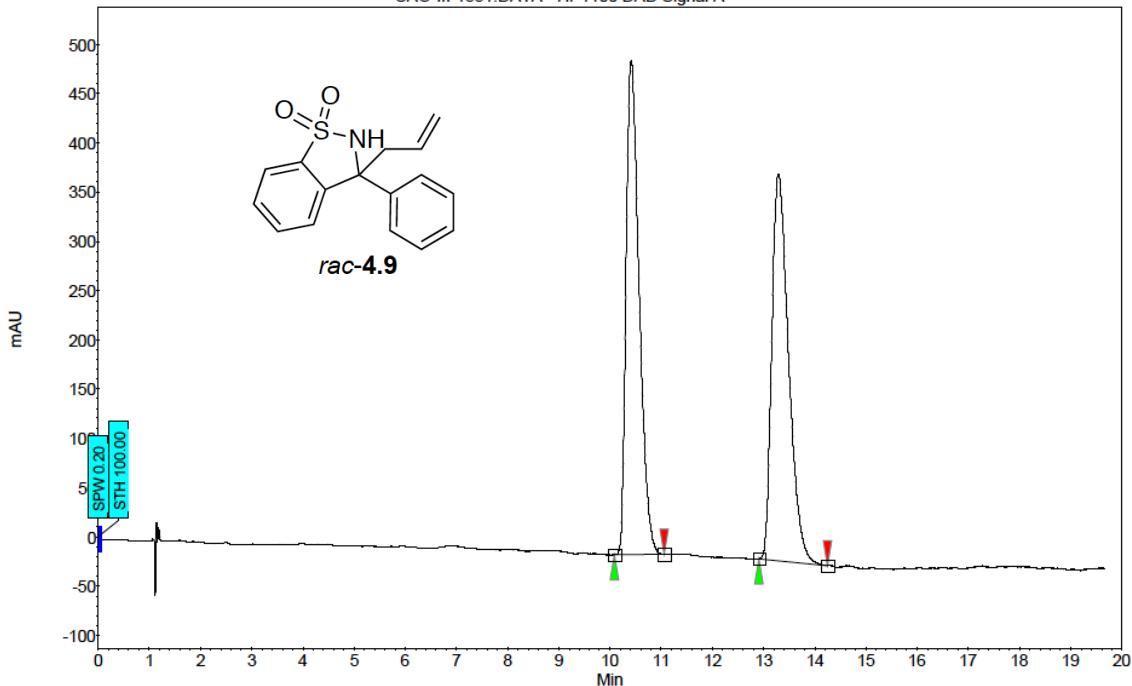
## CAO-III-228B-RAC2.DATA - HP1100 DAD Signal A



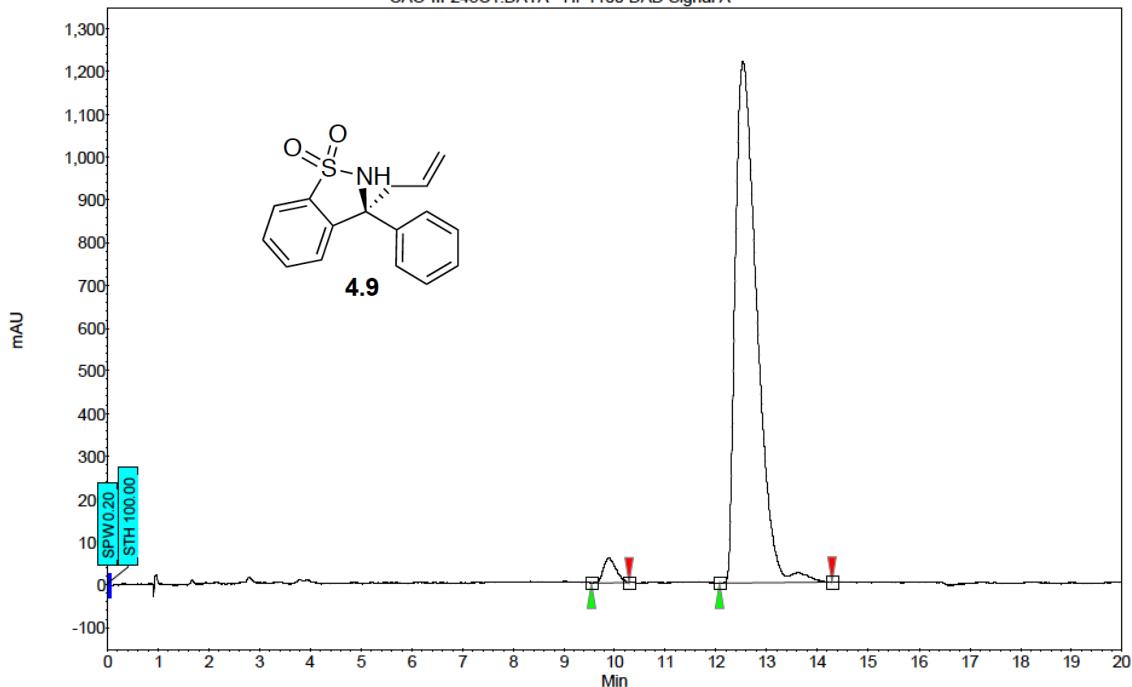
## CAO-III-229B2.DATA - HP1100 DAD Signal A

