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Journal

Journal of Geophysical Research, 104(D20)

ISSN

0148-0227

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Publication Date

1999-10-01

DOI

10.1029/1999JD900797

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Global climatology of abundance and solar absorption of oxygen collision complexes

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Abstract. To improve our understanding of the absorption of solar radiation in the atmosphere we have characterized the spectral absorption, the spatial and temporal abundance, and the radiative forcing of the oxygen collision pairs $O_2 \cdot O_2$ and $O_2 \cdot N_2$ ($O_2 \cdot X \equiv O_2 \cdot O_2 + O_2 \cdot N_2$). The regional, vertical, seasonal, and annual patterns of $O_2 \cdot X$ abundance and radiative forcing are obtained from a general circulation model. We estimate the mean absorption by $O_2 \cdot X$, heretofore neglected in large scale atmospheric models, is $0.75\text{--}1.2 \text{ W m}^{-2}$, or 1–2% of total atmospheric solar absorption. $O_2 \cdot X$ absorption reduces surface insolation by $0.48\text{--}0.78 \text{ W m}^{-2}$ and increases the net radiative flux at the tropopause by $0.32\text{--}0.52 \text{ W m}^{-2}$. These ranges bracket the uncertainties due to spectral absorption cross sections, $O_2 \cdot N_2$ efficiency, $O_2 \cdot X$ abundance, and cloud distribution. Globally averaged, $O_2 \cdot X$ enhances absorption equally in clear and in cloudy skies.

We create a global climatology of well-mixed collision complex abundances by scaling $O_2 \cdot X$ abundance to other complexes such as $N_2 \cdot N_2$ and $O_2 \cdot Ar$. Collision complex abundance depends quadratically on the concentrations of the constituents. This dependence causes a 20% increase in $O_2 \cdot X$ abundance in the Arctic relative to the Tropics for the same sea level pressure. The variations in zonal mean $O_2 \cdot X$ abundance due to surface elevation, the annual mean meridional temperature gradient, and to seasonal temperature variations are 40, 15, and 10%, respectively. $O_2 \cdot X$ heating obeys the weakly absorbing, linear limit so it peaks at the surface in clear skies, but clouds shift this peak up by 200–300 mb on seasonal timescales. Surface albedo and clouds strongly modulate the solar forcing efficiency of $O_2 \cdot X$ (forcing per unit abundance) by altering the mean photon path length. These factors produce annual mean forcing maxima in the subtropics over bright deserts and regions of marine stratus. Inclusion of $O_2 \cdot X$ in models is likely to reduce cold biases in the summertime polar atmosphere, where $O_2 \cdot X$ contributes 2–4% of total solar heating.

1. Introduction

Our ability to balance the solar radiation budget of the atmosphere is uncertain [e.g., *Stephens and Tsay*, 1990; *Kato et al.*, 1997; *Ramanathan and Vogelmann*, 1997; *Collins*, 1998, and references therein]. Until we can perform this fundamental task, our skill at predicting the all sky radiation budget, and hence at predicting climate and climate change, will continue to be highly uncertain [e.g., *Cess et al.*, 1989]. The atmospheric collision complexes of oxygen $O_2 \cdot O_2$ and $O_2 \cdot N_2$ (henceforth $O_2 \cdot X \equiv O_2 \cdot O_2 + O_2 \cdot N_2$) have recently been identified as contributors to discrepancies between modeled and observed surface insolation [*Pfeilsticker et al.*, 1997; *Solomon et al.*, 1998]. $O_2 \cdot X$ presents a timely and interesting case for a radiative budget study because its solar absorption cross section is relatively well known, but its

abundance and absorption have not been characterized at the global scale necessary to fully describe its impact on climate.

The absorption features of oxygen collision complexes have been studied in the laboratory and in the atmosphere for many decades [*Perner and Platt*, 1980; *Solomon et al.*, 1998, and references therein]. Collisions relax the selection rules for dipole-forbidden transitions in one or both O_2 molecules, allowing for absorption during the collision. The detection of discrepancies between modeled and observed solar radiation in cloudy skies [*Cess et al.*, 1995; *Ramanathan and Vogelmann*, 1997] provoked interest in whether $O_2 \cdot X$ contributes to this discrepancy [*Erle et al.*, 1995; *Murcray et al.*, 1997] and in what its global radiative forcing characteristics are.

Pfeilsticker et al. [1997] first estimated the global annual mean solar absorption by $O_2 \cdot O_2$ in clear sky and in all sky (i.e., including clouds) conditions as 0.53 and 0.57 W m^{-2} , respectively. Two newer estimates include the effects of the powerful $1.27 \mu\text{m}$ band and of nitrogen induced absorption ($O_2 \cdot N_2$) in this band. Estimates from these studies suggest that globally and annually averaged, $O_2 \cdot X$ absorbs $0.9\text{--}1.3 \text{ W m}^{-2}$ [*Solomon et al.*, 1998] or 0.84 W m^{-2} [*Mlawer et al.*, 1998]. Thus $O_2 \cdot X$ is thought to contribute 1–

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2% of the total solar atmospheric absorption of 67 W m^{-2} (out of 342 W m^{-2} incoming) [Kiehl and Trenberth, 1997]. These estimates of O₂·X absorption are based on single column line-by-line, multiple-scattering radiative transfer models which have been extensively validated against atmospheric observations by high spectral resolution instruments. The disparity in the latter two estimates is mostly due to uncertainties in absorption cross sections for O₂·X bands in the near infrared and, to a lesser extent, from extrapolating single-column results to global annual means.

We use the term “radiative forcing” to denote various measures of the influence of atmospheric heating by O₂·X on the climate system. Line-by-line estimates of O₂·X absorption are too computationally expensive to convey the rich regional and seasonal behavior of O₂·X radiative forcing caused by variations in temperature, pressure, elevation, clouds, surface albedo, and mixing ratios of other radiatively active atmospheric constituents. A general circulation model (GCM) is best suited to provide self-consistent, time-varying realistic distributions of O₂·X forcing. Employing a GCM allows us to extend prior studies and to provide additional forcing metrics, including clear versus cloudy sky forcing, surface forcing, and net forcing at the tropopause. These metrics allow us to compare the radiative forcing by O₂·X to other absorbers, including greenhouse gases. These comparisons, in turn, help to prioritize implementation of O₂·X physics in large-scale atmospheric models. It is worthwhile noting that in contrast to anthropogenically influenced greenhouse gas concentrations, O₂·X abundance is not changing.

To our knowledge no prior study has constructed a systematic global climatology for any collision complex (or dimer). Moreover, the regional, seasonal, and vertical patterns of O₂·X abundance generalize to all complexes formed from well-mixed gases. Thus these patterns, which depend mainly on the meridional temperature gradient and orographic features, apply to all well-mixed collision complexes, for example, N₂·N₂, O₂·Ar, and O₂·CO₂. The patterns of O₂·X radiative forcing, which depend additionally on insolation, surface albedo, and cloud cover, only apply to other complexes with absorption features similar to those of O₂·X. Thus, by characterizing O₂·X abundance and forcing, we automatically obtain insight into all well-mixed collision complexes.

Establishing the spatial and temporal patterns of O₂·X abundance and forcing will also help future studies ascertain whether unexplained atmospheric solar absorption is consistent with absorption by collision complexes (or dimers) of well-mixed molecules. The existence of complexes besides O₂·X which cause significant solar radiative forcing is highly speculative. All significant structured atmospheric absorption measured in the atmosphere for large portions of the solar spectrum has been convincingly attributed to known absorbers [Pfeilsticker et al., 1997; Solomon et al., 1998; Mlawer et al., 1998; Vogelmann et al., 1998]. Moreover, a number of recent field studies explains all observed broadband clear sky solar absorption to within the uncertainties in the observations and models [Conant et al., 1997;

Zender et al., 1997; Ramanathan and Vogelmann, 1997; Jing and Cess, 1998]. However, Kato et al. [1997] and Halthore et al. [1998] find inexplicable discrepancies between models and observations of the clear sky diffuse field, and many independent studies show enhanced cloudy sky solar absorption relative to model predictions on both local and global scales [Cess et al., 1995; Zender et al., 1997; Collins, 1998]. Since O₂·X has been proposed as a candidate for explaining some of these discrepancies [Erle et al., 1995; Murcray et al., 1997], this study attempts to bound the role of O₂·X in explaining enhanced absorption in clear and cloudy skies.

The remainder of this study is organized as follows: Section 2 presents the nomenclature of collision complexes and summarizes the laboratory measurements of O₂·X cross sections. Section 3 illustrates the modeled spectral characteristics of O₂·X absorption during a field campaign and shows its net vertical heating signature in an idealized atmosphere. Section 4 presents the regional, seasonal, and vertical climatologies of O₂·X abundance and forcing and generalizes O₂·X abundance to other well-mixed collision complexes. Section 5 summarizes the results and discusses their implication for future studies of the atmosphere.

2. Characterization of Abundance and Absorption Cross Section

Quantifying the global radiative forcing of a gaseous molecular species *i* requires characterization of both its concentration $n_i(x, y, z, t)$ molecule cm⁻³ and its absorption cross section $\sigma_i(\lambda)$ cm² molecule⁻¹. The convolution of these two quantities with the instantaneous radiation field yields the instantaneous forcing due to the gas. However, careful laboratory measurements provide convincing evidence that for atmospheric temperatures and solar wavelengths, O₂·X absorption is due to transient collision complexes rather than bound dimers (or van der Waals complexes) [Greenblatt et al., 1990; Orlando et al., 1991]. Since O₂·X is not a bound molecule, we begin by defining properties analogous to n_i and σ_i , but that apply to collision complexes.

