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## Authors

Walker, M S Dietderich, D R

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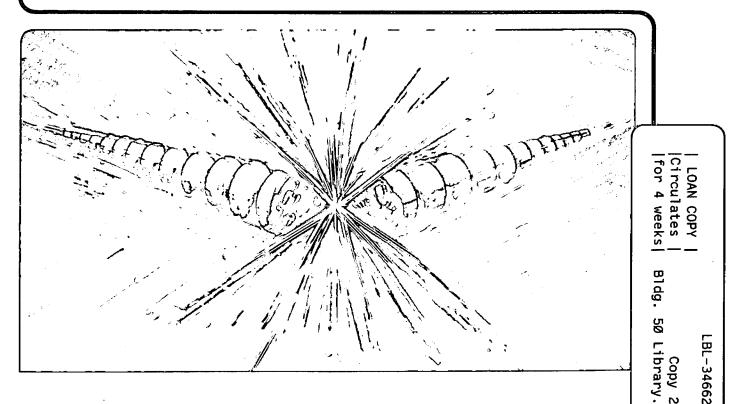
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## **Progress in the Development of BSCCO-2212 Superconductors**

M.S. Walker and D.R. Dietderich

November 1992



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## SC-MAG-436 LBL-34662

## Progress in the Development of BSCCO-2212 Superconductors\*

## M. S. Walker

(Visiting Scientist, Lawrence Berkeley Laboratory) Intermagnetics General Corporation Guilderland, New York

## D. R. Dietderich

Lawrence Berkeley Laboratory University of California Berkeley, California

## November, 1992

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## PROGRESS IN THE DEVELOPMENT OF BSCCO-2212 SUPERCONDUCTORS

M. S. Walker\* Intermagnetics General Corporation, Guilderland, New York

D. R. Dietderich Lawrence Berkeley Laboratory, Berkeley, California

**ABSTRACT** - BSCCO-2212 powders were uniformly dip coated on silver ribbons of 1 meter lengths. Short sample sandwiches of various combinations of silver and the dip coated ribbon were reacted to form possible conductor geometries for characterization and testing. Optical and Scanning Electron Microscope (SEM) examination of surfaces and cross sections showed good but still not ideal microstructures. Representative critical current densities at 6 T and 4.2 K ranged from  $8x10^3$ A/cm<sup>2</sup> for sandwich samples to  $2x10^4$ A/cm<sup>2</sup> for open-faced samples.

#### INTRODUCTION

Due to their high upper critical fields, Bi-based high-T<sub>c</sub> superconductors are promising candidates for use in high magnetic fields at low temperature. The potential candidates include conductors based on the Bi<sub>2</sub>Sr<sub>2</sub>Ca<sub>1</sub>Cu<sub>2</sub>O<sub>8</sub> (BSCCO-2212) or the Bi<sub>2</sub>Sr<sub>2</sub>Ca<sub>2</sub>Cu<sub>3</sub>O<sub>10</sub> (BSCCO-2223) compounds. There are several wire or tape manufacturing methods currently being pursued by various companies and research laboratories. Vacuumschmelze, Furukawa and Westinghouse both have active programs aimed at the development of Ag-sheathed tapes via a powder in a tube method based on the BSCCO-2212 compound.<sup>1,3</sup> Asahi Glass Company in collaboration with the National Research Institute for Metals (NRIM), SUNY Buffalo have been producing tapes of the BSCCO-2212 by the doctor-blade process.<sup>4,7</sup> Recently, Asahi, NRIM and Hitachi Cable have developed a dip-coating method to produce long lengths of conductor which have been wound into coils.<sup>8</sup> The recent critical current density (J<sub>c</sub>) achievements of these groups are greater than 1000 A/mm<sup>2</sup> can be

\*(Visiting Scientist with Lawrence Berkeley Laboratory at time of report)

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obtained for fields greater than 20T at 4.2K. The results also show that the highest critical currents are obtained from material with the best grain alignment, i.e. textured microstructure. For low temperature application (1.8 K or 4.2 K) the BSCCO-2212 tapes seems to be as good as and may be superior to the BSCCO-2223 tapes.