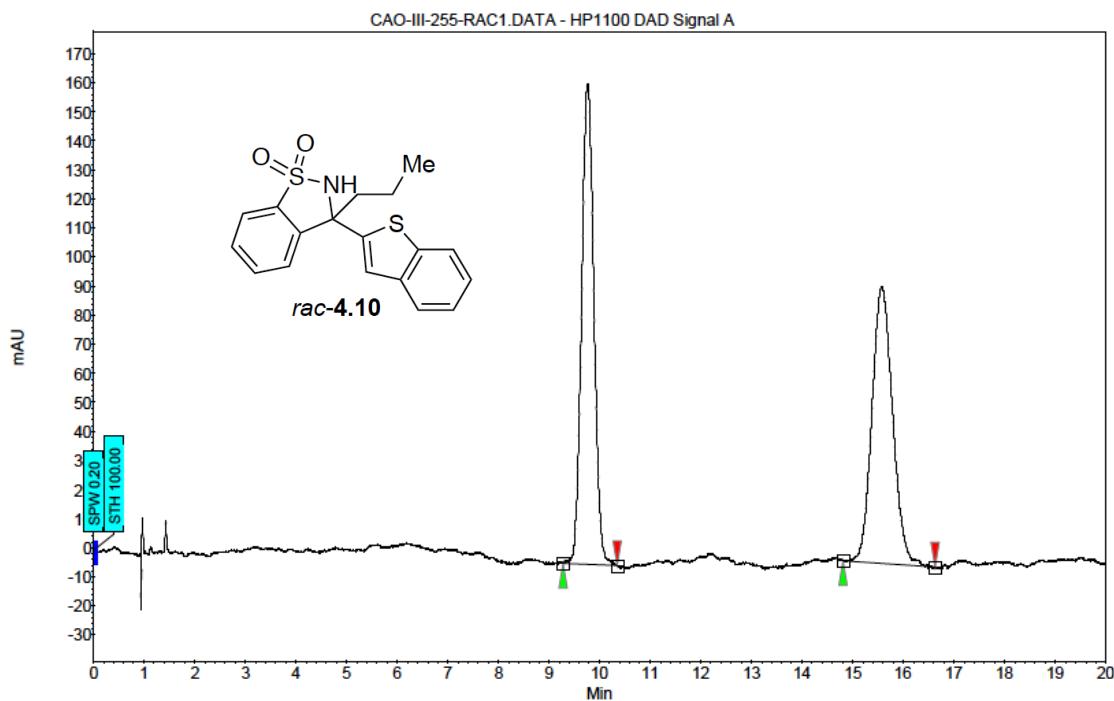


## CAO-III-1331.DATA - HP1100 DAD Signal A

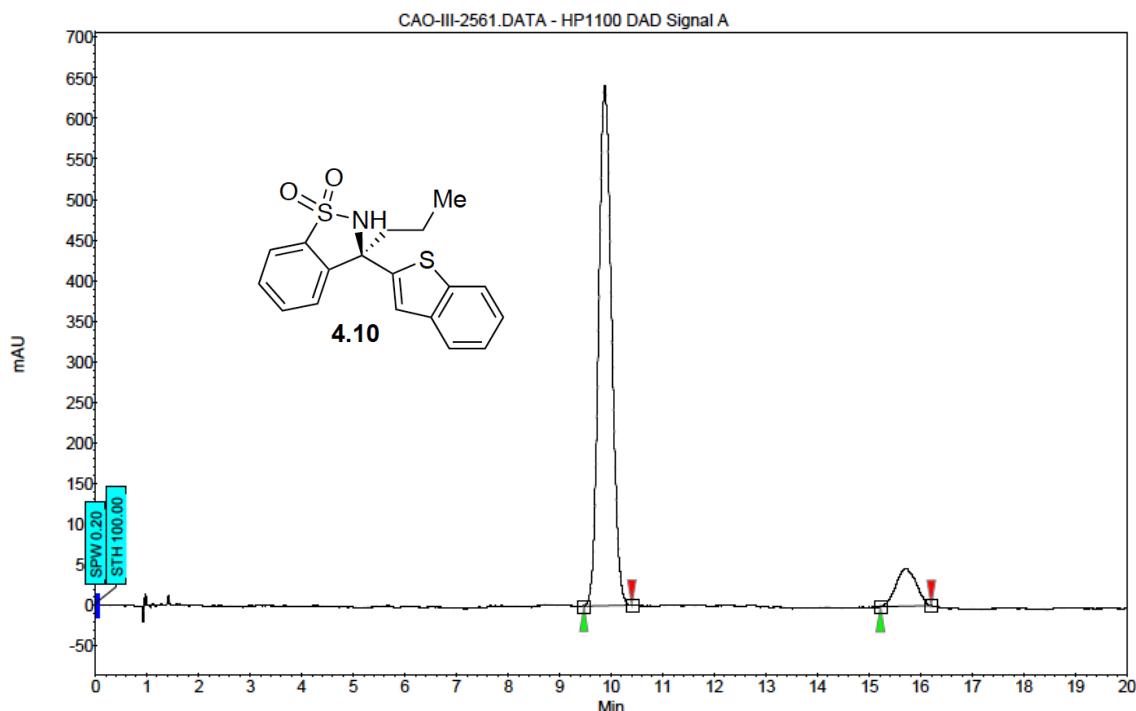


## CAO-III-248C1.DATA - HP1100 DAD Signal A



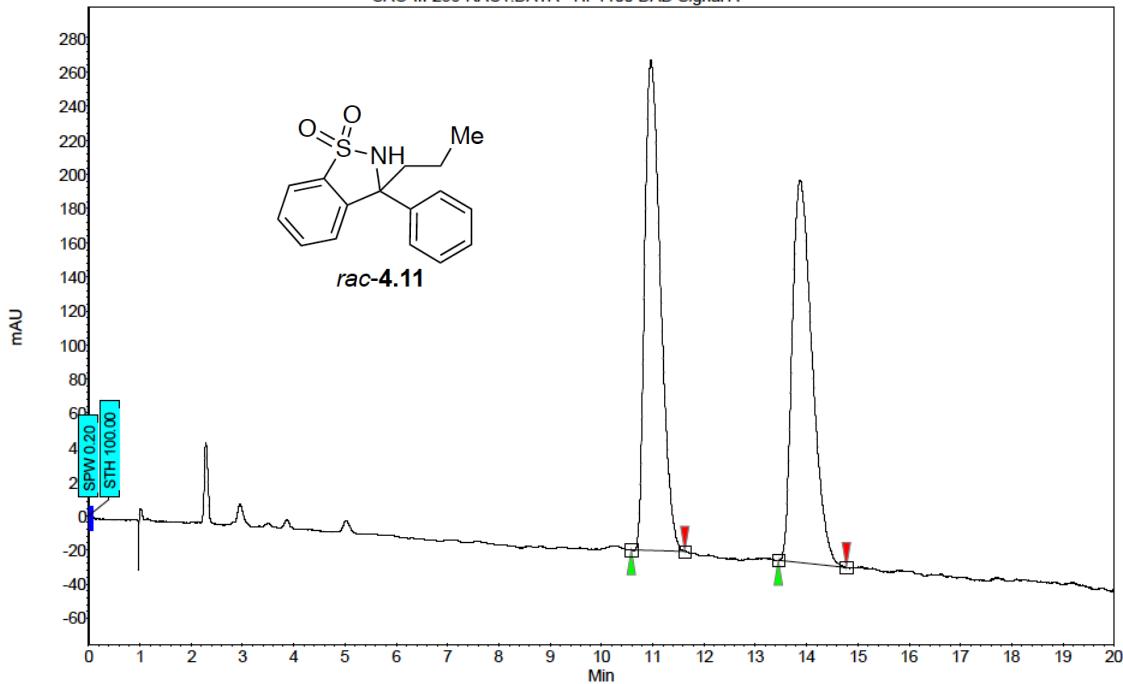


Index	Name	Start [Min]	Time [Min]	End [Min]	RT Offset [Min]	Quantity [% Area]	Height [µV]	Area [µV.Min]	Area [%]
1	UNKNOWN	9.27	9.77	10.35	0.00	50.78	165.4	46.4	50.780
2	UNKNOWN	14.81	15.58	16.63	0.00	49.22	95.4	44.9	49.220
Total						100.00	260.8	91.3	100.000