2.1. Abundance

The “abundance” $n_{X \cdot Y}$ of a collision complex of molecules X and Y is analogous to a traditional number concentration of a molecular species. We define $n_{O_2 \cdot X} \equiv n_{O_2} n_X$ where n_{O_2} and n_X are the number concentrations of O₂ and of its collision partner (N₂ or O₂, unless otherwise specified) in molecule cm⁻³, respectively. Thus the dimensions of $n_{O_2 \cdot X}$ are molecule² cm⁻⁶. By definition, $n_{O_2 \cdot X}$ is linearly proportional to the frequency of collisions of O₂ with X. To graphically depict the total column abundance of O₂·X, we will present the vertical column paths $N_{O_2 \cdot X}$ defined by

$$\begin{aligned} N_{O_2 \cdot O_2} &\equiv \int_0^{\text{TOA}} n_{O_2 \cdot O_2} dz = \int_0^{\text{TOA}} n_{O_2}^2 dz \\ N_{O_2 \cdot N_2} &\equiv \int_0^{\text{TOA}} n_{O_2} n_{N_2} dz = \gamma N_{O_2 \cdot O_2} \end{aligned} \quad (1)$$

where the integration over z (height) is from the surface ($z = 0$) to the top of atmosphere (TOA ≈ 1.5 mb due to GCM vertical discretization). Likewise, the column path of a molecule X, in molecule cm⁻², will be denoted by N_X . We assume in (1) that the volume mixing ratios of N₂ and O₂ are constant in the Earth's atmosphere and that their ratio is the constant $\gamma \equiv n_{N_2}/n_{O_2} = 3.72787$ [Goody and Yung, 1989, p. 9]. Since the vertical profile of $n_{O_2 \cdot X}$ varies with the square of atmospheric density, variations in γ above the stratosphere have negligible impact on the column abundance, and thus forcing, of O₂·X.

2.2. Absorption Cross Sections

The absorption bands of the O₂·O₂ and O₂·N₂ complexes have been studied for decades. Solomon *et al.* [1998] give an excellent review of laboratory research on oxygen collision complex absorption. Here we summarize only the research necessary to understand the physical parameterizations and uncertainties present in this study.

Laboratory measurements of O₂·X absorption infer a binary absorption coefficient $\sigma^b(\lambda)$ from the quadratic dependence of measured absorption on O₂ concentration, n_{O_2} . Newnham and Ballard [1998] provide a detailed description of this type of measurement and its uncertainties. The dimensions of $\sigma^b(\lambda)$ are cm⁵ molecule⁻². Thus, for a given concentration n_{O_2} molecule cm⁻³ and path length Δz centimeters of collision partner (O₂ or N₂), the optical depths $\tau_{O_2 \cdot X}$ due to O₂·X at wavelength λ are

$$\tau_{O_2 \cdot O_2}(\lambda) = \sigma_{O_2 \cdot O_2}^b(\lambda) n_{O_2}^2 \Delta z \quad (2)$$

$$\begin{aligned} \tau_{O_2 \cdot N_2}(\lambda) &= \sigma_{O_2 \cdot N_2}^b(\lambda) n_{O_2} n_{N_2} \Delta z \\ &= \gamma \epsilon_{N_2} \sigma_{O_2 \cdot O_2}^b(\lambda) n_{O_2}^2 \Delta z \end{aligned}$$

$$\tau_{O_2 \cdot N_2}(\lambda) = \gamma \epsilon_{N_2} \tau_{O_2 \cdot O_2}(\lambda) \quad (3)$$

where (3) applies to the 1.27 μm band only (see Plate 1) and ϵ_{N_2} , discussed below, is the efficiency of N₂ (relative to O₂) at inducing O₂ absorption through collisional processes (i.e., $\sigma_{O_2 \cdot N_2}^b = \epsilon_{N_2} \sigma_{O_2 \cdot O_2}^b$).

Plate 1 shows the solar absorption continua for O₂·X measured by Greenblatt *et al.* [1990] at ~ 0.15 nm resolution for $0.335 < \lambda < 1.137 \mu\text{m}$. The 1.27 μm band has not been adequately examined with modern laboratory techniques, so we adopt the cross sections for this band recommended by Solomon *et al.* [1998]. On the basis of a number of earlier measurements, Solomon *et al.* constructed the 1.27 μm band by stretching the measured 1.06 μm band to a full width at half maximum (FWHM) of 180 cm⁻¹ and scaling its integrated band intensity by 1.6. The cross sections of O₃ and O₂ [World Meteorological Organization, 1985], and of NO₂ [Davidson *et al.*, 1988], are shown for comparison. The red curves show that significant O₂·O₂ solar absorption occurs in two spectral regions: $0.3 < \lambda < 0.7 \mu\text{m}$ and $1.0 < \lambda < 1.4 \mu\text{m}$. O₂·O₂ visible absorption is due to simultaneous collision induced transitions in both O₂ molecules, while near-infrared absorption involves only one O₂ transition.

Newnham and Ballard [1998] recently measured $\sigma_{O_2 \cdot O_2}^b$ for $0.455 < \lambda < 0.830 \mu\text{m}$ and compared their measure-

ments to those of Greenblatt *et al.* [1990] (Plate 1). The maximum $\sigma_{O_2 \cdot O_2}^b$ and the integrated band intensities of most bands differed by $< 20\%$ between the two sets of measurements. Greenblatt *et al.* found the integrated band intensity of the bands centered at 360, 380, and 477 nm varied by $< 15\%$ from $T = 296$ K to 196 K. Newnham and Ballard found the integrated band intensity of four bands from 532–630 nm varied by $< 10\%$ from $T = 283$ K to 223 K but that the intensity of the 477 nm band decreased by 30% over this temperature range. Thus the two studies disagree on the temperature dependence of the the only band examined in both studies, the 477 nm band. Why the disagreement at 477 nm exceeds the stated error bars of the two studies is unexplained. Balloon-borne observations of the 447, 532, and 577 nm bands by Osterkamp *et al.* [1998] recorded a 20% decrease in $\sigma_{O_2 \cdot O_2}^b$ as the temperature increased from $T = 200$ K to 300 K. Note that this decrease is in contrast to the increase of $\sigma_{O_2 \cdot O_2}^b$ with T measured by Newnham and Ballard in the 477 nm band. Osterkamp *et al.* conclude that O₂·O₂ may be a metastable complex with some dimer-like properties. For simplicity, our study assumes $\sigma_{O_2 \cdot X}^b$ is constant with temperature. However, we should remain open to the possibility that $\sigma_{O_2 \cdot X}^b$ varies significantly with temperature for some bands.

The parameter ϵ_{N_2} defines the efficiency of N₂ relative to O₂ as a partner for inducing absorption in the O₂ 1.27 μm band. The upper and lower blue curves centered at $\lambda = 1.27 \mu\text{m}$ in Plate 1 correspond to $\sigma_{O_2 \cdot N_2}^b$ assuming $\epsilon_{N_2} = 0.1$ and 0.3, respectively. All prior studies show $0.1 \leq \epsilon_{N_2} \leq 0.3$, and the preponderance suggest $\epsilon_{N_2} = 0.3$ (see discussion by Solomon *et al.* [1998]).

Solomon *et al.* [1998] used differential optical absorption spectroscopy at large zenith angles to show their high spectral resolution (≈ 0.6 nm) in situ measurements agree with the laboratory cross sections of Greenblatt *et al.* [1990] (Plate 1) to within 0.3% for the O₂·O₂ band at 0.63 μm . Mlawer *et al.* [1998] recently inferred $\sigma_{O_2 \cdot X}^b$ for $1.0 < \lambda < 1.57 \mu\text{m}$ from field measurements with a high-resolution (0.6 cm⁻¹) interferometer. They inferred comparable peak intensities in the 1.06 and 1.27 μm O₂·X bands, but narrower FWHMs, than previous investigators [Solomon *et al.*, 1998]. These narrower FWHMs would reduce absorption in the near-infrared O₂·X bands by about 30% relative to Greenblatt *et al.* and Solomon *et al.*

The uncertainties in $\sigma_{O_2 \cdot X}^b(\lambda)$ contribute roughly half of the uncertainty in our estimate of global O₂·X forcing. The uncertainty in $\sigma_{O_2 \cdot X}^b(\lambda)$ is $< 10\%$ for $0.335 < \lambda < 1.137 \mu\text{m}$ [Greenblatt *et al.*, 1990]. Uncertainty in 1.27 μm band absorption is $\sim 30\%$. This was estimated by extrapolating to global scale the difference between single-column simulations employing the continua proposed by Solomon *et al.* [1998] and by Mlawer *et al.* [1998]. For consistency, our study adopts the same cross sections as Solomon *et al.*, but we note two choices that keep our results for total O₂·X absorption closer to those of Mlawer *et al.* First, we use $\epsilon_{N_2} = 0.2$, a somewhat conservative value. Second, we neglect the O₂·O₂ absorption band near 1.58 μm . This band absorbs $\sim 9\%$ as much as the 1.27 μm band [Mlawer *et al.*,

1998]. Appending the 1.58 μm continuum to the Solomon et al. cross sections would increase our computed total O₂·X solar forcing by $\sim 4\%$. Other absorption features of O₂·X are negligible in terms of the total absorbed solar irradiance. These features, and those of many other well-mixed collision complexes, are summarized in Table 1.

3. High Resolution Single Column Simulations

O₂·X complexes are a relatively recent addition to the longstanding group of solar absorbers considered in large-scale atmospheric models: H₂O, O₃, O₂, NO₂, and CO₂. As such, the spectrally and vertically dependent features of O₂·X absorption have not yet been thoroughly characterized in the literature. We now analyze these features in high-resolution, single-column simulations in order to better understand their implications on global-annual scales.

3.1. Spectral Characteristics and Overlap

Two previous studies characterized aspects of the spectral overlaps between O₂·X and more familiar absorbers. Pfeilsticker et al. [1997, Figure 2] show O₂·X spectral features at surface and TOA for $335 < \lambda < 1137 \text{ nm}$. Solomon et al. [1998, Figure 2] show residual surface absorption spectra of O₂·O₂, O₂, and H₂O in the region $610 < \lambda < 680 \text{ nm}$. To complement these studies, our Plate 2 shows the modeled absorption optical depth $\tau_{\text{abs}}(\lambda)$ of all significant gaseous solar absorbers for $200 < \lambda < 1400 \text{ nm}$. The spectra represent local solar noontime (solar zenith angle $\theta = 45.128^\circ$) for October 15, 1995, at the Clouds and Radiation Testbed (CART) site in Oklahoma. This was a pristine day with minimal aerosol burden ($0.04 < \tau_{0.5\mu}^{\text{aer}} < 0.05$, not shown), low precipitable water (11.6 kg m^{-2}), low column O₃ (233 DU), and estimated column NO₂ of $5 \times 10^{19} \text{ m}^{-2}$. Zender et al. [1997] describe the multistream narrow band (10 cm^{-1}) Malkmus model, the experimental uncertainties, and all the input data in more detail.