It has been found that the BSCCO-2223 phase can be aligned by thermomechanical methods, such as rolling and sintering. However, a deformation-andsintering schedule does not produce an aligned BSCCO-2212 microstructure. Aligned structures are only obtained by methods that have come to be called "melt growth" or "partial melt growth" processes.<sup>5</sup> These processes, as their names imply, melt part or all of the 2212 phase, which during cooling reverts to the 2212 phase in such a way as to produce an aligned structure. However, this reaction is not simple; the 2212 phase transforms to a liquid and at least three solid phases which undergo the reverse transformation to the 2212 phase on cooling as will as the loss of Bi due to the long time at high temperature.<sup>9</sup>

Of the several fabrication methods to produce long lengths of BSCCO-2212 we have adopted the dip-coating approach. Preliminary results for various tape geometries and atmospheres to control both differential thermal strain between the Ag and 2212 and Bi loss are reported as well.

#### CONDUCTOR FABRICATION

The conductor samples were fabricated by dip-coating silver ribbons with a BSCCO-2212 powder slurry, variously hanging short samples of the coated ribbons or sandwiching them between comparable additional lengths of silver on alumina platforms, baking out the organic substances, and reacting to form conductor. To prepare the slurry a nominally 99.9+% pure commercial BSCCO-2212 powder from the SSC Corporation was mixed by milling with trichloroethylene as a carrier fluid, sorbitan trioleate as a dispersant and polyvinyl butyral as a binder. The 2212 particles used were platelets approximately 2-3  $\mu$ m in length or width and 0.3  $\mu$ m in thickness, as determined by SEM examination of the green or baked conductors (described later in this section). Silver ribbons over a 1 meter long, approximately 60  $\mu$ m thick and approximately 5 mm wide were first cleaned in trichloroethylene and then dipped through the slurry using the arrangement shown schematically in Figure 1. Strongly adherent, uniform air-dried powder coatings thus obtained ranged in thickness from approximately 50  $\mu$ m to over 150  $\mu$ m per side, depending upon the starting viscosity of the fluid and rate of travel of the ribbon through the slurry. Figure 2 shows pieces from 1 meter lengths, of wider green dip-coated ribbons prepared in a similar way.

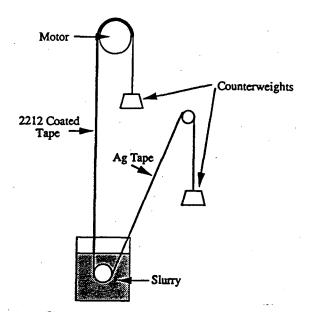


Figure 1 - Schematic of arrangement for dip-coating silver ribbons with a slurry containing BSCCO-2212 powders.

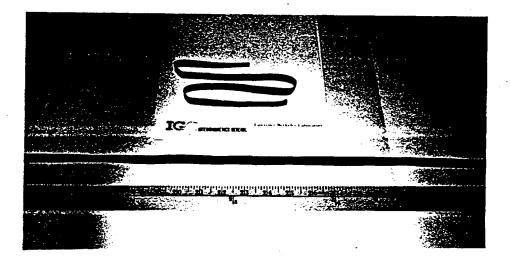
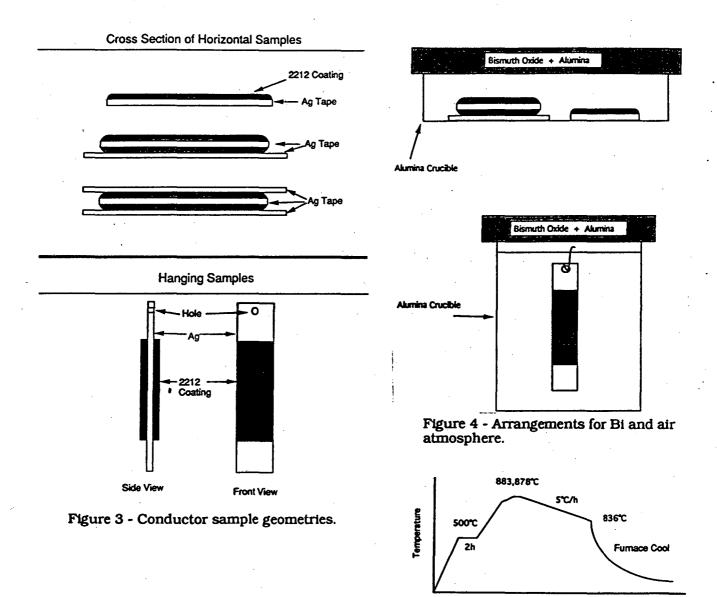
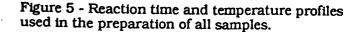


Figure 2 - Pieces from 1 meter lengths of green dip coated 13 mm wide BSCCO-2212 and silver ribbons.