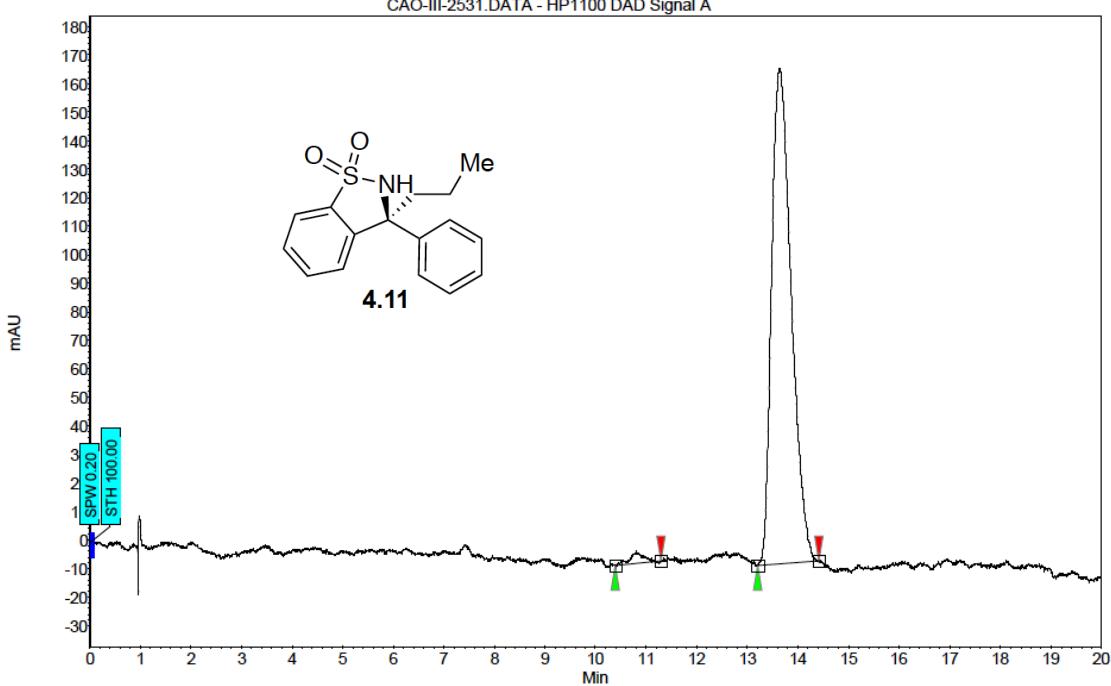


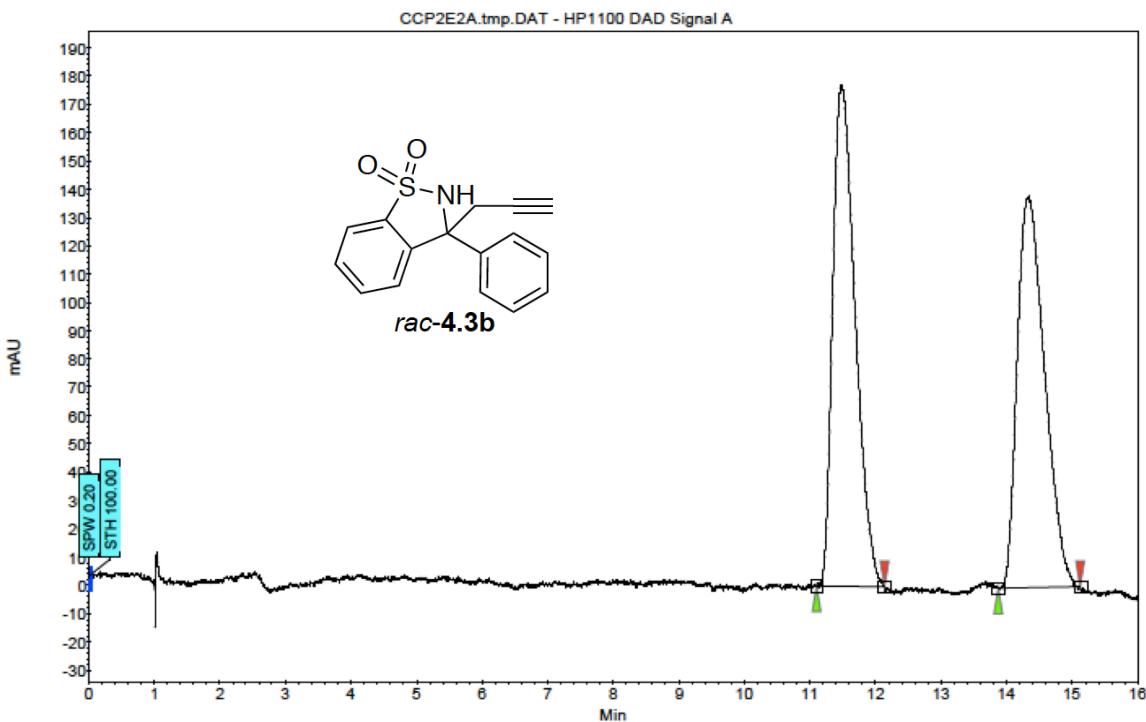
Index	Name	Start [Min]	Time [Min]	End [Min]	RT Offset [Min]	Quantity [% Area]	Height [µV]	Area [µV.Min]	Area [%]
1	UNKNOWN	9.47	9.88	10.40	0.00	89.76	640.8	185.6	89.762
2	UNKNOWN	15.22	15.71	16.20	0.00	10.24	46.3	21.2	10.238
Total						100.00	687.1	206.7	100.000