Plate 2 isolates the solar absorption continuum (using $\sigma_{\text{O}_2\cdot\text{X}}^b$ from Plate 1) of each gaseous absorber considered in the model. O₂·X shares with H₂O the distinction of overlapping with all the significant solar absorbers. The strongest visible bands of O₂·O₂ overlap with the center of the O₃ Chappuis bands near 0.6 μm . The O₂·O₂ band at 0.630 μm overlaps the discrete O₂ γ band at 0.629 μm . The strong 1.06 μm band partially overlaps a saturated water vapor band, but its center is the strongest nonoverlapped feature of O₂·O₂ in the spectrum. Absorption at 1.27 μm shows strong overlap among the oxygen infrared band, O₂·O₂, and O₂·N₂. Although 1.27 μm is in a water vapor window, both wings of the O₂·X bands centered there overlap with weak CO₂ bands and with the wings of H₂O bands. This overlap with H₂O lessens somewhat in dry air masses where column moisture is $< 5 \text{ kg m}^{-2}$ (e.g., Antarctica), but O₂·X absorption is only weakly sensitive to variations in H₂O (see section 4.3). As Plate 2 shows, contributions of O₂·N₂ and O₂·O₂ to the total 1.27 μm continuum are nearly equal if $\epsilon_{\text{N}_2} = 0.2$. Using $\epsilon_{\text{N}_2} = 0.3$ (and weaker $\sigma_{\text{O}_2\cdot\text{X}}^b(1.27 \mu\text{m})$), Mlawer et al. [1998] found O₂·N₂ absorption may exceed O₂·O₂ absorption in this band by 15%.

The O₂·O₂ band at 0.477 μm occupies a relatively transparent spectral region where it is often stronger than any other gaseous absorber, i.e., O₃ or NO₂. This band is discernible in routine optical depth spectra taken at the CART site and other locations [Michalsky et al., 1999]. Note that neglecting O₂·O₂ absorption in its stronger bands could bias operational retrievals of other absorber amounts (e.g., O₃, NO₂, or aerosol), leading to overestimates of order 0.01 optical depths. Fortunately, we are aware of no operational retrievals that are compromised by neglect of O₂·O₂ absorption. For this particular profile, we estimate O₂·X causes 2.3 W m^{-2} , or 1.4%, of the total modeled atmospheric solar absorption of 164 W m^{-2} . Accounting for this absorption reduces the discrepancy between noontime modeled and observed surface insolation (773 and 760 W m^{-2} , respectively) from 13.0 to 11.2 W m^{-2} [Zender et al., 1997].

3.2. Vertical Heating Profile

To illustrate the vertical profile of O₂·X heating, we simulated “climatological” clear and cloudy sky, aerosol-free heating profiles for a 92 level midlatitude summer atmosphere [Briegleb, 1992]. Figure 1 shows the resulting vertical profile of total solar heating for the clear sky profile, the O₂·X contribution to this clear sky heating, and the O₂·X cloudy sky heating. The solar zenith angle ($\theta = 60^\circ$), surface albedo ($A = 0.2$), and cloud properties (100 mb thick liquid stratus cloud centered at 850 mb, condensate path 100 g m^{-2}) represent climatological atmospheric properties. The O₂·X heating profiles are multiplied by a factor of 100 for clarity.

The clear sky O₂·X heating increases rapidly with pressure, peaking at the surface. Thus O₂·X clear sky heating weakly destabilizes the atmospheric column. At moderate zenith angles, O₂·X absorption is relatively weak (i.e., $\tau_{\text{O}_2\cdot\text{X}} < 0.1$) and is not dominated by overlap with other gases (Plate 2). Heating by a well-mixed molecule in this nonoverlapped, optically thin limit is constant with air density. In contrast, it is easy to show that in this limit, heating by a well-mixed collision complex (or dimer) increases linearly with air density. The high correlations of the O₂·X heating in Figure 1 with air density (not shown) confirm that O₂·X heating obeys the weakly absorbing linear limit in both clear and cloudy skies. The absorbed energy actually increases quadratically with dry air density, i.e., linearly with O₂·X abundance (1), but one factor of density (the heat capacity of air in $\text{J m}^{-3} \text{ K}^{-1}$) is removed in converting from a volumetric heating (W m^{-3}) to an absolute heating rate (K d^{-1}). We note that O₂·X is the only significant solar absorber with this vertical signature. The O₂·X heating rates in Figure 1 do not appear linear because they are expressed as functions of pressure rather than density.

Figure 1 shows that clouds significantly enhance O₂·X heating above cloud top and reduce heating beneath cloud top (dotted line). Thus clouds shift the O₂·X heating profile maximum upward to the level of the cloud top. This feature appears prominently in climatological averages of O₂·X heating shown in section 4.4.

In summary, O₂·X bands are present in much of the visible spectrum and in important near-infrared windows. These

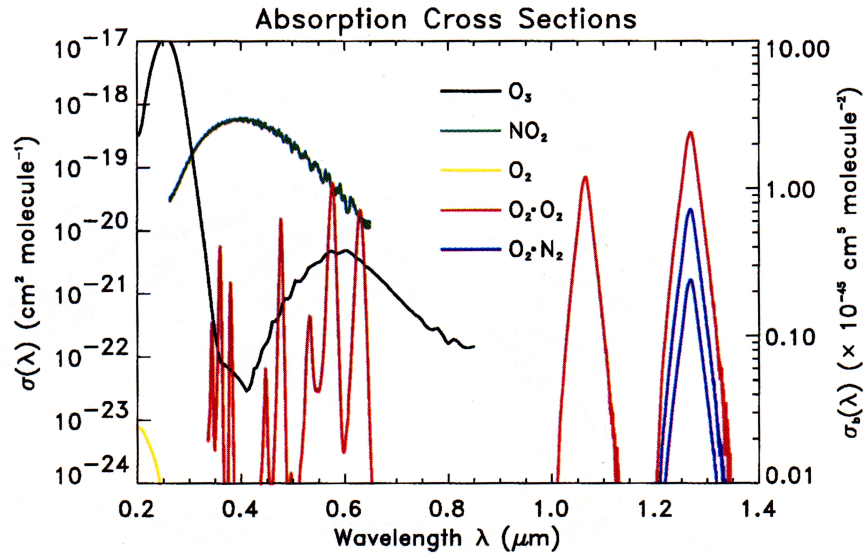


Plate 1. Solar absorption continua of O₂-X, O₃, O₂, and NO₂. Continua of O₂-O₂ (red) and O₂-N₂ (blue) are plotted as binary absorption coefficients $\sigma_{\text{O}_2\text{-X}}^b$ (cm⁵ molecule⁻²) on the right-hand axis. Continua of O₃ (black), NO₂ (green), and O₂ (yellow) are plotted as traditional cross sections σ (cm² molecule⁻¹) on the left-hand axis.

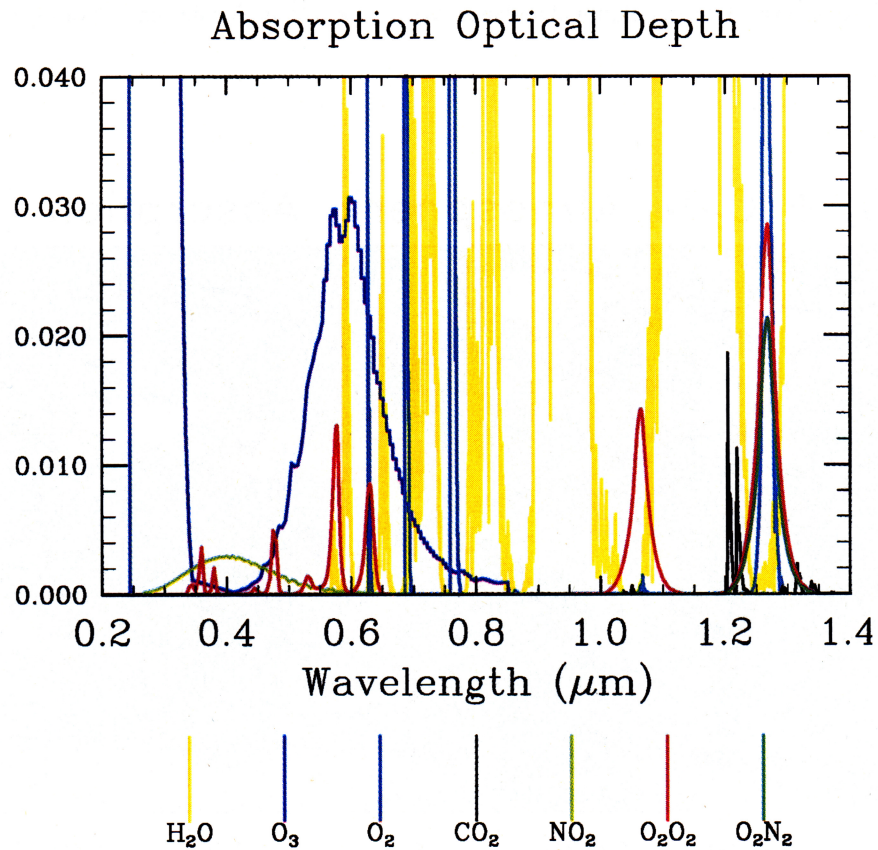


Plate 2. Decomposition of simulated absorption optical depth $\tau_{\text{abs}}(\lambda)$ at the CART site at local solar noon on October 15, 1995. Gaseous absorbers included are O₂-O₂ (red), O₂-N₂ (green), H₂O (yellow), O₃ (blue), O₂ (light blue), NO₂ (light green), and CO₂ (black). The line spectra of H₂O, CO₂, and O₂ are averaged over 10 cm⁻¹.