Samples approximately 3 cm long were cut from the ribbons and formed into different conductor constructions and geometries, as shown in Figure 3, by either hanging them free of other contacts or placing them horizontally on silver strips to form open face or closed sandwiches under light alumina weights for reaction in air only or also in the presence of a source of bismuth vapor, as shown schematically in Figure 4. All samples were given the heat treatments shown in Figure 5. These consisted of (a) a bake out of organics at 500°C followed by, (b) partial melting at 883°C or 878°C and (c) slow cool through the 2212 formation region to 836°C.





Time

### MICROSTRUCTURE

SEM micrographs, such as that shown in Figure 6, of cracked cross sections of the green or baked-out (but not reacted) oxide layers show a relatively porous nearly uniform coating of apparently randomly orientated platelet particles. After melt processing with the heat treatment schedule given above a dense partially aligned 2212 microstructure is produced. The cross-section of a tape, Figure 7, shows that there is grain alignment near the surface and 2212/Ag interfaces, but not throughout the entire oxide thickness.

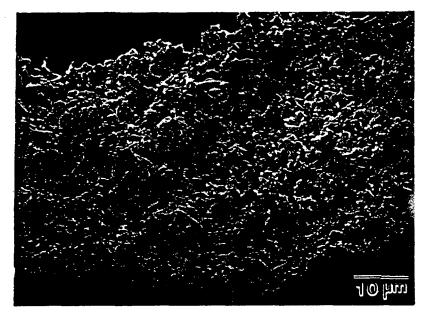


Figure 6 - Typical SEM micrograph of the cracked cross section of green or baked-out BSCCO-2212 dip coated ribbon.

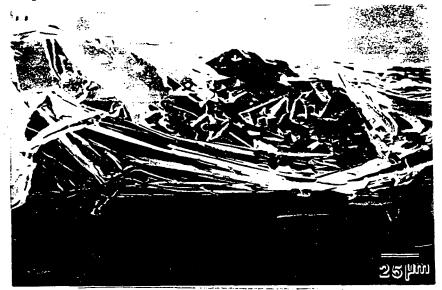


Figure 7 - Typical SEM micrograph of the cracked cross section of reacted BSCCO-2212 dip coated ribbon.

#### CRITICAL CURRENTS

Transport critical currents were measured using a standard four point method with Pb-In-Ag soldered current and voltage contacts. Open faced sandwiches and hanging samples were additionally supported by gluing onto brass strip backing plates before soldering. Figure 8 shows transport critical currents along the conductor long axis for the various sample geometries, at 4.2 K in fields transverse to the current and parallel to the flat surface of the ribbons. Corresponding critical current densities in the oxide layers ranged from  $8 \times 10^3 \text{A/cm}^2$  for sandwich samples to  $2 \times 10^4 \text{A/cm}^2$  for open-faced samples, as shown in Figure 9.

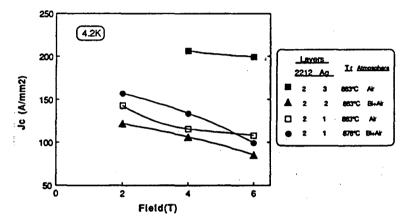


Figure 8 - Transport critical currents in the direction of the long axis of short BSCC-2212 samples of various constructions. Results are for magnetic fields transverse to the long axis and parallel to the flat face of the samples.

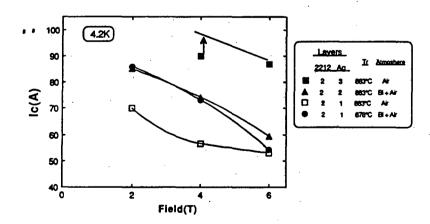


Figure 9 - Transport critical current densities in the direction of the long axis of short BSCC-2212 samples of various constructions. Results are for magnetic fields transverse to the long axis and parallel to the flat face of the samples.

#### CONCLUSIONS

Dip coating processes were established for coating meter long strips of silver with BSCCO-2212 powders. First examinations of conductor geometries showed the open faced double sandwiches to yield highest currents and current densities. There are indications that the presence of Bi atmosphere may increase critical current density in the open faced samples, particularly at the higher melt temperatures.

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LAWRENCE BERKELEY LABORATORY UNIVERSITY OF CALIFORNIA TECHNICAL INFORMATION DEPARTMENT BERKELEY, CALIFORNIA 94720

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