## CAO-III-236.RAC1.DATA - HP1100 DAD Signal A

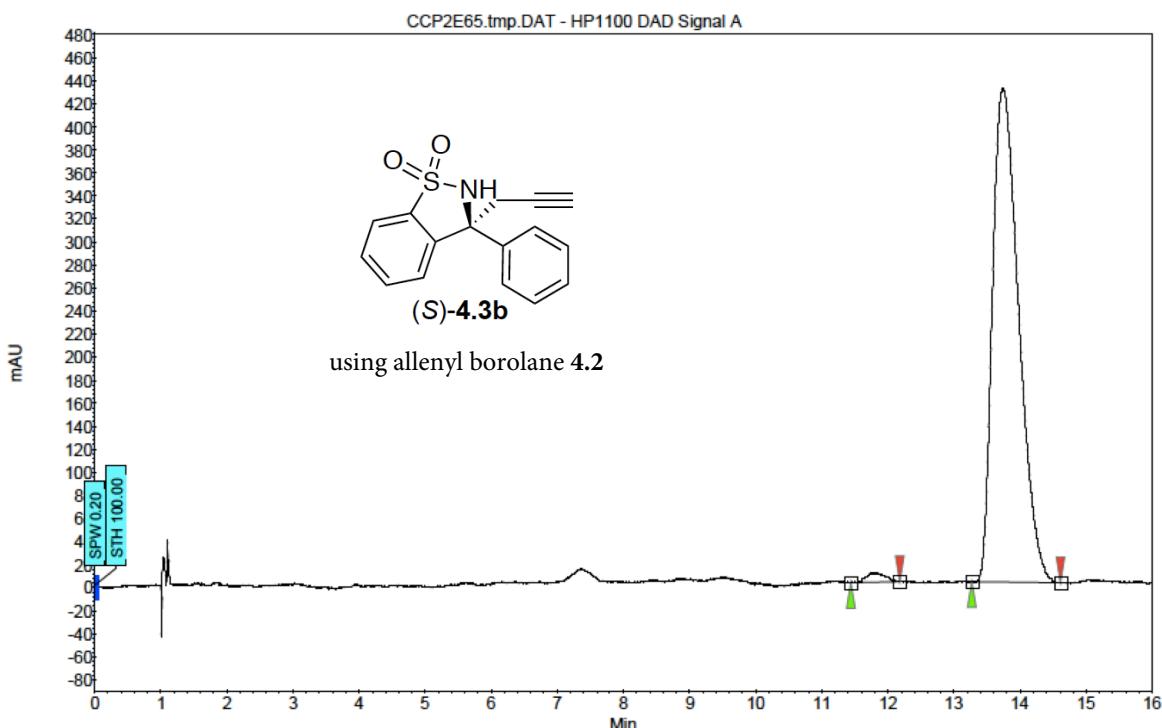


## CAO-III-2531.DATA - HP1100 DAD Signal A

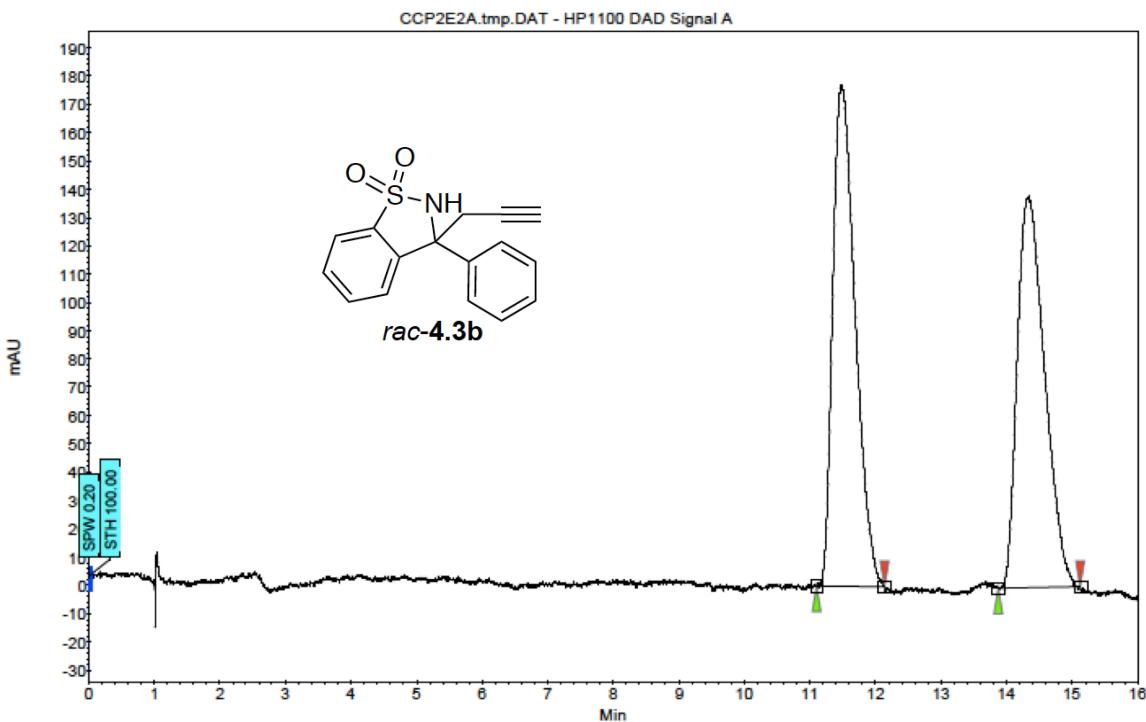




Index	Name	Start [Min]	Time [Min]	End [Min]	RT Offset [Min]	Quantity [% Area]	Height [ $\mu$ V]	Area [ $\mu$ V.Min]	Area [%]
1	UNKNOWN	11.10	11.48	12.14	0.00	51.26	177.2	70.1	51.263
2	UNKNOWN	13.87	14.33	15.12	0.00	48.74	138.2	66.6	48.737
Total					100.00	315.3	136.7	100.000	



Index	Name	Start [Min]	Time [Min]	End [Min]	RT Offset [Min]	Quantity [% Area]	Height [ $\mu$ V]	Area [ $\mu$ V.Min]	Area [%]
1	UNKNOWN	11.44	11.78	12.18	0.00	1.38	8.5	2.7	1.379
2	UNKNOWN	13.27	13.73	14.61	0.00	98.62	428.9	196.4	98.621
Total					100.00	437.3	199.2	100.000	



Index	Name	Start [Min]	Time [Min]	End [Min]	RT Offset [Min]	Quantity [% Area]	Height [ $\mu$ V]	Area [ $\mu$ V.Min]	Area [%]
1	UNKNOWN	11.10	11.48	12.14	0.00	51.26	177.2	70.1	51.263
2	UNKNOWN	13.87	14.33	15.12	0.00	48.74	138.2	66.6	48.737
Total						100.00	315.3	136.7	100.000