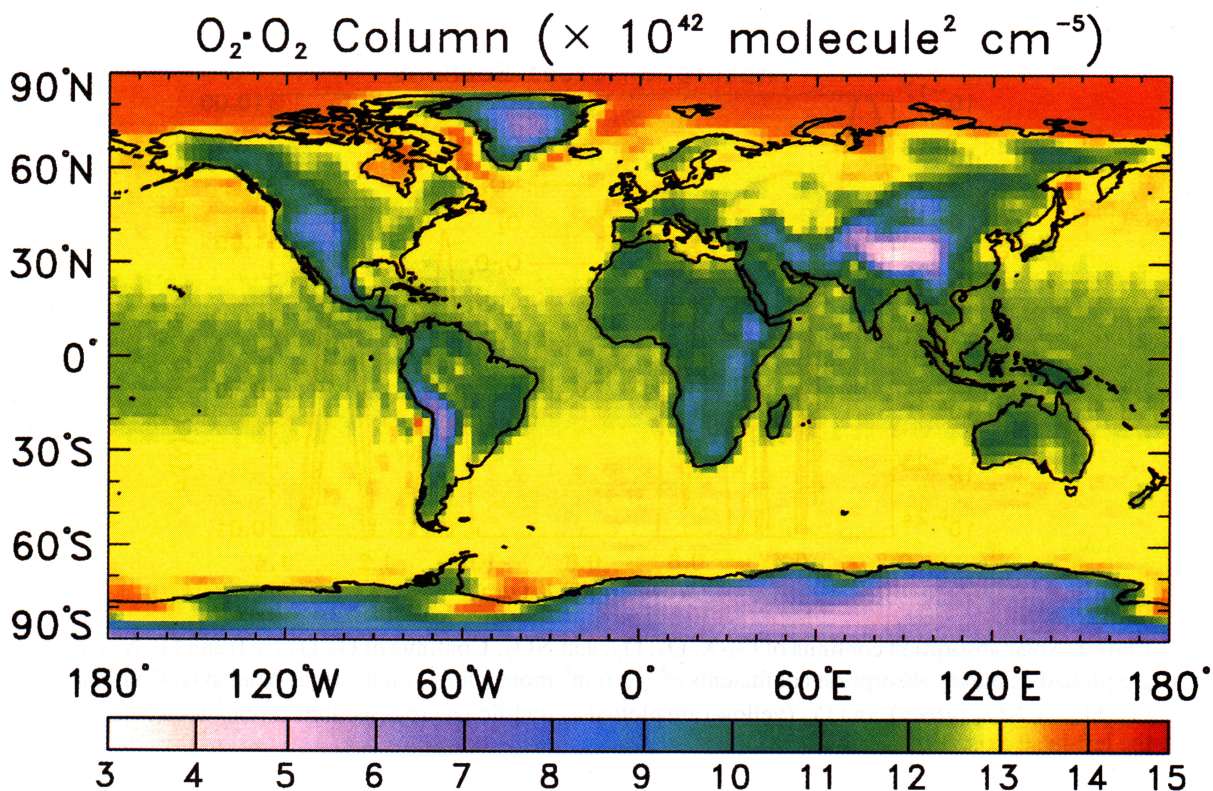


Plate 3. Annual mean column abundance $N_{O_2 \cdot O_2}$ ($\times 10^{42}$ molecule² cm⁻⁵) of O₂·O₂.

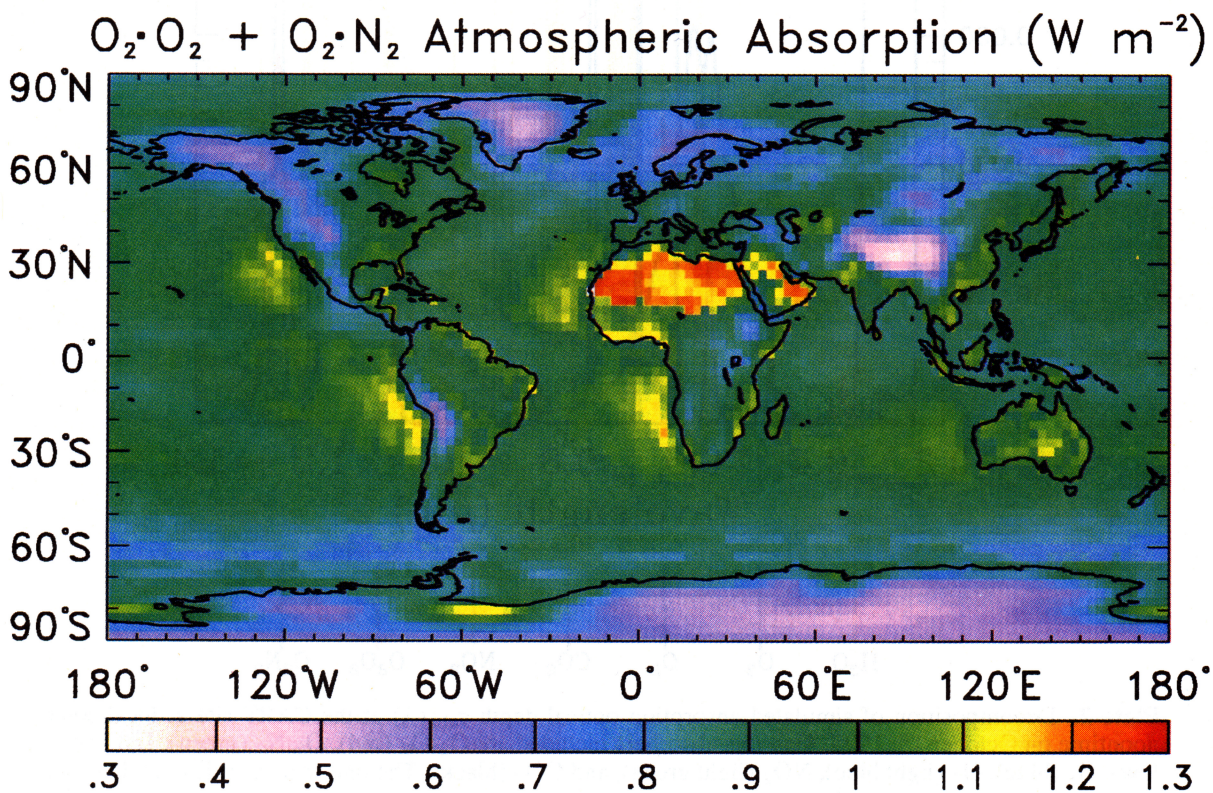


Plate 4. Annual mean increase in column atmospheric absorption (W m⁻²) due to O₂·X.

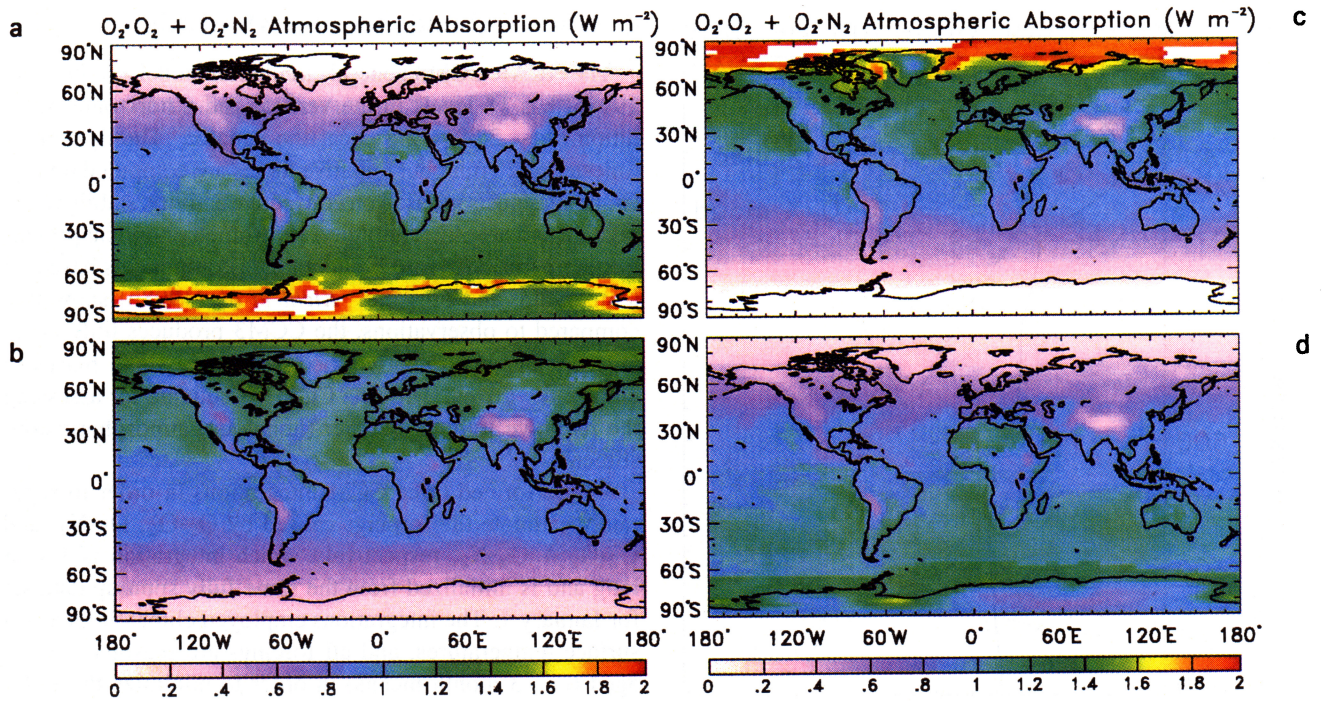


Plate 5. Seasonal mean increase in atmospheric solar absorption (W m^{-2}) due to $\text{O}_2 \cdot \text{X}$. (a) December–February (DJF). (b) March–May (MAM). (c) June–August (JJA). (d) September–November (SON). White regions exceed maximum scale value by $< 0.7 \text{ W m}^{-2}$.

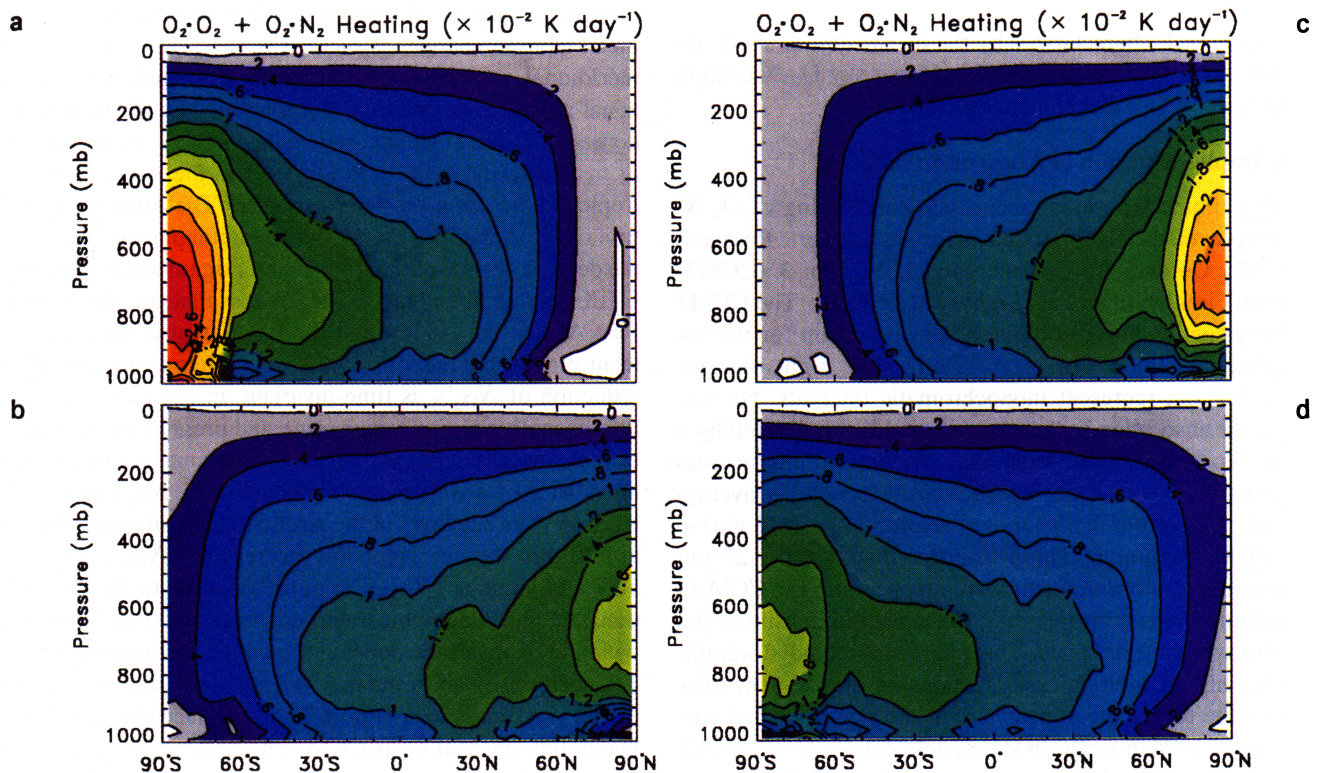


Plate 6. As in Plate 5, but for the vertical profile of the heating ($\times 10^{-2} \text{ K d}^{-1}$) due to $\text{O}_2 \cdot \text{X}$.

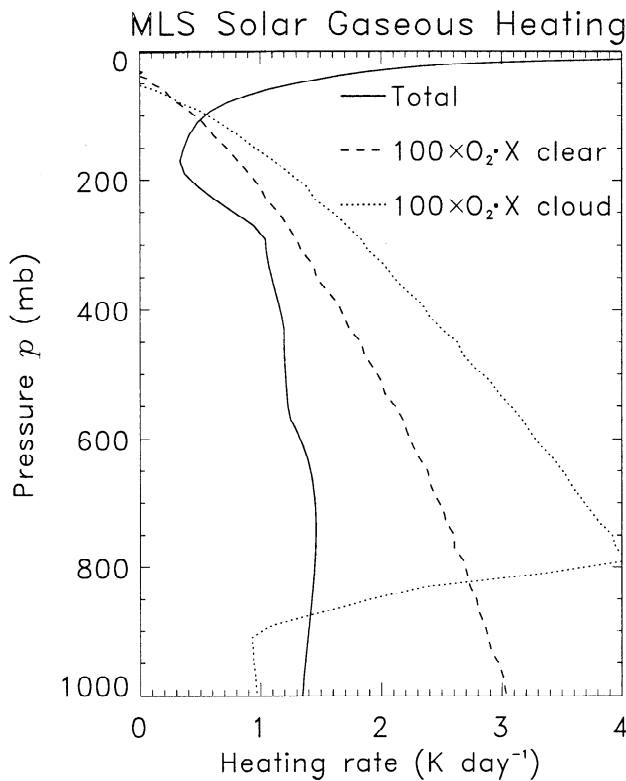


Figure 1. Vertical profile of solar gaseous heating (K d^{-1}) for a standard, climatological midlatitude summer atmosphere. Shown are the total solar heating rate in clear sky (solid) and, exaggerated by a factor of 100, the O₂·X contributions to the heating in clear sky (dashed) and in the presence of a 100 g m^{-2} , 100 mb thick stratus cloud (dotted).

bands are well characterized by the linear limit of small absorber paths. The vertical profile of clear sky O₂·X heating weakly decreases the atmospheric stability.

3.3. Implementation of Absorption in GCM

To estimate the global abundance and forcing of O₂·X, we employed the National Center for Atmospheric Research (NCAR) Community Climate Model, Version 3 (CCM3) general circulation model [Kiehl *et al.*, 1998a]. The CCM3 employs an 18 bin δ -Eddington approximation for the solar spectral region [Briegleb, 1992]. The eight bins for $\lambda < 0.7 \mu\text{m}$ represent monochromatic intervals, but water vapor absorption in the near-infrared is represented by a seven bin k -distribution technique. We created binary cross sections $\bar{\sigma}$ for each CCM spectral bin by spectrally averaging $\sigma_{\text{O}_2 \cdot \text{X}}^b(\lambda)$ from the laboratory resolution ($\sim 0.15 \text{ nm}$) to the CCM resolution. The near-infrared 1.06 and 1.27 μm continuum absorption bands were mapped into the CCM k -distribution bins corresponding to the mean intensity of the nearest water vapor bands. This procedure included weighting the high-resolution cross sections by the incident solar flux at the top of the atmosphere in the more detailed narrow band multistream (NBM) model [Zender *et al.*, 1997]. In offline single-column comparisons, the CCM3 radiation model and the NBM model O₂·X absorption agreed to within 10% for standard atmospheric conditions ranging from Arctic to Tropical, so no tuning of the $\bar{\sigma}$ was required.

4. Global Abundance and Absorption

The NCAR CCM3 general circulation model [Kiehl *et al.*, 1998a] has 18 levels in the vertical and a horizontal resolution of $\sim 2.8^\circ \times 2.8^\circ$ near the equator. The dynamical timestep is 20 min, but radiative fluxes are computed hourly. Cloud diagnosis is based on convective activity, stability, and relative and absolute humidity. Hydrometeor optical properties are partitioned into liquid and ice components based on temperature and pressure. Kiehl *et al.* [1998b] show that compared to observations, the CCM3 produces little or no bias in clear sky radiative fluxes and only small biases in cloudy sky fluxes. Thus the CCM3 is suitable for assessing the radiative forcing of other (noncloud) atmospheric constituents.

We performed three CCM integrations in order to separately estimate the forcings caused by O₂·O₂, O₂·N₂, and O₂·O₂ + O₂·N₂, respectively. Each integration is 1 year long and is initialized from the same fully spun up data set for January 1. The integrations employed climatological sea surface temperatures, and all forcings are seasonally averaged (i.e., 3 month means) in order to minimize the influence of variability on the results. The solar radiation physics were called twice each solar radiation time step (60 min): once with and once without O₂·X heating. O₂·X forcing is the difference between these values. We archived O₂·X abundance and forcing every time step (20 min), but these forcings were not allowed to affect the predicted climate.

4.1. Annual Mean Abundance and Absorption

Collision complex abundance is a statistic of the second moment of dry air density (1). Plate 3 shows the annual mean column abundance of O₂·O₂, $N_{\text{O}_2 \cdot \text{O}_2}$. The major meridional gradient is caused by the poleward decrease in zonal mean temperature. Cold polar air of a given mass is denser than warm tropical air. Since $n_{\text{O}_2 \cdot \text{O}_2}$ is the square of n_{O_2} (1), $N_{\text{O}_2 \cdot \text{X}}$ is $\sim 20\%$ greater in the Arctic than the tropics at the same sea level pressure. The same would be true of the Antarctic, except the Antarctic plateau displaces the densest portion of the troposphere, significantly reducing O₂·X column abundance. The Tibetan Plateau, Greenland, East Africa, the Andes, and the Rockies also show significant orographic reduction in $N_{\text{O}_2 \cdot \text{X}}$. Note that orographic reduction of $N_{\text{O}_2 \cdot \text{X}}$ is time-invariant, in contrast to effects of seasonally varying temperature and pressure in the extratropics, discussed below. Whereas the zonal mean column abundance of a well-mixed molecule (e.g., N_{O_2}) decreases by 30% from sea level in the Arctic to the Antarctic plateau at 700 mb, the corresponding decrease in $N_{\text{O}_2 \cdot \text{X}}$ exceeds 50%. In summary, O₂·X is more sensitive to the thermodynamic and orographic environment than O₂ or N₂ alone, owing to its quadratic dependence on these constituents.

Uncertainties in simulated $N_{\text{O}_2 \cdot \text{X}}$ stem from biases in model pressure p , temperature T , and surface elevation. The maximum seasonal mean CCM temperature bias in the lower troposphere is $< 6 \text{ K}$ [Hack *et al.*, 1998]. The maximum seasonal mean CCM grid point surface pressure bias is $< 1\%$ [Hurrell *et al.*, 1998]. Biases in column O₂ arising from the discretization of surface elevation do not ex-

ceed 1%. Assuming the worst case scenario (errors are additive) yields a conservative uncertainty in seasonal grid point mean N_{O_2} and N_{N_2} of $\pm 4\%$, implying uncertainty bounds for $N_{O_2 \cdot O_2}$ and $N_{O_2 \cdot N_2}$ of $\pm 8\%$. The maximum uncertainty is over the Antarctic shelf and the Tibetan Plateau, regions where variability and differences with climatological analyses are relatively large [Hurrell *et al.*, 1998]. The simulated annual mean surface pressure is 984.5 mb, in excellent agreement with 984.9 mb from National Centers for Environmental Prediction (NCEP) reanalyses. This confirms that grid point biases in $N_{O_2 \cdot X}$ are generally random, not systematic, and we place the global mean bias conservatively at $\pm 2\%$. As will be described below, grid point biases in simulated O₂·X forcing are generally dominated by uncertainties in absorber cross section and cloud vertical distribution, not O₂·X abundance.

The regional abundance of any well-mixed collision complex X·Y is determined by scaling Plate 3 by the product of the ratios n_X/n_{O_2} and n_Y/n_{O_2} . Table 1 summarizes the scale factors (relative to O₂·O₂) and global annual mean abundances of many well-mixed collision complexes. The scale factors were computed using atmospheric abundances of Goody and Yung [1989, p. 9], corrected to present-day values in the case of CO₂ (355 ppm) and CH₄ (1.7 ppm).

The final column of Table 1 lists the approximate centers of the collision induced absorption bands due to each complex, apart from the O₂·X bands shown in Plate 1. Taken together, such bands appear in the spectrum from the Herzberg continuum to the far infrared. In addition to the bands shown in Plate 1, collisions involving O₂·O₂ and O₂·N₂ induce oxygen absorption in the Herzberg continuum from 195 to 250 nm and in the oxygen (forbidden) fundamental vibration band at 6.42 μm [Shardanand and Rao, 1977; Orlando *et al.*, 1991]. Both O₂·N₂ and N₂·N₂ induce absorption in the nitrogen (forbidden) fundamental band at 4.29 μm . Insolation in these spectral regions is too weak to allow significant solar absorption in absolute terms. However, it may be important to account for this absorption when retrieving atmospheric parameters such as temperature from remotely sensed spectra, for example, in the CO₂ 4.3 μm band [Goody and Yung, 1989, p. 192].

Plate 4 shows the annual mean increase in atmospheric absorption due to O₂·X. The annual mean forcing approximately resembles the abundance (Plate 3), though some new features are evident. First note the enhanced atmospheric absorption above bright, low surfaces (desert, stratus clouds, ice). Since the spectral optical depth of most O₂·X bands is < 0.05 (Plate 2), absorption in these bands is in the linear limit where a change in absorber path causes a proportionate change in atmospheric absorption. Reflection from low, bright surfaces increases photon path lengths in the O₂·X-rich lower atmosphere. This increases the mean O₂·X path and thus the O₂·X absorption. However, reflection from high clouds (e.g., tropical cirrus) prevents photons from reaching the lower atmosphere and reduces the mean O₂·X path traversed by these photons. This reduces O₂·X absorption and causes the local minima in O₂·X forcing in the Intertropical Convergence Zone (ITCZ) region. The zonal annual mean atmospheric absorption due to O₂·X is $\sim 0.9 \text{ W m}^{-2}$ and does not vary strongly with latitude (not shown).

It is interesting to note that geographic structure of O₂·N₂ absorption closely resembles that of O₂·O₂ absorption, so that their sum, Plate 4, adequately represents the structure of each. Since O₂·N₂ absorption occurs entirely in the 1.27 μm band, it is enhanced relative to O₂·O₂ over vegetated surfaces but diminished over sea-ice and ocean due differences between the near-infrared and visible albedos of these surfaces. The magnitude of global mean O₂·N₂ absorption is proportional to the assumed N₂ efficiency. This study assumes $\epsilon_{N_2} = 0.2$, which leads to O₂·N₂ forcing being 20% of O₂·O₂ absorption. As discussed in section 2.2, ϵ_{N_2} may be 0.3, in which case O₂·N₂ forcing would be 30% of O₂·O₂ forcing.

It is useful to distinguish atmospheric absorption from other common metrics of radiative forcing. Table 2 summarizes the global mean statistics of O₂·X atmospheric forcing, surface forcing, column radiative forcing, and net radiative forcing at the tropopause. The first row, atmospheric forcing, is the mean increase in atmospheric absorption due to O₂·X (Plate 4), 0.93 W m^{-2} . For comparison, the all sky solar absorption due to H₂O, O₃, O₂, and CO₂ is $\sim 43, 14,$

Table 1. Global Annual Mean Abundances of Well-Mixed Collision Complexes

Complex X·Y	Scale $N_{X \cdot Y}/N_{O_2 \cdot O_2}$	Abundance, $\text{molecule}^2 \text{ cm}^{-5}$	Collision Induced Absorption Bands
O ₂ ·O ₂	1	1.24×10^{43}	HC, 1.58 μm 6.42 μm
O ₂ ·N ₂	3.728	4.62×10^{43}	HC, 4.29 μm , 6.42 μm , 100 cm^{-1}
N ₂ ·N ₂	13.90	1.72×10^{44}	4.29 μm , 90 cm^{-1}
O ₂ ·Ar	0.0446	5.52×10^{41}	HC, 1.27 μm , 6.42 μm
O ₂ ·CO ₂	1.69×10^{-3}	2.09×10^{40}	1.27 μm
O ₂ ·CH ₄	8.1×10^{-6}	1.00×10^{38}	HC
CO ₂ ·CO ₂	2.87×10^{-6}	3.55×10^{37}	60 cm^{-1}

Scale factors relate $N_{X \cdot Y}$ to $N_{O_2 \cdot O_2}$ (1). Band locations are approximate centers of absorption bands. HC denotes Herzberg Continuum ($\sim 200\text{--}250 \text{ nm}$). O₂·O₂ and O₂·N₂ bands considered in this study (Plate 1) are omitted for brevity. Spectral data compiled from Shardanand [1977], Shardanand and Rao [1977], Goody and Yung [1989], Orlando *et al.* [1991], Oshima *et al.* [1995], Solomon *et al.* [1998], and Mlawer *et al.* [1998].

Table 2. Global Annual Mean Forcings of O₂·X

Forcing, W m ⁻²	O ₂ ·O ₂	O ₂ ·N ₂	Total
Atmospheric Absorption	0.78	0.15	0.93
(same, but for clear sky)	(0.78)	(0.17)	(0.94)
Surface Insolation	-0.50	-0.10	-0.60
Surface Absorption	-0.42	-0.08	-0.51
Sfc. + Atm. Absorption	0.36	0.07	0.42
Net Flux at Tropopause	0.34	0.06	0.40

Total forcings may be up to 20% lower or 30% higher than indicated. See text for uncertainty analysis. Sfc, surface; Atm, atmosphere.

2, and 0.5 W m⁻², respectively [Kiehl and Trenberth, 1997] (NO₂ abundance is highly heterogeneous and a reliable estimate of its mean absorption has yet to be made). The second row shows that on average, the clear sky atmospheric absorption by O₂·X slightly exceeds the all sky absorption.

Atmospheric absorption at solar wavelengths reduces surface insolation (third row of Table 2) and thus surface absorption (fourth row). The fifth row shows the column radiative forcing, the sum of atmospheric forcing and surface forcing. This quantity (column radiative forcing) equals the change in net radiative flux at the top of the atmosphere. The impact of atmospheric constituents on tropospheric climate is, by convention, quantified by the change in net radiative flux at the tropopause, shown in row six. More than 97% of O₂·X, and thus 97% of O₂·X absorption, is in the troposphere (see Plate 6). Therefore the column radiative forcing of O₂·X nearly equals the net radiative forcing at the tropopause.

4.2. Uncertainty in Mean O₂·X Forcing

The largest uncertainties in the estimates in Table 2 stem from uncertainties in cross section $\sigma_{O_2 \cdot O_2}^b$, N₂ efficiency ϵ_{N_2} , O₂·X abundance $N_{O_2 \cdot X}$, and cloud vertical distribution. The uncertainty ranges for these quantities are $\sigma_{O_2 \cdot O_2}^b(\lambda)$, -30 to +10%, depending on λ (see section 2.2); ϵ_{N_2} , $\pm 50\%$; and $N_{O_2 \cdot X}$, $\pm 2\%$. Also, appending the O₂·O₂ 1.58 μm continuum of Mlawer *et al.* [1998] to the Solomon *et al.* [1998] cross sections would increase our computed total O₂·X solar forcing by $\sim 4\%$. As discussed in section 2.1, the possible temperature dependence of $\sigma_{O_2 \cdot X}^b(\lambda)$ is not yet fully understood [Greenblatt *et al.*, 1990; Newnham and Ballard, 1998; Osterkamp *et al.*, 1998]. Therefore we have not included uncertainty due to possible temperature dependence of $\sigma_{O_2 \cdot X}^b$ in our error analysis.

The simulated mean clear sky and all sky planetary albedos agree with satellite observations to within 0.005 [Hack *et al.*, 1998], eliminating the possibility of gross surface or cloud reflectance biases in estimated O₂·X forcing. Modeled total cloud fraction, 0.588, is bracketed by the Nimbus 7 and International Satellite Cloud Climatology Project (ISCCP) estimates of 0.522 and 0.625, respectively [Hack *et al.*, 1998]. The vertical distribution of cloudiness generally agrees with observations for clouds below 400 mb but disagrees for cloud top heights above 400 mb, a region where clouds efficiently reduce O₂·X forcing. The CCM

high cloud fraction, 0.34, exceeds ISCCP analyses by 0.20 [Hack *et al.*, 1998]. There is considerable uncertainty in the observations, but if the ISCCP total and high cloud estimates are correct, then we estimate the modeled all sky O₂·X forcing is up to 10% too low because of exaggerated amounts of high-level cloud relative to low-level cloud.

Except for O₂·X abundance, the aforementioned uncertainties are systematic, not random. Summing the uncertainties for a conservative error estimate yields a total uncertainty range for the all sky forcings in Table 2 of -20 to +30%. Thus the uncertainty range for O₂·X all sky absorption is 0.75–1.21 W m⁻². The lower bound forcing corresponds to the 1.06 and 1.27 μm $\sigma_{O_2 \cdot X}^b$ of Mlawer *et al.* [1998] and N₂ efficiency $\epsilon_{N_2} = 20\%$ (3). The upper bound forcing uses $\sigma_{O_2 \cdot X}^b$ from Greenblatt *et al.* [1990] and Solomon *et al.* [1998], $\epsilon_{N_2} = 30\%$, and assumes that 4% of O₂·X absorption occurs at 1.58 μm and that the CCM3 overpredicts high level clouds relative to low-level clouds.

The results in Table 2 appear to contrast with two prior studies. Pfeilsticker *et al.* [1997] estimate O₂·O₂ enhances all sky heating relative to clear sky heating by $\sim 7\%$ (neglecting the 1.27 μm band). Using identical O₂·X cross sections to our study, Solomon *et al.* [1998] estimated the mean atmospheric absorption due to O₂·X is 0.9–1.3 W m⁻². Both these prior studies used the same method to convert single-column results to global mean results. The method assumes 50% of the globe is covered with cloud top heights at 700 mb and that these clouds increase the photon path by 30 km [Pfeilsticker *et al.*, 1997]. Thus their conversion method assumes fewer and lower clouds than predicted by the CCM3. These assumptions both lead to larger forcings than shown in Table 2 and, taken together, suffice to explain the differences between the studies. It is unclear whether assumptions regarding surface albedo and elevation also contribute to these model differences.

4.3. Seasonal Structure of Abundance and Absorption

Owing to the seasonal cycles of insolation and temperature, annual mean forcings do not suffice to characterize the geographic and vertical distribution of O₂·X forcing as it impacts the climate system. To illustrate this point, Figure 2 shows the zonal mean of the column abundance of O₂·O₂ for December–February (DJF) and June–August (JJA). The increases in $N_{O_2 \cdot X}$ from summer to winter in the North and South polar regions are 8–10% and 1–3%, respectively. These increases are entirely due to seasonal changes of pressure and temperature. Although the solar radiative forcing of O₂·X during polar winter is zero, this seasonal cycle in $N_{O_2 \cdot X}$ does affect O₂·X absorption bands in the far infrared (see Table 1) throughout the year. The meridional gradient and seasonal changes in $N_{O_2 \cdot X}$ depicted in Figure 2 apply to all other collision complexes in Table 1 as well.

The strongest modulator of solar radiative forcing is, of course, the seasonal change of insolation. Plate 5 shows the geographic distribution of the seasonal mean atmospheric absorption by O₂·X. O₂·X absorption is greatest in polar regions in summer, where peak insolation coincides with the greatest O₂·X abundances (Figure 2). Plates 5a

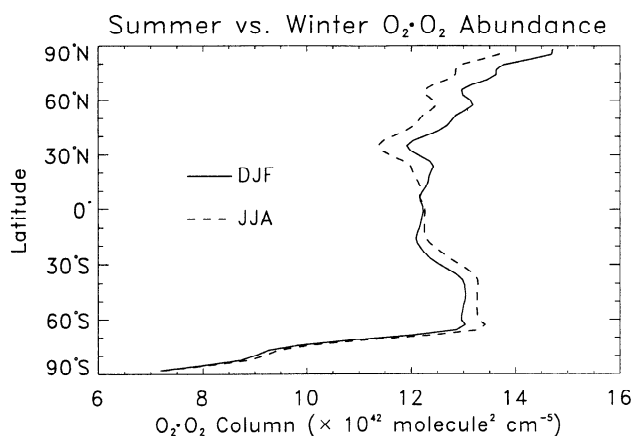


Figure 2. Zonal mean column abundance of O₂·O₂, N_{O₂·O₂} ($\times 10^{42}$ molecule² cm⁻⁵), for DJF (solid line) and JJA (dashed line).

and 5c show southern and northern summer forcings are nearly symmetric about the equator. Arctic forcing is 10–30% stronger than Antarctic forcing, mainly the Antarctic plateau reduces N_{O₂·X}. Most remarkable are the large polar regions where atmospheric O₂·X absorption approaches or exceeds 2.0 W m⁻² on seasonal timescales. Seasonal sea ice boundaries appear in Plates 5a–5d where they demarcate the open ocean from regions of higher absorption. Equinoctial forcings (Plates 5b and 5d) extend farther into the winter hemisphere than solstitial forcings, and are weaker.

Surface albedo, elevation, and clouds all play roles in determining the seasonal forcing maxima near the poles. To help disentangle these roles, we show in Figure 3 the zonal mean of the winter and summer seasonal forcings (Plates 5a and 5c) due to O₂·X. The influence of clouds and surface albedo on O₂·X forcing is clarified by comparing the seasonal forcings in Figure 3 to the seasonal abundances in Figure 2. The zonal mean summertime O₂·X forcing in the Antarctic (70–90°S), 1.69 W m⁻², is 96% of the forcing in the Arctic (70–90°N), 1.76 W m⁻², although the Antarctic has only 74% of the O₂·X. Thus surface and low-level cloud reflectance makes O₂·X forcing 30% more efficient (per unit abundance) in the Antarctic than in the Arctic. Within 5° of the poles, the forcing efficiency of O₂·X is 75% greater in the Antarctic than in the Arctic. The change in forcing efficiency reduces, and nearly eliminates, the surface elevation-induced disparity in forcing between the poles.

The large difference between the forcing efficiencies of O₂·X in the Arctic and the Antarctic is caused by differences in the surface albedo, cloud distribution, and water vapor column between the two polar climate systems. First, the simulated visible surface albedo of Arctic sea ice is ~ 0.6 , much lower than the simulated visible surface albedo of the Antarctic continent, ~ 0.95 [Briegleb and Bromwich, 1998a]. Offline sensitivity studies show that in the absence of clouds, the higher surface reflectance underlying Antarctic O₂·X increases forcing (per unit abundance) by $\sim 25\%$ relative to the Arctic. Observations suggest the modeled surface albedo and low-level cloud fraction in the summertime

Arctic are ~ 0.15 and 0.10 too high, respectively [Briegleb and Bromwich, 1998a]. This suggests the simulated summertime Arctic forcing is too high, which would amplify the disparity between Arctic and Antarctic forcing efficiencies.

A second, but much weaker, effect is due to the change in simulated precipitable water content from the summertime Arctic (11.5 kg m⁻²) to the Antarctic (2.7 kg m⁻²). Both are in fair agreement with European Centre for Medium-Range Weather Forecasts (ECMWF) analyses [Briegleb and Bromwich, 1998a]. The drier Antarctic column makes the wings of the near-infrared water vapor bands (compare Plate 2) more transmissive (less saturated). This increases absorption per unit O₂·X by $\sim 2\%$.

4.4. Vertical Structure

The vertical profile of O₂·X heating is important in determining possible dynamical feedbacks as well as the change in net flux at the tropopause. Plate 6 shows the vertical profile of the seasonally and zonally averaged solar heating due to O₂·X. The vertical axis is the CCM hybrid vertical coordinate times 1000 [Kiehl et al., 1998a]. Thus the panels transition from a pure sigma coordinate representation at the surface to a pure pressure representation near 200 mb. As mentioned previously, virtually all of the heating is confined to the troposphere. O₂·X heating decreases toward the winter hemisphere due to decreasing daylight hours. Above 700 mb (300 mb above the surface) the heating decreases with pressure due to decreasing O₂·X abundance (compare Figure 1). Seasonal O₂·X heating in excess of 0.02 K d⁻¹ extends throughout the polar summer troposphere. This is 2–4% of local solar heating owing to all other solar absorbers (compare Figure 1). The heating rate in the Antarctic summer troposphere (Plate 6a) is nearly 20% stronger than the Arctic summer troposphere (Plate 6c) due to its thinner at-

The heating peaks between 200–300 mb above the surface in all regions every season. This peak does not coincide with the maximum O₂·X abundance (which is always at the surface) even though O₂·X absorption is in the opti-

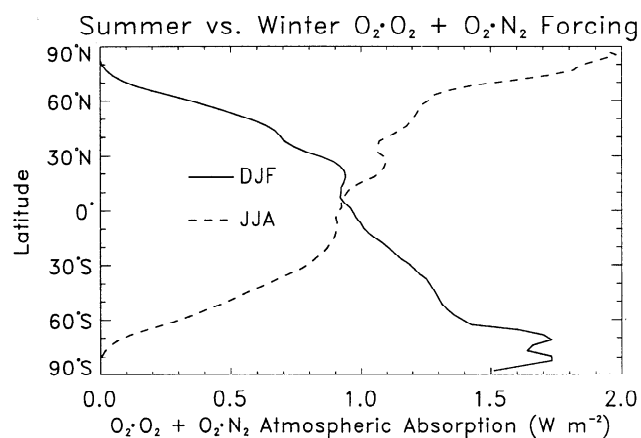


Figure 3. Zonal seasonal mean increase in atmospheric solar absorption (W m⁻²) due to O₂·X: DJF (solid line) and JJA (dashed line).

cally thin limit (Plate 2). As shown in Figure 1, the location of maximum O₂·X heating does coincide with maximum O₂·X abundance except when efficient scatterers such as clouds screen the O₂·X in the lower atmosphere from incoming solar radiation. Thus cloud reflection is responsible for shifting the mean vertical location of maximum O₂·X absorption, and thus heating, upward from the surface by 200–300 mb. Plate 6 supports this by showing that the heating maxima as a function of latitude dips somewhat closer to the surface in the subtropical regions characterized by large-scale subsidence and relatively clear skies.

Radiative forcing in other nonoverlapped, optically thin bands of well-mixed collision complexes (e.g., the N₂·N₂ fundamental band at 4.29 μm, see Table 1), will show qualitatively similar patterns to Plates 4–6. The magnitude of forcing by other bands is, of course, highly sensitive to the spectral location of the particular band.

5. Discussion and Summary

We have characterized the spectral, vertical, regional, and seasonal atmospheric heating by O₂·X. The motivations for modeling the global distribution and forcing of O₂·X were to refine and extend estimates of O₂·X forcing inferred from one-dimensional studies to the global climate scale. Our global simulations indicate that global annual mean atmospheric absorption by O₂·X is 0.75–1.2 W m⁻² (Table 2). This falls between the recent estimates of 0.9–1.3 W m⁻² [Solomon *et al.*, 1998] and 0.84 W m⁻² [Mlawer *et al.*, 1998]. This absorption reduces surface insolation by 0.48–0.78 W m⁻².

O₂·X absorption is highly localized in the troposphere and it increases the net radiative flux at the tropopause by 0.32–0.52 W m⁻². It is useful to compare this absolute radiative forcing by O₂·X (whose abundance is not changing) to greenhouse gas forcing (due to changes in the gases' abundance from preindustrial to present-day levels). This comparison enables us to assess the magnitude of the perturbation to the mean state climate caused by O₂·X. The 0.32–0.52 W m⁻² forcing by O₂·X is less than the anthropogenic forcing of CO₂ (1.75 W m⁻²) but comparable to forcing by CH₄ (0.44 W m⁻²), stratospheric H₂O (0.15 W m⁻²), and N₂O (0.11 W m⁻²) [Shine *et al.*, 1990, p. 57].

By extending the simulations to the global scale, we were able to characterize the spatial and temporal distributions of O₂·X abundance and forcing. O₂·X abundance mimics the concentration of O₂ and N₂, except its quadratic dependence on these constituents accentuates both its spatial and temporal gradients. This dependence causes a 20% increase in O₂·X abundance in the Arctic relative to the Tropics for the same sea level pressure. O₂·X abundance depends most on zonal mean temperature, pressure, and surface elevation. The variations in zonal mean O₂·X abundance due to surface elevation, the annual mean meridional temperature gradient, and to seasonal temperature variations are 40, 15, and 10%, respectively. These results scale to all collision complexes of well-mixed gases, for example, N₂·N₂ and O₂·Ar.

Many features of O₂·X radiative forcing are common to all solar absorbers, but the relatively sharp gradients in

O₂·X abundance enhance these features. Surface and cloud reflectance modulate the solar forcing efficiency of O₂·X (forcing per unit abundance) by altering the mean O₂·X path traversed by the average photon. These factors explain why annual mean collision complex forcing peaks in the subtropics over bright deserts and regions of marine stratus. The seasonal cycle of insolation causes the seasonal mean solar forcing to peak in summertime polar regions. However, these summertime polar peaks are enhanced by bright surfaces and extensive low-level clouds, especially in the Arctic. The forcing efficiency of complexes in the Antarctic is 30–75% greater than in the Arctic.

In clear skies the magnitude of O₂·X heating grows linearly with air density, peaking at the surface. This heating profile tends to weakly destabilize the atmospheric column. On seasonal timescales, clouds shift the height of peak O₂·X heating upward by 200–300 mb and smooth out the vertical decrease of heating that would otherwise follow the rapid decrease of O₂·X abundance with height. O₂·X heating peaks at 0.02–0.03 K d⁻¹ in the polar troposphere in summer. This is 2–4% of local heating due to all other solar absorbers.

Many GCMs have strong cold biases at the summertime polar tropopause, where the CCM3 (not shown) is up to 10–14 K too cold [Hack *et al.*, 1998]. The CCM3 also suffers from a cold bias in the summertime polar troposphere of 2–6 K [Briegleb and Bromwich, 1998b]. Some, but probably not all, of these biases are due to the numerical treatment of dynamics [Williamson and Olson, 1998]. It is likely that allowing O₂·X heating to affect the thermal structure of the atmosphere will ameliorate, though not eliminate, the remainder of these biases. For example, single-column simulations show the radiative relaxation timescale at the polar tropopause is ~ 50 days. Thus the O₂·X heating of ~ 0.01 K d⁻¹ (Plate 6) increases the radiative equilibrium temperature by ~ 1 K in this region.

We quantified some fingerprints that distinguish absorption due to well-mixed collision complexes (and well-mixed dimers) from absorption due to well-mixed monomers. The most significant distinguishing feature is the stronger spatial and temporal gradients of collision complex abundance relative to monomer abundance. Whereas the zonal mean column abundance of a well-mixed molecule such as O₂ decreases by 30% from sea level in the Arctic to the Antarctic plateau at 700 mb, the corresponding decrease in N_{O₂·X} exceeds 50%. In the weakly absorbing, nonoverlapped limit, well-mixed molecules cause an increase in the solar heating rate that is constant with air density, whereas well-mixed collision complexes such as O₂·X cause the solar heating rate to increase linearly with air density.

Zender *et al.* [1997] demonstrated that O₂·O₂ absorption improved agreement between modeled and observed solar absorption in both clear and cloudy skies during the Atmospheric Radiation Measurement (ARM) Enhanced Shortwave Experiment (ARESE). Their total (direct + diffuse) clear sky surface insolation discrepancies (~ 15 W m⁻²) were no larger than instrumental and model uncertainties (e.g., aerosol properties). Working with an independent set of radiometric observations from ARESE, Kato *et al.* [1997] found excellent model agreement with direct beam observa-

tions but reported a 30 W m^{-2} model overestimate of the instantaneous diffuse surface insolation during clear sky conditions over multiple days during ARESE. They did not account for O₂-X absorption. As shown in section 3.1, O₂-X absorption reduced clear sky noontime surface insolation by $\sim 2 \text{ W m}^{-2}$ during ARESE. However, nearly 90% of this O₂-X absorption came from the direct solar beam, and only 10% occurred in the diffuse field. This illustrates the point that unknown or neglected gaseous absorption processes cannot, by themselves (i.e., without invoking aerosol processes), remedy a clear sky diffuse radiation bias because gaseous absorption of the direct beam far exceeds gaseous absorption of the diffuse field.

Our global simulations show that in the annual mean, O₂-X enhances absorption equally in clear and cloudy skies. Thus O₂-X neither exacerbates nor remediates solar absorption discrepancies that are associated with clouds [Stephens and Tsay, 1990; Cess et al., 1995; Ramanathan and Vogelmann, 1997]. The agreement between clear and cloudy sky O₂-X absorption is due to the global mean cancellation of photon and absorber path length changes due to clouds: Low-level clouds increase O₂-X absorption relative to clear skies by increasing path lengths, while middle- and high-level clouds decrease O₂-X absorption by screening photons from the lower atmosphere. We hypothesize that this agreement is predominantly due to the global vertical distributions of cloud reflectivity and well-mixed molecules and has little to do with spectral features particular to O₂-X. There are two reasons to expect that this hypothesis is true. First, O₂-X absorption occurs at many locations in the visible and near-infrared and is not correlated with water vapor or condensate absorption bands. Second, cloud transmission is nearly spectrally uniform for $0.3 < \lambda < 1.5 \mu\text{m}$, so clouds do not preferentially transmit any energetically significant solar wavelength bands to the lower atmosphere. This hypothesis should be tested because, if it is true, it implies that globally averaged, well-mixed collision complexes or dimers do not enhance absorption more in cloudy sky than in clear sky. A similar hypothesis may hold for well-mixed molecules in general.

The radiative importance of other abundant, but not necessarily well-mixed, collision complexes and dimers in the atmosphere should be explored. Solomon et al. [1998] point out that Ar and CO₂ are also relatively efficient partners for inducing O₂ absorption at $1.27 \mu\text{m}$ but note that O₂-Ar and O₂-CO₂ are too scarce (Table 1) to cause significant absorption. Chylek and Geldart [1997] and Solomon et al. [1998] suggest O₂-H₂O and (H₂O)₂ as promising candidates to examine in the search for neglected solar absorbers.

In summary, the recent discovery that the oxygen collision pairs O₂-O₂ and O₂-N₂ absorb a small but significant fraction of the globally incident solar radiation alters the long-standing view that H₂O, O₃, O₂, CO₂, and NO₂ are the only significant gaseous solar absorbers in Earth's atmosphere. This study quantified the spectral, regional, vertical, and seasonal patterns of O₂-X abundance and radiative forcing. We now know that globally and annually averaged, O₂-X absorbs about 1 W m^{-2} of solar radiation, more than CO₂, and that O₂-X does not increase cloudy

sky solar absorption relative to clear sky absorption. Improved measurements of O₂-X absorption cross sections in the near infrared will remove about half of the 25% uncertainty in total O₂-X absorption. Accounting for O₂-X reduces discrepancies between models and measurements of solar absorption. O₂-X absorption should therefore be included in high-resolution radiative transfer models, remote-sensing retrieval algorithms, and in large-scale atmospheric models used to simulate climate and climate change. O₂-X has the most potential to improve the simulated climate in summertime polar regions. Future versions of the CCM will include this absorption.

Acknowledgments. S. Solomon and R. Portmann provided O₂-X cross section data and helpful suggestions at an early stage of this study. J. Burkholder, E. Mlawer, D. Newnham, and K. Pfeilsticker provided useful comments on O₂-O₂ cross section measurements and uncertainties. B. Briegleb, W. Collins, and S. Solomon made numerous suggestions which improved the quality of the manuscript. R. Portmann and one anonymous reviewer provided constructive reviews. The NCAR ASP program provided support and encouragement. This research was supported in part by DOE ARM Program Grant DEFG0593ER61376.

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(Received January 28, 1999; revised July 14, 1999; accepted July 22, 1999.)